

Electronic Supplementary Information

Exploring new hydrated delta type vanadium oxides for lithium intercalation

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Table S1. ICP-Q-MS and elemental analyses (%) of **NaVOx**, **KVOx** and **NHVOx**.

Sample	V exp	V calc	Na exp	Na calc	K exp	K calc	N exp	N calc
Na _{0.35} V ₂ O ₅ ·0.8(H ₂ O) (NaVOx)	49.2(5)	49.86	3.95(4)	3.94	-----	-----	-----	-----
K _{0.36} (H ₃ O) _{0.15} V ₂ O ₅ (KVOx)	51.1(4)	51.25	-----	-----	7.05(5)	7.08	-----	-----
(NH ₄) _{0.37} V ₂ O ₅ ·0.15(H ₂ O) (NHVOx)	53.6(4)	53.27	-----	-----	-----	-----	2.73(5)	2.71

Table S2. Pattern Matching and Rietveld refinement parameters for **NaVOx**, **NaVOx-HT**, **KVOx** and **NHVOx**.

Sample	NaVOx		NaVOx-HT (110°C)		KVOx		NHVOx	
Formula	Na _{0.35} V ₂ O ₅ ·0.8(H ₂ O)		Na _{0.35} V ₂ O ₅ ·0.23(H ₂ O)		K _{0.36} (H ₃ O) _{0.15} V ₂ O ₅		(NH ₄) _{0.37} V ₂ O ₅ ·0.15(H ₂ O)	
M. Weight (g/mol)	204.34		194.07		198.81		191.26	
Z	2		2		2		2	
Analysis Method	Pattern Matching	Rietveld Analysis	Pattern Matching	Rietveld Analysis	Pattern Matching	Rietveld Analysis	Pattern Matching	Rietveld Analysis
a, Å	11.9157(2)	11.925(1)	11.6862(4)	11.863(3)	11.9055(2)	11.9057(5)	11.8798(2)	11.8807(3)
b, Å	3.74152(5)	3.7415(1)	3.7272(1)	3.7160(5)	3.77266(6)	3.7716(2)	3.76855(4)	3.76875(7)
c, Å	14.2111(4)	14.239(1)	12.3832(3)	12.543(2)	13.1062(3)	13.092(1)	12.6074(2)	12.6044(4)
β, °	114.053(3)	114.296(8)	115.844(2)	117.32(2)	117.331(1)	117.354(4)	110.207(2)	110.160(3)
V, Å ³	578.56(2)	579.06(7)	485.42(3)	491.3(2)	522.96(2)	522.15(5)	529.69(1)	529.79(3)
Reflections	739	741	625	628	670	670	678	679
R _{Bragg}	0.11	13.4	0.05	25.2	0.19	17.1	0.04	16.4
R _f	0.504	10.2	0.580	23.0	0.275	14.6	0.197	14.6
R _p	14.5	28.0	19.8	38.1	17.1	30.2	17.0	28.7
R _{wp}	15.0	30.5	19.1	40.1	18.1	32.0	19.0	29.4
R _{exp}	3.70	3.73	4.58	4.58	3.73	3.79	3.46	3.57
χ ²	16.44	66.90	17.38	76.61	23.59	71.21	30.03	67.90

Table S3. Rietveld refined atomic coordinates and equivalent isotropic displacement parameters, B_{iso} (Å²) of **NaVOx**, **NaVOx-HT**, **KVOx** and **NHVOx**.**NaVOx**

Atom	x	y	z	B _{iso} (Å ²)	Occ. F.
V1	0.1249(9)	0.0	0.0992(6)	0.51(6)	0.5
V2	-0.1839(9)	0.0	0.1004(8)	0.51(6)	0.5
O1	0.183(4)	0.0	0.239(3)	0.5(2)	0.5
O2	-0.033(4)	0.0	0.0835(13)	0.5(2)	0.5
O3	0.3163(16)	0.0	0.0855(14)	0.5(2)	0.5
O4	0.1249(18)	0.0	0.1230(20)	0.5(2)	0.5
O5	-0.107(3)	0.0	0.252(3)	0.5(2)	0.5
O6W	0.290(3)	0.0	0.389(2)	2.224(3)	0.5
O7W	0.582(10)	0.0	0.674(4)	2.224(3)	0.17(2)
Na1	0.5	0.0	0.5	2.224(3)	0.235(9)

Table S3. Continuation.**NaVO_x HT (110°C)**

Atom	x	y	z	B_{iso} (Å²)	Occ. F.
V1	0.1294(7)	0.0	0.1247(6)	0.5	0.5
V2	-0.1782(7)	0.0	0.1039(6)	0.5	0.5
O1	0.209(7)	0.0	0.2868(9)	0.5	0.5
O2	-0.0545(7)	0.0	0.0581(8)	0.5	0.5
O3	0.2609(11)	0.0	0.0539(8)	0.5	0.5
O4	0.153(3)	0.5	0.147(4)	0.5	0.5
O5	-0.087(7)	0.0	0.2664(11)	0.5	0.5
O6W	0.33160	0.5	0.43000	0.5	0.5
Na1	0.08760	-0.8970	0.41440	0.5	0.69(1)

KVO_x

Atom	x	y	z	B_{iso} (Å²)	Occ. F.
V3	0.4745(9)	0.0	0.1018(7)	0.51	0.5
V4	0.7978(8)	0.0	0.1120(6)	0.51	0.5
O8	0.596(3)	0.0	0.061(2)	0.51	0.5
O9	0.878(3)	0.0	0.2545(14)	0.51	0.5
O10	0.567(3)	0.0	0.304(2)	0.51	0.5
O11	0.317(3)	0.0	0.1027(15)	0.51	0.5
O12	0.460(3)	0.5	0.124(2)	0.51	0.5
O13W	0.305(3)	0.0	0.371(2)	0.51	0.5
K1	0.2964(15)	-0.396(3)	0.4058(11)	0.51	0.45(1)

NHVO_x

Atom	x	y	z	B_{iso} (Å²)	Occ. F.
V1	0.1144(7)	0.5	0.1177(6)	0.49	0.5
V2	-0.1907(6)	0.5	0.1170(5)	0.49	0.5
O1	0.1698(19)	0.5	0.2701(16)	0.49	0.5
O2	-0.0650(14)	0.5	0.0899(13)	0.49	0.5
O3	0.2719(20)	0.5	0.0886(15)	0.49	0.5
O4	0.1394(19)	1.0	0.1561(16)	0.49	0.5
O5	-0.1328(19)	0.5	0.2904(17)	0.49	0.5
O8W	0.027(2)	0.5	0.3297(17)	0.49	0.5
N1	0.197(2)	1.0	0.314(2)	0.49	0.5

Table S4. Coordination environment bond distances (\AA) for **NaVOx**, **NaVOx-HT**, **KVOx** and **NHVOx** from the Rietveld analysis.

NaVOx

(V1)-(O1) ⁱ	1.82(4)	(V2)-(O2) ⁱ	1.91(5)	(O6W)-(O1) ⁱ	2.73(3)	(O7W)-(O1) ⁱⁱ	2.56(12)
(V1)-(O2) ⁱ	1.80(5)	(V2)-(O3) ⁱⁱ	2.462(19)	(O6W)-(O1) ⁱ	2.73(3)	(O7W)-(O5) ^{vi}	2.11(3)
(V1)-(O2) ⁱⁱ	2.371(18)	(V2)-(O3) ^v	1.883(3)	(O6W)-(O5) ^v	2.70(6)	(O7W)-(O5) ^{vi}	2.11(3)
(V1)-(O3) ⁱ	2.37(2)	(V2)-(O3) ^v	1.883(3)	(O6W)-(O7W) ⁱⁱ	2.79(9)	(O7W)-(Na1) ⁱ	2.93(4)
(V1)-(O4) ⁱ	1.901(6)	(V2)-(O4) ^v	2.43(3)	(O6W)-(Na1) ⁱ	2.35(3)		
(V1)-(O4) ⁱ	1.901(6)	(V2)-(O5) ⁱ	1.97(4)				

NaVOx-HT (110°C)

(V1)-(O1) ⁱ	1.807(12)	(V2)-(O2) ⁱ	1.805(14)	(O6W)-(O1) ⁱ	2.53	(Na1)-(O1) ⁱ	2.63
(V1)-(O2) ⁱ	1.945(11)	(V2)-(O3) ⁱⁱ	1.760(11)	(O6W)-(O1) ⁱ	2.53	(Na1)-(O5) ⁱ	2.09
(V1)-(O2) ⁱⁱ	2.044(12)	(V2)-(O3) ^v	1.988(4)	(O6W)-(O5) ^v	2.64		
(V1)-(O3) ⁱ	2.120(17)	(V2)-(O3) ^v	1.988(4)	(O6W)-(Na1) ^{vi}	1.78		
(V1)-(O4) ⁱ	1.881(5)	(V2)-(O4) ^v	2.30(5)	(O6W)-(Na1) ^{vii}	1.78		
(V1)-(O4) ⁱ	1.881(5)	(V2)-(O5) ⁱ	1.817(15)				

KVOx

(V3)-(O8) ⁱ	1.76(4)	(V4)-(O8) ⁱ	2.18(4)	(O13W)-(O9) ^v	2.80(4)	(K1)-(O9) ^v	2.61(4)
(V3)-(O8) ⁱⁱ	1.90(3)	(V4)-(O9) ⁱ	1.660(17)	(O13W)-(O9) ^v	2.80(4)	(K1)-(O10) ^v	2.46(3)
(V3)-(O10) ⁱ	2.35(3)	(V4)-(O11) ⁱⁱ	2.498(19)	(O13W)-(K1) ⁱ	1.578(16)	(K1)-(O13W) ⁱ	2.335(14)
(V3)-(O11) ⁱ	1.88(4)	(V4)-(O11) ^v	1.910(6)	(O13W)-(K1) ⁱ	2.335(14)	(K1)-(O13W) ⁱ	1.578(16)
(V3)-(O12) ⁱ	1.928(8)	(V4)-(O11) ^v	1.910(6)	(O13W)-(K1) ⁱⁱ	2.335(14)		
(V3)-(O12) ⁱ	1.928(8)	(V4)-(O12) ^v	1.86(4)	(O13W)-(K1) ⁱⁱ	1.578(16)		

NHVOx

(V1)-(O1) ⁱ	1.80(2)	(V2)-(O2) ⁱ	1.643(20)	(O8W)-(O1) ⁱ	2.08(4)	(N1)-(O1) ⁱ	1.959(8)
(V1)-(O2) ⁱ	2.037(19)	(V2)-(O3) ⁱⁱ	2.434(19)	(O8W)-(O2) ⁱ	2.84(3)	(N1)-(O1) ⁱ	1.959(8)
(V1)-(O2) ⁱⁱ	2.476(18)	(V2)-(O3) ^v	1.941(5)	(O8W)-(O5) ⁱ	1.79(3)	(N1)-(O4) ⁱ	1.87(3)
(V1)-(O3) ⁱ	2.03(3)	(V2)-(O3) ^v	1.941(5)	(O8W)-(N1) ⁱ	2.82(3)	(N1)-(O5) ^v	2.14(4)
(V1)-(O4) ⁱ	1.943(5)	(V2)-(O4) ^v	2.24(3)	(O8W)-(N1) ⁱ	2.82(3)		
(V1)-(O4) ⁱ	1.943(5)	(V2)-(O5) ⁱ	2.05(2)				

Symmetry codes: $i = x, y, z$; $ii = -x, y, -z$; $iii = -x, -y, -z$; $iv = x, -y, z$; $v = x+1/2, y+1/2, z$; $vi = -x+1/2, y+1/2, -z$; $vii = -x+1/2, -y+1/2, -z$; $viii = x+1/2, -y+1/2, z$.

Table S5. Assignment of the Raman vibrational spectra of **NHVOx**, **KVOx** and **NaVOx**.

Symmetry species	NHVOx (cm^{-1})	KVOx (cm^{-1})	NaVOx (cm^{-1})
B _{1g} , B _{3g}	164	164	164
B _{1g}	228	235	-----
B _{1g} , B _{3g}	266	266	266
A _g	399	399	414
B _{2g}	511	506	509
B _{1g} , B _{3g}	676	676	686
B _{2g}	943	957	-----
A _g	1017	1010	1000

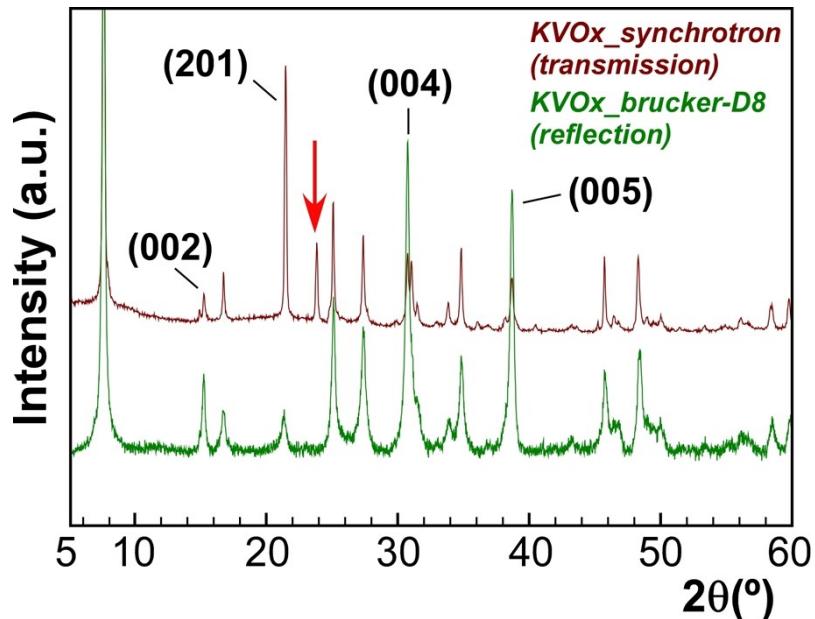


Figure S1. Spurious peak marked with a vertical red arrow observed in the synchrotron X-ray diffraction pattern of **KVOx**. Comparison with the pattern measured in a Brucker D8 Advance diffractometer (CuK α) where the peak does not appear. The wavelength of the synchrotron pattern was changed to CuK α for comparison.

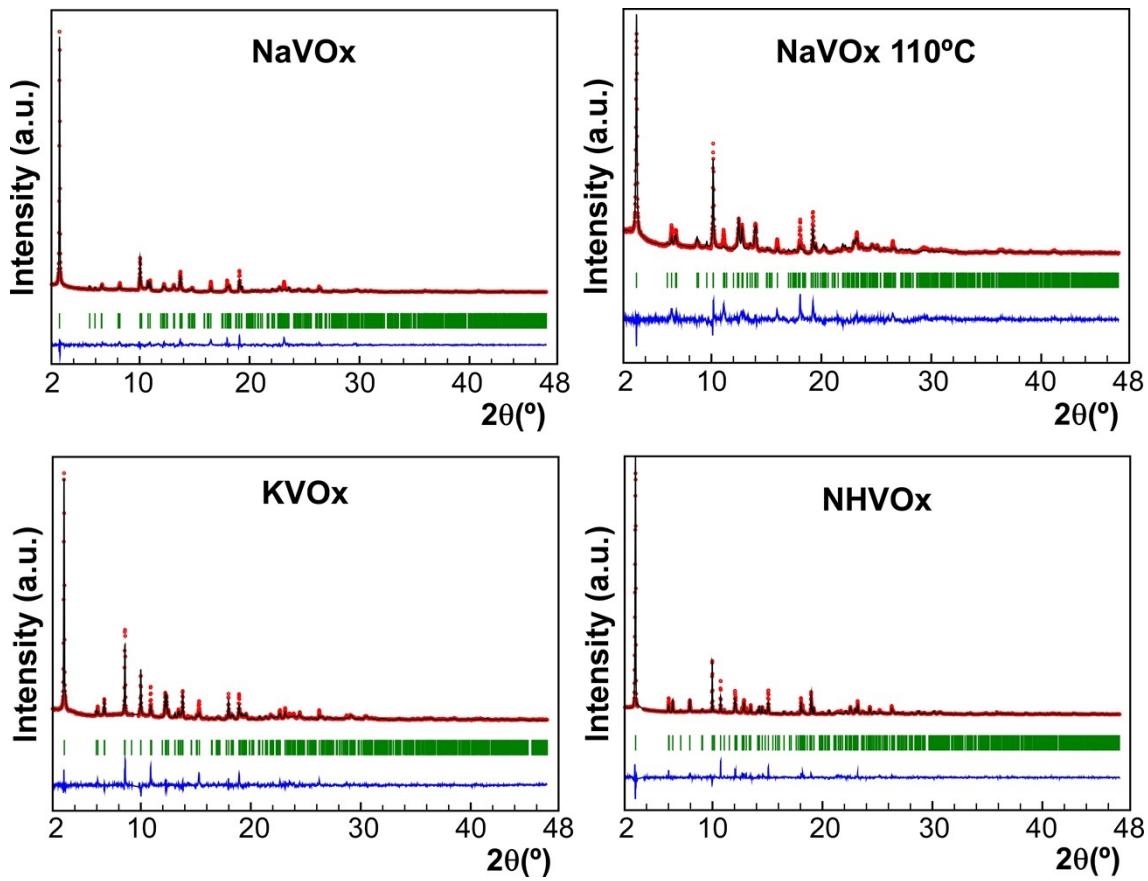


Figure S2. Observed (red dots), calculated (black lines) and difference X-ray powder diffraction patterns (blue lines) for the Rietveld analysis of the phases.

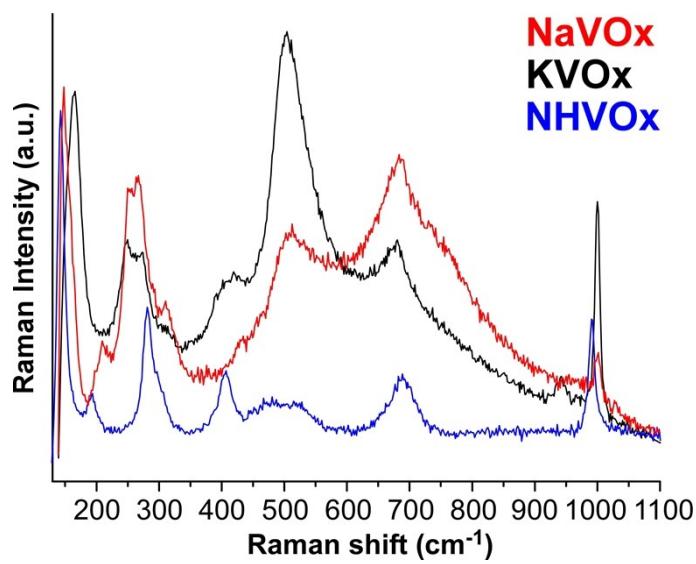


Figure S3. Raman spectra of the phases with the power increased to 50% of the incident laser beam.

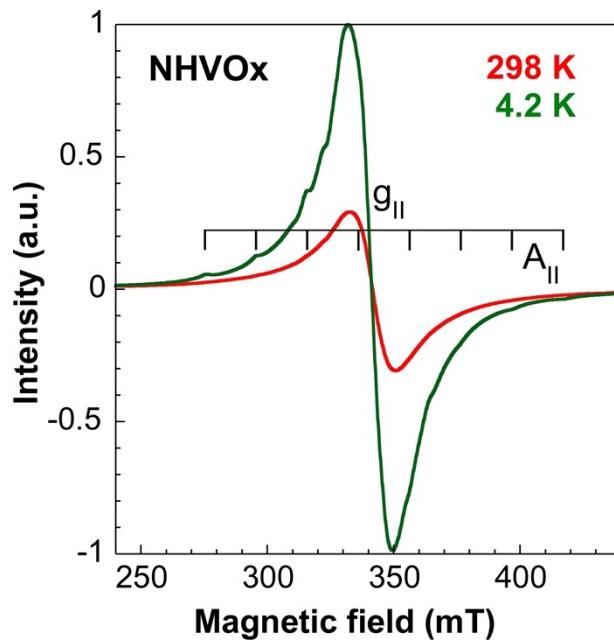


Figure S4. X-band EPR powder spectra at room temperature and 4.2K for NHVOx.

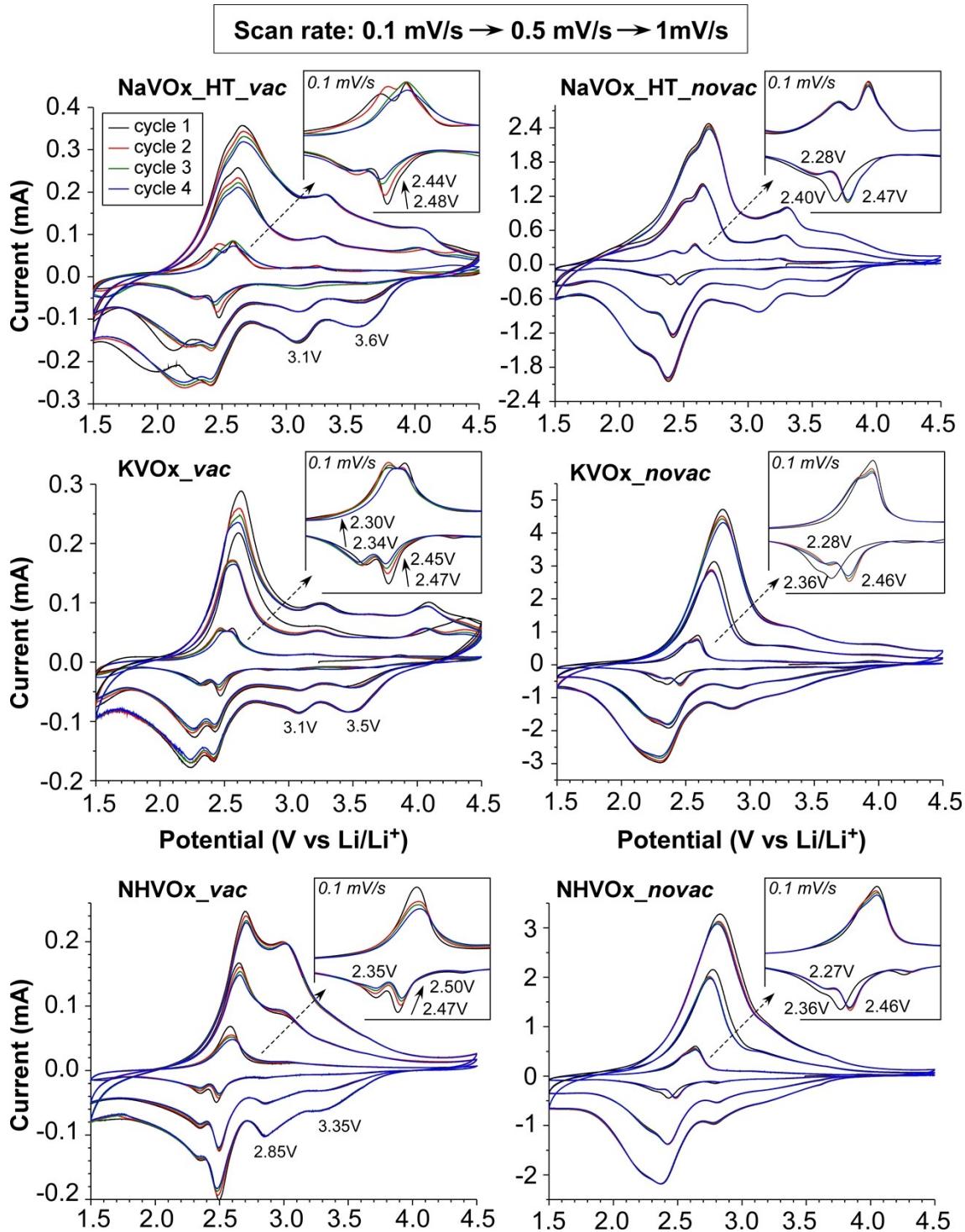


Figure S5. Cyclic voltammetries at different scan rates (0.1 mV/s, 0.5 mV/s and 1mV/s) of the *vac* and *novac* electrodes.

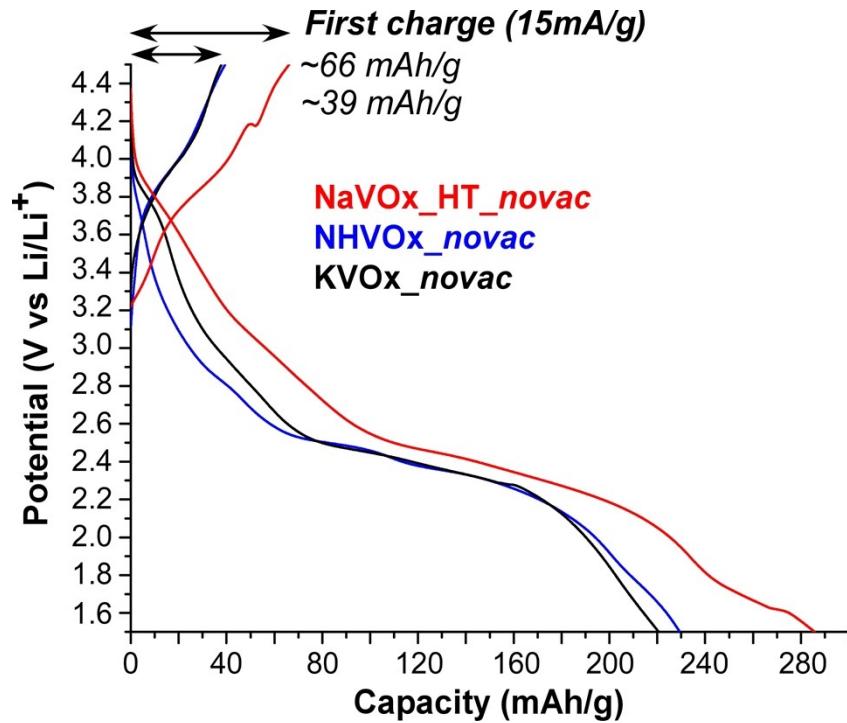


Figure S6. First charge-discharge at 15 mA/g of **NaVOx_x HT novac**, **NHVOx_x novac** and **KVOx_x novac** electrodes.

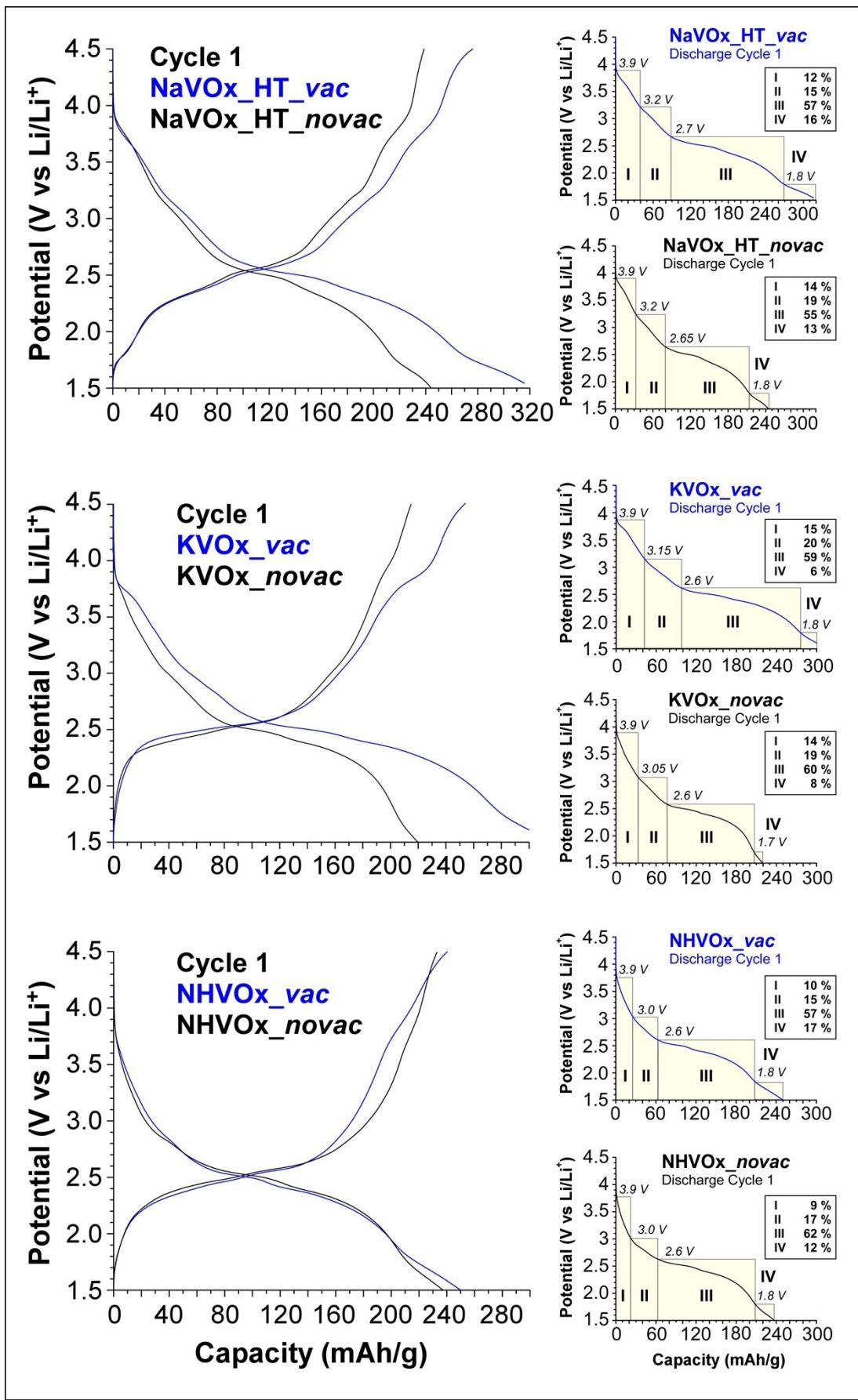


Figure S7. First cycle at 15 mA/g after the cell being discharged to 1.5 V.

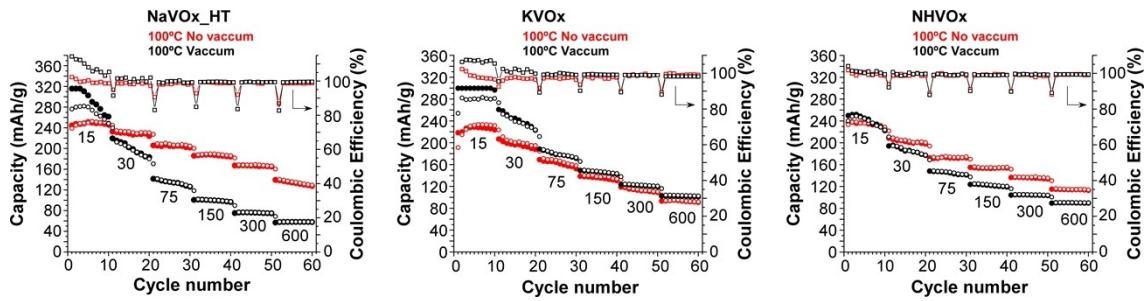


Figure S8. Rate capabilities and the corresponding coulombic efficiencies of the compounds, comparing *_vac* and *_novac* electrodes, recorded at current rates of 15, 30, 75, 150, 300 and 600 mA/g.

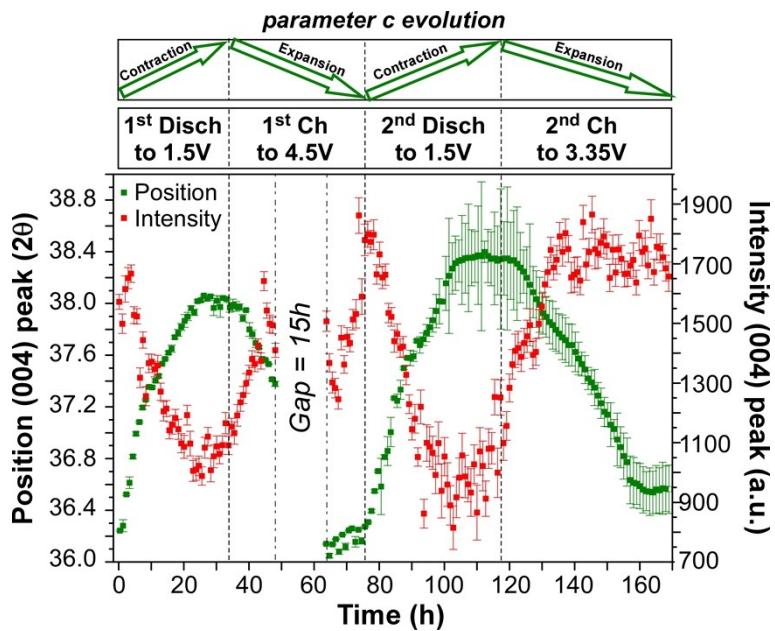


Figure S9. Evolution of the (004) peak position and intensity during the first two discharge-charge cycles of the cell of KVOx.

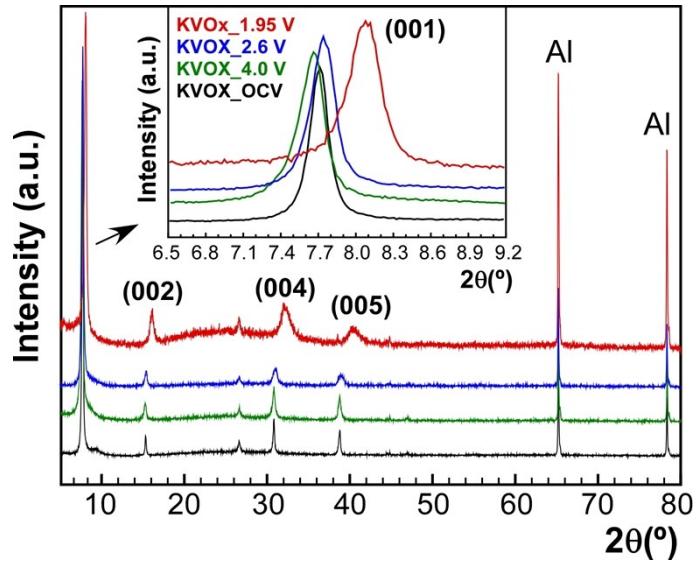


Figure S10. Ex-situ X-ray diffraction patterns of **KVO_x** after charging at 15 mA/g to 4.0 V and discharging to 2.6 V and 1.95 V. The patterns were recorded with CuK α .

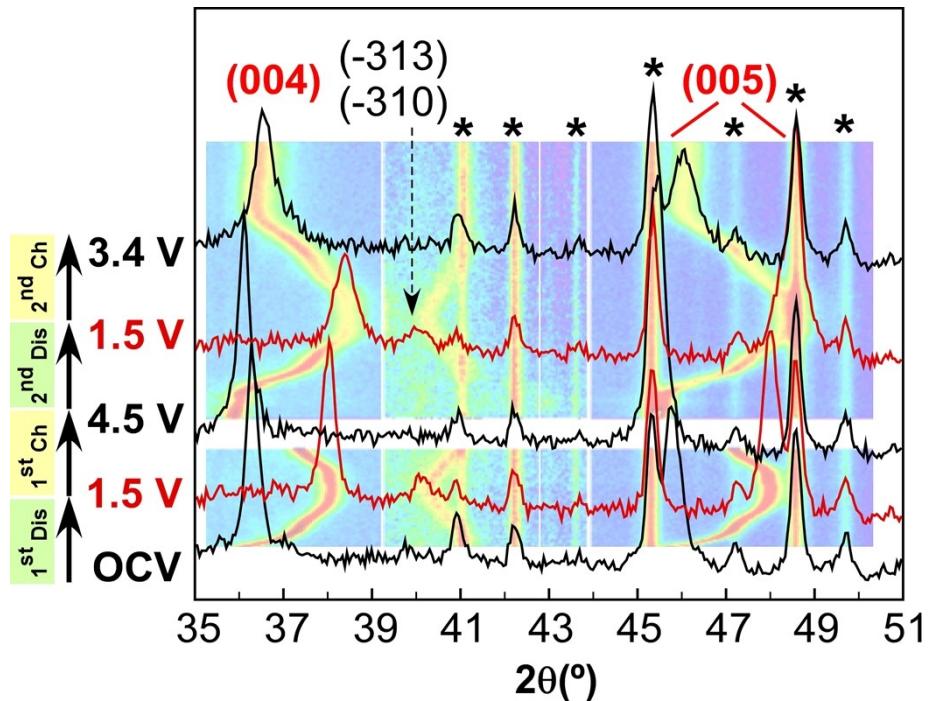


Figure S11. Detailed plot of the XRD patterns between $35 \leq 2\theta \leq 51^\circ$ during the first two cycles of the cell of **KVO_x**. The asterisks represent artifacts of the cell.

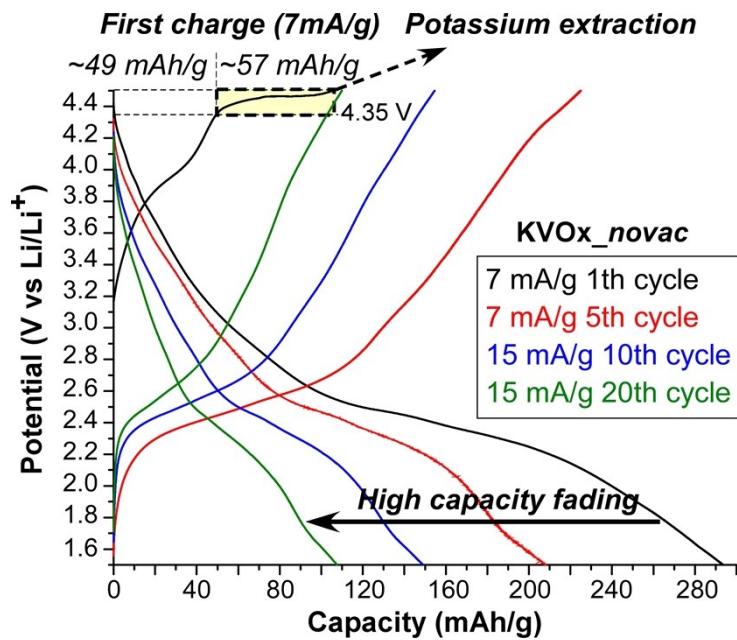


Figure S12. Potassium extraction from the first cycle at very low current density of 6 mA/g for **KVO_x_novac**.

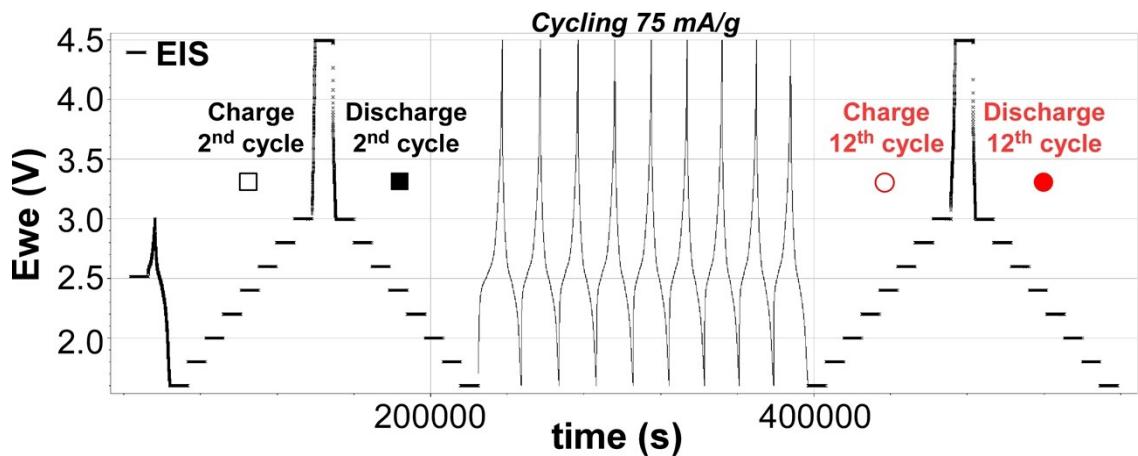


Figure S13. Scheme of the EIS measurement.

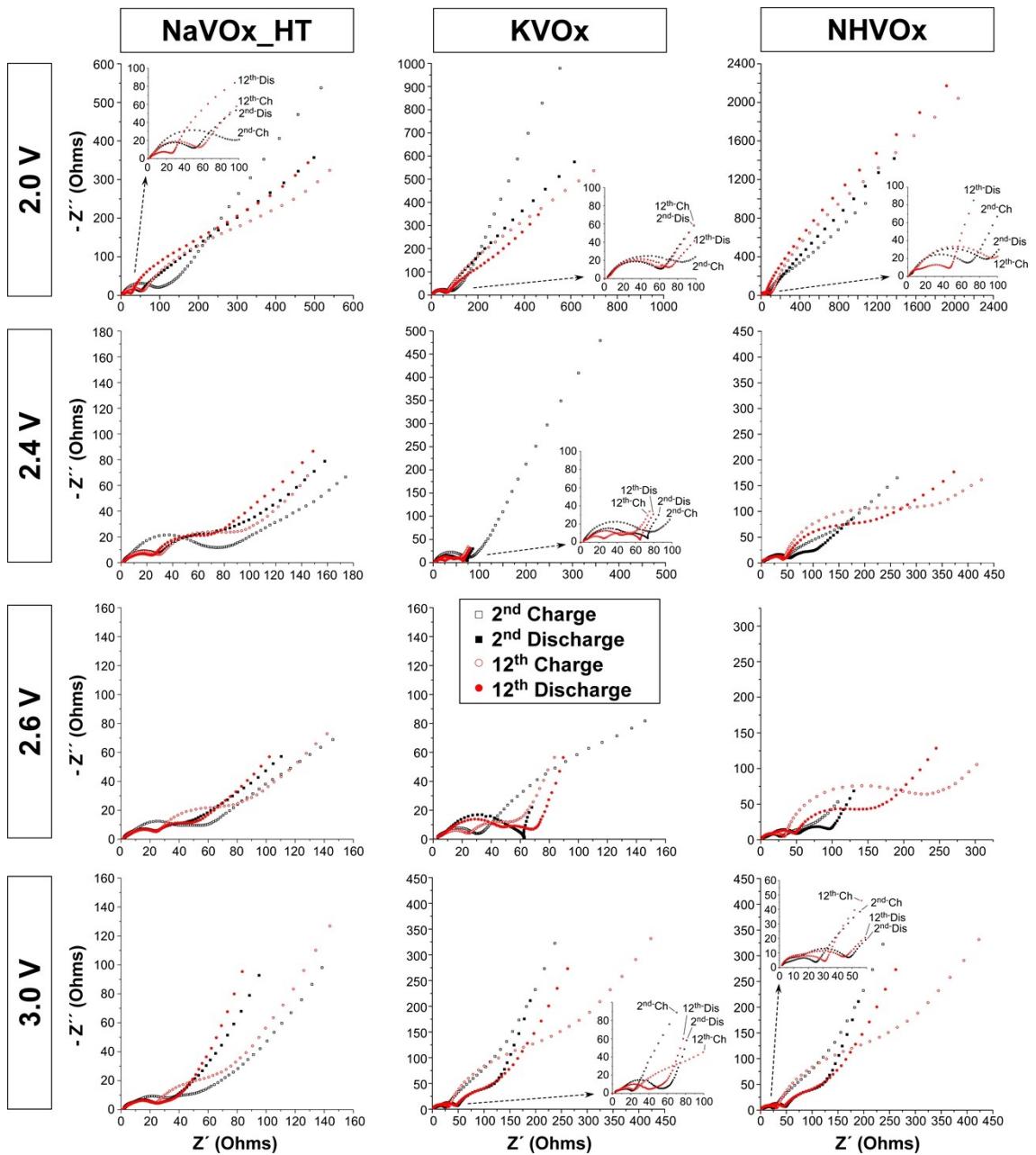


Figure S14. EIS of the phases.

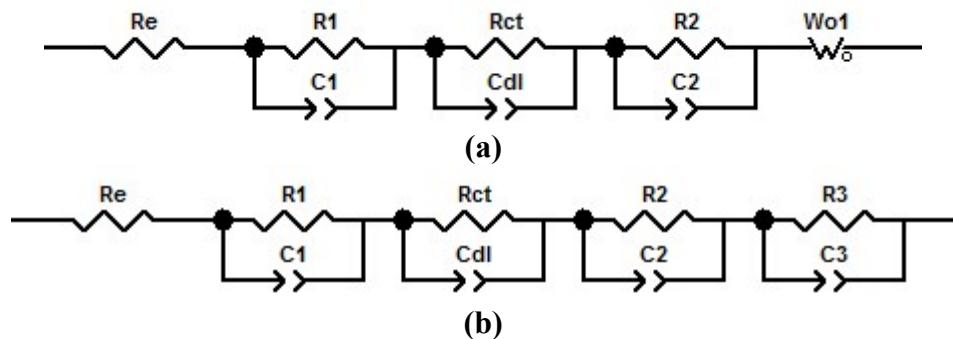


Figure S15. EIS Equivalent circuits for the phases.