Supporting information Proton-conducting layered structures based on transition metal oxo-clusters supported by Sb(III)tartrate scaffolds

Wen Ma,^{a,b} Bing Hu,^{a,*} Kaiqiang Jing,^{a,b} Zhong Li,^c Jiance Jin,^{a,b} Shoutian Zheng^c and Xiaoying Huang^{a,*}

^a State Key Laboratory of Structure Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou 350002, China.

^b University of Chinese Academy of Sciences, Beijing, 100049, China.

^c State Key Laboratory of Photocatalysis on Energy and Environment, College of Chemistry, Fuzhou University, Fuzhou, 350108, China

+ E-mail: xyhuang@fjirsm.ac.cn; hubing@fjirsm.ac.cn





Element	Weight percentage	Atom percentage	Atomic ratio absvd(calcd)	
Co K	17.35	29.90	0.54(0.57)	
Cd L	16.78	15.16	0.28(0.29)	
Sb L	65.87	54.95	1.00(1.00)	

Element	Weight percentage	Atom Percentage	Atomic ratio absvd(calcd)	
КК	4.64	11.17	0.28(0.28)	
Fe K	14.38	24.26	0.61(0.67)	
Cd L	29.54	24.76	0.62(0.61)	
Sb L	51.45	39.81	1.00(1.00)	

Fig. S1 Energy dispersive spectroscopy (EDS) results of compounds 1 (a) and 2 (b).

Compound 1	ICP(%)	Sb	Cd	Со	
	Weight percentage(calcd.)	32.65	8.61	9.03	
	Weight percentage	31.92	6.99	9.14	
Compound 2	ICP(%)	Sb	Cd	Fe	К
	Weight percentage(calcd.)	27.17	15.33	8.31	2.42
	Weight percentage	25.20	14.85	8.57	2.22



Fig. S2 Schematic views of the coordination modes of the tartrate ligands in 1. (a) mode A, (b) mode B, (c) mode C, (d) mode D, (e) mode E and (f) mode F.



Fig. S4 Schematic views of the coordination modes of the tartrate ligands in 2. (a) mode A, (b) mode B, (c) mode C, (d) mode D, (e) mode E, (f) mode F and (g) mode G.







Fig. S9 Packing diagram of 2 viewed along the *b*- and *c*-axis, respectively. The disordered Cd(1B) and Cd(1C) are omitted for clarity.



Fig. S10 Simulated and observed powder X-ray diffraction (PXRD) patterns of 1.







Fig. S15 PXRD patterns for samples of as-synthesize 1 and that after water absorption measurement.







Fig. S17 Impedance spectra of complex 1 under different relative humidities at 25 °C and different temperature conditions at 98% RH.





Fig. S18 Impedance spectra of complex 2 under different relative humidities at 30 °C and different temperature conditions at 98% RH.

Compound 1						
RH %	RH % 55 70 85 98					
σ (S cm ⁻¹)	1.85×10 ⁻⁷	5.74*10 ⁻⁷	1.29*10 ⁻⁶	6.50*10 ⁻⁶		
Compound 2						
RH %	55	70	85	98		
σ (S cm ⁻¹)	1.41×10 ⁻⁶	2.04×10 ⁻⁶	5.68×10 ⁻⁶	3.66×10 ⁻⁵		

Table S3 Proton conductivity (σ) values of 1 and 2 under 98% RH and at different temperatures.

Compound 1								
T/°C	25		40		55	70		85
σ (S cm ⁻¹)	6.50*10	-6 5	5.72*10 ⁻⁵ 2.83*10 ⁻⁴		1	25*10 ⁻³	2.43*10 ⁻³	
Compound 2								
T/°C	30	40		50	60		70	85
σ (S cm ⁻¹)	3.66*10 ⁻⁵	1.37*1	.0 ⁻⁴ 3.	.66*10 ⁻⁴	8.97*1	0-4	1.89*10-3	2.95*10 ⁻³