

Supporting Information

Designing of Novel Zinc(II) Schiff Base Complexes Having Acyl Hydrazone Linkage: Study of Phosphatase and Anti-Cancer Activity

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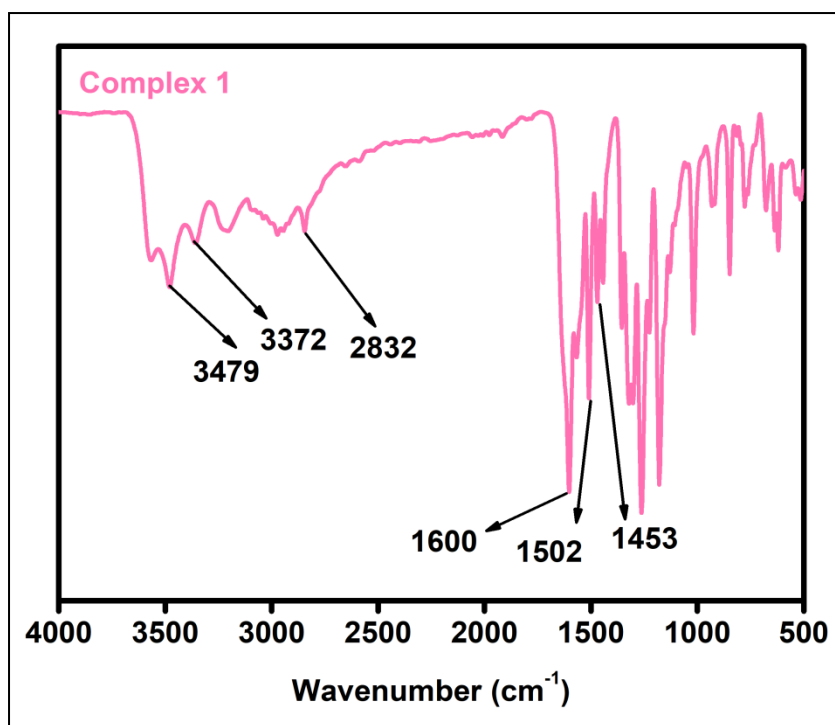


Fig. S1 FT-IR spectrum of complex 1.

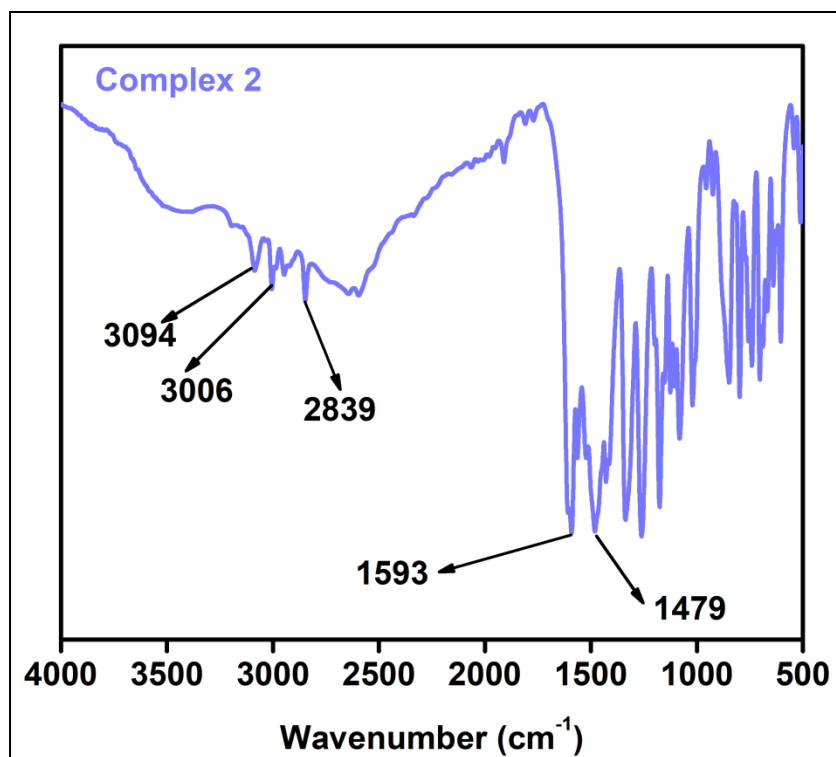


Fig. S2 FT-IR spectrum of complex 2.

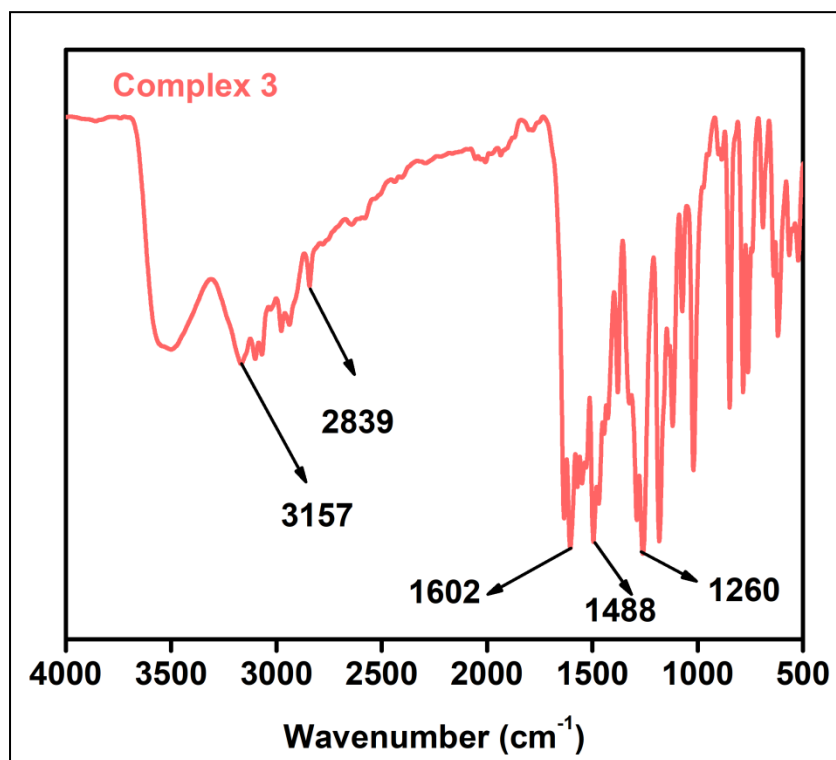


Fig. S3 FT-IR spectrum of complex 3.

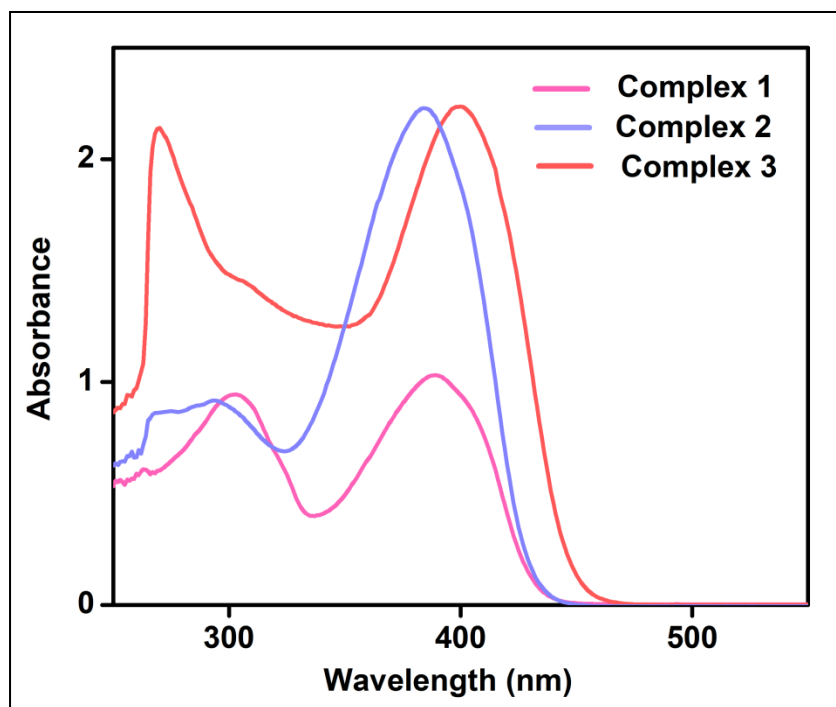


Fig. S4 UV-Vis spectra of complexes **1-3**.

Table S1. Selected Bond Lengths (Å) and Angles (°) of complex **1**

Zn(1)-Cl(1)	2.2392(6)	Zn(1)-N(1)	2.1848(13)
Zn(1)-Cl(2)	2.2268(6)	Zn(1)-N(2)	2.0961(12)
Zn(1)-O(2)	2.2636(12)		
Cl(1)-Zn(1)-Cl(2)	116.11(2)	Cl(2)-Zn(1)-N(1)	101.29(4)
Cl(1)-Zn(1)-O(2)	98.94(3)	Cl(2)-Zn(1)-N(2)	135.51(4)
Cl(1)-Zn(1)-N(1)	103.09(4)	O(2)-Zn(1)-N(1)	143.74(4)
Cl(1)-Zn(1)-N(2)	107.78(3)	O(2)-Zn(1)-N(2)	71.45(4)
Cl(2)-Zn(1)-O(2)	94.27(3)	N(1)-Zn(1)-N(2)	74.55(5)

Table S2. Selected Bond Lengths (Å) and Angles (°) of complex 3

Zn(1)-Cl(1)	2.2519(7)	Zn(1)-N(1)	2.132(2)
Zn(1)-Cl(2)	2.2259(8)	Zn(1)-N(2)	2.1284(19)
Zn(1)-O(1)	2.1700(17)		
Cl(1)-Zn(1)-Cl(2)	118.07(3)	Cl(2)-Zn(1)-N(1)	98.93(6)
Cl(1)-Zn(1)-O(1)	98.44(6)	Cl(2)-Zn(1)-N(2)	120.08(5)
Cl(1)-Zn(1)-N(1)	98.17(6)	O(1)-Zn(1)-N(1)	145.82(7)
Cl(1)-Zn(1)-N(2)	121.86(5)	O(1)-Zn(1)-N(2)	72.51(7)
Cl(2)-Zn(1)-O(1)	99.24(5)	N(1)-Zn(1)-N(2)	73.35(7)

Table S3. H-bonding parameters for complex 1

D-H	d(D-H)	d(H..A)	d(D..A)	d(D..A)	A	Symmetry
N(3)-H(3)	0.808(19)	1.977(19)	2.7730(18)	168.4(19)	O(3)	
O(3)-H(3B)	0.8500	2.4600	3.2820(17)	163.00	Cl(1)	1-x,1-y,1-z
O(3)-H(3C)	0.8500	1.8800	2.720(2)	172.00	O(4)	
O(4)-H(4A)	0.8500	2.0900	2.918(2)	165.00	O(2)	x,-1+y, z
O(4)-H(4B)	0.8500	2.3500	3.1494(18)	156.00	Cl(2)	1-x,1-y,-z

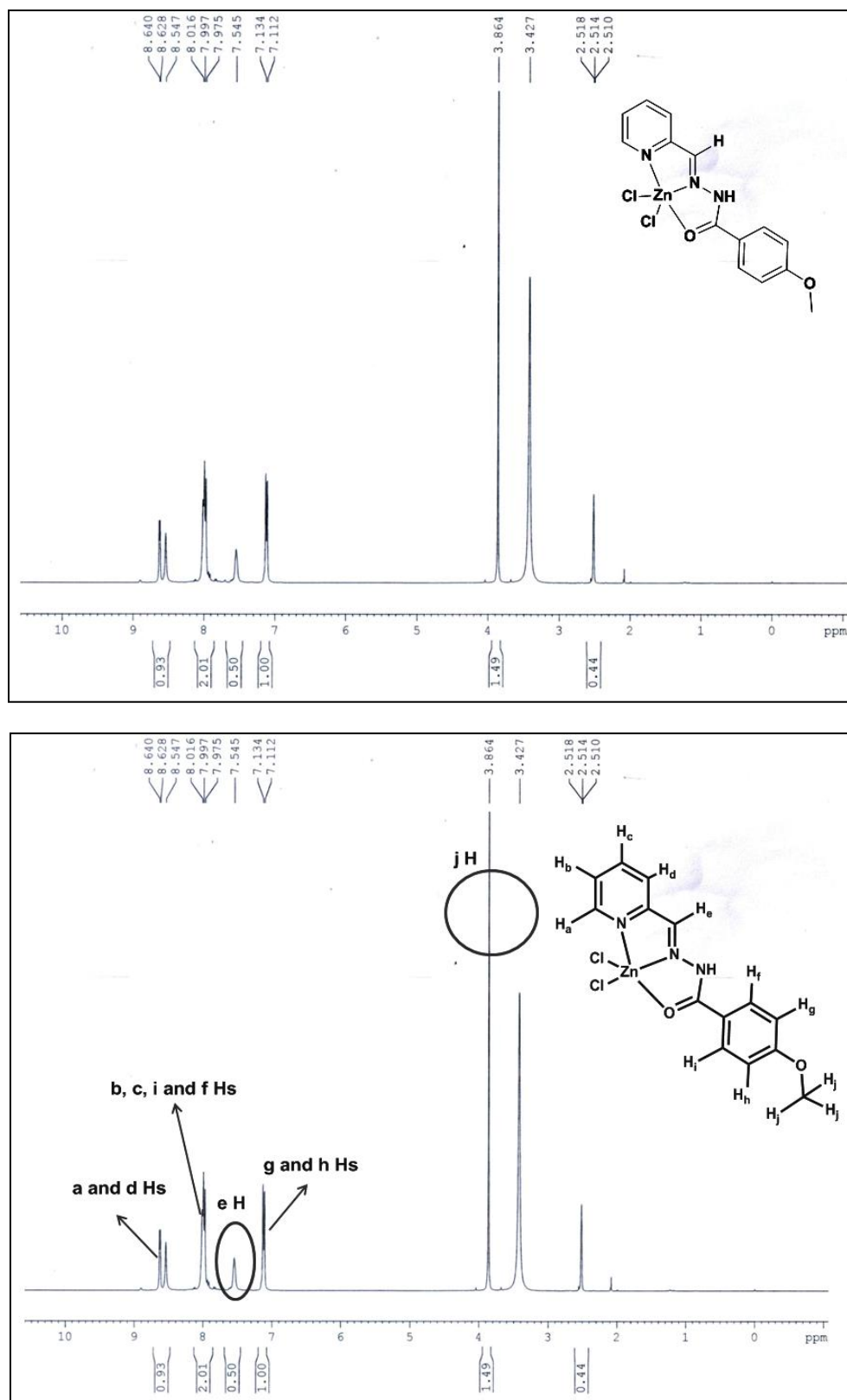


Fig. S5 ^1H NMR spectrum of complex **1** recorded in DMSO-d_6 solvent.

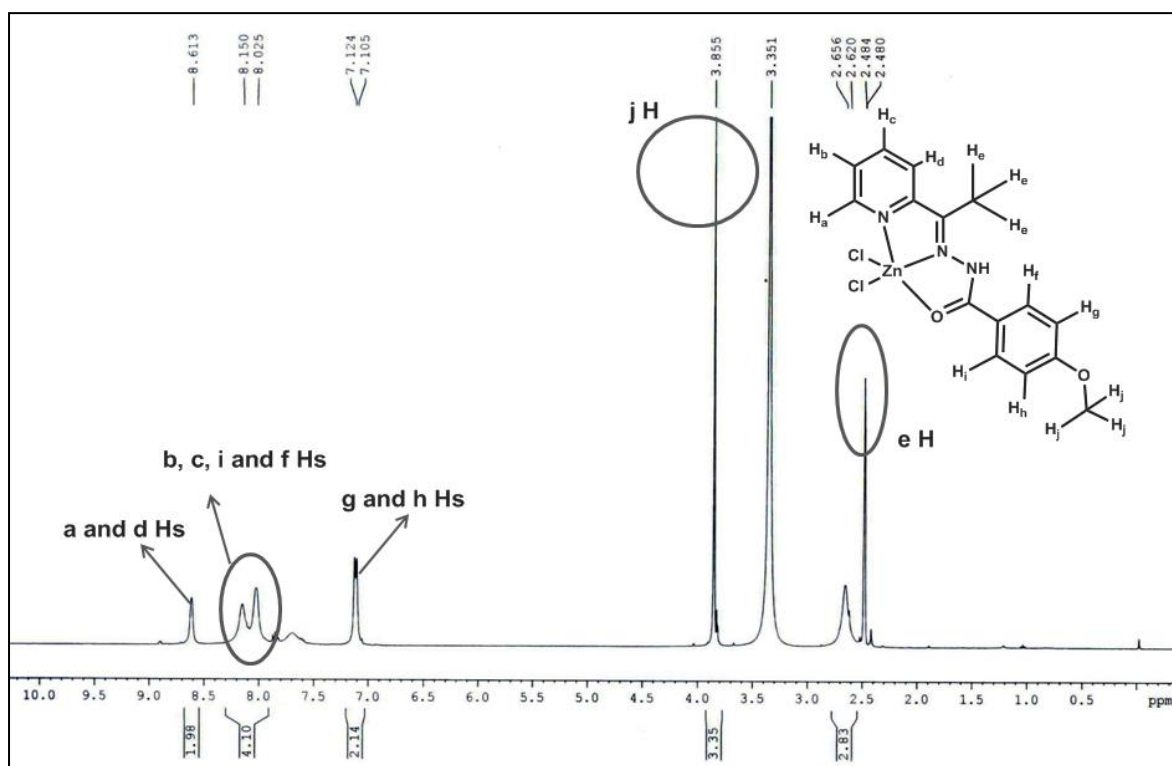
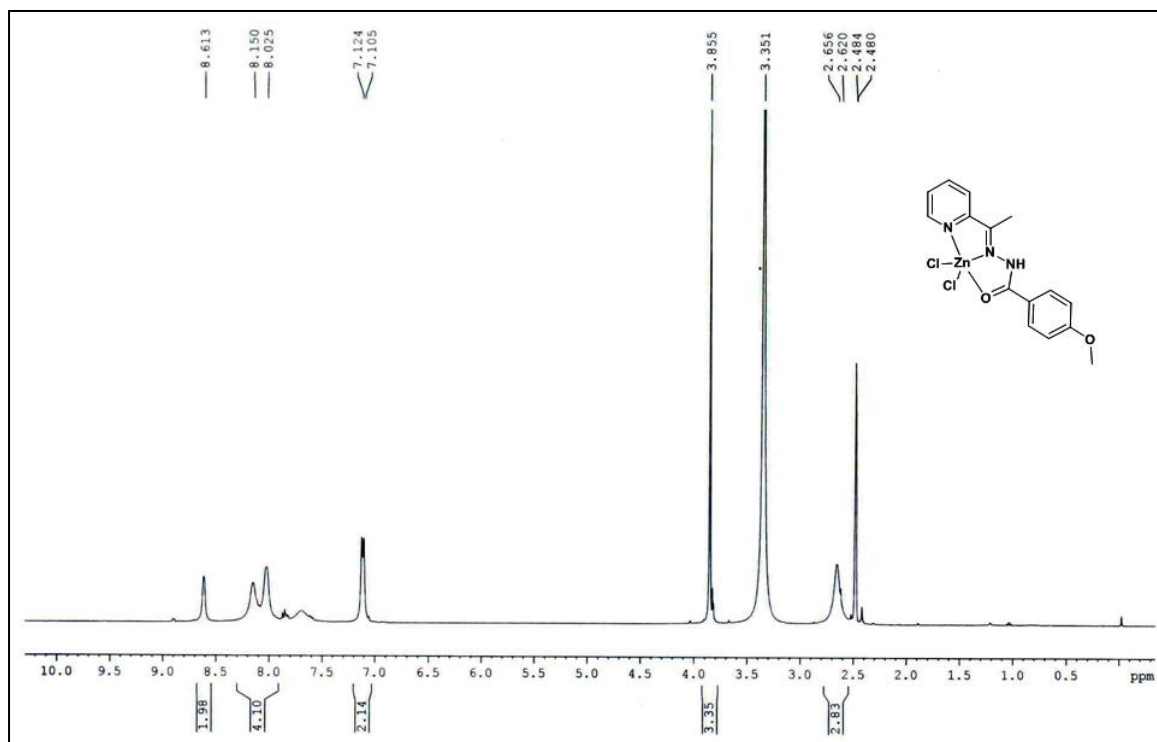


Fig. S6 ¹H NMR spectrum of complex 2 recorded in DMSO-d₆ solvent.

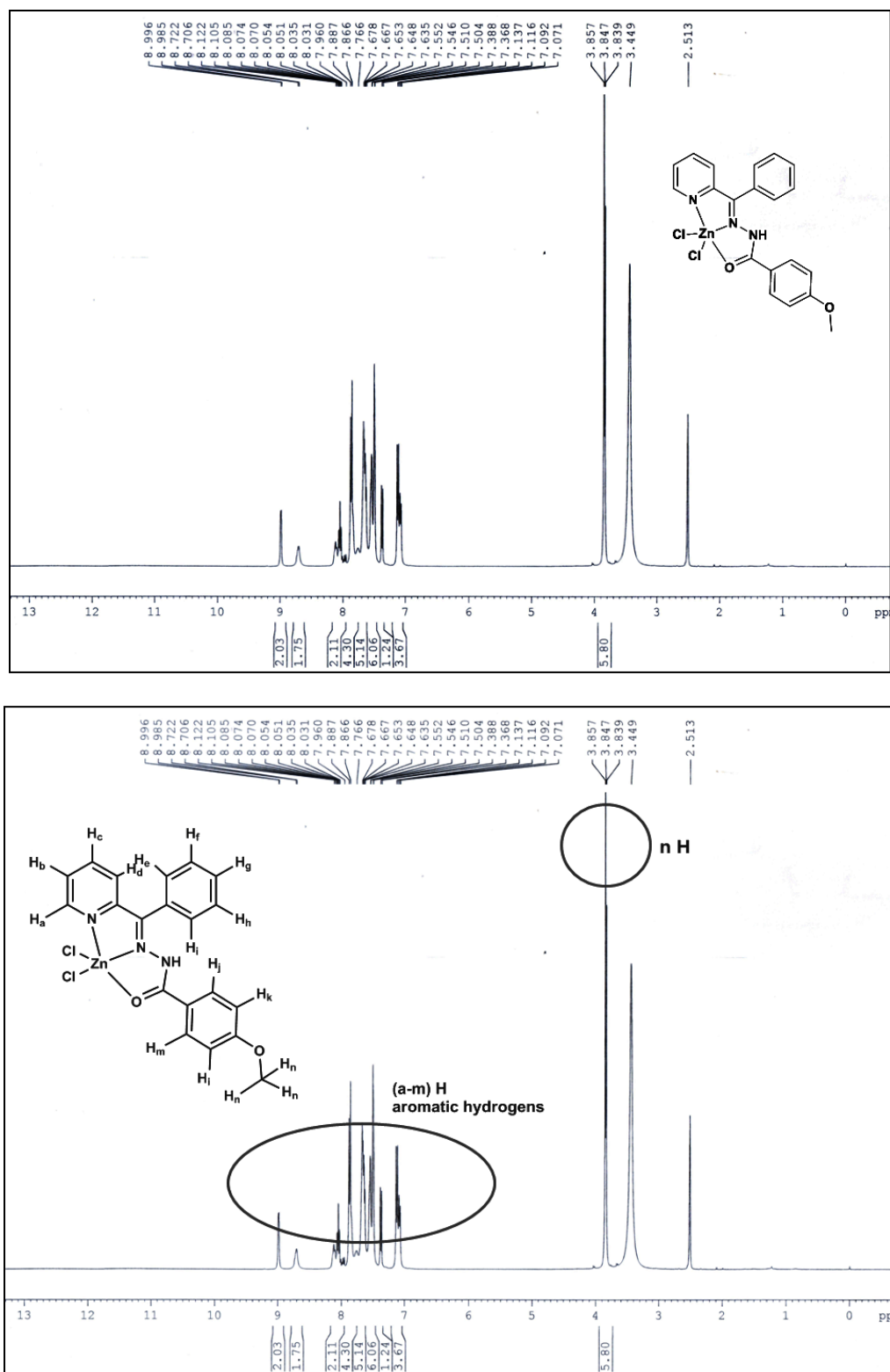


Fig. S7 ^1H NMR spectrum of complex **3** recorded in $\text{DMSO}-d_6$ solvent.

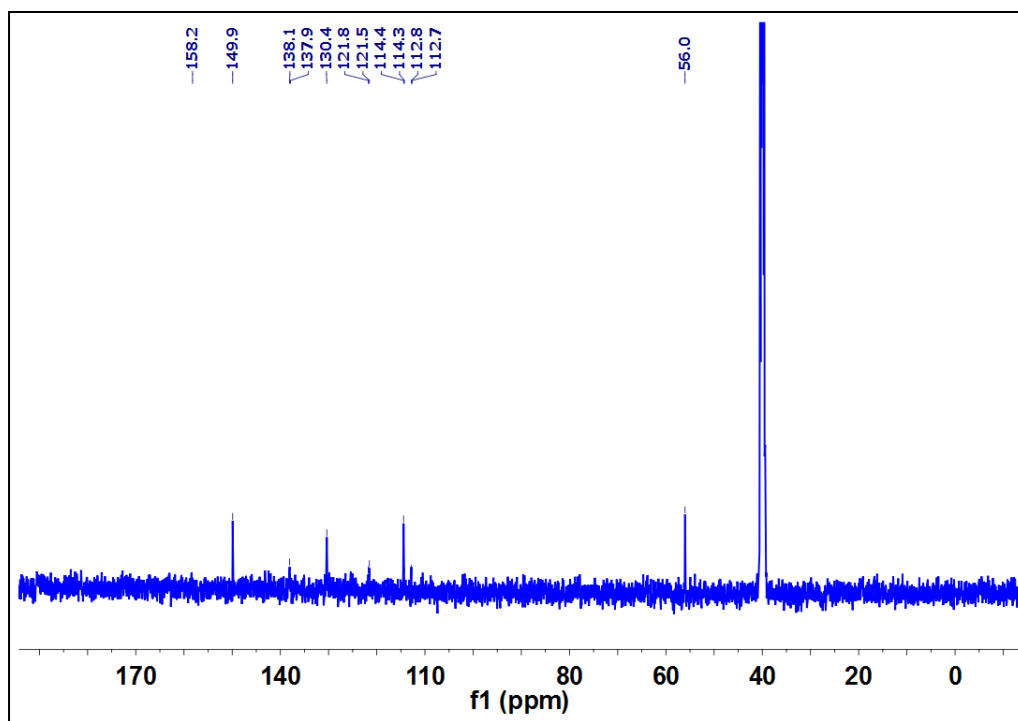


Fig. S8 ^{13}C NMR spectrum of complex **1** recorded in DMSO- d_6 solvent.

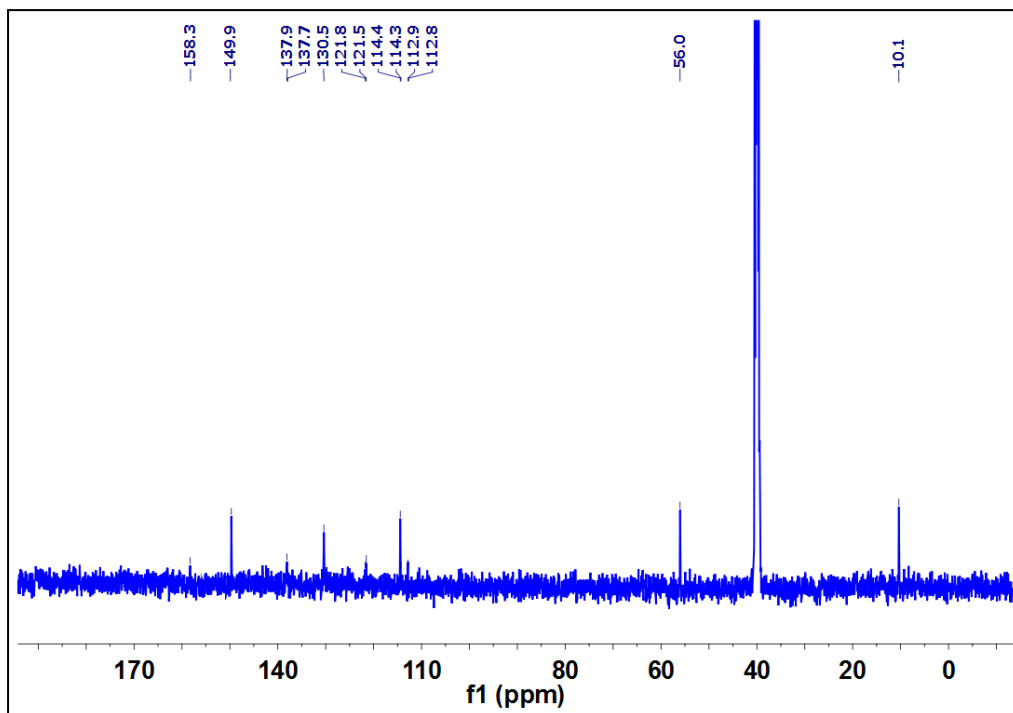


Fig. S9 ^{13}C NMR spectrum of complex **2** recorded in DMSO- d_6 solvent.

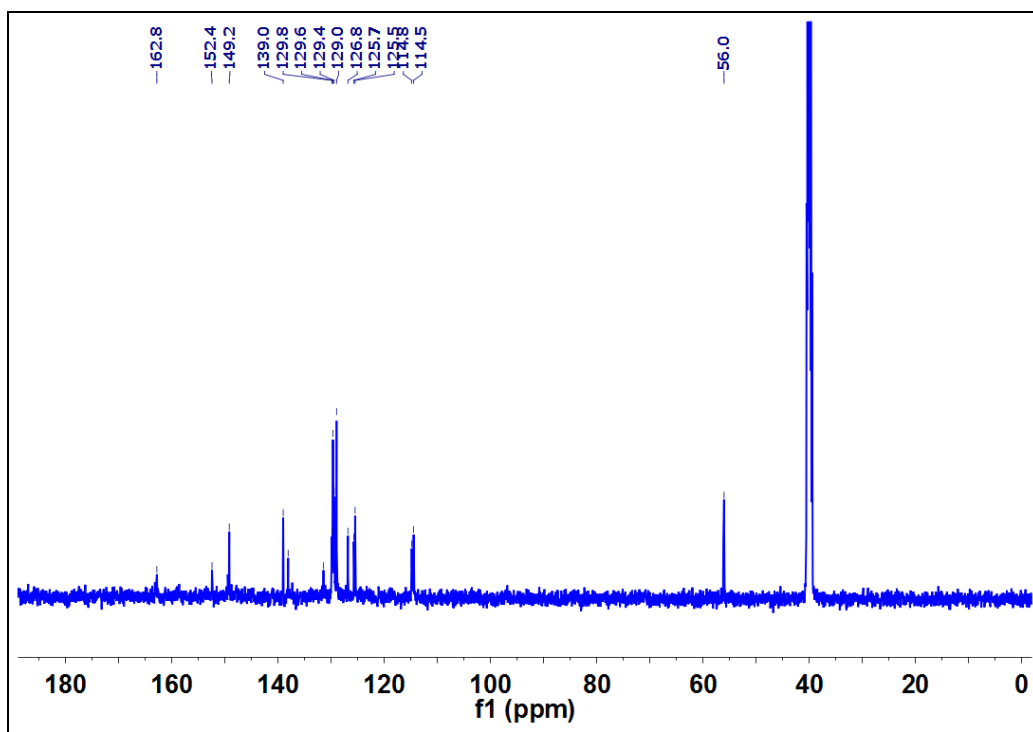


Fig. S10 ^{13}C NMR spectrum of complex 3 recorded in DMSO- d_6 solvent.

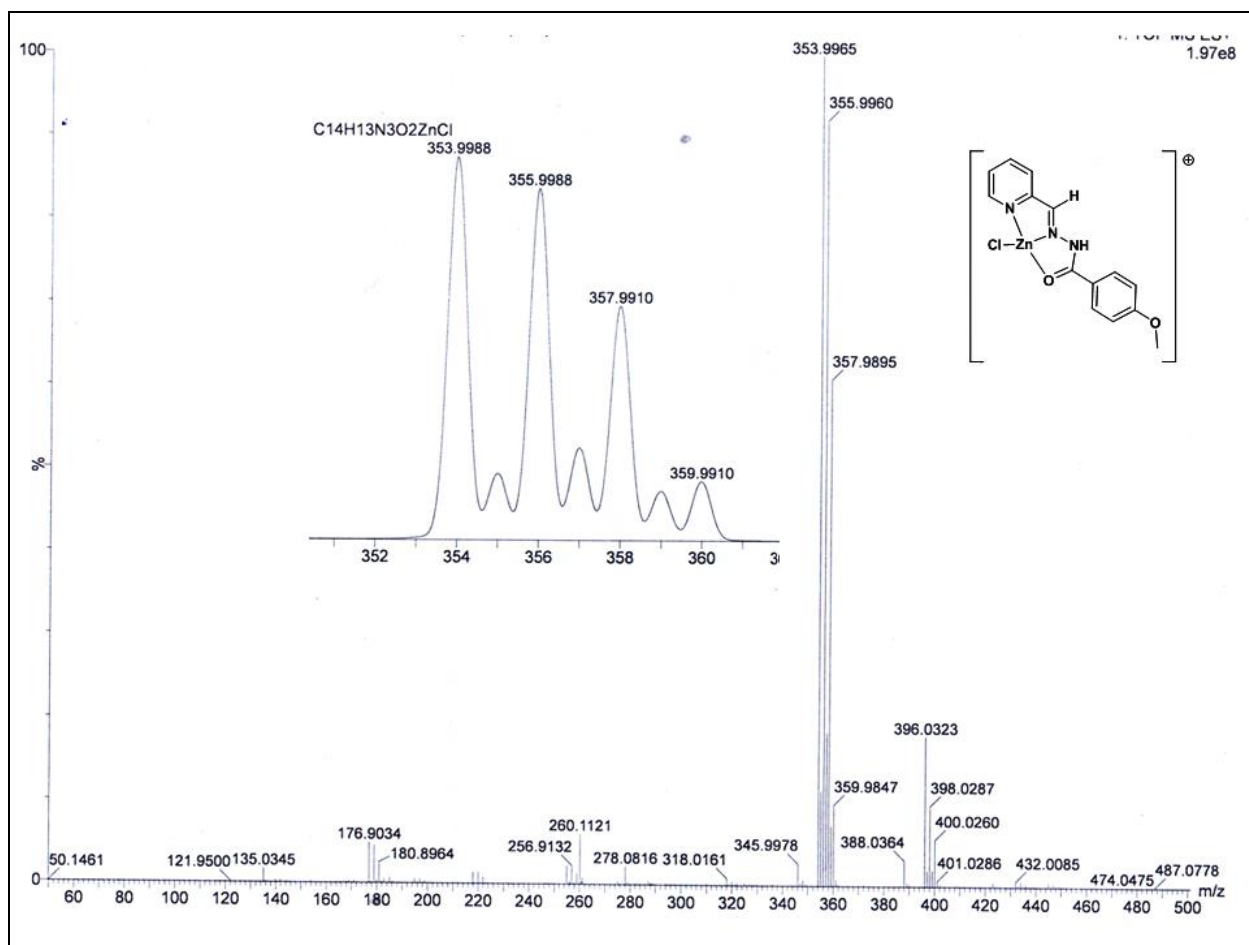


Fig. S11 ESI-MS spectrum of complex **1** in acetonitrile medium recorded in positive mode.

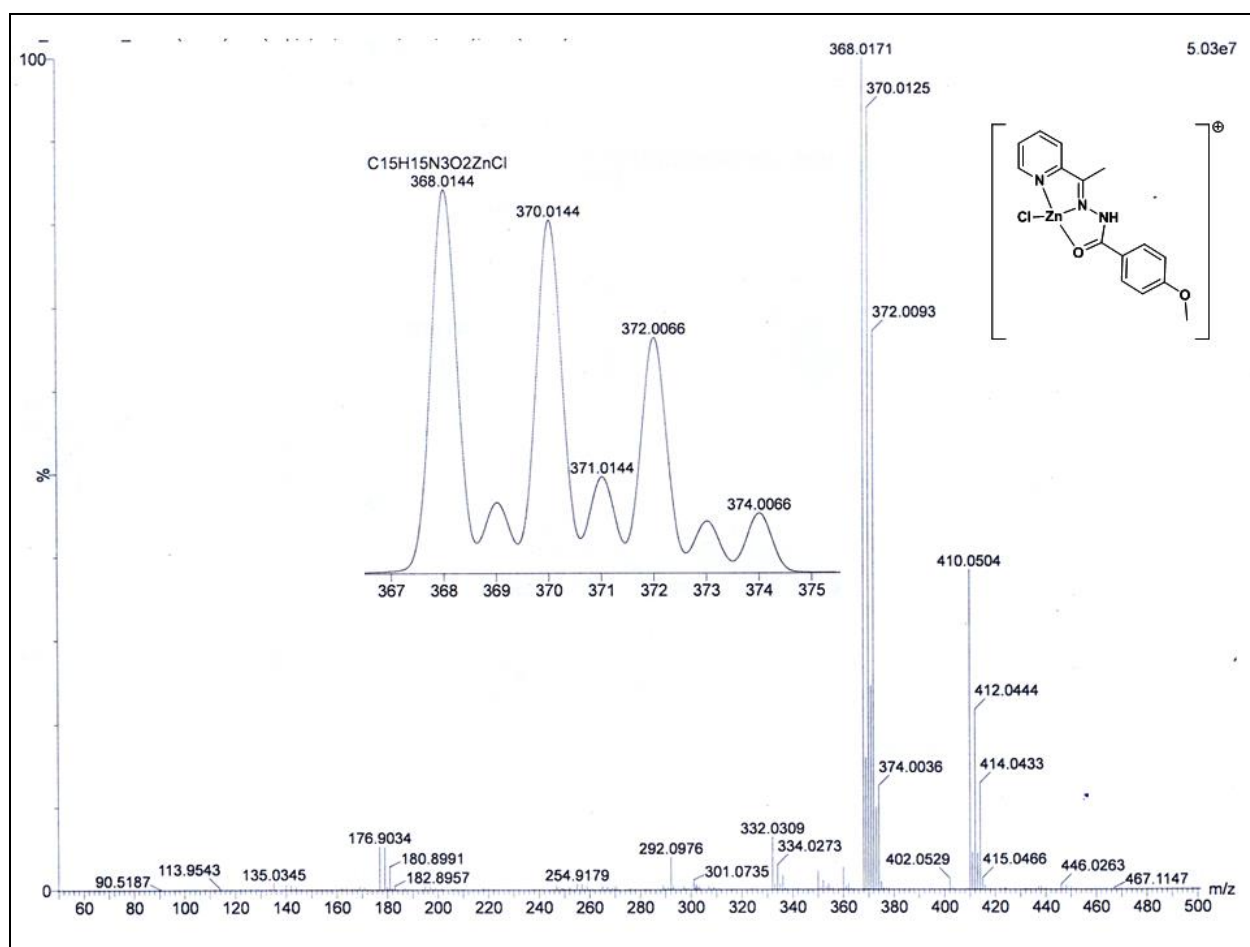


Fig. S12 ESI-MS spectrum of complex 2 in acetonitrile medium recorded in positive mode.

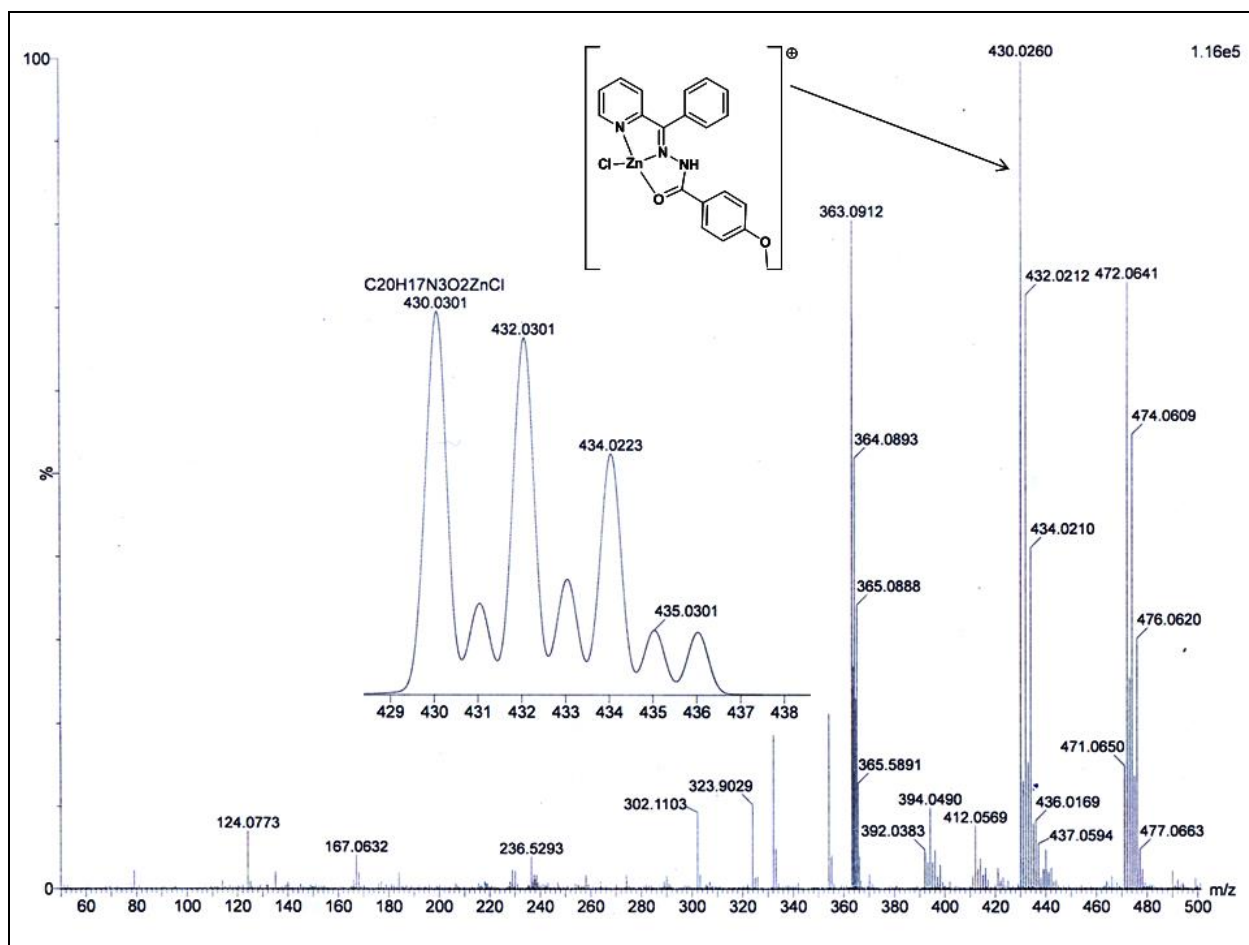


Fig. S13 ESI-MS spectrum of complex **3** in acetonitrile medium recorded in positive mode.

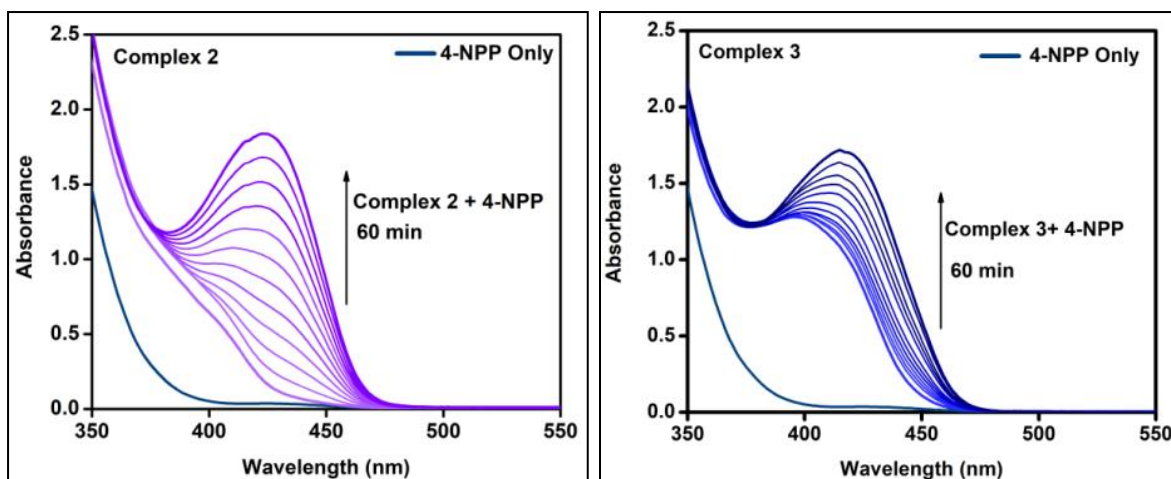


Fig. S14 Wavelength scan for hydrolysis of 4-NPP catalyzed by complex **2** and complex **3** (substrate: catalyst = 20:1) in DMSO-water mixture recorded at 25 °C at intervals of 5 min. [4-NPP] = 1×10^{-3} M; [complex] = 5×10^{-5} M

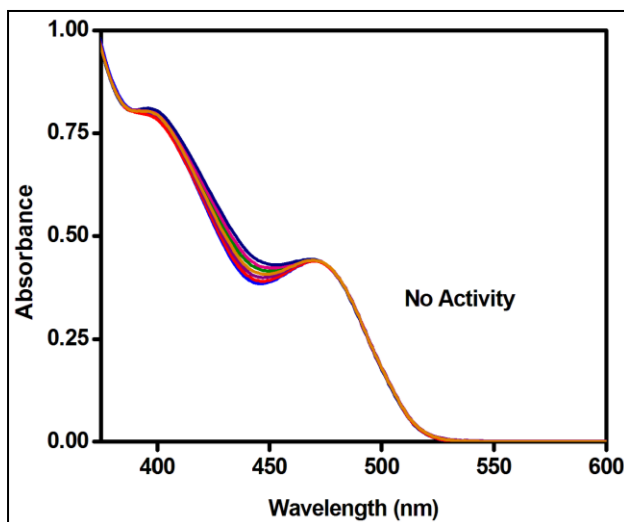


Fig. S15 Controlled experiment of phosphatase activity for zinc(II) chloride in DMSO-water medium at 25°C.

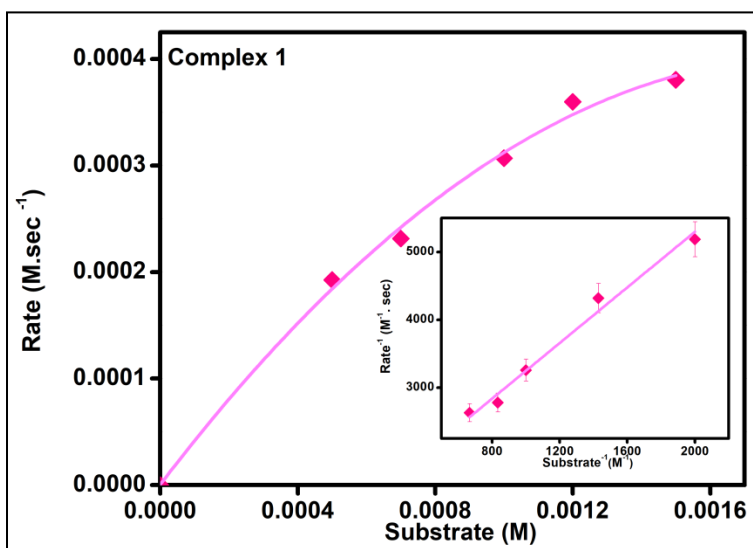


Fig. S16 Dependence of rate of reaction on substrate concentration for complex 1 and inset contains Lineweaver-Burk plots for complex 1 at 25°C in DMSO-water medium for hydrolysis of 4-nitrophenylphosphate.

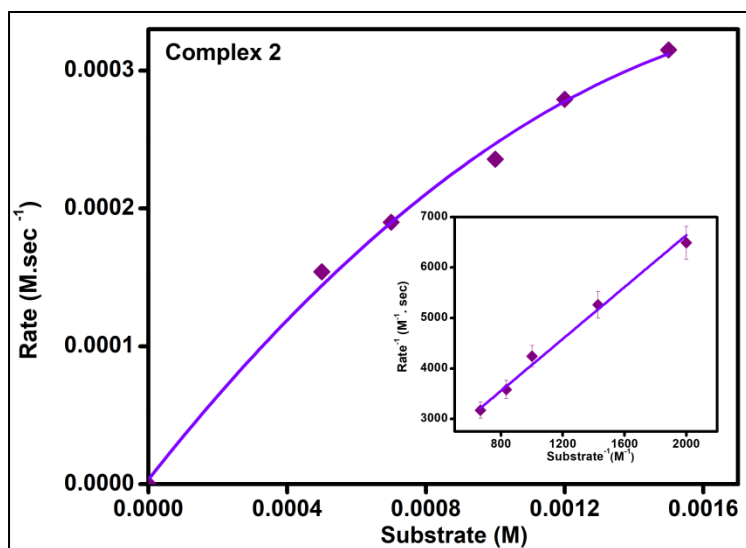


Fig. S17 Dependence of rate of reaction on substrate concentration for complex 2 and inset contains Lineweaver-Burk plots for complex 2 at 25°C in DMSO-water medium for hydrolysis of 4-nitrophenylphosphate.

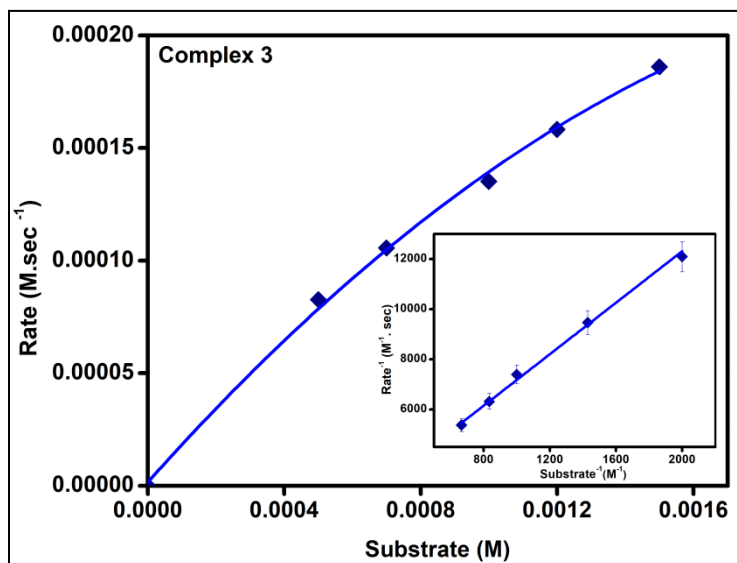


Fig. S18 Dependence of rate of reaction on substrate concentration for complex 3 and inset contains Lineweaver-Burk plots for complex 3 at 25°C in DMSO-water medium for hydrolysis of 4-nitrophenylphosphate.

Table S4. Comparison Table of phosphatase activity of some dinuclear bivalent metal complexes described in literature

Catalyst ^a	Substrate	k_{cat} (s ⁻¹) (Solvent)	Ref. ^{year}
1) [Zn(L1)Cl ₂].2H ₂ O	PNPP	16.15 s ⁻¹ (DMSO/H ₂ O)	Present work
2) [Zn(L2)Cl ₂]	PNPP	12.52 s ⁻¹ (DMSO /H ₂ O)	Present work
3) [Zn(L3)Cl ₂]	PNPP	9.116 s ⁻¹ (DMSO/H ₂ O)	Present work
4) [Zn ₃ (L ¹) ₂ (OAc) ₄]	PNPP	7.11× 10 ⁻² s ⁻¹ (MeOH/H ₂ O)	S1 ²⁰¹⁴
5) [Zn(L ²)(OAc)]	PNPP	4.841× 10 ⁻³ s ⁻¹ (MeOH/H ₂ O)	S1 ²⁰¹⁴
6) [Zn ₂ L ¹ ₂ Cl ₂]	PNPP	9.35 s ⁻¹ (DMF/H ₂ O)	22 ²⁰¹⁴
7) [Zn ₂ L ¹ ₂ Br ₂]	PNPP	8.80 s ⁻¹ (DMF/H ₂ O)	22 ²⁰¹⁴
8) [Zn ₂ (L) ₂ Cl ₂]	PNPP	9.97 s ⁻¹ (DMF/H ₂ O)	23 ²⁰¹⁵
9) [Zn ₂ (L) ₂ Br ₂]	PNPP	9.47 s ⁻¹ (DMF/H ₂ O)	23 ²⁰¹⁵
10) [Zn ₂ (L) ₂ I ₂]	PNPP	11.62 s ⁻¹ (DMF/H ₂ O)	23 ²⁰¹⁵
11) [Zn ₂ (L)(SCN) ₃]	PNPP	17.01×10 ⁻² s ⁻¹ (DMF/H ₂ O)	S2 ²⁰¹⁷
12) [Zn ₂ (LH ₋₂)]	BNPP	2.24×10 ⁻⁶ s ⁻¹ (H ₂ O)	S3 ²⁰⁰²
13) [Zn ₂ (μ-L ^{Cl} O)(μ ₂ OAc) ₂] (PF ₆)	BDNPP	0.48×10 ⁻³ s ⁻¹ (MeCN/H ₂ O)	S4 ²⁰¹⁶
14) [Ni ₂ (μ-L ^{Cl} O)(μ ₂ -OAc) ₂](PF ₆).3H ₂ O	BDNPP	2.80×10 ⁻³ s ⁻¹ (MeCN/H ₂ O)	S4 ²⁰¹⁶
15) [Zn ₂ (HL1)(μ-OAc)](PF ₆)	BNPP	1.26×10 ⁻⁶ s ⁻¹ (MeCN/H ₂ O)	S5 ²⁰⁰⁸
16) [Zn ₂ (L ¹ H ₋₁)(OH)](ClO ₄) ₂	BNPP	4.9×10 ⁻⁶ s ⁻¹ (DMSO/H ₂ O)	S6 ²⁰⁰⁵
17) [Zn ₂ (L ² H ₋₁) (MeOH) (OH)] (ClO ₄) ₂	BNPP	2.3×10 ⁻⁵ s ⁻¹ (DMSO/H ₂ O)	S6 ²⁰⁰⁵
18) [Zn ₂ (L ² H ₋₁) (MeOH) (OH)] (ClO ₄) ₂	BNPP	1.9×10 ⁻⁶ s ⁻¹ (DMSO/H ₂ O)	S7 ²⁰⁰⁴
19) [Zn ₂ (L ⁴ H ₋₁)]	BNPP	4.2×10 ⁻⁵ s ⁻¹ (DMSO/H ₂ O)	S7 ²⁰⁰⁴
20) [Zn ₂ (BPMP)(μ-OH)] ²⁺	HPNP	6.4×10 ⁻⁴ s ⁻¹ (DMSO/H ₂ O)	S8 ²⁰⁰⁷
21) [Cu ₂ (L)(μ-OH)(H ₂ O)](ClO ₄) ₂	PNPP	1.69 s ⁻¹ (DMSO/H ₂ O)	S9 ²⁰¹⁵
22) [Cu ₂ (L ¹)(μ-O ₂ CMe) ₂][NO ₃]	HPNP	14.5 × 10 ⁻⁴ s ⁻¹ (MeOH/H ₂ O)	S10 ²⁰¹⁷
23)[Cu ₄ (L) ₂ (HL) ₂ (ClO ₄) ₂].CH ₃ OH	PNPP	9.306 s ⁻¹ (DMF/H ₂ O)	25 ²⁰¹⁹
24)[Cu ₄ (L) ₂ (HL) ₂ (NO ₃) ₂].C ₂ H ₅ OH.0.5H ₂ O	PNPP	7.773 s ⁻¹ (DMF/H ₂ O)	25 ²⁰¹⁹
25)[Cu ₄ (L) ₂ (HL) ₂ (OAc) ₂].CH ₃ CN.H ₂ O	PNPP	3.365 s ⁻¹ (DMF/H ₂ O)	25 ²⁰¹⁹
26) [Ni ₂ L ¹ (CH ₃ COO) ₂ (SCN)].(H ₂ O) ₂ .(0.5CH ₃ OH)	PNPP	8.08×10 ⁻² s ⁻¹ (DMSO/H ₂ O)	26 ²⁰¹⁸
27) [Ni ₂ L ² (CH ₃ COO)(SCN) ₂ (CH ₃ OH)].(CH ₃ OH)	PNPP	5.18×10 ⁻² s ⁻¹ (DMSO/H ₂ O)	26 ²⁰¹⁸

$^a\text{HL}^1(4) = \{4\text{-Chloro-2-}[(2\text{-morpholin-4-ylethylimino})\text{-methyl}]\text{-phenol}\}$
 $\text{HL}^2(5) = \{4\text{-Chloro-2-}[(3\text{-morpholin-4-yl-propylimino})\text{-methyl}]\text{-phenol}\}$
 $\text{HL}^1(6,7) = 2\text{-}[(\text{bis}(2\text{-methoxyethyl})\text{amino})\text{methyl}]\text{-4-methylphenol}$
 $\text{HL}(8\text{-}10) = 2\text{-}[\text{bis}(2\text{-methoxyethyl})\text{aminomethyl}]\text{-4-isopropylphenol}$
 $\text{HL}(11) = 4\text{-tert-Butyl-2,6-bis-}[(2\text{-pyridin-2-yl-ethylimino})\text{-methyl}]\text{-phenol}$
 $\text{LH}(12) = 1,1'\text{-}(1H\text{-pyrazole-3,5-diyl})\text{bis}(\text{methylene})\text{bis}[\text{octahydro-1}H\text{-1,4,7-triazonine}]$
 $\text{HL}^{\text{Cl}}(13\text{-}14) = 2,6\text{-bis}[\text{bis}(2\text{-pyridylmethyl})\text{aminomethyl}]\text{-4-chlorophenol}$
 $\text{H}_3\text{L}(15) = N\text{-}(2\text{-hydroxy-3-}\{[(2\text{-hydroxyethyl})(\text{pyridin-2-ylmethyl})\text{amino}]\text{methyl}\}\text{-5-methylbenzyl})\text{-}N\text{-}(\text{pyridin-2-ylmethyl})\text{aminoacetic acid}$
 $\text{L}^1(16) = N,N'\text{-}(4H\text{-pyrazole-3,5-diyl})\text{bis}(\text{methylene})\text{bis}\{2\text{-}(\text{pyridin-2-yl})\text{-}N\text{-}[2\text{-}(\text{pyridin-2-yl})\text{ethyl}]\text{ethanamine}\}$
 $\text{L}^2(17) = N,N'\text{-}(4H\text{-pyrazole-3,5-diyl})\text{bis}(\text{methylene})\text{bis}\{1\text{-}(\text{pyridin-2-yl})\text{-}N\text{-}[2\text{-}(\text{pyridin-2-yl})\text{ethyl}]\text{methanamine}\}$
 $\text{L}^2(18) = N1,N1'\text{-}(4H\text{-pyrazole-3,5-diyl})\text{-bis}(\text{methylene})\text{bis}\{N1\text{-}[2\text{-}(\text{diethylamino})\text{ethyl}]\text{-}N2,N2\text{-diethylethane-1,2-diamine}\}$
 $\text{L}^4(19) = N1,N1'\text{-}(4H\text{-pyrazole-3,5-diyl})\text{bis}(\text{methylene})\text{bis}(N1,N2,N2\text{-trimethylethane-1,2-diamine})$
 $\text{HBPMMP}(20) = 2,6\text{-bis}\{[\text{bis}(\text{pyridin-2-ylmethyl})\text{amino}]\text{methyl}\}\text{-4-methylphenol}$
 $\text{HL}(21) = 4\text{-methyl-2,6-bis}((E)\text{-}(6\text{-methyl-1,4-diazepan-6-ylimino})\text{methyl})\text{phenol}$
 $\text{HL}^1(22) = 4\text{-methyl-2,6-bis}((3\text{-}(\text{pyridin-2-yl})\text{-1}H\text{-pyrazol-1-yl})\text{methyl})\text{phenol}$
 $\text{H}_2\text{L}(23\text{-}25) = 4\text{-bromo-2-}[(2\text{-hydroxy-1,1-dimethyl-ethylamino})\text{-methyl}]\text{-phenol}$
 $\text{HL}^1(26) = 2,6\text{-bis}((E)\text{-}(2\text{-morpholinoethylimino})\text{methyl})\text{-4-tert-butylphenol}$
 $\text{HL}^2(27) = 2,6\text{-bis}((E)\text{-}(2\text{-piperidin-1-yl})\text{ethylimino})\text{methyl})\text{-4-tert-butylphenol}$

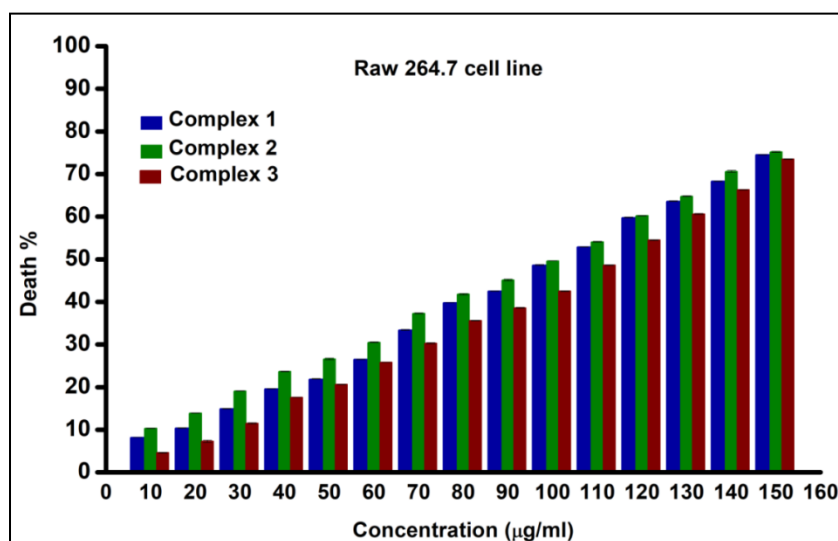
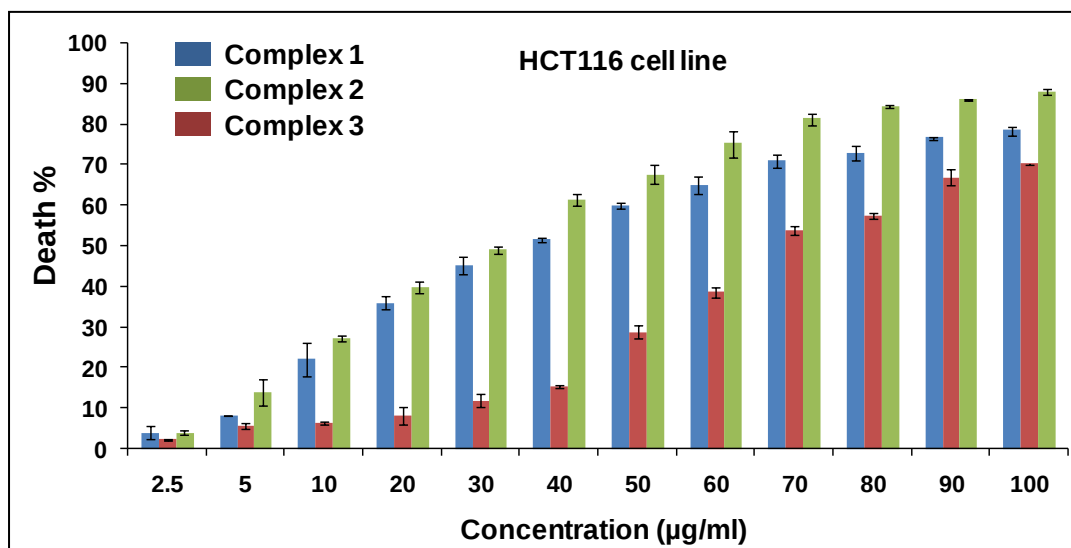
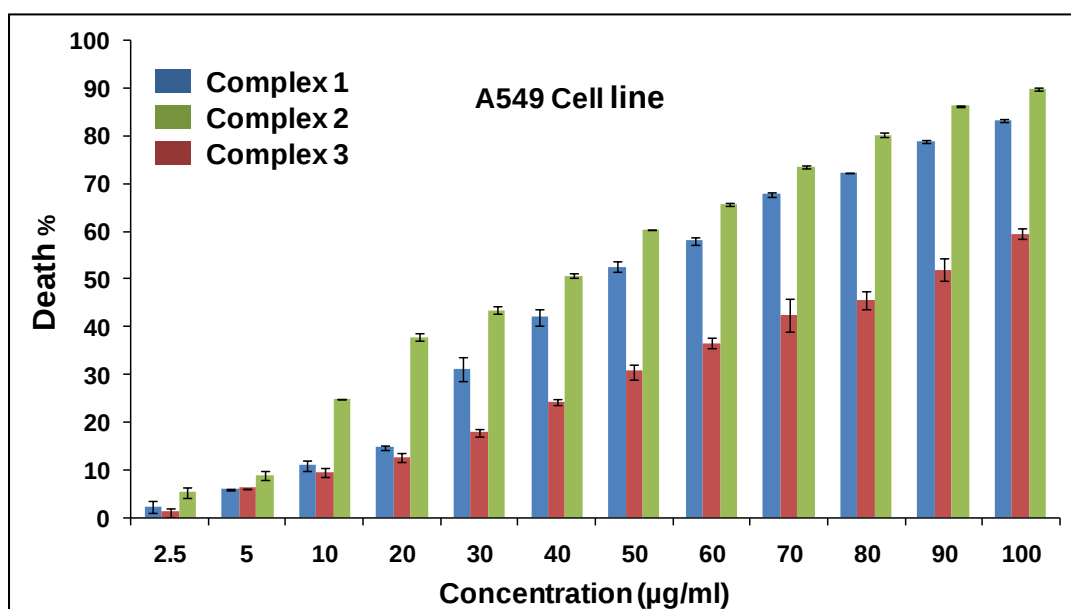


Fig. S19. Cell viability with different concentration (10-150μg/mL) in macrophage cell line (Raw 264.7) for 24 h and cell viability was assayed. Data are delegate as the mean $\pm$ SD and are the collective results of three independent experiments.



(a)



(b)

Fig. S20. Cell viability with different concentration (2.5-100µg/mL) in different cell lines (a) HCT116 (human colorectal carcinoma) and (b) A549 (human non-small lung carcinoma) for 24 h and cell viability was assayed. Data are delegate as the mean $\pm$ SD and are the collective results of three independent experiments.

Table S5. Comparison Table of IC₅₀ of some previously reported complexes described in literature

Metallo drug ^a	Cell Line	IC₅₀	Ref.^{year}
1) [Zn(L1)Cl ₂].2H ₂ O	HepG2	31.53 ± 1.6 µg/mL (81.1 µM)	Present work
	A549	48.26 ± 1.7 µg/mL (125 µM)	
	HCT116	43.70 ± 0.56 µg/mL (112.4 µM)	
2) [Zn(L2)Cl ₂]	HepG2	19.83 ± 1.6 µg/mL (49.2 µM)	Present work
	A549	41.85 ± 0.57 µg/mL (103.8 µM)	
	HCT116	38.66 ± 0.91 µg/mL (95.9 µM)	
3) [Zn(L3)Cl ₂]	HepG2	66.24 ± 0.38 µg/mL (142.2 µM)	Present work
	A549	87.63 ± 1.8 µg/mL (188.2 µM)	
	HCT116	70.44 ± 1.18 µg/mL (151.3 µM)	
4) <i>Cis</i> -platin	HepG2	9.6 ± 2.3 µM	42 ²⁰¹⁸
	A549	30 ± 5.0 µM	S11 ²⁰¹⁶
	HCT116	25.7 ± 6.3 µM	42 ²⁰¹⁸
5) [Zn ₃ L ² (µ-O ₂ CCH ₃) ₂ -(CH ₃ OH) ₄]	HepG2	70 ± 0.1 µM	S12 ²⁰¹⁴
6) Zn(L1)	HepG2	706 µg/mL	S13 ²⁰¹⁴
7) aqua[2,2'-[4-methyl-1,2-henylenebis(nitromethylidene)]dinaphthalato-κ ⁴ O,N,N',O']zinc(II)	HepG2	9.05 µM	S14 ²⁰¹³
8) [Zn(ityp)(OAc)]-OAc	A549	9.05 µM	S15 ²⁰¹⁴
6) [{Zn(µ ² -H ₂ O) _{0.5} (5N ₃ -IPA)(2,2'-bpe)}] _∞	HCT116	31.12 ± 1.78 µg/mL	S16 ²⁰¹⁸
7) [{Zn(µ ² -H ₂ O) _{0.5} (5N ₃ -IPA)(1,10-phen)}] _∞	HCT116	25.56 ± 2.14 µg/mL	S16 ²⁰¹⁸
8) [{Zn(5N ₃ -IPA)(1,2-bpe)}] _∞	HCT116	25.75 ± 1.14 µg/mL	S16 ²⁰¹⁸
9) [{Zn(5N ₃ -IPA)(1,2-bpey)}] _∞	HCT116	31.96 ± 1.54 µg/mL	S16 ²⁰¹⁸
10) [{Zn(H ₂ O) (5N ₃ -IPA) (4,4'-tme)} (H ₂ O) _{0.5}] _∞	HCT116	33.98 ± 1.47 µg/mL	S16 ²⁰¹⁸
11) [Ru(hmb)(L1)Cl]	HCT116	115 ± 25 µM	42 ²⁰¹⁸
	HepG2	>200 µM	42 ²⁰¹⁸
12) [Ru(cym)(L1)Cl]	HCT116	113 ± 40 µM	42 ²⁰¹⁸
	HepG2	>200 µM	42 ²⁰¹⁸
13) [Ru(hmb)(L2)Cl]	HCT116	>200 µM	42 ²⁰¹⁸
	HepG2	>200 µM	42 ²⁰¹⁸
14) [Ru(cym)(L2)Cl]	HCT116	175 ± 10 µM	42 ²⁰¹⁸

	HepG2	>200 μ M	42 ²⁰¹⁸
15)[Ru(hmb)(L3)Cl]	HCT116	>200 μ M	42 ²⁰¹⁸
	HepG2	>200 μ M	42 ²⁰¹⁸
16)[Ru(cym)(L3)Cl]	HCT116	163 $\pm$ 20 μ M	42 ²⁰¹⁸
	HepG2	>200 μ M	42 ²⁰¹⁸
^a H ₂ L (5) = 2-[(2-hydroxyphenylimino)methyl]-6-methoxyphenol L1 (6) = 4-chloro-2-(((4-morpholinophenyl)imino)methyl)phenol itpy (8) = 4'-(1H-imidazol-2-yl)-2,2':6',2''-terpyridine 5N ₃ -IPA (5-9) = 5-azidoisophthalic acid HL1(11-12) = 2-((5-hydroxo-3-methyl-1-phenyl-1H-pyrazol-4-yl)(phenyl)methylene)-1-(2,4-nitrophenyl)hydrazine HL2 (13-14) = 2-((5-hydroxo-3-methyl-1-phenyl-1H-pyrazol-4-yl) (phenyl)methylene)-1-(4-nitrophenyl)-Hydrazine HL3 (15-16)= 2-((5-hydroxo-3-methyl-1-phenyl-1Hpyrazol-4-yl)(phenyl)methylene)-1-(pyridin-2-yl)hydrazine)			

References.

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