

Electronic Supplementary Information

Synthesis and structures of mono- and di-nuclear aluminium and zinc complexes bearing α -diimine and related ligands, and their use in the ring opening polymerization of cyclic esters

Lin Xiao,^a Yanxia Zhao,^{a*} Sijie Qiao,^a Ziyue Sun,^a Orlando Santoro^b and Carl Redshaw^{a,b*}

^a College of Chemistry and Material Science, Northwest University, 710069 Xi'an, China

^b Department of Chemistry & Biochemistry, The University of Hull, Cottingham Rd, Hull, HU6 7RX, U.K.

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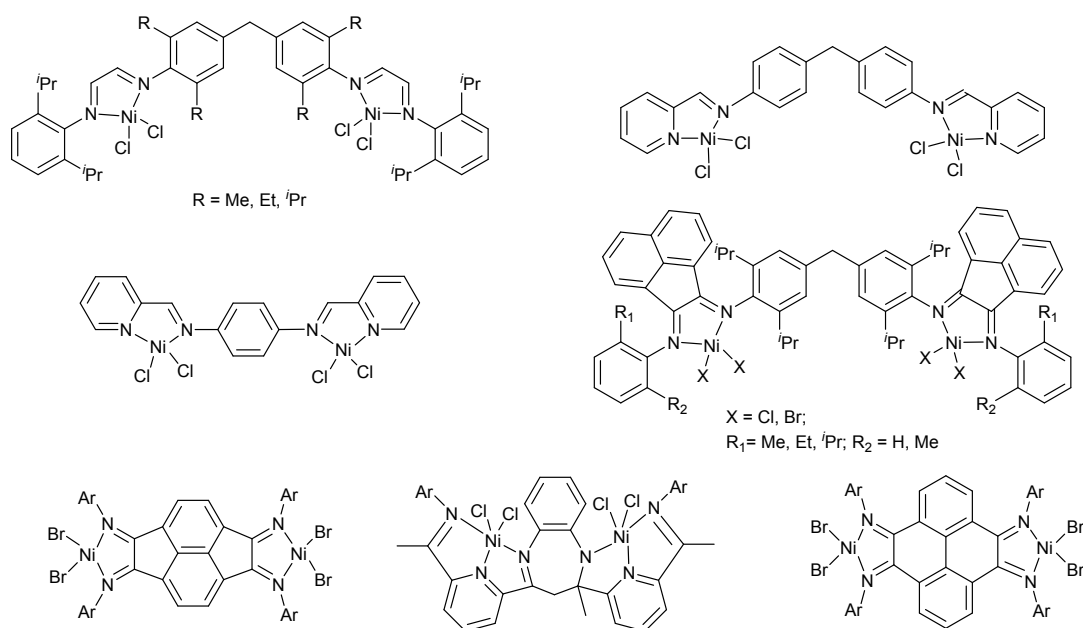
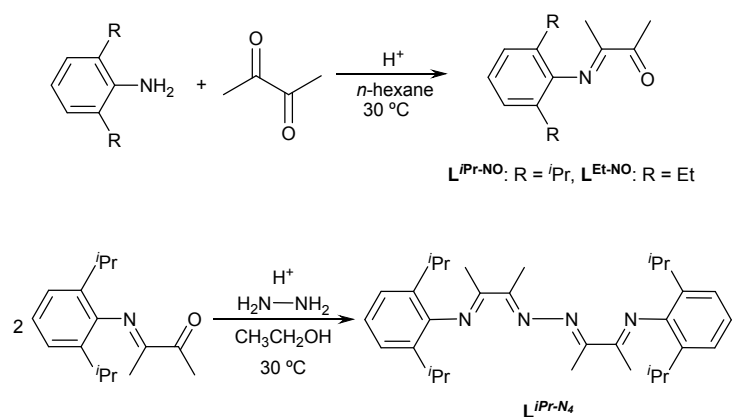


Chart S1. Imine-based frameworks utilized in bi-nickel-based olefin polymerization chemistry. [1]



Scheme S1. Synthesis of the ligands $\text{L}^{i\text{Pr-NO}}$, $\text{L}^{\text{Et-NO}}$ and $\text{L}^{i\text{Pr-N}_4}$.

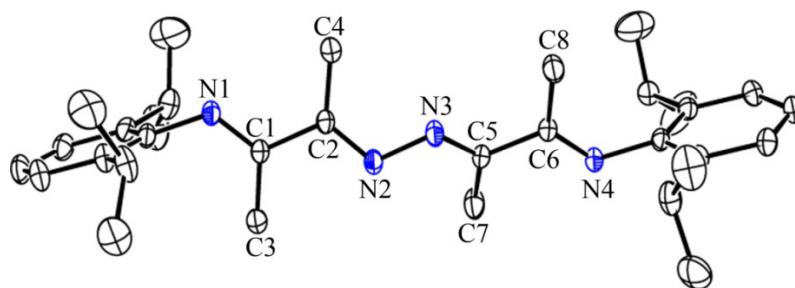


Figure S1. The molecular structure of $\text{L}^{i\text{Pr-N}_4}$ (thermal ellipsoids are set at the 20% probability level; H atoms are omitted for clarity): C1–N1 1.269(3), C1–C2 1.498(3), C1–C3 1.497(4), C2–N2 1.277(3), C2–C4 1.483(4), N2–N3 1.405(3), C5–N3 1.279(3), C5–C7 1.479(4), C5–C6 1.496(3), C6–N4 1.274(3), C6–C8 1.496(4); N1–C1–C2 116.0(2), C1–C2–N2 115.6(2), C2–N2–N3 115.2(2), N2–C3–N5 114.7(2), N3–C5–C6 115.6(2), C5–C6–N4 116.2(2).

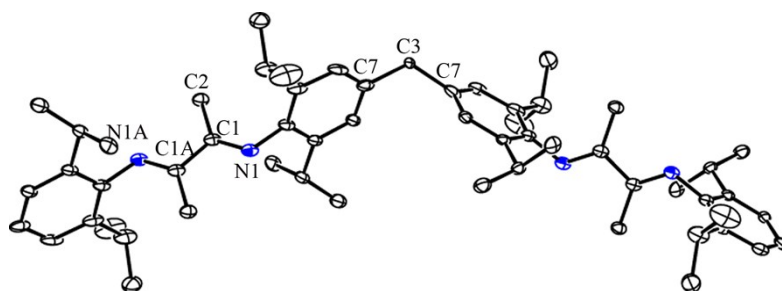


Figure S2. The molecular structure of $L^{iPr-N_2-ArCH_2Ar-N_2}$ (thermal ellipsoids are set at the 20% probability level; H atoms are omitted for clarity): C1–N1 1.291(4), C1–C2 1.505(4), C1–C1A 1.518(4), N1–C1–C2 126.5(3), N1–C1–C1 116.4(4), C2–C1–C1 117.1(4), C7–C3–C7 121.7(5).

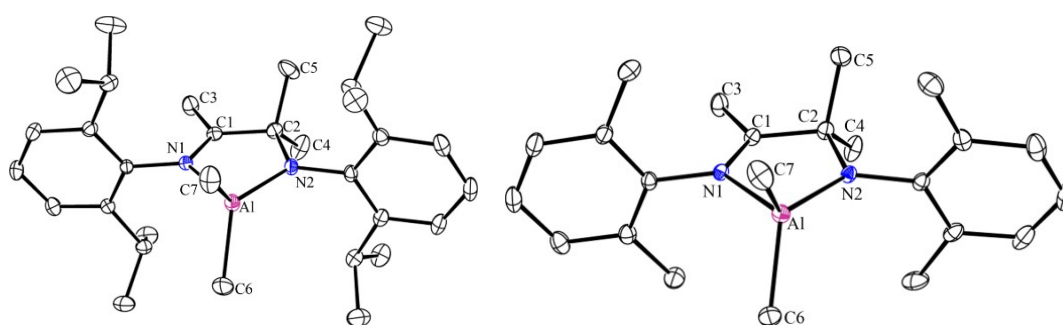


Figure S3. The molecular structure of **1** (left), **3** (right) (thermal ellipsoids are set at the 20% probability level; H atoms are omitted for clarity).

Theoretical Calculations

Structure optimization for the model compounds $[ArN-C(Me)_2C(Me)=NAlMe_2]_2$ (**4**) and its isomers **4a** and **4b**, were carried out at the DFT (B3LYP) level with a 6-31G*^{2,3} basis set using the Gaussian 09 program.⁴ Figure S4 shows the optimized geometries. Here DFT results show that **4** exhibits a slightly lower energy than the imino-amido isomer **4a** ($\Delta E = 18.0 \text{ kJ mol}^{-1}$) and **4b** (30.3 kJ mol^{-1}).

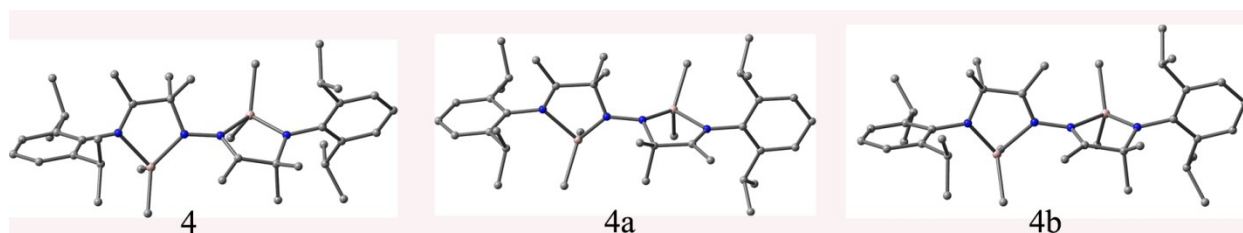


Figure S4. Optimized structures of **4** and its isomers **4a** and **4b**.

Table S1. Relative energies of compounds **4** and **4a** and **4b**.

Compound	Energy (E, a.u.)	ΔE (kJ/mol)
4	-2190.58568031	0
4a	-2190.57882952	18.0
4b	-2190.57413253	30.3

Table S2. Crystallographic data and refinement details for compounds L^{iPr-N4} , **1–3**.

Compound	L^{iPr-N4}	1	2	3
Empirical formula	$C_{32}H_{46}N_4$	$C_{31}H_{49}AlN_2$	$C_{27}H_{41}AlN_2$	$2^*C_{23}H_{33}AlN_2$
Fw	486.73	476.70	420.60	728.98
Crystal system	Triclinic	Triclinic	monoclinic	monoclinic
Space group	$P-1$	$P-1$	Cc	$P2_1/c$
$a / \text{\AA}$	10.030(4)	8.4984(18)	9.425(2)	14.239(13)
$b / \text{\AA}$	12.557(4)	9.799(2)	37.074(9)	8.148(3)
$c / \text{\AA}$	13.499(5)	19.426(4)	8.102(2)	38.13(2)
$\alpha / ^\circ$	66.078(4)	82.115(3)	90	90
$\beta / ^\circ$	84.299(5)	80.860(2)	111.579(3)	91.22(3)
$\gamma / ^\circ$	82.021(5)	69.630(2)	90	90
$V / \text{\AA}^3$	1537.3(9)	1491.4(5)	2632.9(11)	4423(5)
Z	2	2	4	4
$D_{\text{calc}} / \text{g cm}^{-3}$	1.051	1.062	1.061	1.095
$F(000)$	532	524	920	1584
μ / mm^{-1}	0.062	0.088	0.092	0.100
θ range	1.785–24.970	2.132–24.997	2.197–24.958	2.719–25.242
Reflns collected	9657	9538	8102	21974
Independent reflns	5712	5095	8102	7701
Reflns [$I > 2\sigma(I)$]	3705	3943	7632	5035
R_{int}	0.0326	0.0255	0.0285	0.0705
$R_1; wR_2$ [$I > 2\sigma(I)$]	0.0746; 0.1580	0.0608; 0.1741	0.0396; 0.1134	0.0589; 0.1038
$R_1; wR_2$ (all data)	0.0981; 0.1694	0.0782; 0.1873	0.0426; 0.1163	0.1043; 0.1256
GOF (F^2)	1.042	1.155	1.023	1.121

Table S3. Crystallographic data and refinement details for compounds 4–7.

Compound	4	5	6	7·0.5toluene
Empirical formula	C ₃₈ H ₆₄ Al ₂ N ₄	C ₃₄ H ₅₆ Al ₂ N ₂ O ₂	C ₃₂ H ₅₀ ZnN ₂	C ₅₄ H ₈₃ Zn ₃ N ₃ O ₃ • 0.5toluene
Fw	630.89	578.76	528.11	1064.49
Crystal system	Monoclinic	Monoclinic	monoclinic	Orthorhombic
Space group	P21/c	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>c</i>	Pca21
<i>a</i> /Å	13.900(5)	8.505(3)	8.5697(19)	28.2584(16)
<i>b</i> /Å	18.370(6)	13.440(5)	19.226(4)	15.9307(9)
<i>c</i> /Å	16.078(6)	14.919(7)	18.509(4)	13.3344(6)
α /°	90	90	90	90
β /°	106.950(14)	94.584(13)	93.238(8)	90
γ /°	90	90	90	90
<i>V</i> /Å ³	3927(2)	1699.8(11)	3044.6(11)	6002.8(6)
<i>Z</i>	4	2	4	4
<i>D</i> _{calc} /g cm ⁻³	1.067	1.131	1.152	1.127
<i>F</i> (000)	1384	632	1144	2168
μ /mm ⁻¹	0.103	0.116	0.827	1.227
θ range	1.727–24.997	2.669–25.342	2.388 – 26.364	2.459–26.389
Reflns collected	37227	18303	17326	39300
Independent reflns	6900	3091	6030	11494
Reflns [<i>I</i> > 2 σ (<i>I</i>)]	5479	2722	322	9673
<i>R</i> _{int}	0.0720	0.0285	0.0351	0.0485
<i>R</i> ₁ ; <i>wR</i> ₂ [<i>I</i> > 2 σ (<i>I</i>)]	0.0714; 0.2088	0.0330; 0.0731	0.0518; 0.1020	0.0399; 0.0749
<i>R</i> ₁ ; <i>wR</i> ₂ (all data)	0.0877; 0.2088	0.0394; 0.0770	0.0696; 0.1093	0.0527; 0.0796
GOF (<i>F</i> ²)	1.005	1.029	1.130	1.038

Table S4. Crystallographic data and refinement details for compounds **LⁱPr-N₂-ArCH₂Ar-N₂** and **8**.

Compound	LⁱPr-N₂-ArCH₂Ar-N₂	8 ·3.5toluene
Empirical formula	C ₅₇ H ₈₀ N ₄	C ₅₇ H ₈₀ Cl ₄ N ₄ Zn ₂ ·3.5 toluene
Fw	821.25	1410.59
Crystal system	monoclinic	monoclinic
Space group	C2/c	P21/n
<i>a</i> /Å	16.4092(9)	13.65(3)
<i>b</i> /Å	9.5095(9)	25.08(5)
<i>c</i> /Å	17.0073(10)	24.33(5)
<i>α</i> /°	90	90
<i>β</i> /°	108.349(3)	92.47(7)
<i>γ</i> /°	90	90
<i>V</i> /Å ³	2518.9(3)	8319(31)
<i>Z</i>	2	4
<i>D</i> _{cal} /g cm ⁻³	1.083	0.873
<i>F</i> (000)	900	2312
<i>μ</i> /mm ⁻¹	0.062	0.732
<i>θ</i> range	2.51–25.10	2.32–21.79
Reflns collected	10737	14853
Independent reflns	5944	9937
Reflns [<i>I</i> > 2σ(<i>I</i>)]	1603	7649
<i>R</i> _{int}	0.0463	0.0819
<i>R</i> ₁ ; <i>wR</i> ₂ [<i>I</i> > 2σ(<i>I</i>)]	0.0936; 0.1672	0.0740; 0.1906
<i>R</i> ₁ ; <i>wR</i> ₂ (all data)	0.1268; 0.1849	0.1308; 0.2107
GOF (<i>F</i> ²)	1.058	1.024

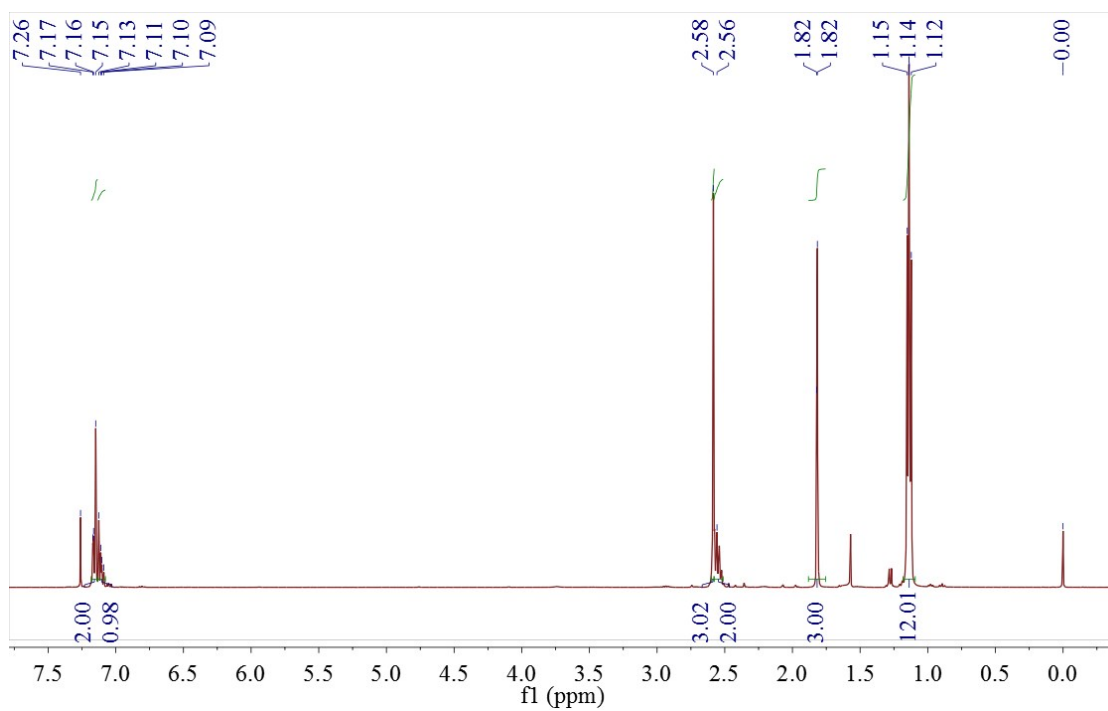


Figure S5. ^1H NMR (400 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{iPr-NO}}$.

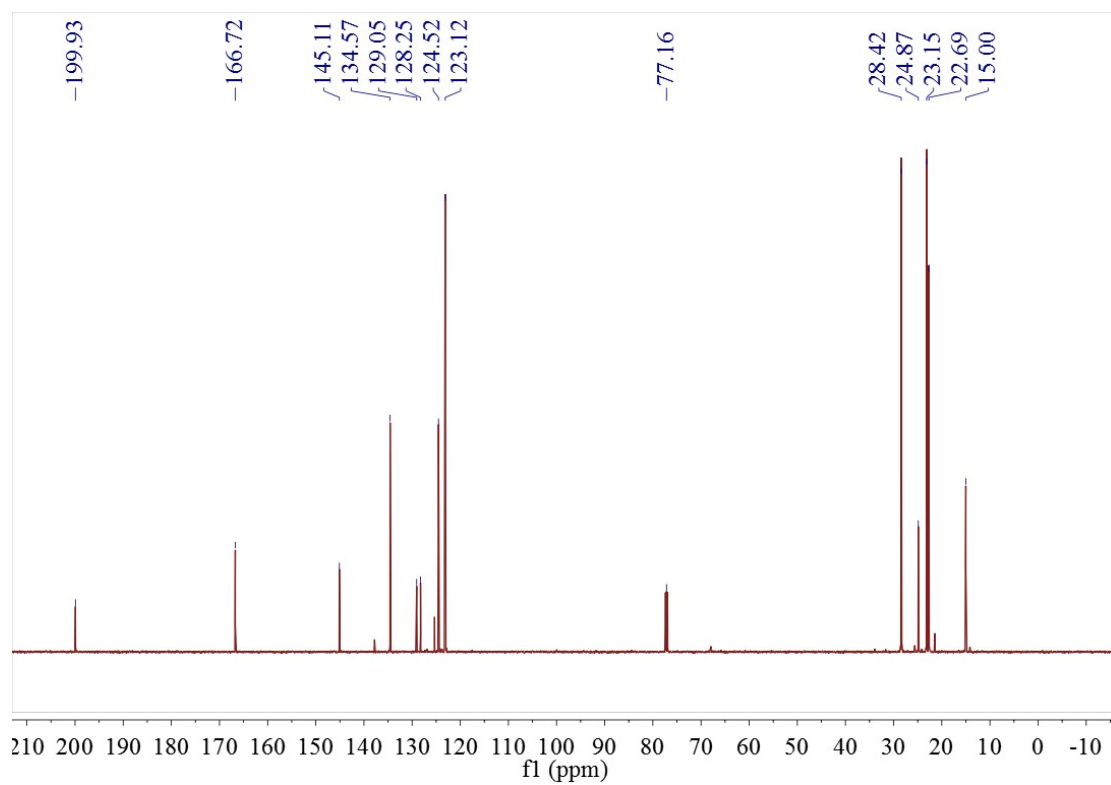
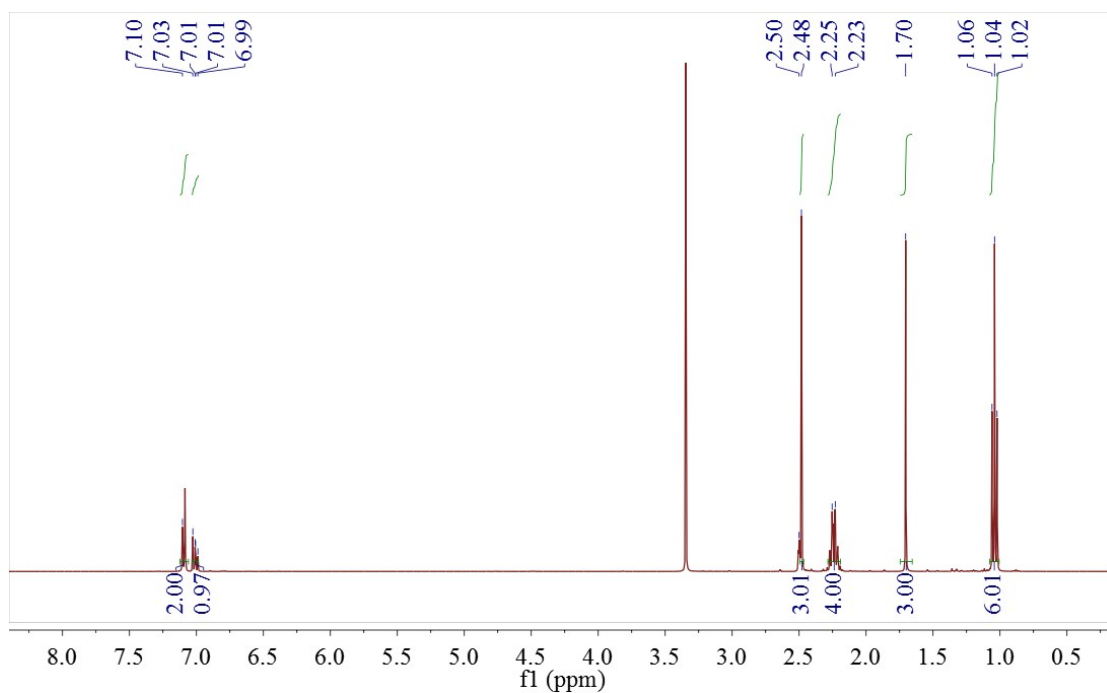


Figure S6. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{iPr-NO}}$.



Fig

ure S7. ^1H NMR (400 MHz DMSO 298K) spectrum for ligand $\text{L}^{\text{Et-NO}}$.

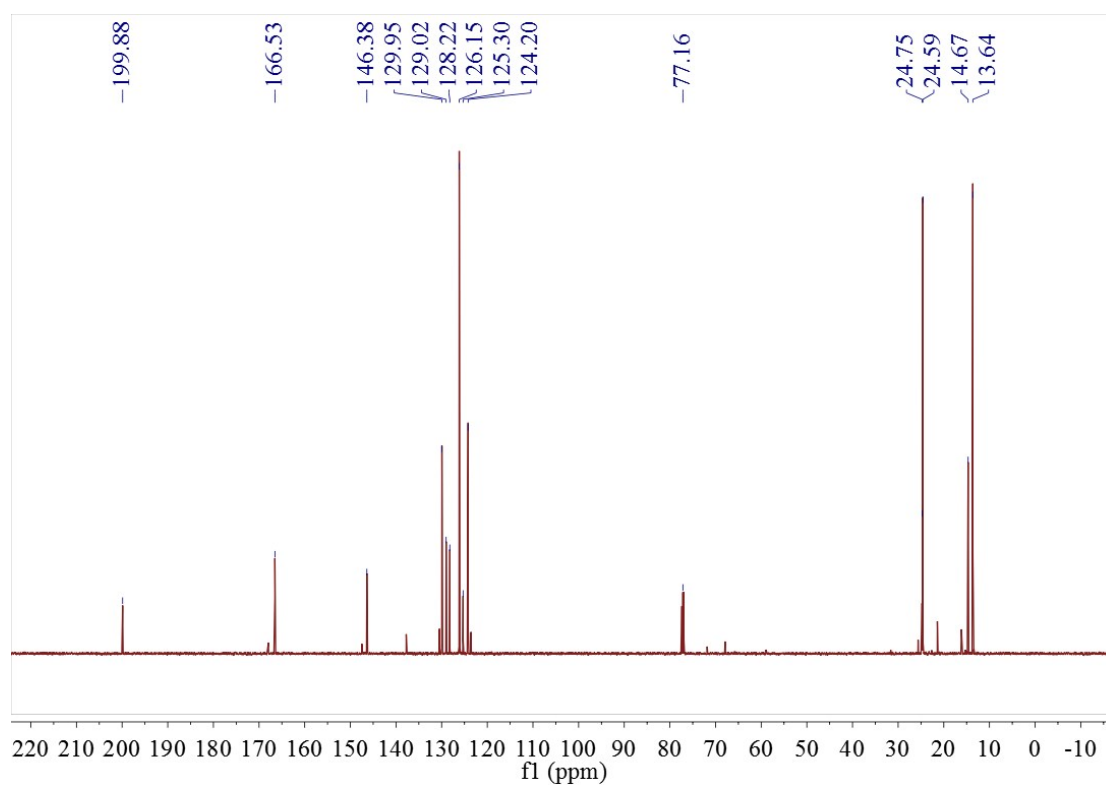
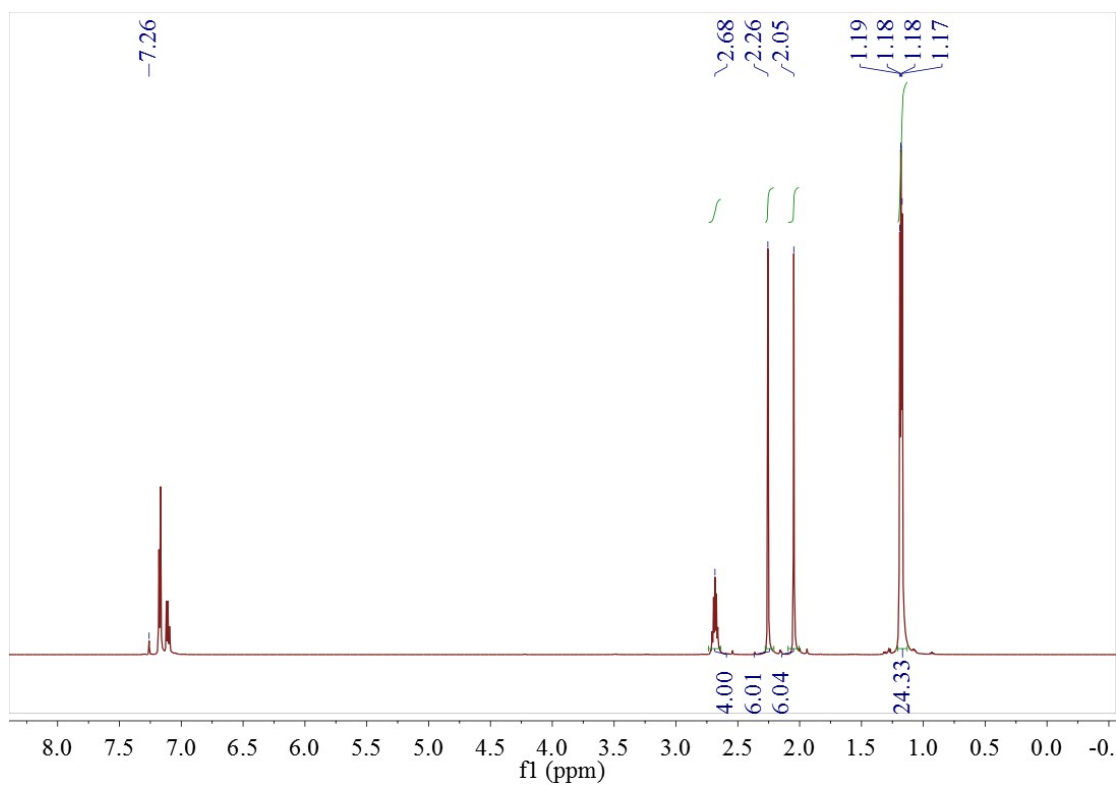


Figure S8. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{Et-NO}}$.



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ure S9. ^1H NMR (400 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{ipr-N}_4}$.

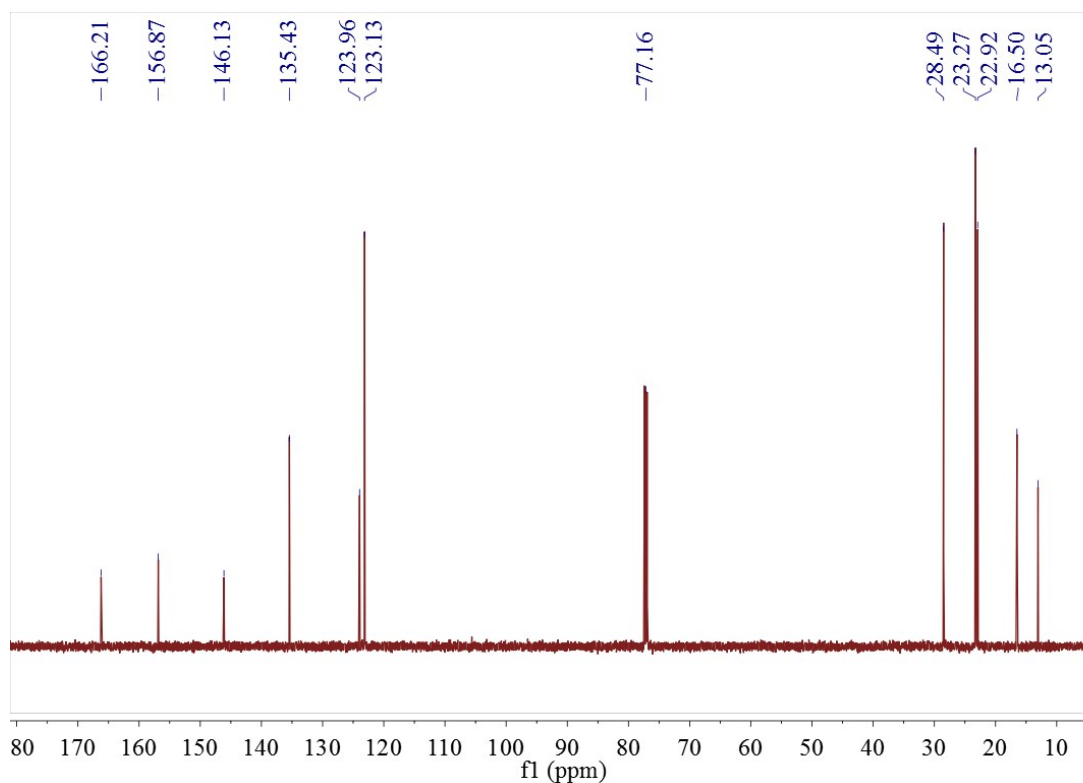


Figure S10. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{ipr-N}_4}$.

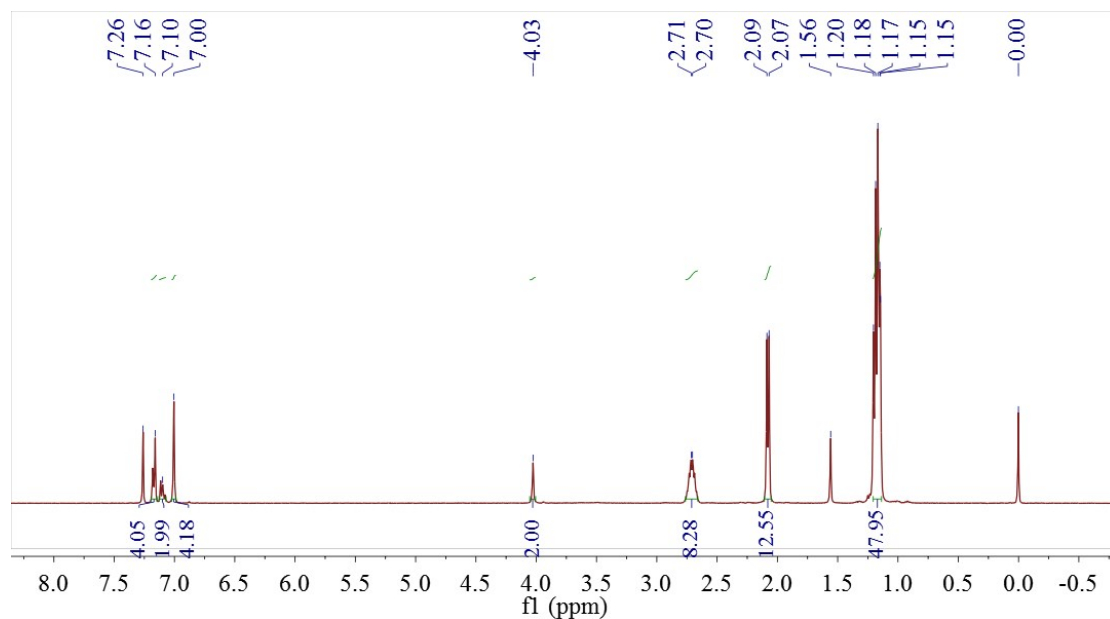


Figure S11. ^1H NMR (400 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{iPr-N}_2\text{-ArCH}_2\text{Ar-N}_2}$

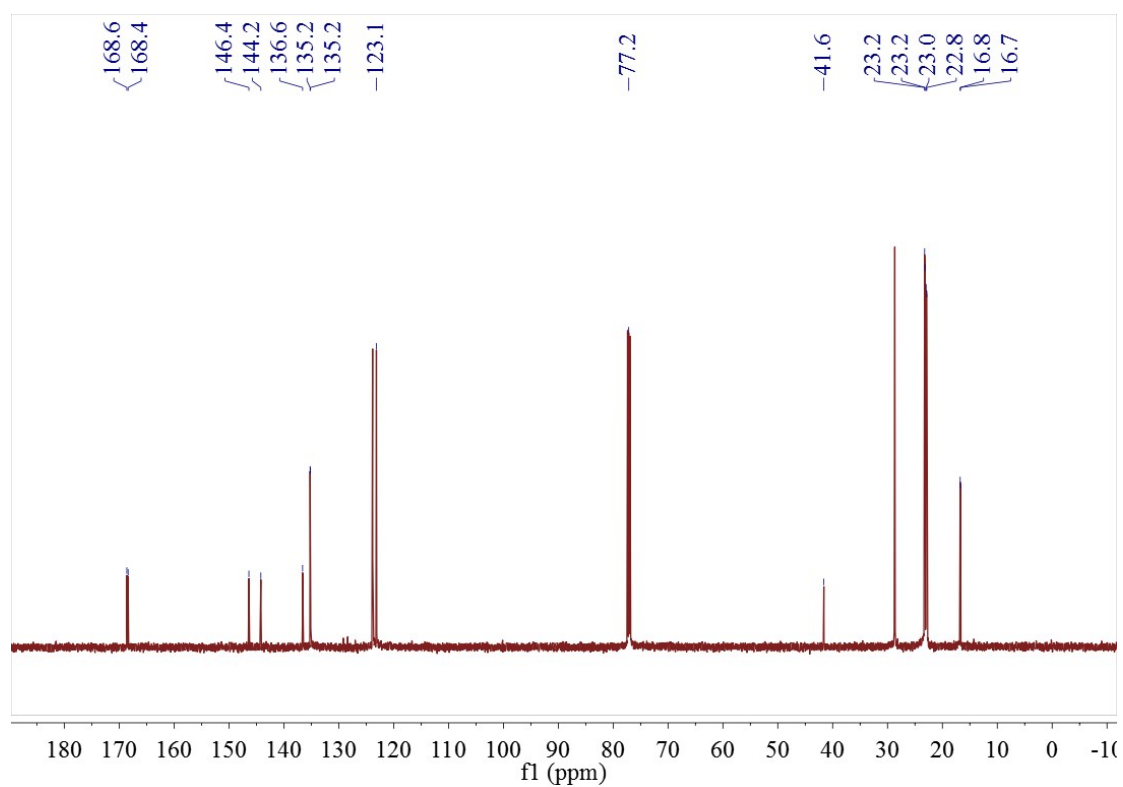
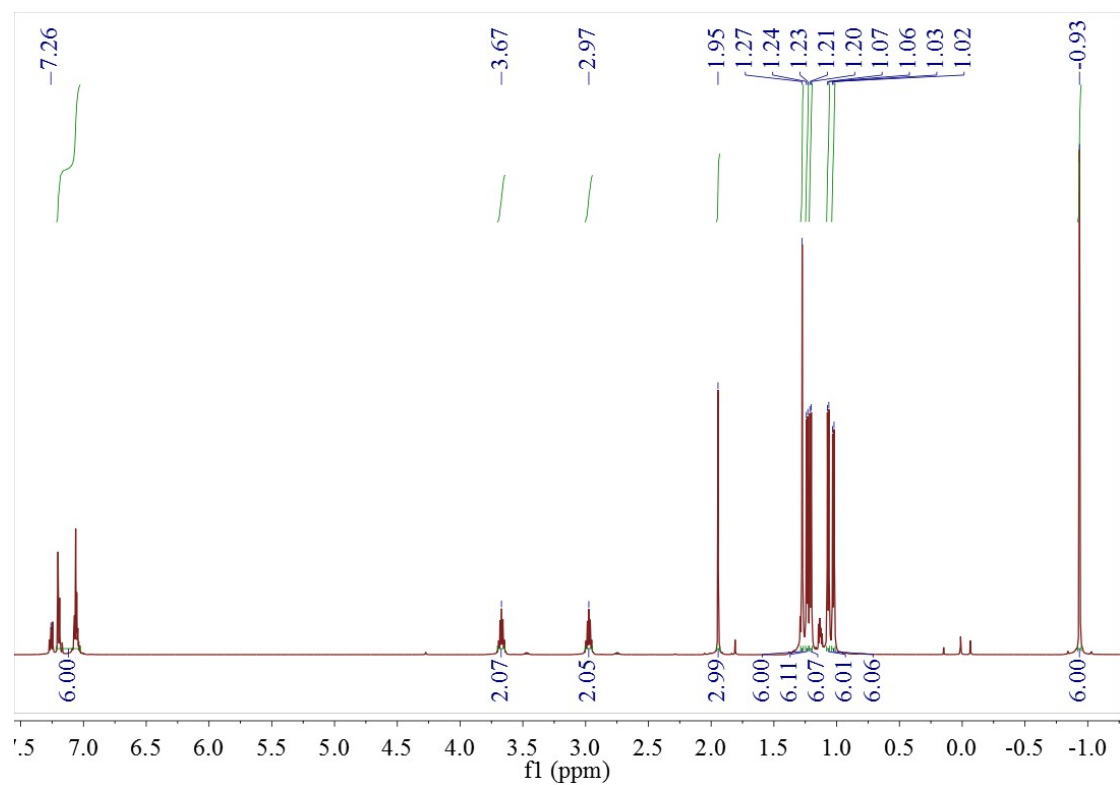


Figure S12. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for ligand $\text{L}^{\text{iPr-N}_2\text{-ArCH}_2\text{Ar-N}_2}$



ure S13. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound **1**.

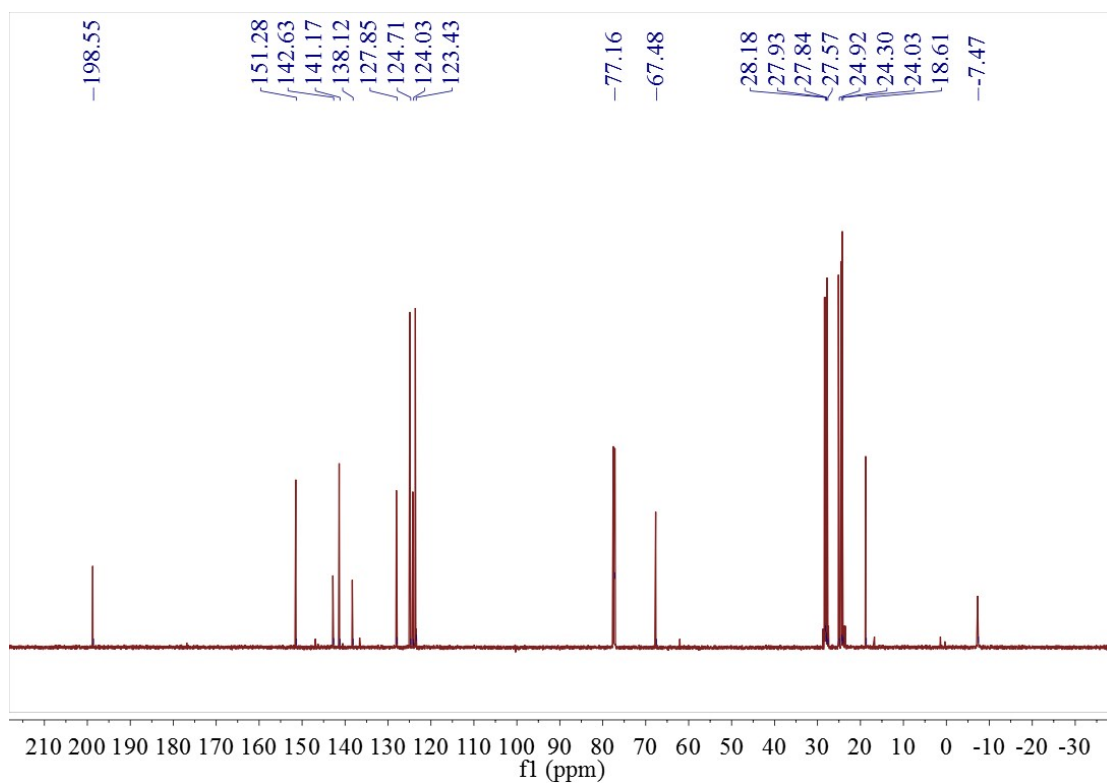


Figure S14. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound **1**.

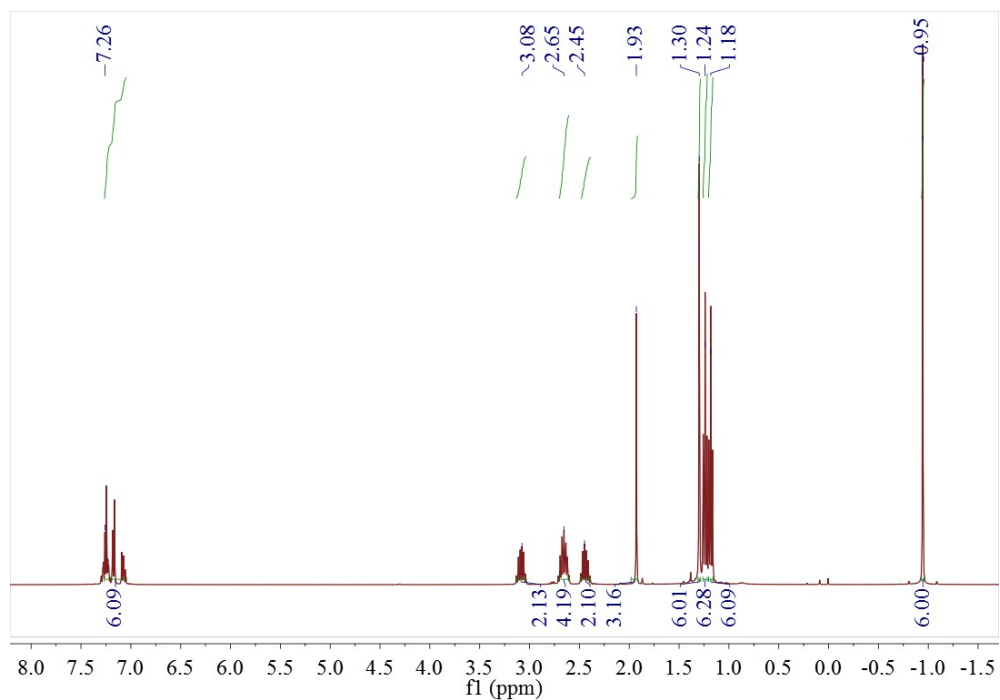


Figure S15. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound 2.

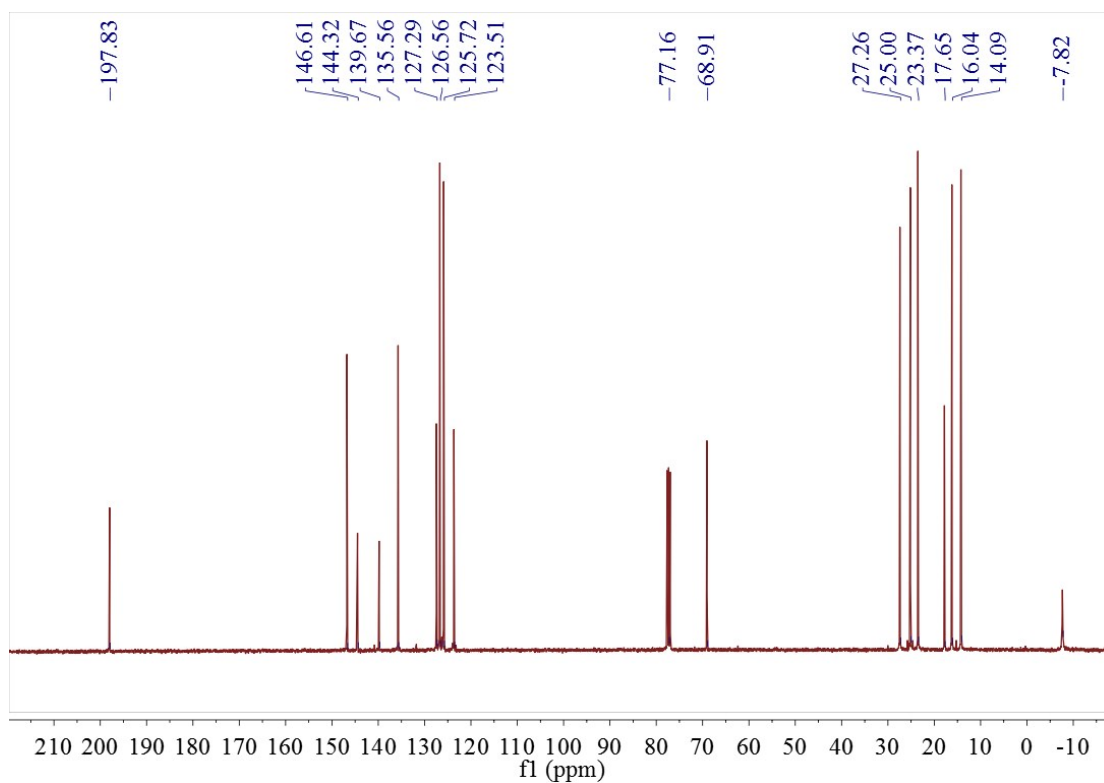
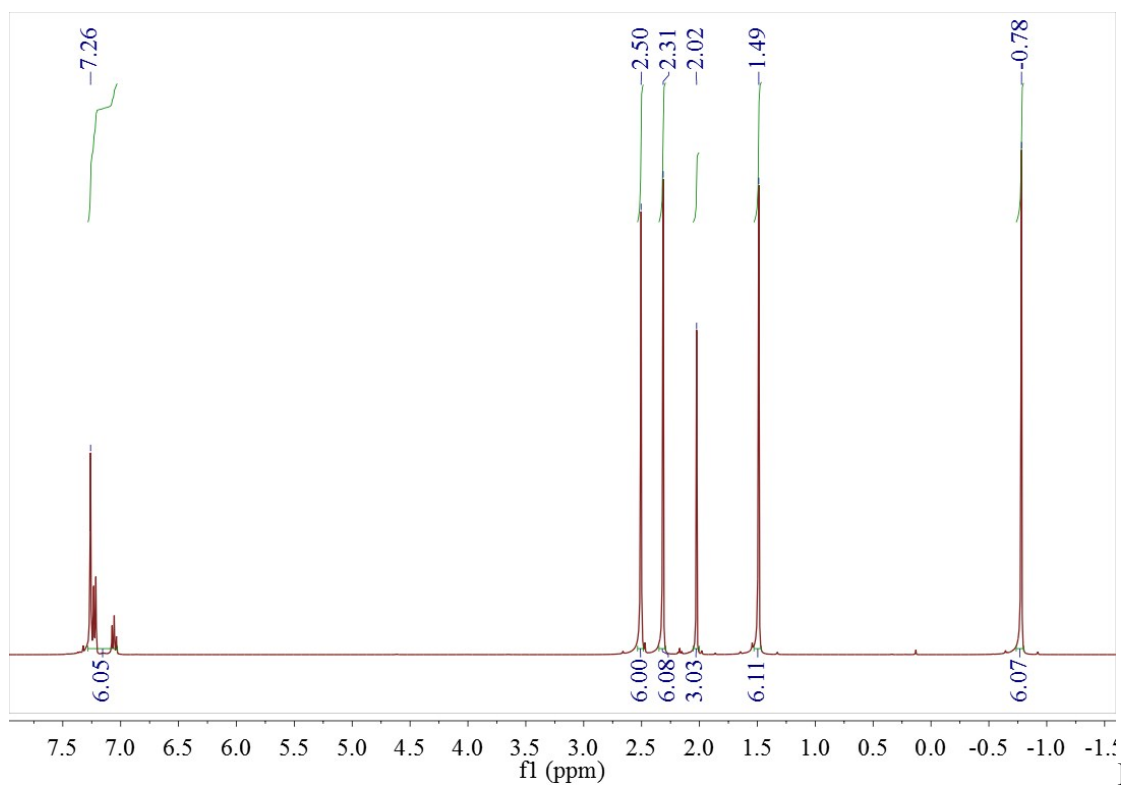


Figure S16. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound 2.



Fig

ure S17. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound **3**.

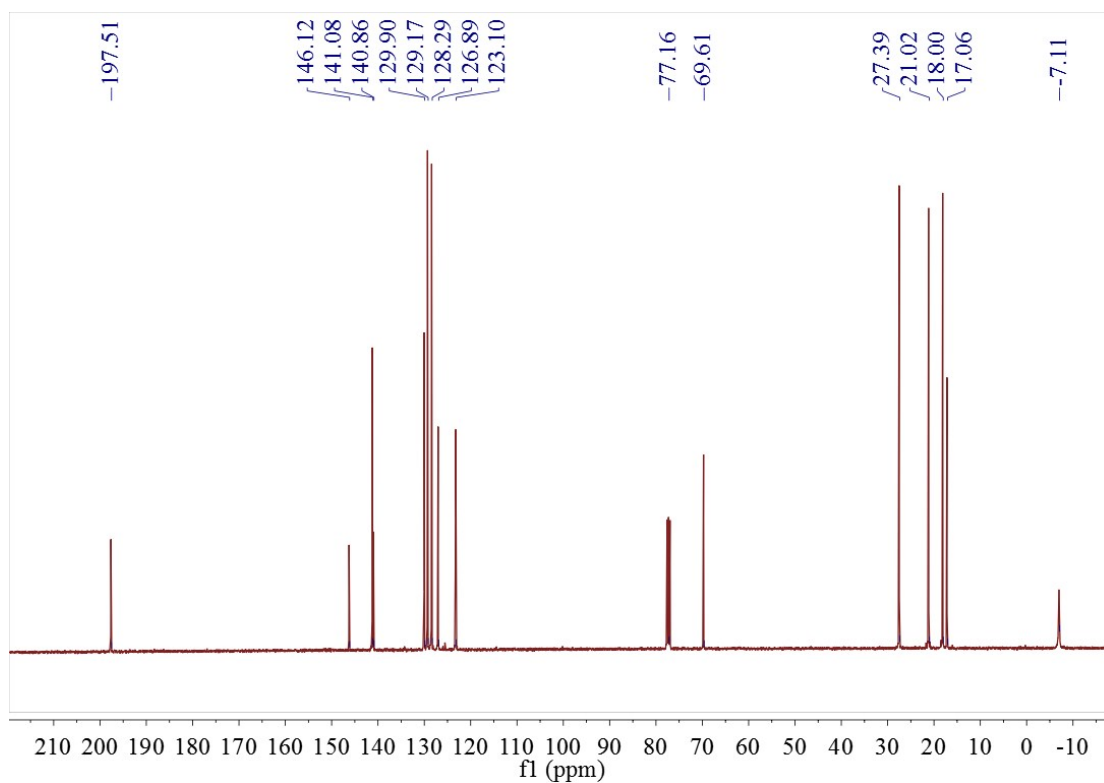


Figure S18. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound **3**.

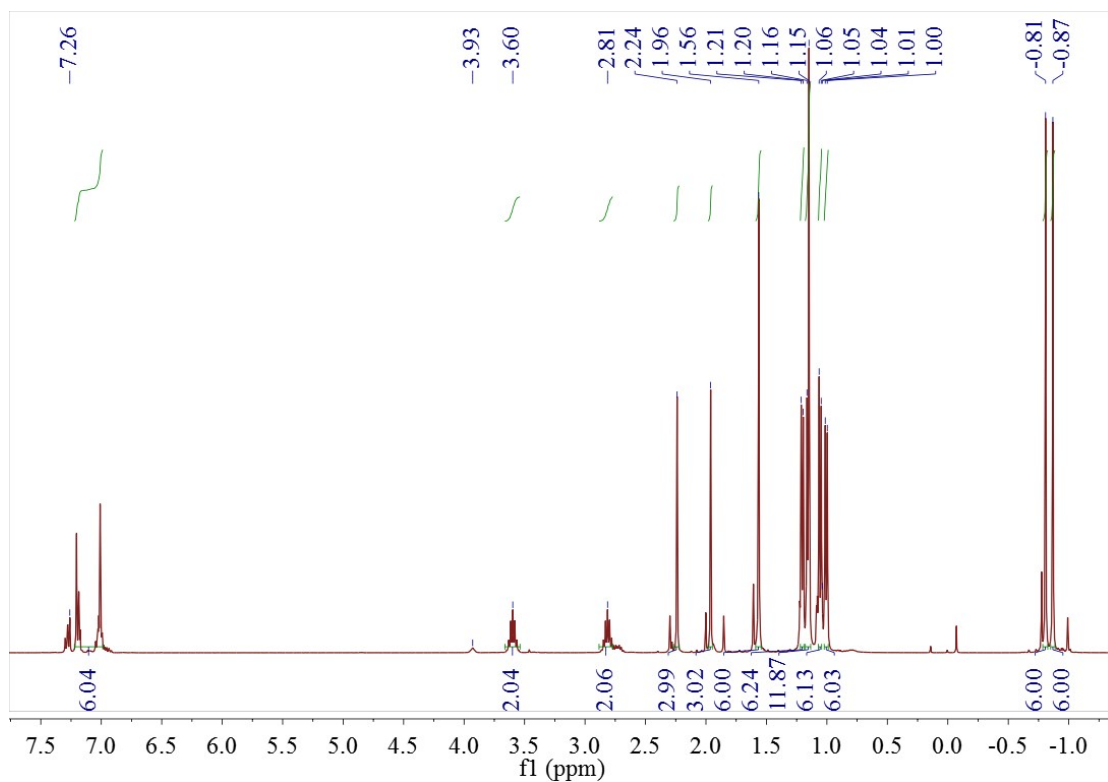


Figure S19. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound 4.

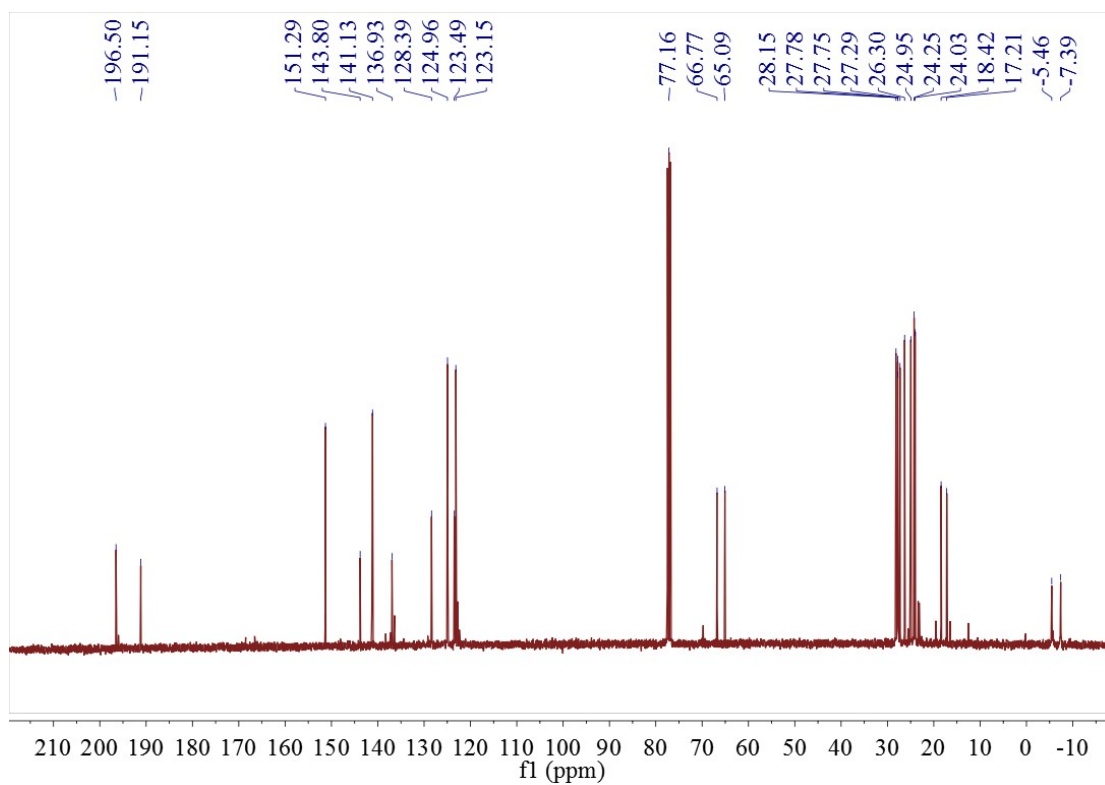
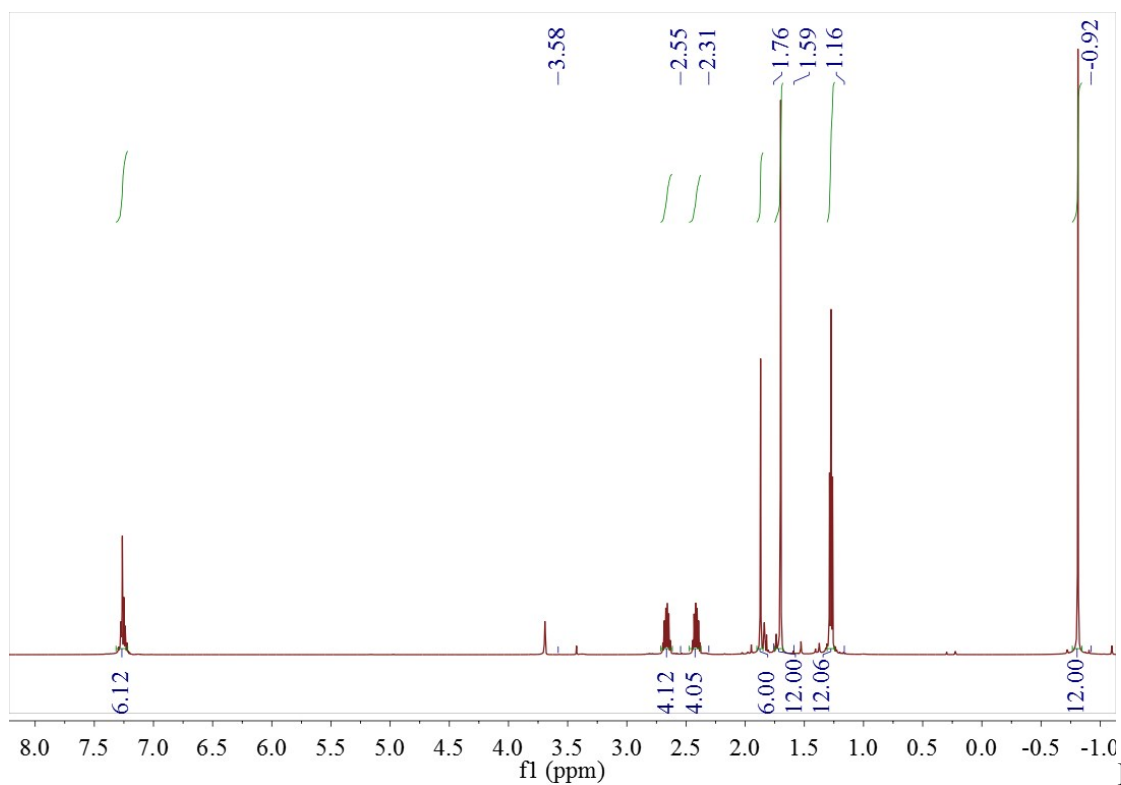


Figure S20. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound 4.

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Figure S21. ^1H NMR (400 MHz THF-d8 298K) spectrum for compound **5**.

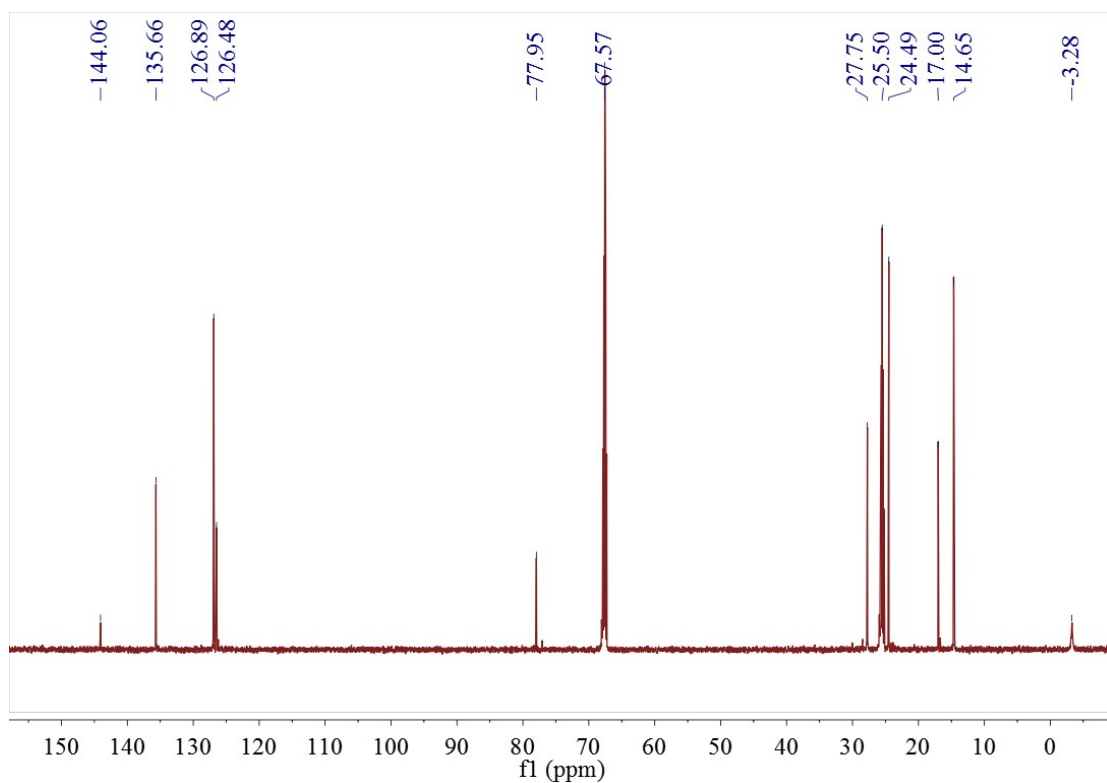


Figure S22. ^{13}C NMR (100.6 MHz THF-d8 298K) spectrum for compound **5**.

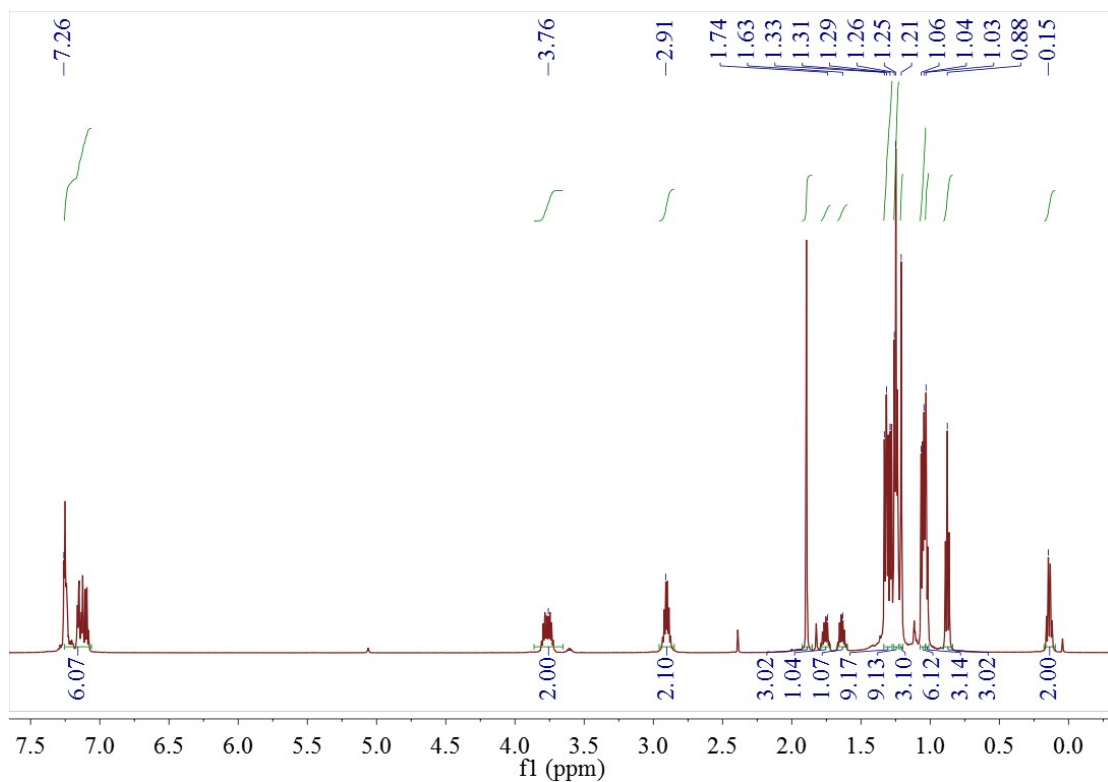


Figure S23. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound **6**.

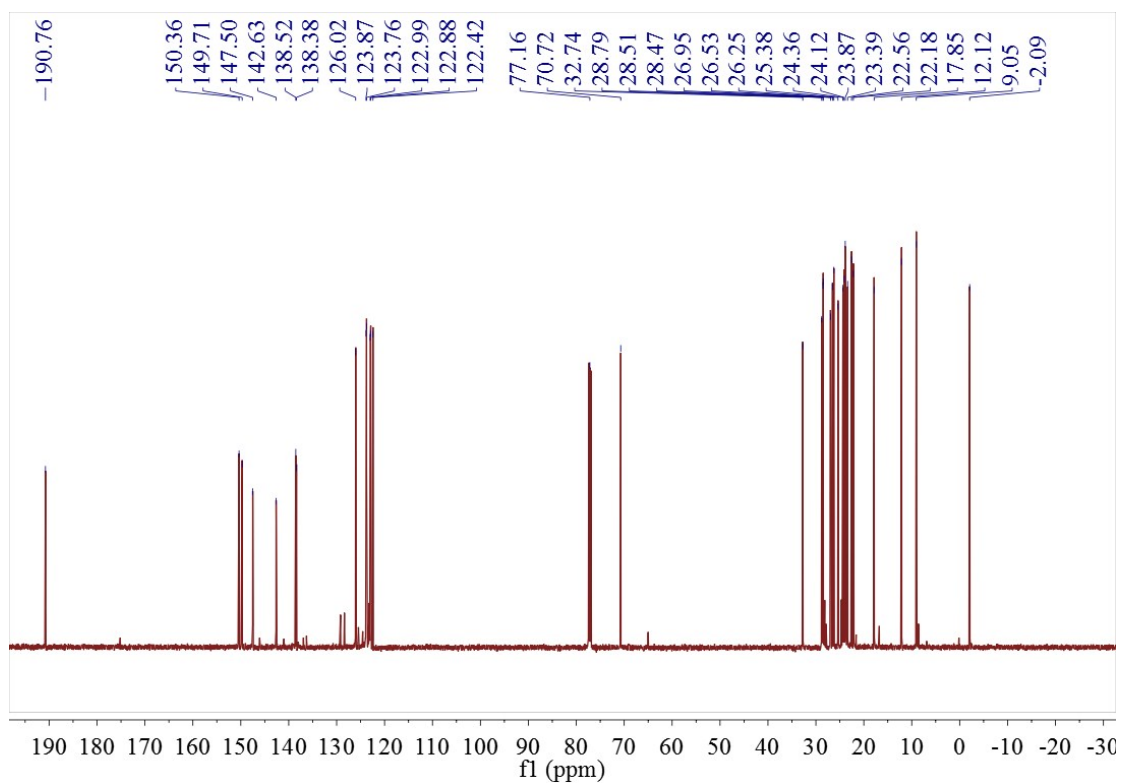


Figure S24. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound **6**.

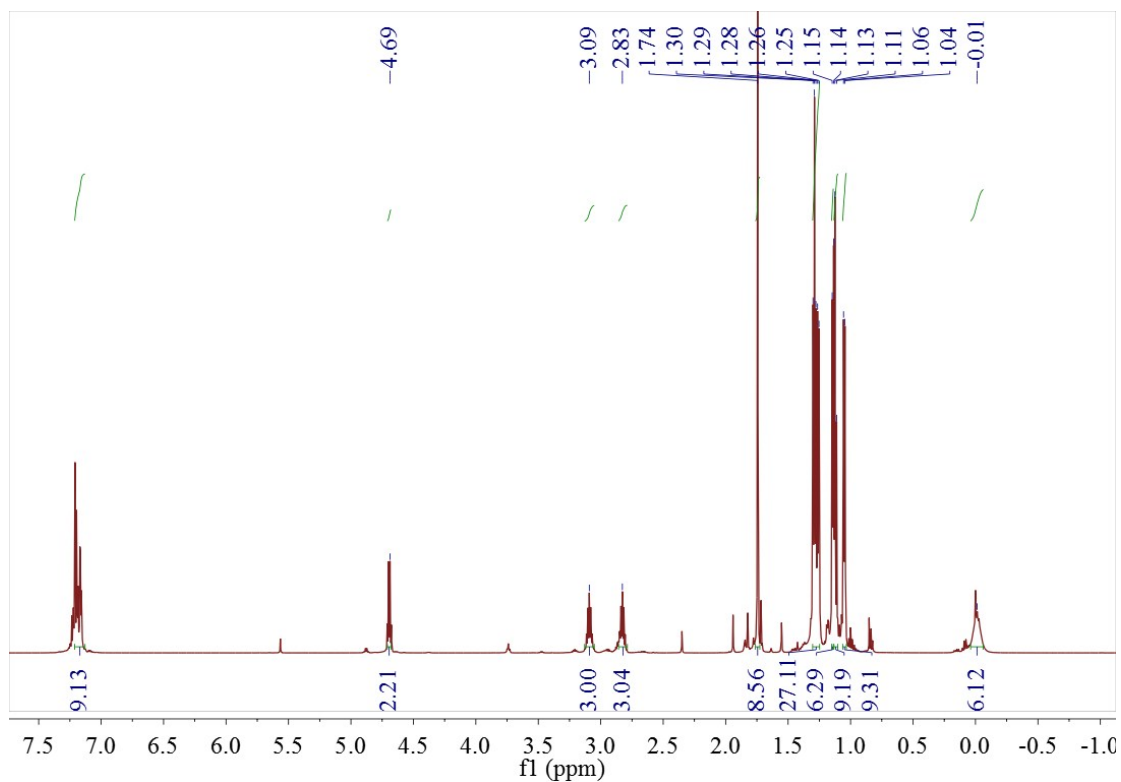


Figure S25. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound 7.

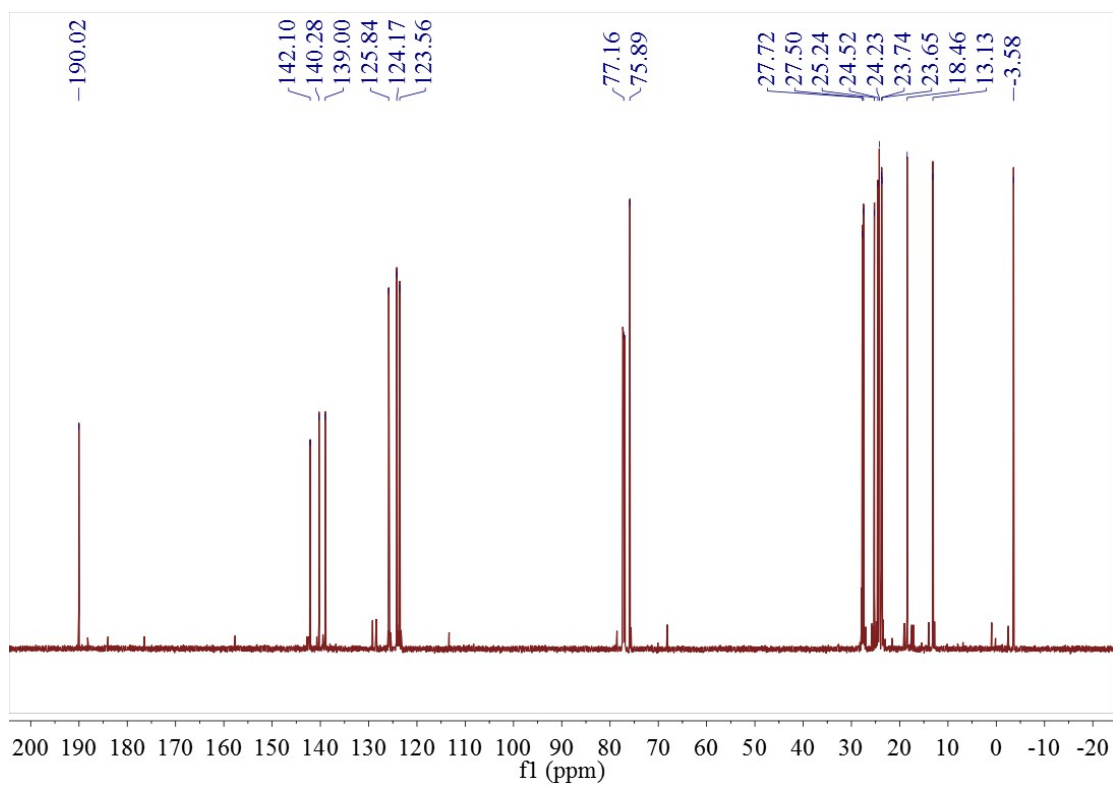


Figure S26. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound 7.

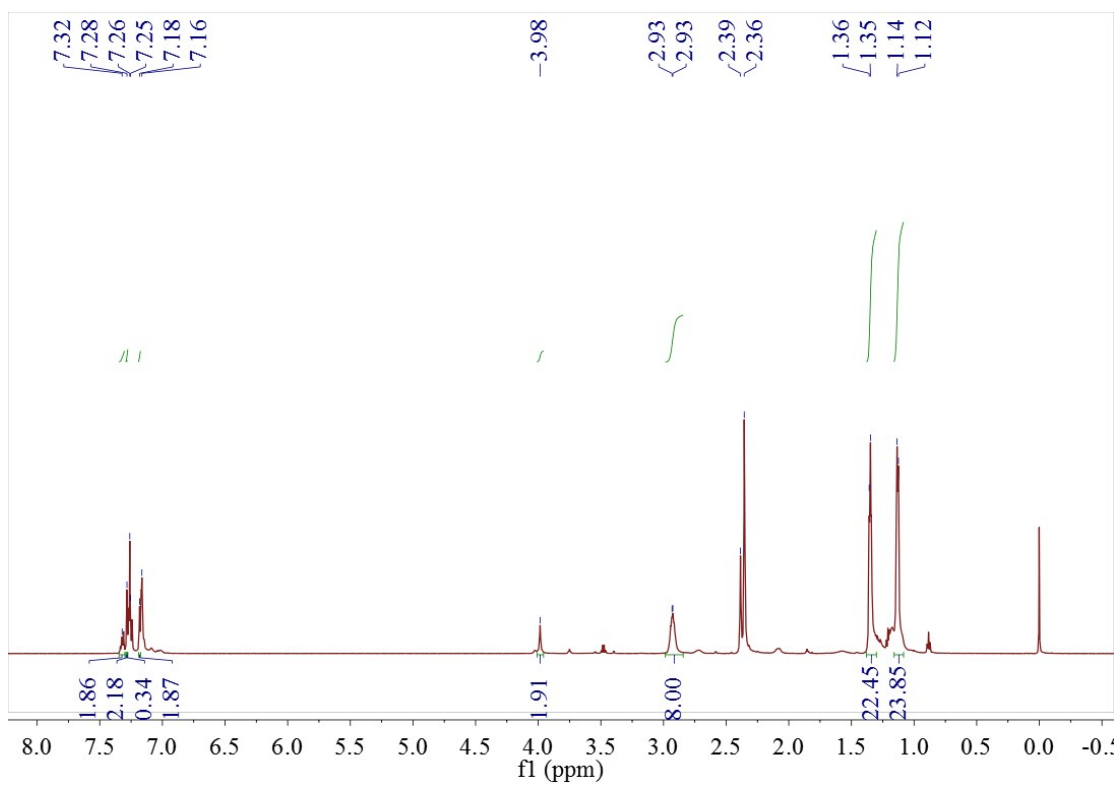


Figure S27. ^1H NMR (400 MHz CDCl_3 298K) spectrum for compound **8**.

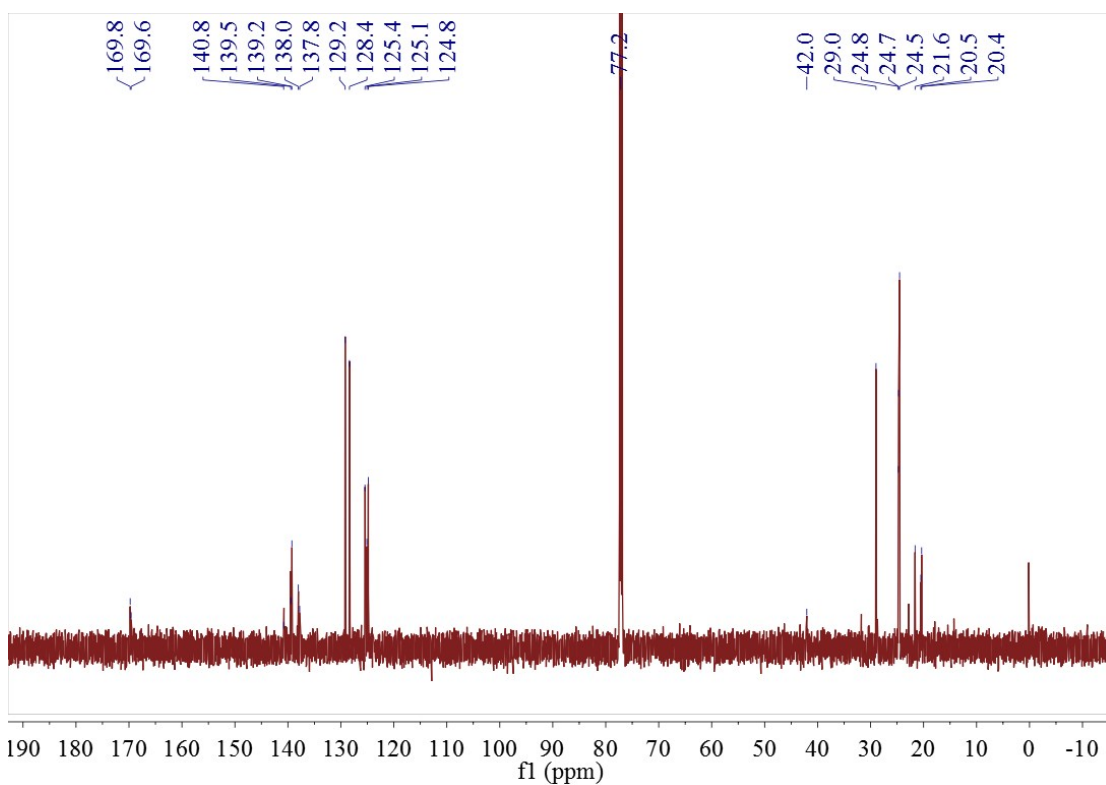


Figure S28. ^{13}C NMR (100.6 MHz CDCl_3 298K) spectrum for compound **8**.

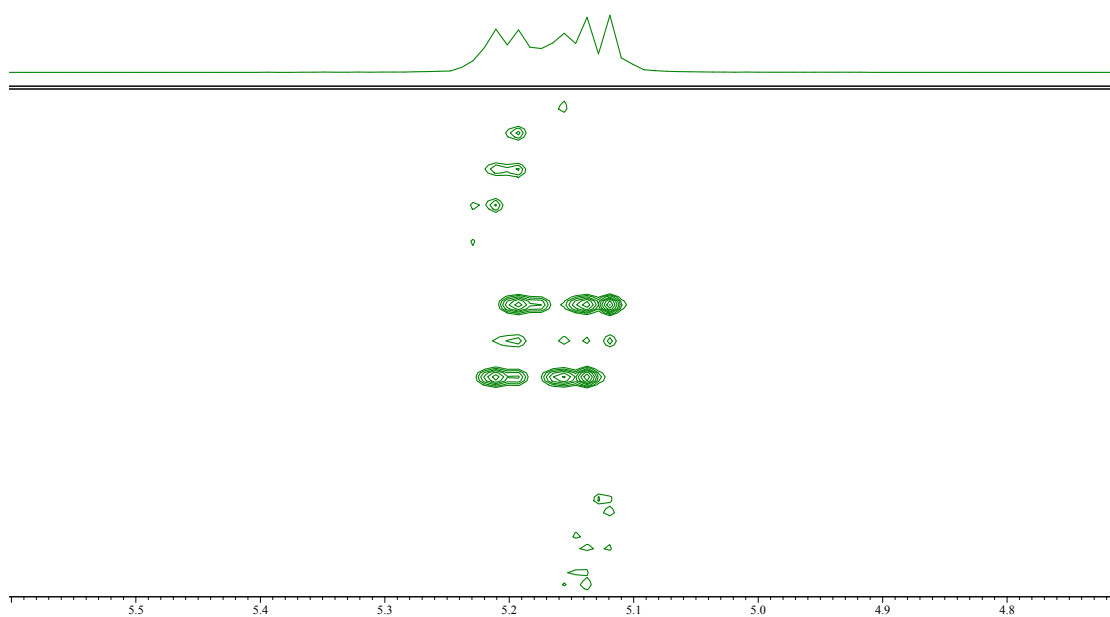


Figure S29. 2D J-resolved ¹H NMR (CDCl₃, 400 MHz, 298 K) spectrum of the PLA synthesized with **6** at 80 °C (table 6, entry 2).

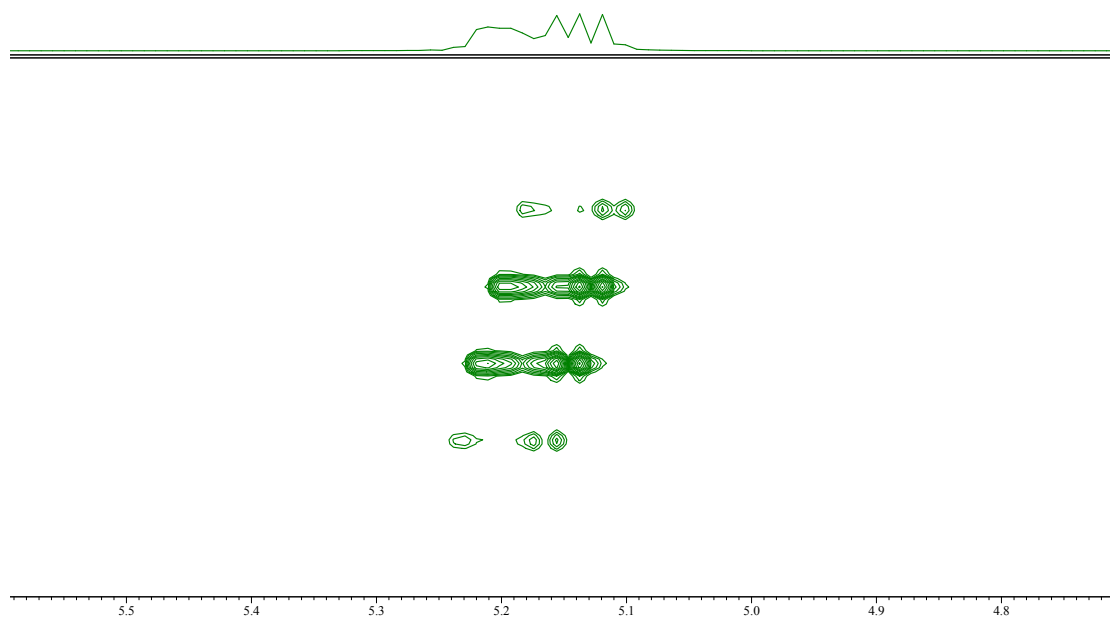


Figure S30. 2D J-resolved ¹H NMR (CDCl₃, 400 MHz, 298 K) spectrum of the PLA synthesized with **7** at 30 °C (table 6, entry 3).

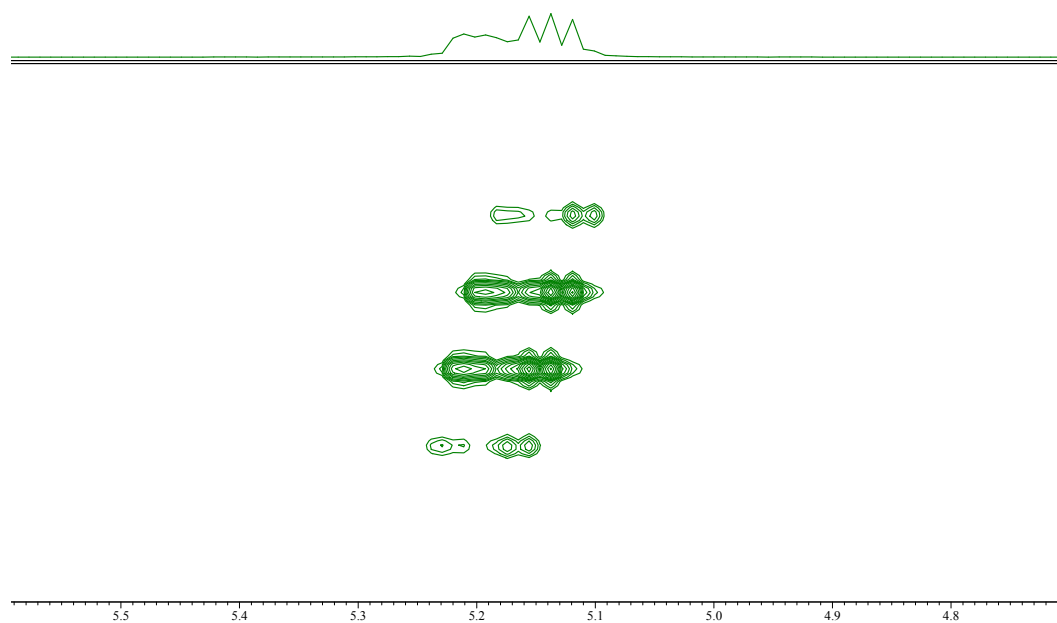


Figure S31. 2D J-resolved ^1H NMR (CDCl_3 , 400 MHz, 298 K) spectrum of the PLA synthesized with **7** at 80 $^\circ\text{C}$ (table 6, entry 4).

References

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