## **Supporting Information**

## Novel phthiocol based organometallics with tridentate coordination sphere and their unexpected cytotoxic behavior

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#### Materials and methods

2-Hydroxy-3-methyl-1,4-dione  $L^{1,2}$ ,  $[RuCl_2(p-cym)]_2^3$ ,  $[OsCl_2(p-cym)]_2^4$  and 4-amino-1*H*pyrazole<sup>5</sup> were synthesized according to literature known procedures. Following chemicals and solvents were used without further purification for the syntheses: ruthenium(III) chlorid hydrate, osmium tetroxide (Johnson Matthey), 1*H*-pyrazole, triethylamine (Acros Organics), 2-methyl-naphthalene-1,4-dione, 4-methyl-1*H*-pyrazole, Pd on activated charcoal, 6-amino-1*H*indazole (Aldrich), 4-nitro-1*H*-pyrazole (Flourochem),  $\alpha$ -terpinene (Alfa Aesar), sodium carbonate (Merck), hydrazine dihydrochloride (Sigma-Aldrich), hydrogen (Messer), hydrochloric acid (37%), sulfuric acid (95%), silica gel (mesh 40-63 µm), ethyl acetate (Reag.Ph.Eur.ACS), *n*hexane (Reag.Ph.Eur.ACS), dichloromethane (stabilized with 0.2% ethanol) (VWR), methanol (HPLC grade), ethanol absolute (Laboratory reagent grade) (Fluka), ammonia hydroxide solution (20-24%) (W.Neubus Enkel GmbH) and 1*H*-indazole (Polivalent-95). Microwave reactions were performed with a Biotage® Initiator+ system. Elemental analysis were conducted by the microanalytical laboratory of the faculty of chemistry of the University of Vienna with a Perkin Elmer 2400 CHN elemental analyzer

<sup>1</sup>H-, <sup>13</sup>C- and 2D-NMR spectra were recorded at 298 K on a Bruker Avance III HD 700 MHz or Bruker Avance III 600 MHz spectrometers at 600.25/700.40 MHz (<sup>1</sup>H) and 150.95/176.13 MHz (<sup>13</sup>C). UV-Vis stability measurements were performed on a Perkin Elmer lambda 35 photometer with PTP (Peltier Temperature Programmer) and Julabo AWC 100 recirculating cooler.

**Cell culture.** In this study, the following human cancer cell lines were used: A549 (non-small cell lung carcinoma), SW480 and HCT-116 (both colon carcinoma) were kindly provided by Brigitte Marian, Institute of Cancer Research, Department of Medicine I, Medical University of Vienna. The cell line CH1/PA-1 (ovarian teratocarcinoma) was kindly provided by Lloyd R. Kelland (CRC Centre for Cancer Therapeutics, Institute of Cancer Research, Sutton, UK). HCT-15 (colon carcinoma, CCL-247<sup>TM</sup>) were obtained from ATCC®.

All cell culture media (including supplements) and reagents were obtained from Sigma-Aldrich, and all plasticware from StarLab, unless stated otherwise. A549, CH1/PA-1 and SW480 cells were grown in MEM supplemented with 10% fetal calf serum (FCS; from BioWest), 1 mM sodium pyruvate, 4 mM L-glutamine and 1% v/v nonessential amino acids (from 100× solution) and L-glutamine. HCT-116 and HCT-15 cells were maintained in McCoy's 5a and RPMI 1640 medium,

respectively, each supplemented with 10% FCS and L-glutamine. All cells were cultured as adherent monolayers in 75 cm<sup>2</sup> flasks and kept in a humidified incubator at 37 °C with 5% CO<sub>2</sub>.

**MTT assay.** Cytotoxicity of the compounds was determined by using the colorimetric MTT assay (MTT = 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide).  $1 \times 10^3$  CH1/PA-1,  $2 \times 10^3$  SW480 and  $3 \times 10^3$  A549 cells were seeded in 100 µL per well into 96-well microculture plates. After 24 h, test compounds were dissolved in DMSO (Fisher Scientific), serially diluted in complete MEM (to final DMSO content not exceeding 0.5% v/v) and added in 100 µL per well. After 96 h, the drug-containing medium was replaced with 100 µL of RPMI 1640/MTT mixture [6 parts of RPMI 1640 medium (supplemented with 10% heat-inactivated fetal bovine serum and 4 mM L-glutamine), 1 part of MTT solution in phosphate-buffered saline (5 mg/mL)]. After incubation for 4 h, the MTT-containing medium was replaced with 150 µL DMSO per well to dissolve the formazan product formed by viable cells. Optical densities at 550 nm (and at a reference wavelength of 690 nm) were measured with a microplate reader (ELx808, Bio-Tek). The 50% inhibitory concentrations (IC<sub>50</sub>) relative to untreated controls were interpolated from concentration–effect curves. At least three independent experiments were performed, each with triplicates per concentration level.

**Spheroid formation.** For spheroid generation, HCT-116, HCT-15, CH1/PA-1 and A549 cells were harvested from culture flasks by trypsinization, resuspended in their respective supplemented medium and seeded in ultra-low attachment round-bottom 96-well plates (Nunclon Sphera<sup>TM</sup>, Thermo Fisher Scientific) at a density of 500 viable cells per well. Plates were incubated at 37 °C with 5% CO<sub>2</sub> for 96 hours to allow spheroid formation and then used for the experiments.

Alamar Blue assay. The test compounds were first dissolved in DMSO, and stock solutions were prepared in appropriate medium according to the cell line and diluted stepwise to obtain a serial dilution. 100  $\mu$ l of the respective dilutions were added to each well, and the plates were incubated for 96 hours at 37 °C with 5% CO<sub>2</sub>. A 440  $\mu$ M resazurin sodium salt solution in PBS was prepared and 20  $\mu$ l were added to each well. The plates were incubated for 16 hours at 37 °C with 5% CO<sub>2</sub>. Fluorescence was measured and recorded with a Synergy HT reader (BioTek). All the results originate from at least three technical and biological replicates.

#### Synthesis of 2-hydroxy-3-methylnaphthalene-1,4-dione (L)



Scheme S1: Synthetic pathway to phthiocol (L)

The 2-hydroxy-3-mehtylnaphtalene-1,4-dione (Phthiocol, L) was prepared as described in literature with minor modifications.<sup>1</sup>

2-Methyl-1,4-naphthoquinone (1.54 g, 8.933 mmol, 1 eq.) was suspended in 100 mL methanol and cooled with an ice bath. Sodium carbonate (0.28 g, 2.680 mmol, 0.3 eq.) and hydrogen peroxide solution (30%, 1.72 mL, 517 mg, 15.186 mmol, 1.7 eq.) were dissolved in 10 mL water and added to the yellow suspension. The mixture was stirred for 0.5 h with ice cooling and for further 3 h at room temperature. The mixture was diluted with water and the volume of methanol was reduced by evaporation. The white precipitate was separated, washed with water and dried in vacuo. The epoxide was suspended in THF and ca. 4 g of silica gel and conc. H<sub>2</sub>SO<sub>4</sub> (1.76 mL, 3.24 g, 33.052 mmol, 3.7 eq.) were added. The suspension was evaporated with 500 mbar and 70 °C until dryness. The formed yellow solid was dissolved in dichloromethane, filtrated and washed with saturated sodium bicarbonate solution. The dark red aqueous layers were combined and acidified with concentrated HCl. The yellow suspension was extracted with dichloromethane, dried over Na<sub>2</sub>SO<sub>4</sub>, evaporated and dried in vacuo. Yield: 1.33 g yellow powder (7.068 mmol, 79 %). <sup>1</sup>H NMR (500.10 MHz, CDCl<sub>3</sub>)  $\delta$  8.13 (dd, J = 7.8, 1.1 Hz, 1H, H<sub>5/8</sub>), 8.08 (dd, J = 7.7, 1.7 Hz, 1H, H<sub>5/8</sub>), 7.75 (ddd, J = 7.6, 7.5, 1.4 Hz, 1H, H<sub>6/7</sub>), 7.68 (ddd, J = 7.6, 7.5, 1.3 Hz, 1H, H<sub>6/7</sub>), 7.29 (s, 1H, H<sub>OH</sub>), 2.11 (s, 3H, H<sub>9</sub>). Anal. Calc. for C<sub>11</sub>H<sub>8</sub>O<sub>3</sub>: C 70.21%, H 4.29%. Found: C: 69.82%, H 4.23%

#### Synthesis of complexes (1a-e, 2a-c)

General procedure



Scheme S2: Synthetic pathway to Ru<sup>II</sup> and Os<sup>II</sup> complexes (1a-c, 2a-c)

The respective metal dimer (1eq.), 1,2-diazole (1.9-2.2 eq.), 1,4-naphthoquinone (L) (2.1-2.2 eq.) and sodium methoxide (4.5 eq.) or triethylamine (10 eq.) were dissolved in 12 mL methanol and stirred under microwave irradiation at 50-60 °C for 6-12 minutes. Afterwards, the mixture was evaporated, the remaining residue was dissolved in dichloromethane and filtered. The solution was purified by column chromatography (Eluent: EtOAC/*n*-hexane/NEt<sub>3</sub> or EtOAc/MeOH/NH<sub>4</sub>OH). The fractions were combined, evaporated and the oily residue was dissolved in dichloromethane. For precipitation, excess of diethyl ether and *n*-hexane were added until precipitation started and the mixture was stored at 4 °C overnight for complete precipitation. The solid was separated and washed twice with diethyl ether and once with *n*-hexane. The yellowish powder was dried *in vacuo*. Yield: 41–84%.

# [3-Methyl-4-oxo-(1*H*- $\kappa N^2$ -pyrazol-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)ruthenium(II)] (1a)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^{6}$ -p-cymene)ruthenium(II) (150 mg, 0.245 mmol, 1 eq.), 1*H*-pyrazole (35 mg, 0.514 mmol, 2.1 eq.), 2-hydroxy-3-methylnaphthalene-1,4-dione (**L**) (97 mg, 0.514 mmol, 2.1 eq.), sodium methoxide (60 mg, 1.102 mmol, 4.5 eq.). The mixture was stirred at 60 °C for 8 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NEt<sub>3</sub> (90/5/5). Yield: 99 mg yellow powder(0.202 mmol, 41%). <sup>1</sup>H NMR (600.25 MHz, MeOD) & 8.33 (dd, *J* = 2.2, 0.7 Hz, 1H, H<sub>3</sub>·), 8.12 – 8.07 (m, 1H, H<sub>arom</sub>), 7.62 – 7.57 (m, 3H, H<sub>arom</sub>), 6.68 (dd, *J* = 2.6, 0.7 Hz, 1H, H<sub>5</sub>·), 6.34 (t, *J* = 2.4 Hz, 1H, H<sub>4</sub>·), 5.98 (d, *J* = 5.8 Hz 1H, H<sub>c</sub>), 5.87 (d, *J* = 5.8 Hz, 1H, H<sub>c</sub>), 5.63 – 5.58 (m, 2H, H<sub>b</sub>), 2.86 (hept, *J* = 6.9 Hz, 1H, H<sub>e</sub>), 2.33 (s, 3H, H<sub>g</sub>), 1.74 (s, 3H, H<sub>9</sub>), 1.33 (dd, *J* = 6.9, 1.3 Hz, 3H, H<sub>f</sub>). <sup>13</sup>C NMR (150.95 MHz, MeOD)  $\delta$  183.9 (C2), 183.6 (C4), 141.5 (C3<sup>+</sup>), 137.8 (Carom.), 134.1 (Carom.), 132.3 (Carom.), 131.2 (Carom.), 127.9 (C5<sup>+</sup>), 127.7 (Carom.), 127.1 (Carom.), 108.7 (C4<sup>+</sup>), 105.6 (C3), 101.0 (Cd), 98.7 (Ca), 94.8 (C1), 83.3 (Cc), 83.1 (Cc), 80.2 (Cb), 80.1 (Cb), 32.7 (Ce), 23.0 (Cf), 22.8 (Cf), 18.3 (Cg), 8.1 (C9). Anal. Calc. for C<sub>24</sub>H<sub>24</sub>N<sub>2</sub>O<sub>3</sub>Ru: C 58.88%, H 4.94%, N 5.72%. Found: C: 58.53%, H 5.16%.

## [3-Methyl-4-oxo-(1*H*- $\kappa N^2$ -indazolyl-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)ruthenium(II)] (**1b**)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^6$ -pcymene)ruthenium(II) (40 mg, 0.065 mmol, 1 eq.), 1H-indazole (15 mg, 0.130 mmol, 2 eq.), 2hydroxy-3-methylnaphthalene-1,4-dione (L) (27 mg, 0.143 mmol, 2.2. eq), triethylamine (90.5 µL, 66 mg, 0.650 mmol, 10 eq.). The mixture was stirred at 50 °C for 12 minutes under microwave irradiation. Flash column chromatography with EtOAc/n-hexane/NEt<sub>3</sub> (90/5/5). Yield: 44 mg yellow powder (0.082 mmol, 63%). <sup>1</sup>H NMR (600.25 MHz, DMSO- $d_6$ )  $\delta$  9.18 (d, J = 0.9 Hz, 1H, H<sub>3</sub>, 8.07 - 8.04 (m, 1H, H<sub>arom</sub>), 7.80 - 7.77 (m, 1H, H<sub>arom</sub>), 7.66 - 7.60 (m, 2H, Harom), 7.56 – 7.53 (m, 1H, Harom), 7.07 – 7.03 (m, 1H, Harom), 7.01 – 6.97 (m, 1H, Harom), 6.14  $(d, J = 5.9 \text{ Hz}, 1\text{H}, \text{H}_{b}), 6.08 (d, J = 5.9 \text{ Hz}, 1\text{H}, \text{H}_{b}), 5.77 (d, J = 5.9 \text{ Hz}, 1\text{H}, \text{H}_{c}), 5.68 (d, J = 5.9 \text{ Hz}, 1\text{H}, 1\text$ 5.9 Hz, 1H, H<sub>c</sub>), 5.22 (dd, J = 8.7, 0.9 Hz, 1H, H<sub>arom</sub>), 2.83 (hept, J = 6.9 Hz, 1H, H<sub>c</sub>), 2.28 (s, 3H,  $H_g$ ), 1.54 (s, 3H,  $H_9$ ), 1.27 (d, J = 6.9 Hz, 6H,  $H_f$ ). <sup>13</sup>C NMR (150.95 MHz, DMSO)  $\delta$  180.8 (C2), 180.2 (C4), 137.0 (C<sub>arom.</sub>), 135.2 (C3'), 135.1 (C<sub>arom.</sub>), 132.7 (C<sub>arom.</sub>), 130.6 (C<sub>arom.</sub>), 129.8 (C<sub>arom.</sub>), 127.8 (Carom.), 127.0 (Carom.), 125.4 (Carom.), 124.2 (Carom.), 121.7 (Carom.), 121.5 (Carom.), 109.5 (C<sub>arom</sub>), 102.6 (C3), 98.6 (Cd), 97.3 (Ca), 95.8 (C<sub>1</sub>), 82.6 (Cb), 82.3 (Cb), 79.0 (Cc), 78.2 (Cc), 30.9 (Ce), 22.6 (Cf), 22.3 (Cf), 17.6 (Cg), 8.1 (C9). Anal. Calc. for C<sub>28</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>Ru • 0.25 H<sub>2</sub>O: C 61.80%, H 4.91%, N 5.15%. Found: C 61.62%, H 5.16%, N 5.41%.

[3-Methyl-4-oxo-(4-methyl-1*H*- $\kappa N^2$ -pyrazol-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)ruthenium(II)] (1c)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^6$ -*p*-cymene)ruthenium(II) (50 mg, 0.082 mmol, 1 eq.), 4-methyl-1*H*-pyrazole (13.64 µL, 14 mg, 0.164 mmol, 2 eq.), 2-hydroxy-3-methylnaphthalene-1,4-dione (**L**) (34 mg, 0.180 mmol, 2.2 eq.), triethylamine (114.3 µL, 83 mg 0.820 mmol, 10 eq.). The mixture was stirred at 50 °C for 10 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NH<sub>4</sub>OH (88/10/2). Yield: 69 mg yellow powder (0.137 mmol, 84%).

<sup>1</sup>H NMR (600.25 MHz, MeOD)  $\delta$  8.14 (s, 1H, H<sub>3'</sub>), 8.11 – 8.07 (m, 1H, H<sub>arom</sub>), 7.62 – 7.55 (m, 3H, H<sub>arom</sub>), 6.47 (s, 1H, H<sub>5'</sub>), 5.94 (d, *J* = 5.9 Hz, 1H, H<sub>c</sub>), 5.85 (d, *J* = 5.9 Hz, 1H, H<sub>c</sub>), 5.60 – 5.55 (m, 2H, H<sub>b</sub>), 2.86 (hept, *J* = 6.9 Hz, 1H, H<sub>e</sub>), 2.32 (s, 3H, H<sub>g</sub>), 1.97 (s, 3H, H<sub>6'</sub>), 1.74 (s, 3H, H<sub>9</sub>), 1.34 (d, *J* = 2.1 Hz, 3H, H<sub>f</sub>), 1.33 (d, *J* = 2.1 Hz, 3H; H<sub>f</sub>). <sup>13</sup>C NMR (151 MHz, MeOD)  $\delta$  183.9 (C2), 183.8 (C4), 141.3 (C3'), 137.9 (C<sub>arom</sub>), 134.1 (C<sub>arom</sub>), 132.2 (C<sub>arom</sub>), 131.1 (C<sub>arom</sub>), 127.8 (C<sub>arom</sub>), 127.0 (C<sub>arom</sub>), 126.9 (C5'), 119.9 (C4'), 105.5 (C3), 101.0 (Cd), 98.6 (Ca), 94.5 (C1), 83.3 (Cc), 83.0 (Cc), 80.2 (Cb), 80.0 (Cb), 32.7 (Ce), 23.0 (Cf), 22.8 (Cf), 18.3 (Cg), 8.9 (C6'), 8.1 (C9). Anal. Calc. for C<sub>25</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>Ru: C 59.63%, H 5.20%, N 5.56%. Found: C: 59.68%, H 5.44%, N 5.64%.

#### [3-Methyl-4-oxo-(4-amino-1*H*- $\kappa N^2$ -pyrazol-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)ruthenium(II)] (1d)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^{6}$ -*p*-cymene)ruthenium(II) (50 mg, 0.082 mmol, 1 eq.), 4-amino-1*H*-pyrazole (14 mg, 0.164 mmol, 2 eq.), 2-hydroxy-3-methylnaphthalene-1,4-dione (**L**) (34 mg, 0.180 mmol, 2.2 eq.), triethylamine (114.3 µL, 83 mg 0.820 mmol, 10 eq.). The mixture was stirred at 50 °C for 12 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NH<sub>4</sub>OH (88/10/2). Yield: 46 mg brown powder (0.091 mmol, 55%) <sup>1</sup>H NMR (600.25 MHz, MeOD)  $\delta$  8.10 – 8.05 (m, 1H, <sub>Harom</sub>), 7.91 (d, *J* = 0.9 Hz, 1H, H<sub>3</sub>·), 7.63 – 7.55 (m, 3H, H<sub>arom</sub>), 6.21 (d, *J* = 0.9 Hz, 1H, H<sub>5</sub>·), 5.92 (d, *J* = 5.9 Hz, 1H, H<sub>c</sub>), 5.83 (d, *J* = 5.8 Hz 1H, H<sub>c</sub>), 5.56 (dd, *J* = 6.4 Hz,6.3 Hz, 2H, H<sub>b</sub>), 2.86 (hept, *J* = 6.9 Hz, 1H, H<sub>e</sub>), 2.32 (s, 3H, H<sub>g</sub>), 1.73 (s, 3H, H<sub>9</sub>), 1.35 – 1.32 (m, 6H, H<sub>f</sub>). <sup>13</sup>C NMR (150.95 MHz, MeOD)  $\delta$  183.9 (C4), 183.7 (C2), 138.0 (Carom.), 134.1 (Carom.), 132.9 (C4'), 132.3 (C3'), 132.1 (Carom.), 131.1 (Carom.), 127.8 (Carom.), 127.0 (Carom.), 116.9 (C5'), 105.4 (C3), 100.9 (Cd), 98.6 (Ca), 94.6 (C1), 83.3 (Cc), 82.9 (Cc), 80.1 (Cb), 80.0 (Cb), 32.7 (Ce), 23.0 (Cf), 22.8 (Cf), 18.3 (Cg), 8.1 (C9). C<sub>24</sub>H<sub>25</sub>N<sub>3</sub>O<sub>3</sub>Ru • 0.3 H<sub>2</sub>O: C 56.53%, H 5.06%, N 8.26%. Found: C 56.21%, H 4.99%, N 8.64%.

### [3-Methyl-4-oxo-(6-Amino-1*H*- $\kappa N^2$ -indazolyl-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)ruthenium(II)] (1e)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^6$ -pcymene)ruthenium(II) (100 mg, 0.163 mmol, 1 eq.), 6-amino-1H-indazole (41 mg, 0.310 mmol, 1.9 eq.), 2-hydroxy-3-methylnaphthalene-1,4-dione (L) (64 mg, 0.342 mmol, 2.1 eq.). triethylamine (227.3 µL, 165 mg 1.630 mmol, 10 eq.). The mixture was stirred at 50 °C for 6 minutes under microwave irradiation. Flash column chromatography with EtOAc/nhexane/NH<sub>4</sub>OH (88/10/2). Yield: 77 mg yellow/greenish crystals (0.091 mmol, 45%). <sup>1</sup>H NMR  $(600.25 \text{ MHz}, \text{MeOD}) \delta 8.66 \text{ (d}, J = 0.9 \text{ Hz}, 1\text{H}, \text{H}_{7'}), 8.17 \text{ (ddd}, J = 7.9, 1.4, 0.5 \text{ Hz}, 1\text{H}, \text{H}_5), 7.72$  $(ddd, J = 7.6, 1.4, 0.5 Hz, 1H, H_8), 7.66 (ddd, J = 7.7, 7.6, 1.4 Hz, 1H, H_{arom}), 7.59 (ddd, J = 7.5, 1.4 Hz, 1H, H_{arom}), 7.59 (dddd, J = 7.5, 1.$ 7.5, 1.4 Hz, 1H, H<sub>arom</sub>), 7.40 (dd, J = 8.8, 0.7 Hz, 1H, H<sub>4</sub>), 6.50 (dd, J = 8.8, 1.8 Hz, 1H, H<sub>5</sub>), 5.98  $(d, J = 5.6 Hz, 1H, H_b), 5.87 (d, J = 5.6 Hz, 1H, H_b), 5.61 (dd, J = 5.60, 5.61 Hz, 2H, H_c), 4.47 (dd, J = 5.6 Hz, 1H, H_b), 5.87 (d, J = 5.6 Hz, 1H,$  $J = 1.8, 0.9 \text{ Hz}, 1\text{H}, \text{H}_{3'}$ , 2.88 (hept,  $J = 6.9 \text{ Hz}, 1\text{H}, \text{H}_{e}$ ), 2.34 (s, 3H, H<sub>g</sub>), 1.71 (s, 3H, H<sub>9</sub>), 1.37 -1.33 (m, 6H, H<sub>e</sub>).<sup>13</sup>C NMR (150.95 MHz, MeOD) δ 184.4 (C2), 184.2 (C4), 150.5 (C3a'), 139.8 (C6'), 138.4 (C7a'), 137.0 (C7'), 134.3 (C8a), 132.2 (Carom.), 131.3 (Carom.), 128.47 (C8), 127.1 (C5), 122.7 (C4'), 119.3 (C4a), 115.8 (C5'), 106.8 (C3), 101.1 (Cd), 98.7 (Ca), 94.0 (C1), 92.1 (C3'), 83.6 (Cb), 83.3 (Cb), 80.6 (Cc), 80.5 (Cc), 32.69 (Ce), 23.1 (Cf), 22.9 (Cf), 18.3 (Cg), 8.0 (C9). C<sub>28</sub>H<sub>27</sub>N<sub>3</sub>O<sub>3</sub>Ru • 0.5 CH<sub>2</sub>Cl<sub>2</sub>: C 57.33%, H 4.73%, N 7.04%. Found: C 56.99%, H 4.97%, N 7.09%.

## [3-Methyl-4-oxo-(1*H*- $\kappa N^2$ -pyrazol-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)osmium(II)] (2a)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^{6}$ -*p*-cymene)osmium(II) (200 mg, 0.253 mmol, 1 eq.), 1*H*-pyrazole (34 mg, 0.506 mmol), 2-hydroxy-3-methylnaphthalene-1,4-dione (L) (105 mg, 0.557 mmol, 2.2 eq.), sodium methoxide (60 mg, 1.113 mmol, 4.4 eq.). The mixture was stirred at 60 °C for 8 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NEt<sub>3</sub> (90/5/5). Yield: 190 mg yellow powder (0.328 mmol, 65 %). <sup>1</sup>H NMR (700.40 MHz, MeOD)  $\delta$  8.29 (d, *J* = 2.2 Hz, 1H, H<sub>3</sub>·), 8.17 – 8.11 (m, 1H, H<sub>arom.</sub>), 7.66 – 7.60 (m, 3H, H<sub>arom.</sub>), 6.87 (d, *J* = 2.6 Hz, 1H, H<sub>5</sub>·), 6.39 (t, *J* = 2.4 Hz, 1H, H<sub>4</sub>·), 6.19 (d, *J* = 5.5 Hz, 1H, H<sub>c</sub>), 6.09 (d, *J* = 5.5 Hz, 1H, H<sub>c</sub>), 5.85 (d, *J* = 5.4 Hz, 1H, H<sub>b</sub>), 5.80 (d, *J* = 5.5 Hz, 1H, H<sub>b</sub>), 2.74 (hept, *J* = 6.9 Hz, 1H, H<sub>c</sub>), 2.40 (s, 3H, H<sub>g</sub>), 1.75 (s, 3H, H<sub>9</sub>), 1.31 (d, *J* = 6.9 Hz, 6H, H<sub>f</sub>).<sup>13</sup>C NMR (176.13 MHz, MeOD)  $\delta$  184.6 (C2), 183.1 (C4), 141.4 (C3<sup>·</sup>), 136.7 (C<sub>arom.</sub>), 134.3 (C<sub>arom.</sub>), 132.5 (C<sub>arom.</sub>), 131.5 (C<sub>arom.</sub>), 127.8 (C<sub>arom.</sub>), 127.6 (C5<sup>·</sup>), 127.3 (C<sub>arom.</sub>), 109.4 (C4<sup>·</sup>), 105.4 (C3), 98.1 (C1), 90.8 (Cd), 89.1 (Ca), 73.9 (Cc), 73.4 (Cc), 70.4 (Cb), 70.3 (Cb), 33.1 (Ce), 23.4 (Cf), 23.2 (Cf), 18.6 (Cg), 8.1 (C9). Anal. Calc. for C<sub>24</sub>H<sub>24</sub>N<sub>2</sub>O<sub>3</sub>Os•0.25CH<sub>2</sub>Cl<sub>2</sub>: C 48.55%, H 4.12%, N 4.67%. Found: C 48.40%, H 4.12%, N 4.67%.

# [3-Methyl-4-oxo-(1*H*- $\kappa N^2$ -indazolyl-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)osmium(II)] (**2b**)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^6$ -*p*-cymene)osmium(II) (200 mg, 0.253 mmol, 1 eq.), 1*H*-indazol (60 mg, 0.506 mmol), 2-hydroxy-3-methylnaphthalene-1,4-dione (L) (105 mg, 0.557 mmol, 2.2 eq.), sodium methoxide (60 mg, 1.113 mmol, 4.4 eq.). The mixture was stirred at 60 °C for 8 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NEt<sub>3</sub> (90/5/5). Yield: 178 mg yellow powder (0.283 mmol 56 %). <sup>1</sup>H NMR (700.40 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  9.17 (s, 1H, H3'), 8.11 (dd, *J* = 7.8, 1.4 Hz, 1H, H<sub>arom</sub>), 7.83 – 7.80 (m, 1H, H<sub>arom</sub>), 7.70 – 7.65 (m, 2H, H<sub>arom</sub>), 7.62 – 7.57 (m, 1H, H<sub>arom</sub>), 7.13 – 7.02 (m, 2H, H<sub>arom</sub>), 6.33 (d, *J* = 5.4 Hz, 1H, Hc), 6.26 (d, *J* = 5.5 Hz, 1H, Hc), 5.98 (d, *J* = 5.4 Hz, 1H, Hc), 5.29 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>), 2.72 (hept, *J* = 6.9 Hz, 1H, H<sub>e</sub>), 2.34 (s, 3H, H<sub>g</sub>), 1.56 (s, 3H, H<sub>9</sub>), 1.27 – 1.23 (m, 6H, H<sub>f</sub>). <sup>13</sup>C NMR (176.13 MHz, DMSO)  $\delta$  181.0 (C4), 180.3 (C2), 135.9 (C<sub>arom</sub>), 135.5 (C3'), 134.4 (C<sub>arom</sub>), 132.8 (C<sub>arom</sub>), 130.9 (C<sub>arom</sub>), 130.1 (C<sub>arom</sub>), 128.5 (C<sub>arom</sub>), 127.0 (C<sub>arom</sub>), 125.6 (Ca<sub>arom</sub>), 124.6 (C<sub>arom</sub>), 122.1 (C<sub>arom</sub>), 121.9 (C<sub>arom</sub>), 109.6 (C<sub>arom</sub>), 127.0 (C3), 98.8 (C1), 88.8 (Cd), 87.9 (Ca), 72.9 (Cc), 72.6 (Cc), 69.2 (Cb), 68.5 (Cb), 31.3 (Ce), 23.0 (Cf), 22.6 (Cf), 17.8 (Cg), 8.2 (C9). Anal. Calc. for C<sub>28</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>Os•0.1H<sub>2</sub>O: C 53.33%, H 4.19%, N 4.44%. Found: C 52.92%, H 4.15%, N 4.39%.

### [3-Methyl-4-oxo-(4-methyl-1*H*- $\kappa N^2$ -pyrazol-1-yl)-1,4-dihydronaphtalene-1,2-bis(olato)- $\kappa O^1$ - $\kappa O^2$ )( $\eta^6$ -*p*-cymene)osmium(II)] (2c)



The reaction was performed according to the general procedure, using bis[dichlorido( $\eta^6$ -*p*-cymene)osmium(II) (200 mg, 0.253 mmol, 1 eq.), 4-methyl-1*H*-pyrazole (41.9 µL, 42 mg, 0.506 mmol), 2-hydroxy-3-methylnaphthalene-1,4-dione (L) (105 mg, 0.557 mmol, 2.2 eq.), sodium methoxide (60 mg, 1.113 mmol, 4.4 eq.). The mixture was stirred at 60 °C for 8 minutes under microwave irradiation. Flash column chromatography with EtOAc/*n*-hexane/NEt<sub>3</sub> (90/5/5). Yield: 137 mg yellow powder (0.231 mmol 46 %). <sup>1</sup>H NMR (700.40 MHz, MeOD)  $\delta$  8.16 – 8.11 (m, 1H, H<sub>arom</sub>.), 8.10 (s, 1H, H<sub>3</sub>·), 7.66 – 7.59 (m, 3H, H<sub>arom</sub>.), 6.67 (s, 1H, H<sub>5</sub>·), 6.15 (d, *J* = 5.4 Hz, 1H, H<sub>b</sub>), 6.07 (d, *J* = 5.4 Hz, 1H, H<sub>b</sub>), 5.82 (d, *J* = 5.5 Hz, 1H, H<sub>c</sub>), 5.77 (d, *J* = 5.5 Hz, 1H, H<sub>c</sub>), 2.73 (hept, *J* = 6.9 Hz, 1H, H<sub>e</sub>), 2.39 (s, 3H, H<sub>g</sub>), 1.98 (s, 3H, H<sub>6</sub>·), 1.75 (s, 3H, H<sub>9</sub>), 1.31 (d, *J* = 6.9 Hz, 6H, H<sub>f</sub>).<sup>13</sup>C NMR (176.12 MHz, MeOD)  $\delta$  183.2 (C4), 181.9 (C2), 139.8 (C3'), 135.5 (C<sub>arom</sub>.), 132.9 (C<sub>arom</sub>.), 131.0 (C<sub>arom</sub>.), 130.0 (C<sub>arom</sub>.), 126.4 (C<sub>arom</sub>.), 125.8 (C<sub>arom</sub>.), 125.2 (C5'), 119.2 (C4'), 103.8 (C3), 96.5 (C1), 89.4 (Cd), 87.6 (Ca), 72.4 (Cb), 71.9 (Cb), 69.0 (Cc), 68.8 (Cc), 31.7 (Cg), 22.0 (Cf), 21.8 (Cf), 17.2 (Ce), 7.5 (C6'), 6.7 (C9). Anal. Calc. for C<sub>25</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>Os: C 50.66%, H 4.42%, N 4.73%. Found: C 50.46%, H 4.40%, N 4.72%.

## NMR Spectra



Figure S1: <sup>1</sup>H-NMR spectrum of compound L



Figure S2: <sup>1</sup>H-NMR spectrum of compound 1a



Figure S3: <sup>13</sup>C-NMR spectrum of compound 1a



Figure S4: <sup>1</sup>H-NMR spectrum compound **1b** 



Figure S5: <sup>13</sup>C-NMR spectrum of compound **2b** 



Figure S6: <sup>1</sup>H-NMR spectrum of compound 1c



Figure S7: <sup>13</sup>C-NMR spectrum of compound 1c



Figure S8: <sup>1</sup>H-NMR spectrum of compound 1d



Figure S9: <sup>13</sup>C-NMR spectrum of compound 1d



Figure S10: <sup>1</sup>H-NMR spectrum of compound 1e



Figure S11: <sup>13</sup>C-NMR spectrum of compound **1e** 



Figure S12: <sup>1</sup>H-NMR spectrum of compound 2a



Figure S13: <sup>13</sup>C-NMR spectrum of compound **2a** 



Figure S14: <sup>1</sup>H-NMR spectrum of compound **2b** 



Figure S15: 13C-NMR spectrum of compound 2b



Figure S16: <sup>1</sup>H-NMR spectrum of compound **2c** 



Figure S17: <sup>13</sup>C-NMR spectrum of compound **2c** 

#### X-ray analysis

The X-ray intensity data were measured on Bruker D8 Venture diffractometer equipped with multilayer monochromators, Cu/Mo K/a INCOATEC micro focus sealed tubes and Oxford cooling system. The structures were solved by *direct methods or charge flipping* and refined by full-matrix least-squares techniques. Non-hydrogen atoms were refined with anisotropic displacement parameters. Hydrogen atoms were inserted at calculated positions and refined with riding model. The following software was used: Bruker SAINT software package<sup>6</sup> using a narrowframe algorithm for frame integration, SADABS7 for absorption correction, OLEX28 for structure solution, refinement, molecular diagrams and graphical user-interface, Shelxle9 for refinement and graphical user-interface SHELXS-2015<sup>10</sup> for structure solution, SHELXL-2015<sup>11</sup> for refinement, *Platon*<sup>12</sup> for symmetry check and  $\pi$ - $\pi$  Interactions calculations. Experimental data and CCDC-Codes Experimental data and CCDC-Code (Available online: http://www.ccdc.cam.ac.uk/conts/retrieving.html) can be found in Table S1. Crystal data, data collection parameters, and structure refinement details are given in Table S2 to Table S17. Crystal structures and  $\pi$ - $\pi$  Interactions are visualized in Figure S18 to Figure S27.

Sample	Machine	Source	Temn	Detector	Time/	#Frames	Frame	CCDC
Sampie	Machine	Source	remp.	Distance	Frame	#Praines	width	ССРС
	Bruker		[K]	[mm]	[s]		[°]	
417 <b>1</b> a	D8	Мо	100	30	15	752	0.400	1955180
576 1b	D8	Мо	100	30	2	765	0.400	1955183
542 1c	D8	Cu	100	30	26	2037	0.500	1955182
640 1d	D8	Мо	100	30	5	816	1.000	1955187
636 1e	D8	Cu	100	30	20	2533	1.000	1955186
510 2a	D8	Мо	100	40	1.5	1214	0.500	1985181
579 2b	D8	Мо	100	30	0.5	1200	0.500	1955184
580 2c	D8	Cu	100	30	3	2281	1.000	1955185

Table S1: Experimental parameter and CCDC-Codes.



Figure S18: Crystal structure of 1a. C-C- Bond precision: 0.0068 Å. Solvent and counter ion omitted for clarity.

Table S2: Sample and crystal data (1a).

Chemical formula	C27H32N2O5Ru	Crystal system	monoclinic		
Formula weight [g/mol]	565.61	Space group	C2/c		
Temperature [K]	100	Z	8		
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	5052.2(4)		
Radiation (Wavelength [Å])	MoK $\alpha$ ( $\lambda = 0.71073$ )	Unit cell dimensions [Å] and [°]	32.8559(11)	90	
Crystal size / [mm <sup>3</sup> ]	$0.316\times0.053\times0.034$		8.4989(4)	96.9960(17)	
Crystal habit	clear yellow needle		18.2283(7)	90	
Density (calculated) / [g/cm <sup>3</sup> ]	1.487	Absorption coefficient / [mm <sup>-1</sup> ]	0.661		
Abs. correction Tmin	0.6285	Abs. correction Tmax	0.746		
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]		2336	

Index ranges	$\begin{array}{c} -39 \leq h \leq 30, \ -8 \leq k \leq 10, \\ -21 \leq l \leq 21 \end{array}$	Theta range for data collection [°]	4.502 to 50.692		
Reflections number	10516	Data / restraints / parameters	4622/0/325		
Refinement method	Least squares	Einal D indiana	all data	R1 = 0.0658, wR2 = 0.1218	
Function minimized	$\Sigma \mathrm{w}(\mathrm{F_o}^2 - \mathrm{F_c}^2)^2$	Final K mulces	I>2σ(I)	R1 = 0.0478, wR2 = 0.1130	
Goodness-of-fit on F <sup>2</sup>	1.048	Weighting scheme $w=1/[\sigma_2(Fo_2)+(0.0001P)_2+32.97]$ where $P=(F_o^2+2F_c^2)/3$		2)+(0.0001P)2 +32.9719P]	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	2.44/-1.33			ere $P=(F_o^2+2F_c^2)/3$	

Table S3: Data collection and structure refinement (1a).



Figure S19: Crystal structure of **1b**. C-C- Bond precision: 0.0073Å. Second independent moiety B and solvent omitted for clarity. Squeeze was used to cut the volume and the corresponding electron densities, because small electron densities in excluded volumes could not be matched. Details see cif-code.



Figure S20: Analysis of  $\pi$ - $\pi$  Interactions. The distance of Cg-Cg amounts 3.809Å and 3.734 Å. Solvent and hydrogen omitted for clarity.

Chemical formula	C28.7H27.75N2O3.17Ru	Crystal system	triclinic		
Formula weight [g/mol]	552.55	Space group	P-1		
Temperature [K]	100	Z	4		
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	2687.88(9)		
Radiation (Wavelength [Å])	MoKα ( $\lambda$ = 0.71073)	Unit cell dimensions [Å] and [°]	11.4410(2)	107.7827(7)	
Crystal size / [mm <sup>3</sup> ]	$0.2\times0.06\times0.04$		15.6073(3)	99.4000(7)	
Crystal habit	clear yellow needle		17.4184(3)	108.5729(6)	
Density (calculated) / [g/cm <sup>3</sup> ]	1.365	Absorption coefficient / [mm <sup>-1</sup> ]	0.615		
Abs. correction Tmin	0.2162	Abs. correction Tmax	0.259		
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]		1133	

Table S4: Sample and crystal data (1b).

Index ranges	$\begin{array}{c} -13 \leq h \leq 13,  -18 \leq k \leq \\ 18,  -20 \leq l \leq 20 \end{array}$	Theta range for data collection [°]	4.652 to 50.68		
Reflections number	28022	Data / restraints / parameters	9774/40/662		
Refinement method	Least squares	Final D indiana	all data	R1 = 0.0651, wR2 = 0.1049	
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$	Final K mulces	I>2σ(I)	R1 = 0.0455, wR2 = 0.0934	
Goodness-of-fit on F <sup>2</sup>	1.039		w=1/[σ2(Fo	o2)+(0.0260P)2 +3.1126P]	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	0.95/-0.68	Weighting scheme	where $P=(F_o^2+2F_c^2)/3$		

## Table S5: Data collection and structure refinement (1b).



Figure S21: Crystal structure of **1c**. C-C- Bond precision: 0.0105 Å. Second independent moiety B omitted for clarity. Squeeze was used to cut the volume and the corresponding electron densities, because small electron densities in excluded volumes could not be matched. Details see cif-code.

Table S6: Sample and crystal data (1c).

Chemical formula	C25H26N2O3Ru	Crystal system	triclinic		
Formula weight [g/mol]	503.55	Space group	P-1		
Temperature [K]	100	Z	4		
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	2233.61(9)		
Radiation (Wavelength [Å])	$CuK\alpha (\lambda = 1.54178)$	Unit cell dimensions [Å] and [°]	9.6758(2)	71.6959(13)	
Crystal size / [mm <sup>3</sup> ]	$0.05\times0.025\times0.025$		14.2466(3)	84.0849(14)	
Crystal habit	clear yellow block		17.8729(4)	72.7405(13)	
Density (calculated) / [g/cm <sup>3</sup> ]	1.497	Absorption coefficient / [mm <sup>-1</sup> ]	5.914		
Abs. correction Tmin	0.0245	Abs. correction Tmax	0.1163		
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]		1032	

Index ranges	$\begin{array}{c} \text{-10} \leq h \leq 11,  \text{-16} \leq k \leq \\ 16,  \text{-20} \leq l \leq 18 \end{array}$	Theta range for data collection [°]	5.208 to 127.366		
Reflections number	15128	Data / restraints / parameters	7170/0/569		
Refinement method	Least squares	Final D indiana	all data	R1 = 0.0730, wR2 = 0.1754	
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$	Final R Indices	I>2σ(I)	R1 = 0.0650, wR2 = 0.1663	
Goodness-of-fit on F <sup>2</sup>	1.058		w=1/[σ2(Fo2)+(0.1130P)2+3.2250P]		
Largest diff. peak and hole [e Å- <sup>3</sup> ]	2.84/-0.42	Weighting scheme	where $P = (F_o^2 + 2F_c^2)/3$		

Table S/: Data collection and structure refinement (Ic
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Figure S22: Crystal structure of **1d**. C-C- Bond precision: 0.0135 Å. Second independent molecule omitted for clarity. Squeeze was used to cut the volume and the corresponding electron densities, because small electron densities in excluded volumes could not be matched. Details see cif-code.

Table S8: Sample and crystal data (1d).

Chemical formula	C24H25N3O3Ru	Crystal system	monoclinic		
Formula weight [g/mol]	504.54	Space group	P21/n		
Temperature [K]	100	Z	8		
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	4641.7(6)		
Radiation (Wavelength [Å])	MoK $\alpha$ ( $\lambda = 0.71073$ )	Unit cell dimensions [Å] and [°]	11.3136(9)	90	
Crystal size / [mm <sup>3</sup> ]	$0.1\times0.05\times0.02$		23.5155(17)	102.414(3)	
Crystal habit	clear orange plate		17.8647(15)	90	
Density (calculated) / [g/cm <sup>3</sup> ]	1.444	Absorption coefficient / [mm <sup>-1</sup> ]	0.705		
Abs. correction Tmin	0.0635	Abs. correction Tmax	0.0916		
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]		2064	

Index ranges	$\begin{array}{c} -13 \leq h \leq 13,  -28 \leq k \leq \\ 28,  0 \leq l \leq 21 \end{array}$	Theta range for data collection [°]	4.768 to 50.77		
Reflections number	16802	Data / restraints / parameters	8522/6/569		
Refinement method	Least squares	Final D indiana	all data	R1 = 0.1232, wR2 = 0.2108	
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$	Final K mulces	I>2σ(I)	R1 = 0.0884, wR2 = 0.1915	
Goodness-of-fit on F <sup>2</sup>	1.07	w=1/[σ2(Fo2)+(0.0714P)2+3		2)+(0.0714P)2 +38.3513P]	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	1.34/-0.93	Weighting scheme	where $P = (F_o^2 + 2F_c^2)/3$		

## Table S9: Data collection and structure refinement (1d).



Figure S23: Crystal structure of **1e**. C-C- Bond precision: 0.0144 Å. Squeeze was used to cut the volume and the corresponding electron densities, because small electron densities in excluded volumes could not be matched. Details see cif-code.

Table S10: Sample and crystal data (1e).

Chemical formula	C28H27N3O3Ru	Crystal system	monoclinic		
Formula weight [g/mol]	554.59	Space group	P21/c		
Temperature [K]	100	Z	4		
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	2853.5(4)		
Radiation (Wavelength [Å])	$CuK\alpha (\lambda = 1.54178)$	Unit cell dimensions [Å] and [°]	10.9871(9)	90	
Crystal size / [mm <sup>3</sup> ]	$0.1\times0.1\times0.03$		9.5021(7)	98.361(3)	
Crystal habit	clear yellow block		27.626(2)	90	
Density (calculated) / [g/cm <sup>3</sup> ]	1.291	Absorption coefficient / [mm <sup>-1</sup> ]	4.691		
Abs. correction Tmin	0.0625	Abs. correction Tmax	0.1665		
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]		1136	

Index ranges	$\begin{array}{c} \text{-13} \leq h \leq 13,  \text{-10} \leq k \leq \\ 11,  \text{-32} \leq l \leq 33 \end{array}$	Theta range for data collection [°]	8.134 to 136.47	
Reflections number	38659	Data / restraints / parameters	5217/0/320	
Refinement method	Least squares	Final D indiana	all data	R1 = 0.0979, wR2 = 0.2271
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$	Final K mulces	I>2σ(I)	R1 = 0.0974, wR2 = 0.2269
Goodness-of-fit on F <sup>2</sup>	1.244		w=1/[σ2(Fo2)+37.2588P]	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	1.80/-1.65	Weighting scheme	where $P = (F_o^2 + 2F_c^2)/3$	

## Table S11: Data collection and structure refinement (1e).



Figure S24: Crystal structure of **2a**. C-C- Bond precision: 0.0070Å. Counter ion omitted for clarity Table S12: Sample and crystal data (**2a**).

Chemical formula	C24H26N2O4Os	Crystal system	monoclinic	
Formula weight [g/mol]	596.67	Space group		P21/n
Temperature [K]	100	Z		4
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	2052.02(17)	
Radiation (Wavelength [Å])	MoKa ( $\lambda = 0.71073$ )	Unit cell dimensions [Å] and [°]	11.8354(6) 90	
Crystal size / [mm <sup>3</sup> ]	$0.41 \times 0.22 \times 0.2$		14.6547(7) 110.4966(15)	
Crystal habit	clear yellow block		12.6306(6) 90	
Density (calculated) / [g/cm <sup>3</sup> ]	1.931	Absorption coefficient / [mm <sup>-1</sup> ]	6.25	
Abs. correction Tmin	0.0887	Abs. correction Tmax	0.2651	
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]	1168	

Index ranges	$-14 \le h \le 14, -17 \le k \le 17, -15 \le l \le 15$	Theta range for data collection [°]	4.424 to 50.698	
Reflections number	22678	Data / restraints / parameters	3762/3/287	
Refinement method	Least squares	Einel Dindiese	all data	R1 = 0.0323, wR2 = 0.0803
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$	Final R Indices	I>2σ(I)	R1 = 0.0306, wR2 = 0.0792
Goodness-of-fit on F <sup>2</sup>	1.103		eme $w=1/[\sigma 2(Fo2)+(0.0343P)2+4.0645P]$ where P=(F <sub>o</sub> <sup>2</sup> +2F <sub>c</sub> <sup>2</sup> )/3	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	2.16/-1.12	Weighting scheme		

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Figure S25: Crystal structure of **2b**. C-C- Bond precision: 0.0142 Å. Second independent moiety B omitted for clarity. Squeeze was used to cut the volume and the corresponding electron densities, because small electron densities in excluded volumes could not be matched. Details see cif-code.



Figure S26: Analysis of  $\pi$ - $\pi$  Interactions. The distance of Cg-Cg amounts 3.800Å and 3.750 Å. Hydrogen omitted for clarity.

Chemical formula	C28H26N2O3Os	Crystal system	triclinic	
Formula weight [g/mol]	628.71	Space group	P-1	
Temperature [K]	100	Z	4	
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]	2702.76(13)	
Radiation (Wavelength [Å])	MoKα ( $\lambda$ = 0.71073)	Unit cell dimensions [Å] and [°]	11.4724(3) 107.6878(10)	
Crystal size / [mm <sup>3</sup> ]	$0.2\times0.15\times0.1$		15.5715(4) 99.3524(10)	
Crystal habit	clear yellow block		17.4570(5) 108.3467(10)	
Density (calculated) / [g/cm <sup>3</sup> ]	1.545	Absorption coefficient / [mm <sup>-1</sup> ]	4.747	
Abs. correction Tmin	0.0294	Abs. correction Tmax	0.099	
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]	1232	

Table S14: Sample and crystal data (2b).

Table S15: Data collection and structure refinement (2b).

Index ranges	$\begin{array}{c} \text{-14} \leq h \leq 14,  \text{-19} \leq k \leq \\ 19,  \text{-22} \leq l \leq 22 \end{array}$	Theta range for data collection [°]	4.6 to 54.206	
Reflections number	72850	Data / restraints / parameters	11905/0/621	
<b>Refinement method</b>	Least squares	Einal D indiana	all data	R1 = 0.0623, wR2 = 0.1242
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$	r mai K muices	I>2σ(I)	R1 = 0.0509, wR2 = 0.1186
Goodness-of-fit on F <sup>2</sup>	1.052	w=1/[σ2(Fo2)+(0.0237P)2 +45.539		2)+(0.0237P)2 +45.5397P]
Largest diff. peak and hole [e Å <sup>-3</sup> ]	3.65/-1.69	Weighting scheme	where $P=(F_o^2+2F_c^2)/3$	



Figure S27: Crystal structure of 2c. C-C- Bond precision: 0.0054Å

Table S16: Sample and crystal data (2c).

Chemical formula	C25H26N2O3Os	Crystal system	monoclinic	
Formula weight [g/mol]	592.68	Space group	P21/c	
Temperature [K]	100	Z	4	
Measurement method	f and w scans	Volume [Å <sup>3</sup> ]		2131.7(2)
Radiation (Wavelength [Å])	$CuK\alpha (\lambda = 1.54178)$	Unit cell dimensions [Å] and [°]	16.3690(9) 90	
Crystal size / [mm <sup>3</sup> ]	$0.1\times0.05\times0.02$		7.6784(4) 110.5999(13)	
Crystal habit	clear yellow block		18.1186(10) 90	
Density (calculated) / [g/cm <sup>3</sup> ]	1.847	Absorption coefficient / [mm <sup>-1</sup> ]	11.544	
Abs. correction Tmin	0.042	Abs. correction Tmax	0.1665	
Abs. correction type	multiscan	F(000) [e <sup>-</sup> ]	1160	

Index ranges	$\begin{array}{c} -19 \leq h \leq 19,  -8 \leq k \leq 9, \\ -21 \leq l \leq 21 \end{array}$	Theta range for data collection [°]	10.432 to 137.18	
Reflections number	31965	Data / restraints / parameters	3888/0/285	
Refinement method	Least squares	Final D indiana	all data	R1 = 0.0301, wR2 = 0.0796
Function minimized	$\Sigma w (F_o^2 - F_c^2)^2$	Final K mulces	I>2σ(I)	R1 = 0.0298, wR2 = 0.0794
Goodness-of-fit on F <sup>2</sup>	1.11		w=1/[ $\sigma$ 2(Fo2)+(0.0451P)2 +2.5141P] where P=(F <sub>o</sub> <sup>2</sup> +2F <sub>c</sub> <sup>2</sup> )/3	
Largest diff. peak and hole [e Å- <sup>3</sup> ]	1.81/-0.80	Weighting scheme		

## Table S17: Data collection and structure refinement (2c).

#### **UV-Vis stability measurements**

DMSO/DMF was obtained from VWR. PBS buffer solution was adjusted with NaOH or HCl to pH 7.4. Stock solutions with a concentration of 10 mM in DMSO/DMF were prepared. 1980  $\mu$ L PBS buffer solution, 12  $\mu$ L DMSO/DMF and 8  $\mu$ L of the stock solution were added to a quartz cuvette (d=1cm) to obtain a concentration of 40  $\mu$ M with 1% DMSO/DMF. The UV-Vis spectra were recorded in the range of 200-750 nm at 20 °C on a Perkin Elmer lambda 35 photometer with PTP (Peltier Temperature Programmer) and Julabo AWC 100 recirculating cooler.



Figure S28: UV-Vis spectrum of phthiocol L, with 40 µM in PBS/1% DMF (pH= 7.4), 24 h



Figure S29: UV-Vis spectrum of KP2048 1, with 40  $\mu$ M in PBS/1% DMF (pH= 7.4), 24 h. Red: t = 0; black: t = 24 h



Figure S30. UV-Vis spectrum of 1a, with 40  $\mu$ M in PBS/1% DMSO (pH= 7.4), 72 h



Figure S31: UV-Vis spectrum of **1b**, with 40  $\mu$ M in PBS/1% DMSO (pH= 7.4), 72 h



Figure S32: UV-Vis spectrum of 1c, with 40  $\mu$ M in PBS/1% DMSO (pH= 7.4), 72 h



Figure S33: UV-Vis spectrum of 1d, with 40  $\mu$ M in PBS/1% DMSO (pH= 7.4), 72 h



Figure S34: UV-Vis spectrum of 1e, with 40 µM in PBS/1% DMSO (pH= 7.4), 72 h



Figure S35: UV-Vis spectrum of 2a, with 40 µM in PBS/1% DMSO (pH= 7.4), 72 h



Figure S36: UV-Vis spectrum of 1a, with 40  $\mu$ M in PB/1% DMSO (20mM, pH = 5.8), 48 h



Figure S37: UV-Vis spectrum of 1a, with 40  $\mu$ M in PB PB/1% DMSO (20mM, pH = 6.2), 48 h



Figure S38: UV-Vis spectrum of 1a, with 40  $\mu$ M in PB PB/1% DMSO (20mM, pH = 6.7), 48 h



Figure S39: UV-Vis spectrum of 1a, with 40  $\mu$ M in PB PB/1% DMSO (20mM, pH = 7.9, 48 h



Figure S40: Absorption vs. time at 363 nm at different pH values (5.8–7.9) of compound 1a

#### **Cytotoxicity in Cell Cultures**

Compound	IC <sub>50</sub> [μM]				
	A549	SW480	CH1/PA-1		
HPz	> 200	> 200	> 200		
HInd	> 200	> 200	> 200		
4-MeHPz	> 200	> 200	> 200		
4-NH <sub>2</sub> -HPz	> 200	> 200	> 200		
6-NH <sub>2</sub> -HInd	> 200	> 200	> 200		

Table S18: In vitro anticancer activity (IC50 values) of 1,2-diazoles



Figure S41: Concentration–effect curves of Ru compounds 1a-1e in monolayer cultures of the cancer cell lines A549 (top), PA-1/CH1 (middle) and SW480 (bottom) in the MTT assay (exposure time: 96 h). Values are means  $\pm$  standard deviations of at least three independent experiments.



Figure S42: Concentration–effect curves of Os compounds 2a-2c in monolayer cultures of the cancer cell lines A549 (top), PA-1/CH1 (middle) and SW480 (bottom) in the MTT assay (exposure time: 96 h). Values are means  $\pm$  standard deviations of at least three independent experiments.

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