

Supporting Information

Fluorination and reduction of CaCrO_3 by topochemical methods

Christian A. Juillerat,^{1, 2*} Yoshihiro Tsujimoto,^{1, 3*} Akira Chikamatsu,⁴ Yuji Masubuchi,⁵ Tetsuya Hasegawa,⁴ Kazunari Yamaura^{1, 3}

¹ *Research Center for Functional Materials, National Institute for Materials Science, 1-1 Namiki, Tsukuba, Ibaraki 305-0044, Japan*

² *Department of Chemistry and Biochemistry, University of South Carolina, Columbia, South Carolina 29208, USA*

³ *Graduate School of Chemical Sciences and Engineering, Hokkaido University, North 13 West 8, Kita-ku, Sapporo 060-0808, Japan*

⁴ *Department of Chemistry, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-0033, Japan*

⁵ *Faculty of Engineering, Hokkaido University, North 13 West 8, Kita-ku, Sapporo 060-8628, Japan*

*Corresponding authors

Email: juillerc@email.sc.edu, TSUJIMOTO.Yoshihiro@nims.go.jp

Table S1. Crystallographic Parameters Refined from Synchrotron X-ray Powder Diffraction Data Collected from the Fluorinated CaCrO_3 at Room Temperature.

atom	site	x	y	z	g	$B_{\text{iso}}/\text{\AA}^2$
Ca	4c	0.0068(4)	0.0411(1)	1/4	1	0.347(14)
Cr	4b	1/2	0	0	1	0.559(11)
X1	8d	0.7096(6)	0.2949(5)	0.0294(4)	1	0.25(3)
X2	4c	0.0825(8)	0.4857(6)	1/4	1	0.25

Space group $Pbnm$ (No. 62): $a = 5.34098(9) \text{ \AA}$, $b = 5.40324(9) \text{ \AA}$, $c = 7.53180(10) \text{ \AA}$. R indices are $R_{\text{wp}} = 2.01\%$, $R_{\text{B}} = 3.73\%$, and $R_{\text{F}} = 2.40\%$. No deficiencies were found at all site occupancies (g). All of the anion sites (X) were assumed to be O. B_{iso} values for X1 and X2 sites were constrained to the same.

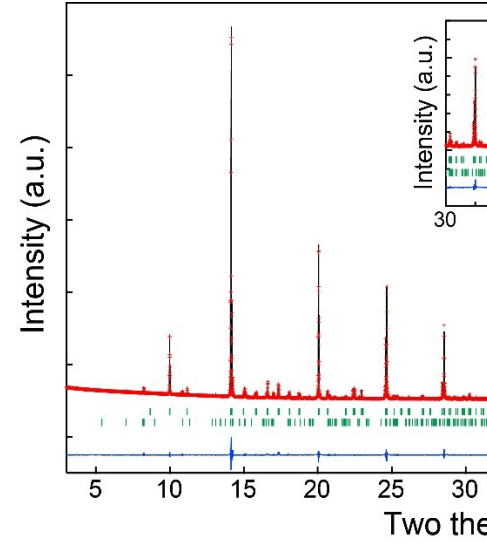


Figure S2. Rietveld refinement against the SXRD data collected from CaCrO_3 at room temperature. The observed (red crosses), calculated (black solid line), and difference (blue solid line) plots are shown. The vertical lines represent CaCrO_3 (87%) and CaCr_2O_4 (13%) from top to bottom. The inset shows an enlarged plot in a high 2θ region.

Table S2. Crystallographic Parameters Refined from Synchrotron X-ray Powder Diffraction Data Collected from CaCrO_3 at Room Temperature.

atom	site	x	y	z	g	$B_{\text{iso}}/\text{\AA}^2$
Ca	4c	-0047(2)	0.03029(8)	1/4	1	0.511(7)
Cr	4b	1/2	0	0	1	0.235(6)
O1	8d	0.7132(2)	0.2863(2)	0.0331(2)	1	0.31(2)
O2	4c	0.0611(3)	0.4896(3)	1/4	1	0.38(3)

Space group $Pbnm$ (No. 62): $a = 5.28912(1) \text{ \AA}$, $b = 5.31796(1) \text{ \AA}$, $c = 7.48677(1) \text{ \AA}$. R indices are $R_{\text{wp}} = 4.06\%$, $R_{\text{B}} = 3.21\%$, and $R_{\text{F}} = 2.40\%$. All site occupancies were fixed to unity.

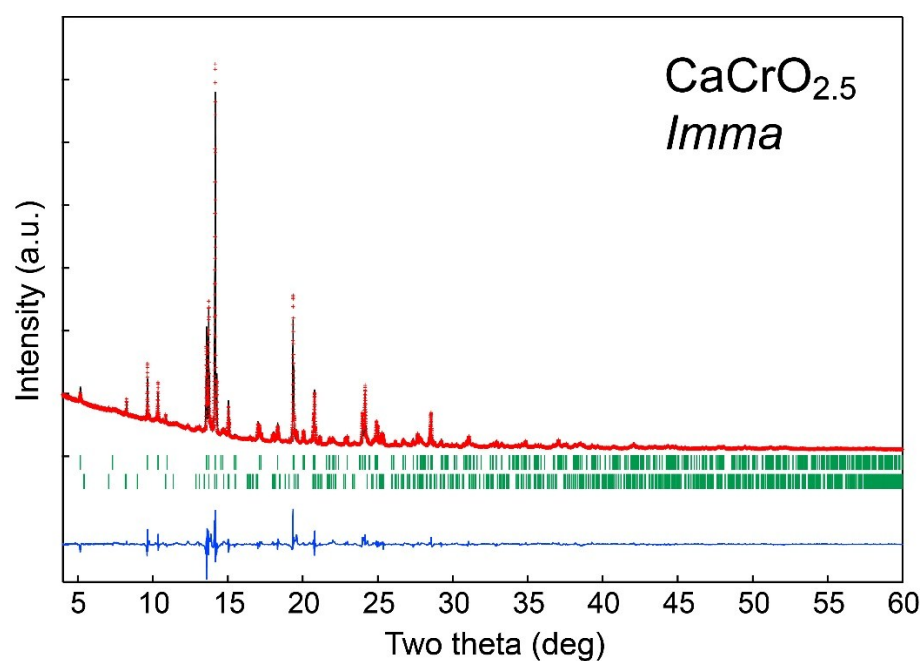


Figure S3. Rietveld refinement against the SXRD data collected from $\text{CaCrO}_{2.5}$ at room temperature. The observed (red crosses), calculated (black solid line), and difference (blue solid line) plots are shown. The vertical lines represent $\text{CaCrO}_{2.5}$ (86%) and CaCr_2O_4 (14%) from top to bottom.

Table S3. Crystallographic Parameters Refined from Synchrotron X-ray Powder Diffraction Data Collected from $\text{CaCrO}_{2.5}$ at Room Temperature.

atom	site	x	y	z	g	$B_{\text{iso}}/\text{\AA}^2$
Ca	8h	0	0.61380(15)	0.4980(7)	1	1.19(5)
Cr1	4a	0	0	0	1	0.59(5)
Cr2	8i	0.4273(4)	1/4	0.5114(8)	0.5	0.17(7)

O1	8g	1/4	0.0065(7)	1/4	1	0.94(9)
O2	8h	0	0.1399(3)	0.0608(9)	1	0.94
O3	8i	0.197(2)	1/4	0.673(2)	0.5	0.94

Space group *Imma* (No. 74): $a = 5.52141(6)$ Å, $b = 14.48419(13)$ Å, $c = 5.46196(5)$ Å. R indices are $R_{\text{wp}} = 6.20\%$, $R_{\text{B}} = 9.83\%$, and $R_{\text{F}} = 6.93\%$. All site occupancies were fixed to unity or half. B_{iso} values for O1, O2, and O3 sites were constrained to the same value.

Table S4. Bond Valence Sums for different anion ordering models.

Structure	Ca	Cr	X1	X2
CaCrO ₃	2.28	3.98	2.08	2.11
Expected values	2	3.5	2	2
CaCrO _{2.5} F _{0.5} (Full anion disorder)	2.15	3.34	1.8	1.88
Expected values	2	3.5	1.83	1.83
CaCrO _{2.5} F _{0.5} (Occupation of X1 by fluorine)	2.14	3.34	1.75	1.98
Expected values	2	3.5	1.75	2
CaCrO _{2.5} F _{0.5} (Occupation of X2 by fluorine)	1.89	3.34	1.68	1.89
Expected values	2	3.5	1.5	2

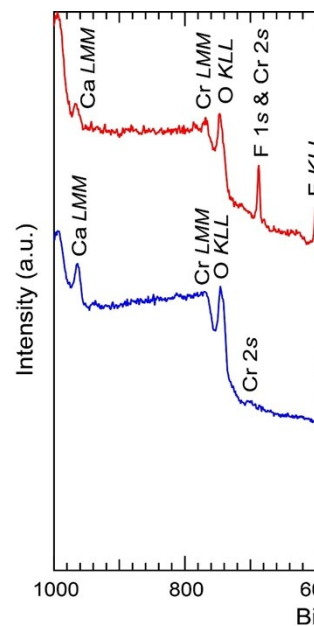


Figure S3. Wide scan X-ray

photoelectron spectra collected from CaCrO_3 and its fluorinated phase at 300 K.

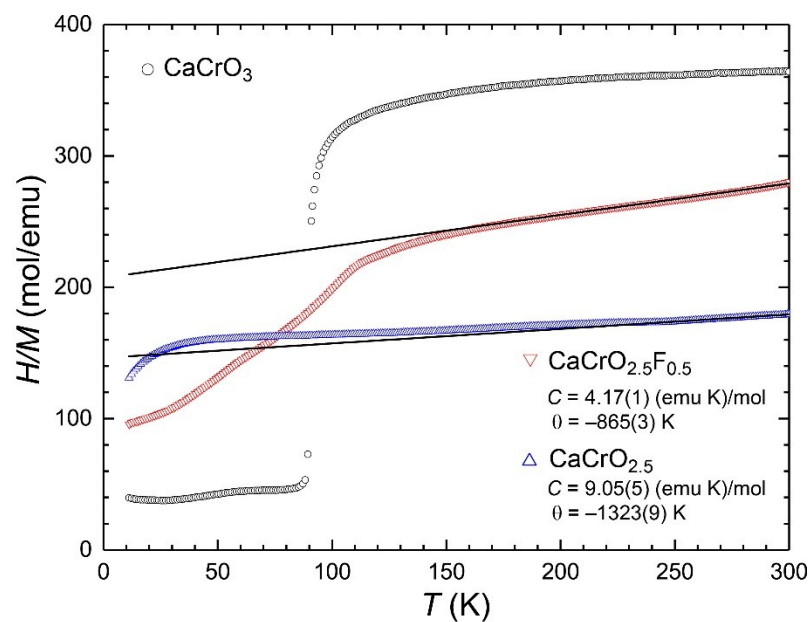


Figure S4. Temperature dependence of the inverse susceptibility of CaCrO_3 , $\text{CaCrO}_{2.5}$, and $\text{CaCrO}_{2.5}\text{F}_{0.5}$. The Curie-Weiss fit to the data for $\text{CaCrO}_{2.5}$ and $\text{CaCrO}_{2.5}\text{F}_{0.5}$ gave physically meaningless Weiss temperatures and much higher Curie constants than those expected from the oxidation states of chromium ions.