

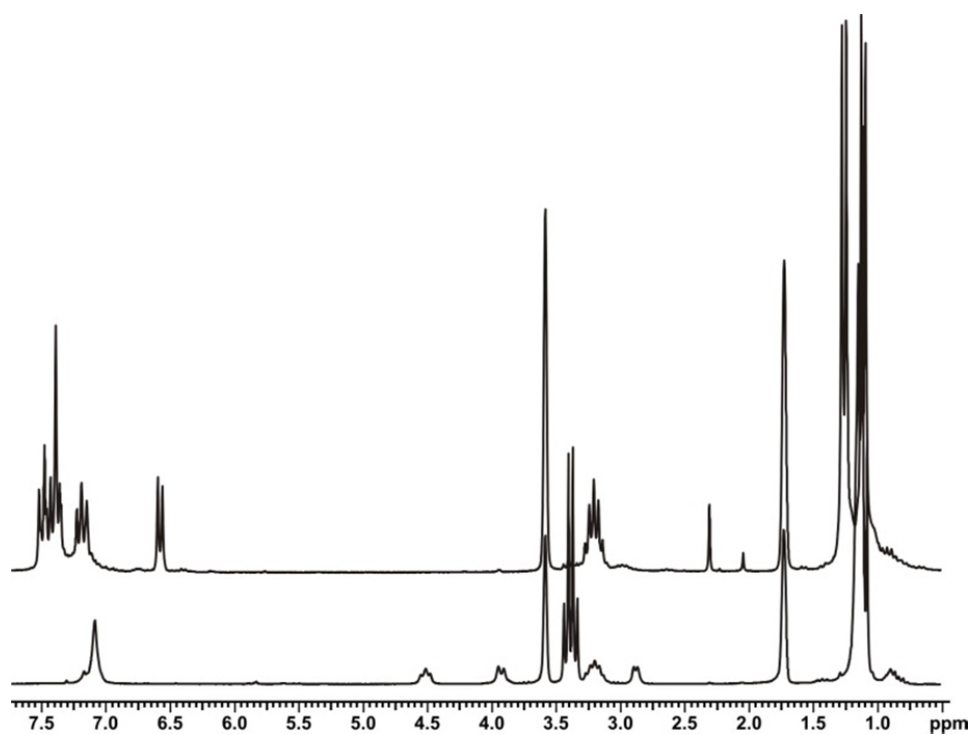
## Alkali Metal Reduction of 1,3,2-Diazaborol and 1,3,2-Diazagermol Derivatives Based on 1,2-Bis[(2,6-diisopropylphenyl)imino]acenaphthene

Daria A. Lukina,<sup>a</sup> Alexandra A. Skatova,<sup>a</sup> Anton N. Lukoyanov,<sup>a</sup> Ekaterina A. Kozlova<sup>a</sup> and Igor L. Fedushkin\*<sup>a</sup>  
G. A. Razuvaev Institute of Organometallic Chemistry of the Russian Academy of Sciences, Tropinina Str. 49,  
Nizhny Novgorod 603137, Russian Federation

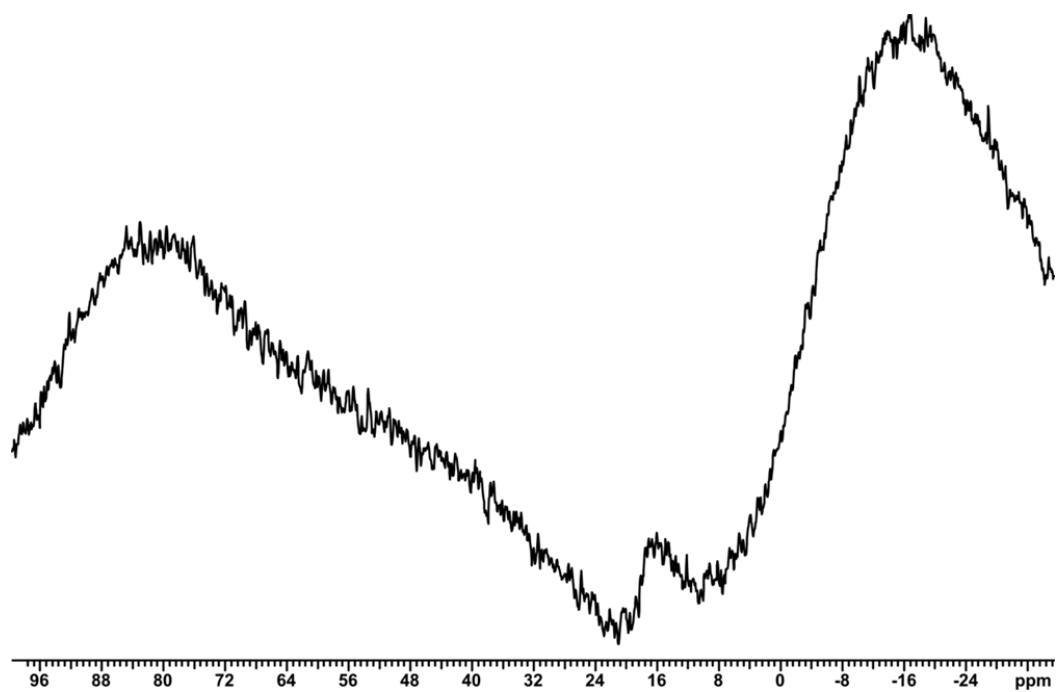
**X-ray crystallography.** The X-ray data for **3**, **4** and **8** were collected at 100(2) K on a Bruker AXS D8 Quest Photon and Agilent Xcalibur E (MoK $\alpha$ -radiation,  $\omega$ -scan technique,  $\lambda = 0.71073 \text{ \AA}$ ). The structures were solved by direct methods and were refined by full-matrix least squares on  $F^2$  using SHELXTL [Sheldrick G.M. (2014). SHELXTL v. 2014/7, Structure Determination Software Suite, Bruker AXS, Madison, Wisconsin, USA]. All hydrogen atoms were placed in calculated positions and were refined in the riding model. SCALE3 ABSPACK [SCALE3 ABSPACK: Empirical absorption correction, CrysAlisPro Software Package Agilent Technologies, 2012] for **3**, **8** and SADABS [Sheldrick G.M. (2012). SADABS v.2012/1, Bruker/Siemens Area Detector Absorption Correction Program, Bruker AXS, Madison, Wisconsin, USA] for **4** were used to perform area-detector scaling and absorption corrections. Crystallographic data and structural refinement details are given in Table 1. CCDC 1957529 (**3**), 1957531 (**4**), 1957530 (**8**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

**Table 1.** Crystal Data and Structure Refinement Details for Compounds **3**, **4** and **8**.

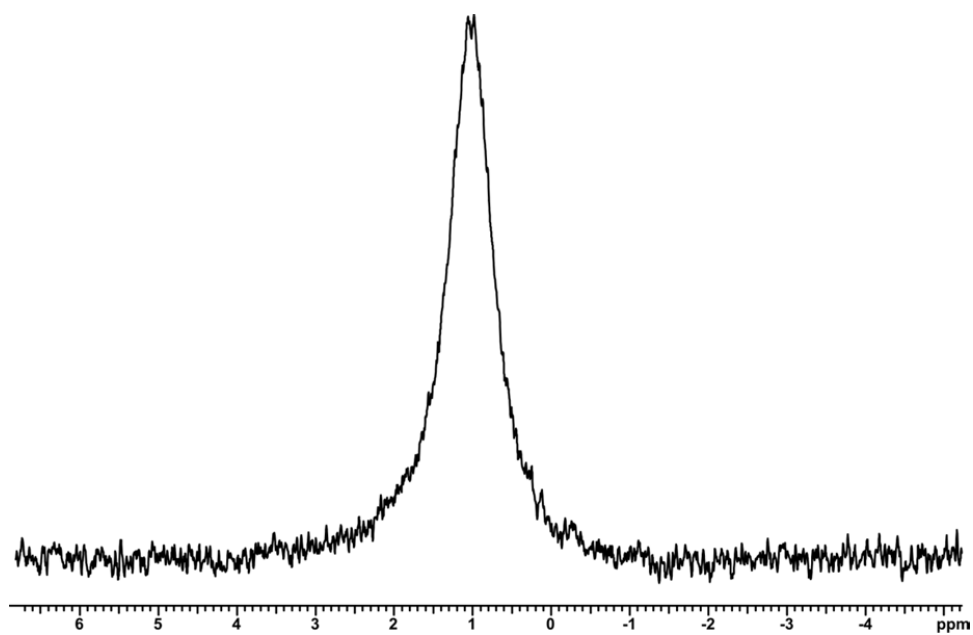
	<b>3</b> · 0.875 C <sub>10</sub> H <sub>8</sub>	<b>4</b>	<b>8</b>
Formula	C <sub>96.75</sub> H <sub>127</sub> B <sub>2</sub> Br <sub>2</sub> Li <sub>4</sub> N <sub>4</sub> O <sub>4</sub>	C <sub>79</sub> H <sub>88</sub> B <sub>2</sub> K <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	C <sub>48</sub> H <sub>71</sub> GeN <sub>2</sub> NaO <sub>6</sub>
Formula weight	1619.22	1225.35	867.64
Temperature/K	100(2)	100(2)	100(2)
Crystal system	Triclinic	Tetragonal	Orthorhombic
Space group	P-1	I-4	P2(1)2(1)2(1)
<i>a</i> /Å	13.2924(4)	33.394(3)	12.205(3)
<i>b</i> /Å	14.4434(3)	33.394(3)	19.4474(18)
<i>c</i> /Å	24.8139(7)	12.3370(9)	20.653(6)
$\alpha$ /deg	101.192(2)	90	90
$\beta$ /deg	102.617(2)	90	90
$\gamma$ /deg	91.829(2)	90	90
<i>V</i> /Å <sup>3</sup>	4546.7(2)	13758(2)	4902.1(18)
<i>Z</i>	2	8	4
density/g/cm <sup>3</sup>	1.183	1.183	1.176
$\mu$ /mm <sup>-1</sup>	0.942	0.187	0.679
<i>F</i> (000)	1719	5232	1856
Crystal size /mm	0.700 × 0.435 × 0.330	0.350 × 0.250 × 0.200	0.510 × 0.270 × 0.110
$\theta$ -range/deg	3.038–26.000	0.862–24.999	3.556–25.999
Index ranges	–16 ≤ <i>h</i> ≤ 16 –17 ≤ <i>k</i> ≤ 17 –30 ≤ <i>l</i> ≤ 30	–39 ≤ <i>h</i> ≤ 38 –20 ≤ <i>k</i> ≤ 39 –14 ≤ <i>l</i> ≤ 14	–15 ≤ <i>h</i> ≤ 13 –22 ≤ <i>k</i> ≤ 23 –15 ≤ <i>l</i> ≤ 25
Reflections collected	73871	38965	17664
Independent reflections	17853 [ <i>R</i> <sub>int</sub> = 0.0719]	12049 [ <i>R</i> (int) = 0.0594]	9511 [ <i>R</i> <sub>int</sub> = 0.0424]
Goodness-of-fit on $F^2$	1.004	1.018	1.011
<i>R</i> <sub>1</sub> / <i>wR</i> <sub>2</sub> ( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))	0.0551 / 0.1285	0.0658 / 0.1604	0.0478 / 0.0906
<i>R</i> <sub>1</sub> / <i>wR</i> <sub>2</sub> (all parameters)	0.0997 / 0.1456	0.0959 / 0.1753	0.0733 / 0.0995
Largest diff peak/hole [e Å <sup>-3</sup> ]	0.880 / –0.848	0.808 / –0.293	0.306 / 0.303



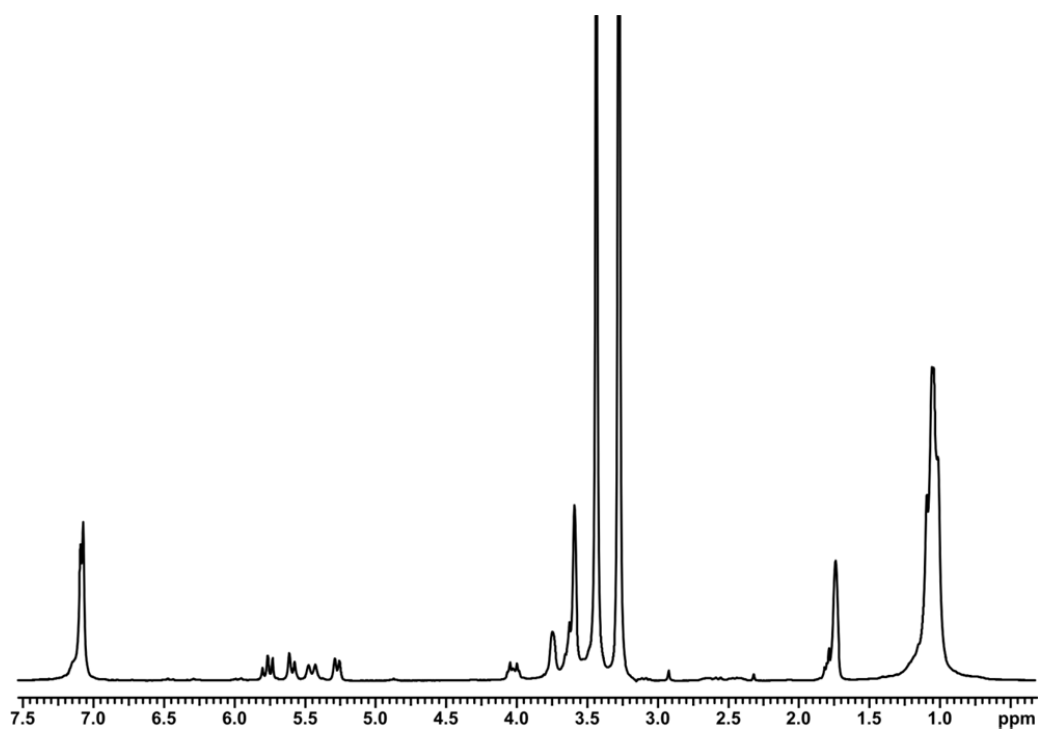
**Figure 1S.**  $^1\text{H}$  NMR spectra of the compounds **1** (top) and **3** (bottom) (200 MHz, 300 K,  $\text{thf-d}_8$ ).



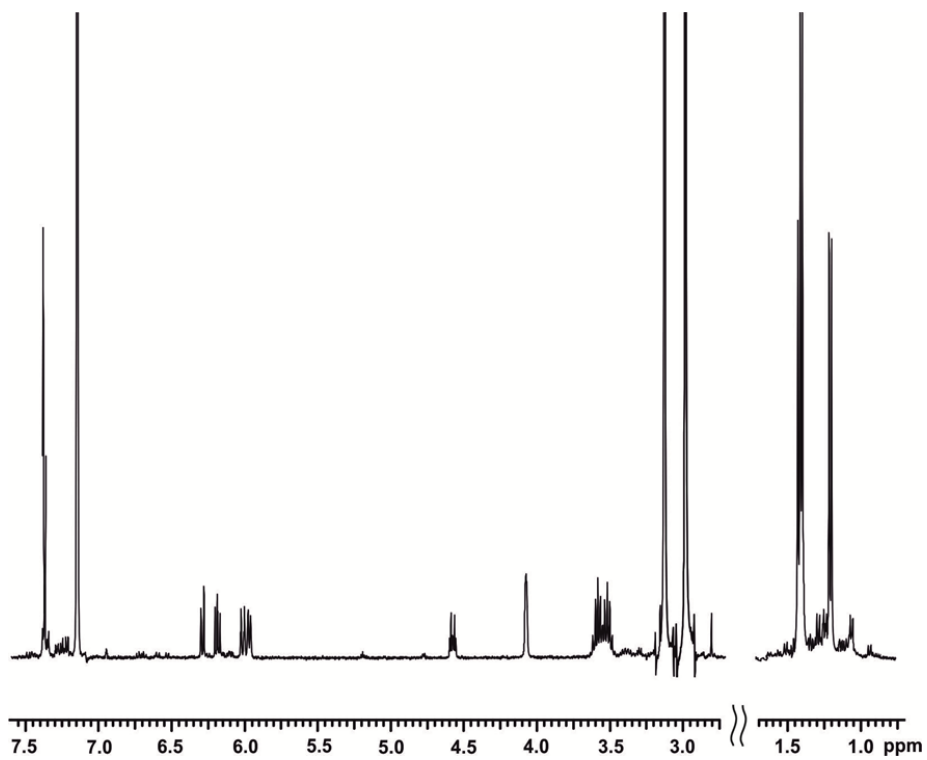
**Figure 2S.**  $^{11}\text{B}$  NMR spectrum of the compound **3** (64.21 MHz, 300 K,  $\text{thf-d}_8$ ).



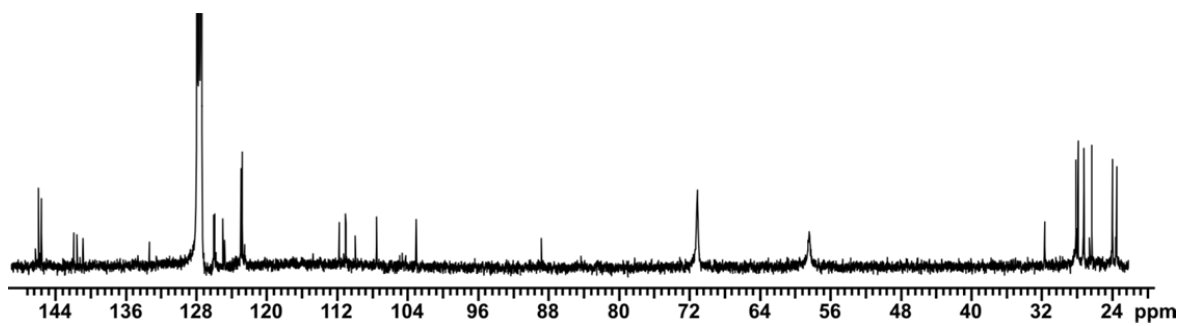
**Figure 3S.**  $^7\text{Li}$  NMR spectrum of the compound **3** (77.78 MHz, 300 K, thf- $d_8$ ).



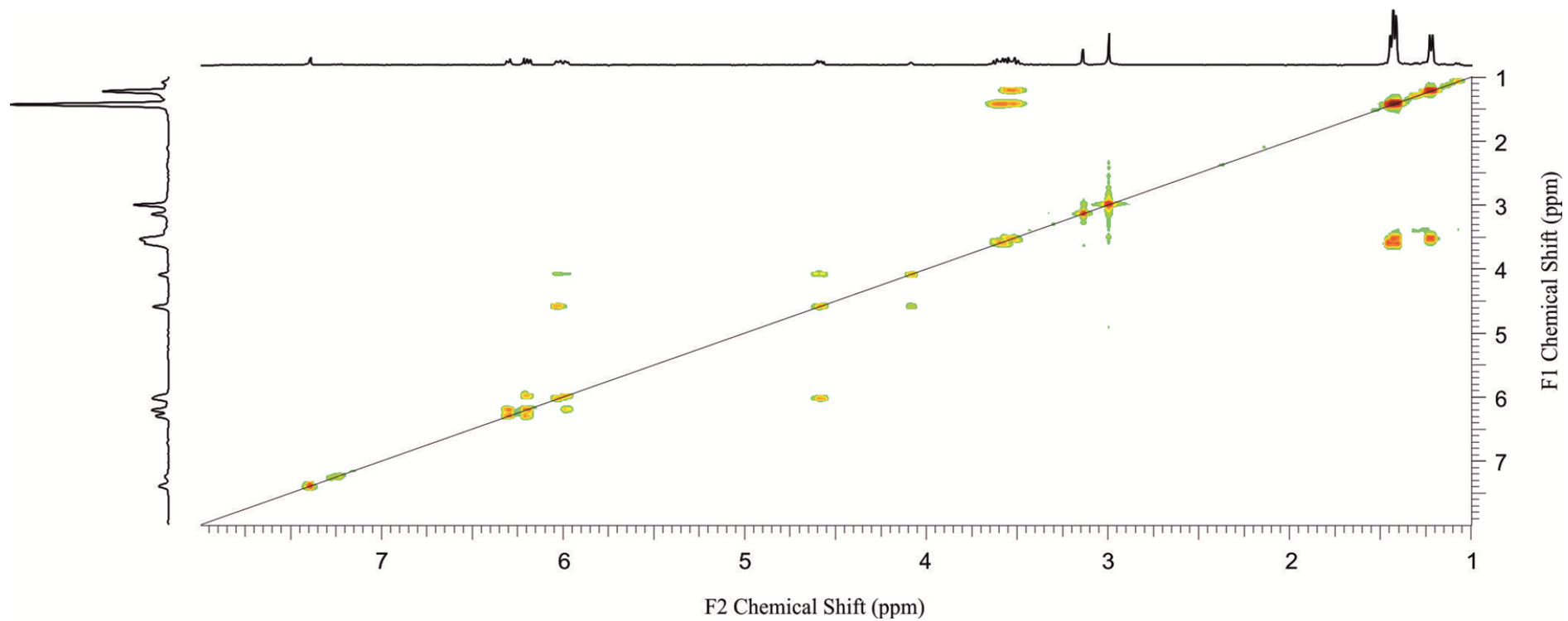
**Figure 4S.**  $^1\text{H}$  NMR spectrum of the compound **6** (200 MHz, 298 K, thf- $d_8$ )



**Figure 5S.**  $^1\text{H}$  NMR spectrum of the compound **8** (400 MHz, 296 K,  $\text{C}_6\text{D}_6$ )



**Figure 6S.**  $^{13}\text{C}$  NMR spectrum of the compound **8** (100 MHz, 296 K,  $\text{C}_6\text{D}_6$ )



**Figure 7S.**  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of **8** (400 MHz, 296 K,  $\text{C}_6\text{D}_6$ ).