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Supporting Information for

The impact of metal cations on photochemical properties of hybrid heterostructures with infinite alkaline-earth metal oxide clusters

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1. Single-crystal X-ray diffraction analysis

Complex	1	2
Empirical formula	C24CaH26N4O9	C24H26N4O9Sr
Formula weight	554.57	602.11
Temperature/K	293(2)	293(2)
Crystal system	tetragonal	tetragonal
Space group	I4 ₁ /a	I4 ₁ /a
a/Å	28.1585(17)	28.614(4)
b/Å	28.1585(17)	28.614(4)
c/Å	13.4560(16)	13.749(3)
$\alpha /^{\circ}$	90	90
β/°	90	90
$\gamma/^{\circ}$	90	90
Volume/Å ³	10669.3(18)	11258(4)
Z	16	16
$\rho_{calc}g/cm^3$	1.381	1.421
µ/mm ⁻¹	0.293	1.970
F(000)	4640	4928.0
Radiation	Mo Ka ($\lambda = 0.71073$)	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/°	3.655 to 25.027	3.130 to 25.023
Reflections collected	23227	42452
Independent reflections	$4698 \ [R_{int} = 0.1125]$	4971 [$R_{int} = 0.0945$]
Data/restraints/parameters	4698/0/303	4971/339/302
Goodness-of-fit on F ²	0.992	1.029
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0702, \ wR_2 = 0.1540$	$R_1 = 0.0563, wR_2 = 0.1491$
Final R indexes [all data]	$R_1=0.1356,wR_2=0.1773$	$R_1 = 0.0869, wR_2 = 0.1652$
Largest diff. peak/hole / e Å ⁻³	0.243/-0.195	0.776/-0.550

Table S1 Crystal data and structure refinements for hybrids 1 and 2.

2. Infrared spectral analysis



Figure S1. Infrared spectrum of hybrid 1



Figure S2. Infrared spectrum of hybrid 2

3. Electron paramagnetic resonance (EPR) spectra



Figure S3. EPR spectra of hybrid 1 before and after irradiation by Xenon Lamp



Figure S4. EPR spectra of hybrid 2 before and after irradiation by Xenon Lamp

4. Powder X-ray diffraction analysis



Figure S5. PXRD patterns of hybrids 1 and 2 before and after the photocatalytic degradation of MB.