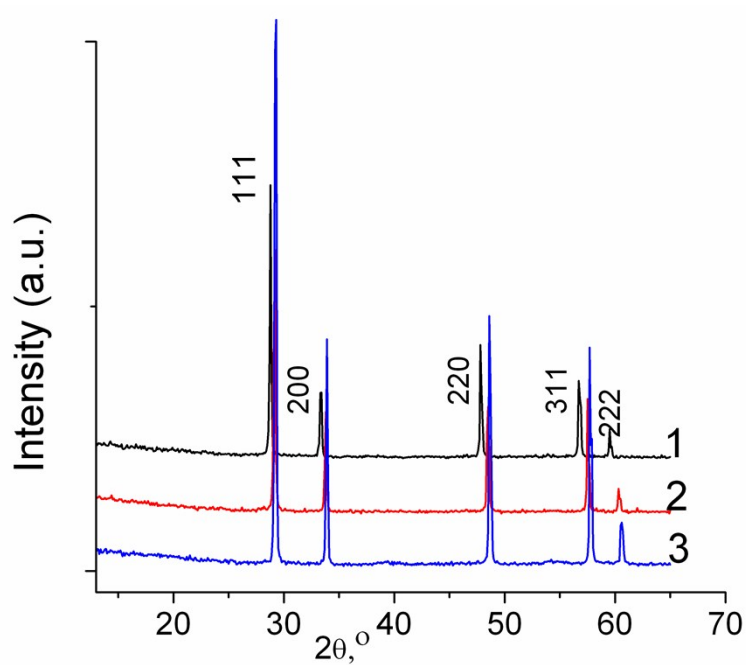
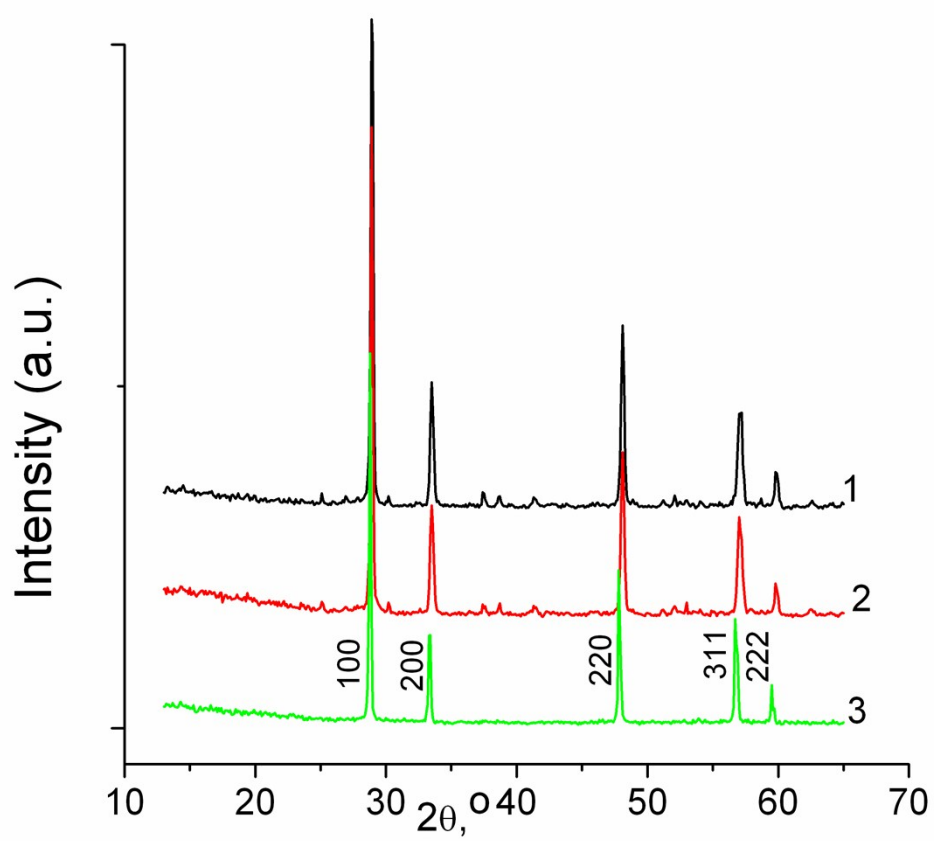


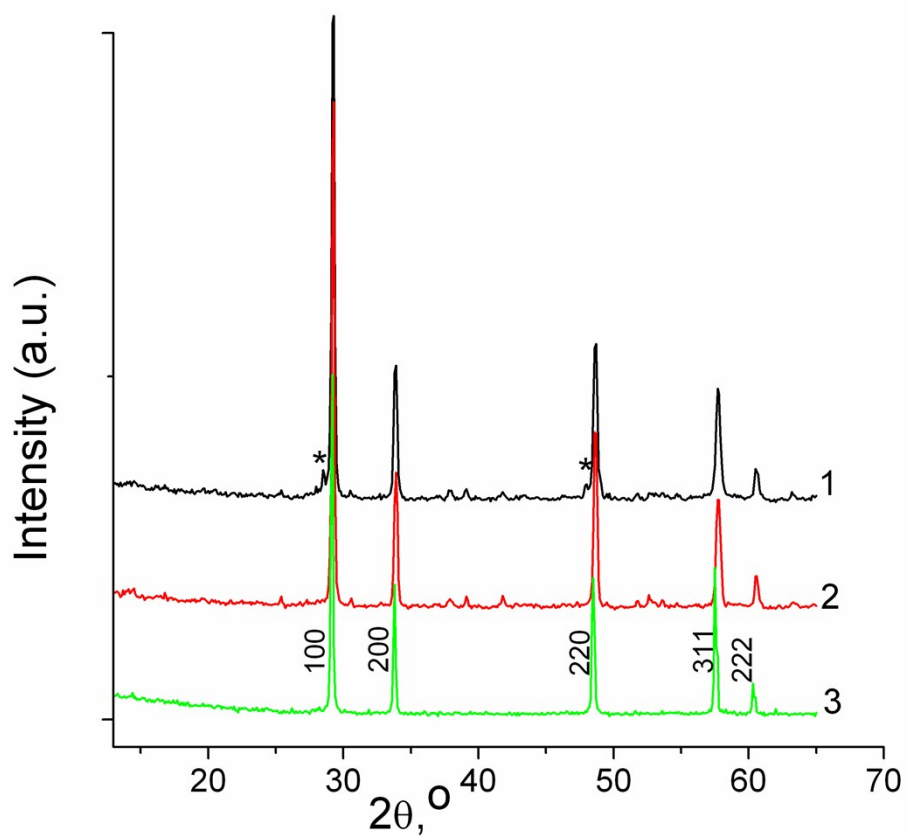
## Figures



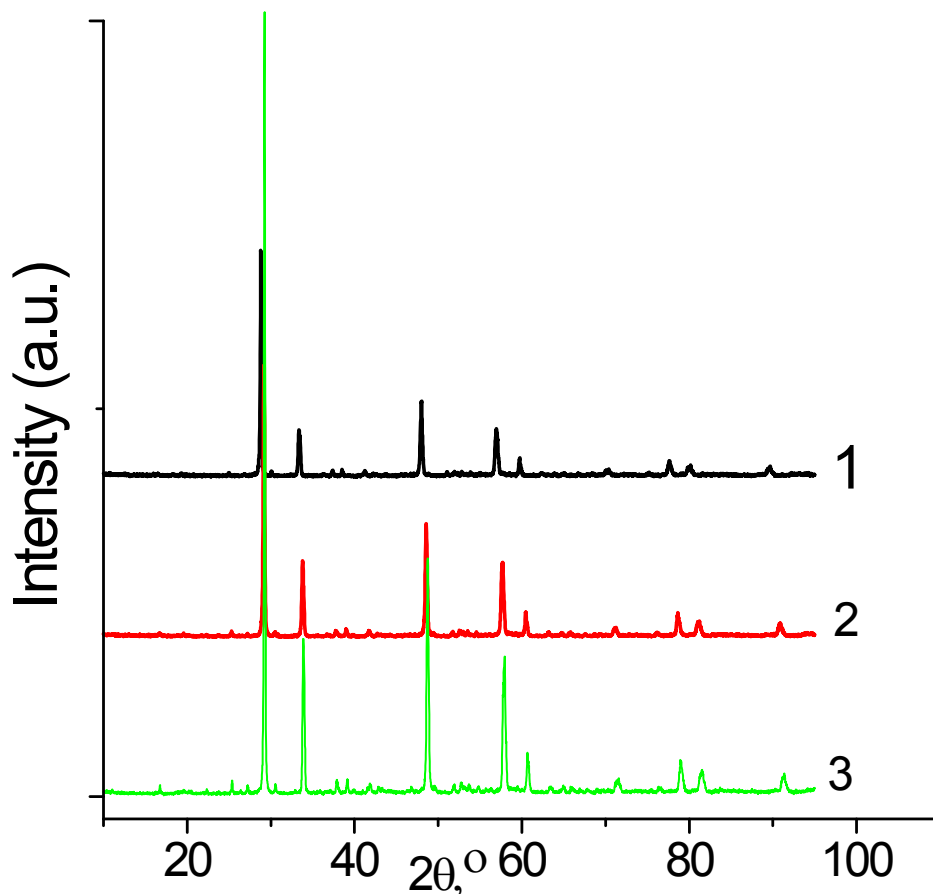
**Fig.S1.** XRD of high – temperature fluorites: (1)  $\text{Gd}_{10}\text{Mo}_2\text{O}_{21}$ , (2)  $\text{Dy}_{10}\text{Mo}_2\text{O}_{21}$ , (3)  $\text{Ho}_{10}\text{Mo}_2\text{O}_{21}$ .



**Fig.S2** XRD data illustrating the phase changes of  $\text{Gd}_{10}\text{Mo}_2\text{O}_{21}$ : (1) 1200°C, 40 h; (2) 1200°C, 80 h; (3) 1600°C, 3 h.



**Fig.S3** XRD data illustrating the phase changes of Dy<sub>10</sub>Mo<sub>2</sub>O<sub>21</sub>: (1) 1200°C, 40 h; (2) 1200°C, 80 h; (3) 1600°C, 3 h (asterisks mark reflections from Dy<sub>2</sub>O<sub>3</sub>).



**Fig.S4** XRD of low – temperature phases: (1)  $Gd_{10}Mo_2O_{21}$ , (2)  $Dy_{10}Mo_2O_{21}$ , (3)  $Ho_{10}Mo_2O_{21}$ .

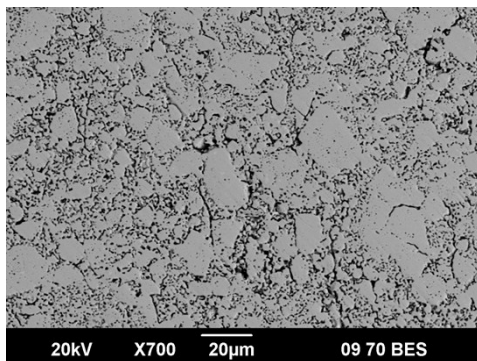
#### Supplement A.

The microstructure of the  $LnMoO$ -1200 ( $Ln = Gd, Dy, Ho$ ) ceramics produced at 1200°C and  $LnMoO$ -1600 ( $Ln = Gd, Dy, Ho$ ) ceramics produced at 1600 °C is illustrated in **Figs. S5 – S7**. **Figs. S5a, S6a** present cross-section of the ceramics of the orthorhombic  $LnMoO$ -1200 ( $Ln = Gd, Dy$ ) molybdates, obtained in single-phase form at 1200 °C. They are similar in microstructure. The orthorhombic  $LnMoO$ -1200 ( $Ln = Gd, Dy$ ) ceramics consists mainly of smaller grains, 0.2 to 0.5  $\mu m$  in size, but there are also grains  $\sim 35 \mu m$  in size (**Figs. S5a, S6a**). The orthorhombic  $HoMoO$ -1200 ceramics is more porous than the  $LnMoO$ -1200 ( $Ln = Gd, Dy$ ) ceramics and was investigated by SEM earlier [44].

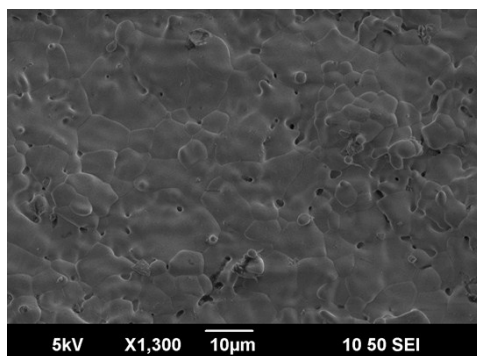
**Figs. S5b, S6b** show the fresh surface of the fluorite  $LnMoO$ -1600 ( $Ln = Gd, Dy$ ) ceramics, obtained in single-phase form at 1600 °C. SEM of ceramics cross-section of the high-

temperature fluorite modifications after polishing and thermal etching at 1450°C for 0.5 h are shown in **Figs. S5c, S6c**. The grain size in the gadolinium molybdate fluorite sample is 10–15  $\mu\text{m}$ . Note the unusual microstructure of the surface of the fluorite DyMoO-1600 ceramics (**Fig. S6b**). Some of its grains have the form of the square plates ( $10\times 10\times 4\ \mu\text{m}$ ), and the size of some grains cannot be determined (**Fig. S6b**). The fluorite HoMoO-1600 ceramics is more inhomogeneous than the gadolinium molybdate and dysprosium molybdate fluorite ceramics (**Fig. S7**). Its grains range in size up to 5  $\mu\text{m}$ , and some are even 25  $\mu\text{m}$  in size.

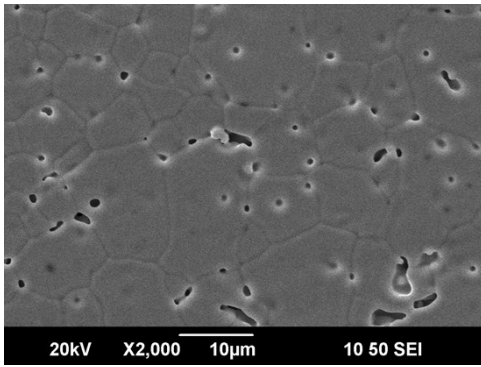
As a rule, in the series of rare-earth tungstates [40] and molybdates [31], the density decreases significantly with a decrease in the ionic radius of lanthanides following the same trend in the binary RE oxides.



**a**

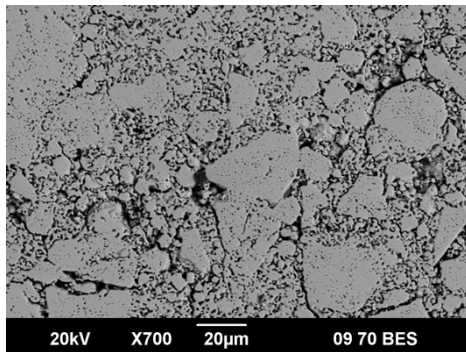


**b**

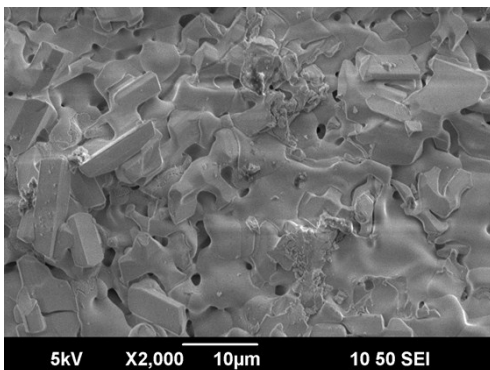


**c**

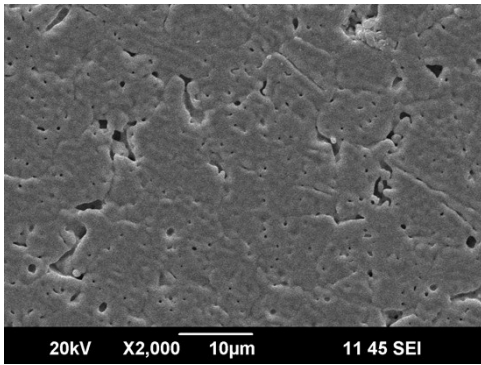
**Fig.S5** Microstructure of the gadolinium ceramics prepared by firing (a) at 1200 °C for 80 h, then polished and thermally etched at 1050°C (ceramics cross-section surface; back-scattered electron image), (b) at 1600 °C for 3 h, and (c) at 1600 °C for 3 h, then polished and thermally etched at 1450°C (ceramics cross-section surface).



**a**

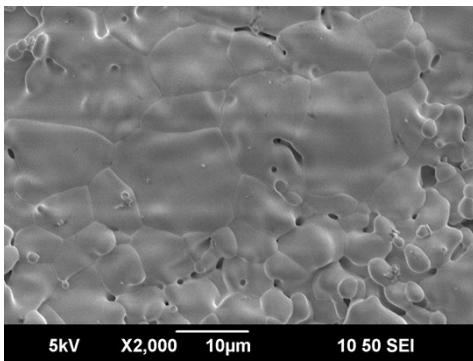


**b**



**c**

**Fig.S6** Microstructure of the dysprosium ceramics prepared by firing (a) at 1200 °C for 80 h, then polished and thermally etched at 1050°C (ceramics cross-section surface; back-scattered electron image), (b) at 1600 °C for 3 h, and (c) at 1600 °C for 3 h, then polished and thermally etched at 1450°C (ceramics cross-section surface).



**Fig.S7** Microstructure of the holmium ceramics prepared by firing at 1600 °C for 3 h.

**Table S1. Crystal structural parameters for  $\text{Ho}_{10}\text{Mo}_2\text{O}_{21}$  refined against NPD data collected at room temperature. Space group Pbcn (# 60). Numbers in parentheses are standard deviations of the last significant digit.  $a = 15.8751(6)$  Å,  $b = 10.5318(3)$  Å,  $c = 10.5672(4)$  Å,  $V = 1766.76(11)$  Å<sup>3</sup>, overall isotropic temperature factor  $U_{\text{iso}} = 0.0047(3)$  Å<sup>2</sup>**

Atom	Wyckoff site	x	y	z
Mo1	8d	0.5868(10)	0.2438(25)	0.1193(12)
Ho1	8d	0.0732(9)	-0.0047(12)	0.1365(13)
Ho2	8d	0.2720(6)	0.2437(20)	0.1322(13)
Ho3	8d	0.0798(11)	0.4934(12)	0.1210(16)
Ho4	8d	0.6001(7)	0.7495(20)	0.1447(11)
Ho5	8d	0.7456(15)	0.0067(18)	0.1410(13)
O1	8d	0.3408(11)	0.3857(20)	-0.0026(22)
O2	8d	0.5473(10)	0.3710(17)	0.0176(19)
O3	8d	0.3579(12)	0.3570(19)	0.2678(22)
O4	8d	0.3209(15)	0.6172(21)	0.2186(17)
O5	8d	0.5147(12)	0.1126(21)	0.0537(19)
O6	8d	0.3246(14)	0.1975(16)	0.4721(17)
O7	8d	0.6694(14)	0.9074(18)	0.0043(19)
O8	8d	0.1434(12)	0.6331(21)	0.2525(20)
O9	8d	0.1825(15)	0.3847(20)	0.2170(15)
O10	4c	0.5	0.3023(25)	0.25
O11	4c	0.5	0.6350(26)	0.25
O12	4c	0.0	0.3940	0.25