

## **Supporting Information**

# **Room-temperature conversion of Cu<sub>2-x</sub>Se to CuAgSe nanoparticles to enhance the photocatalytic performance of their composites with TiO<sub>2</sub>**

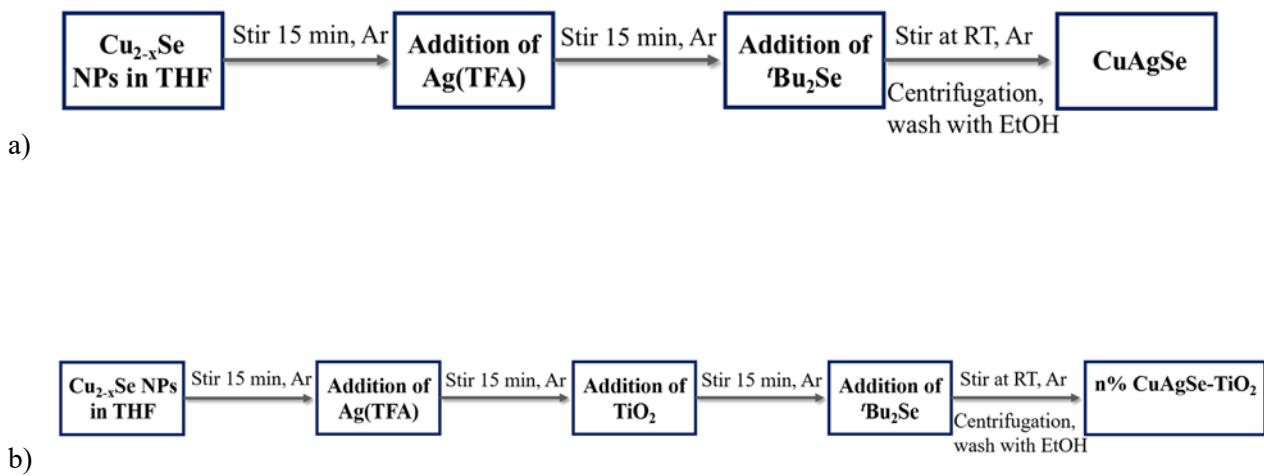
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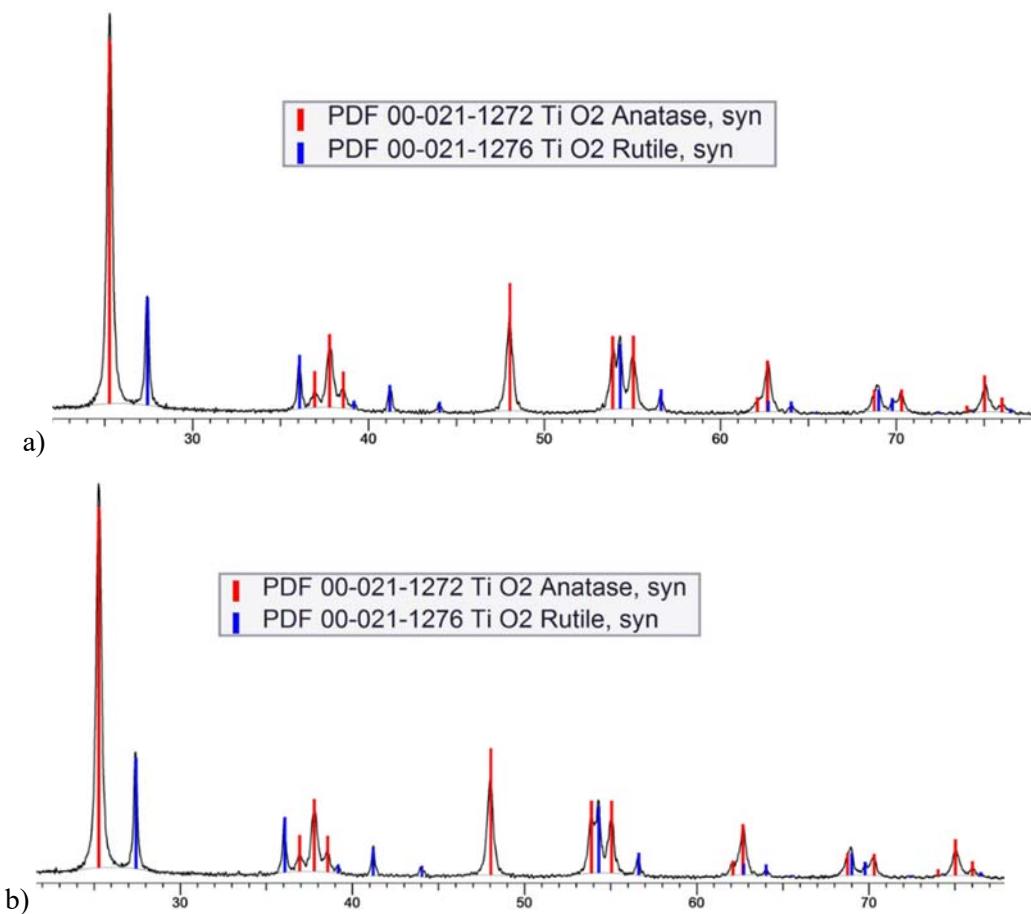
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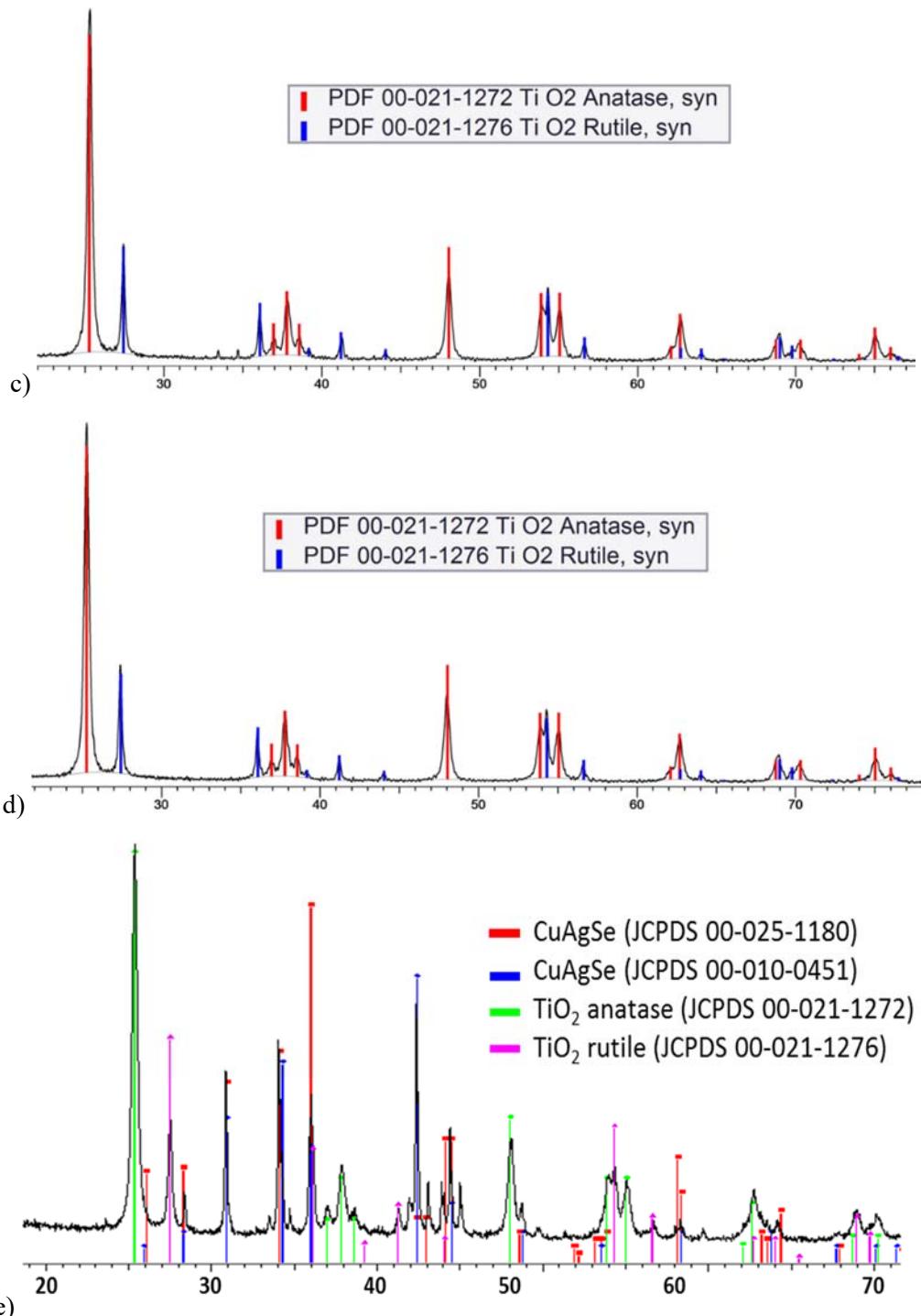
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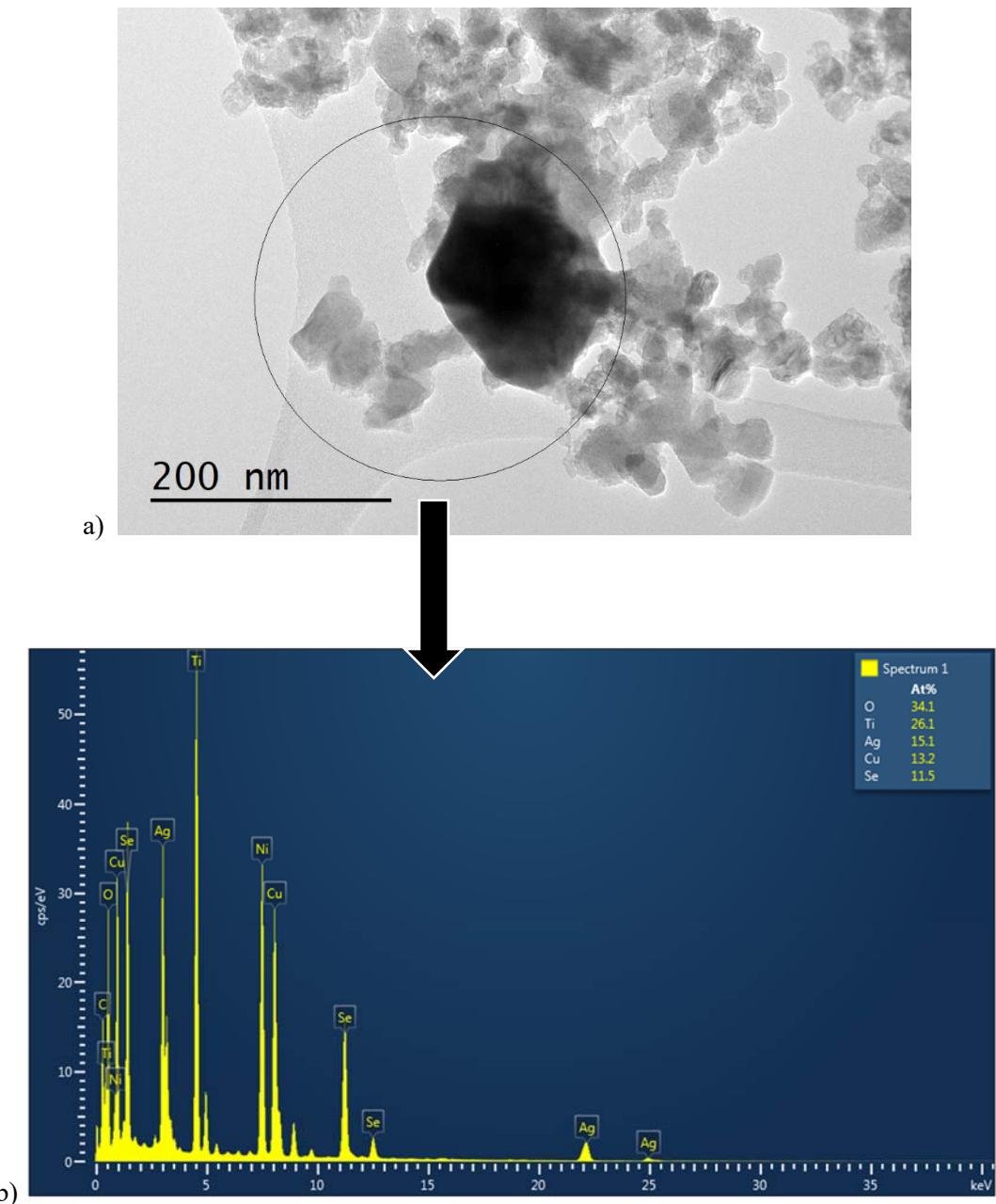


**Figure S1.** Schematic representation of the synthesis of the a) CuAgSe NPs and b)  $n\%$  CuAgSe-TiO<sub>2</sub> nanocomposites.

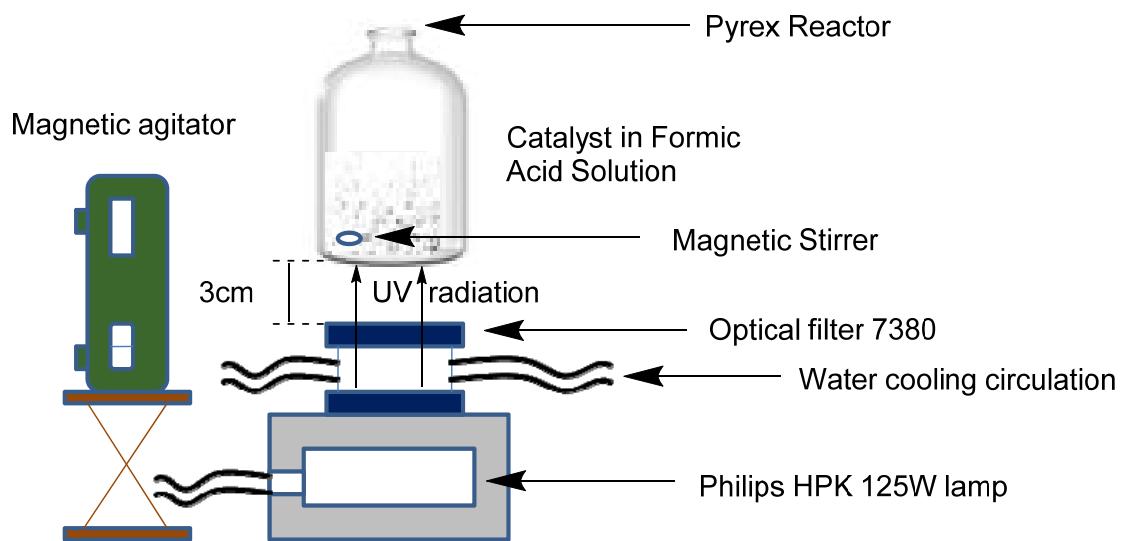




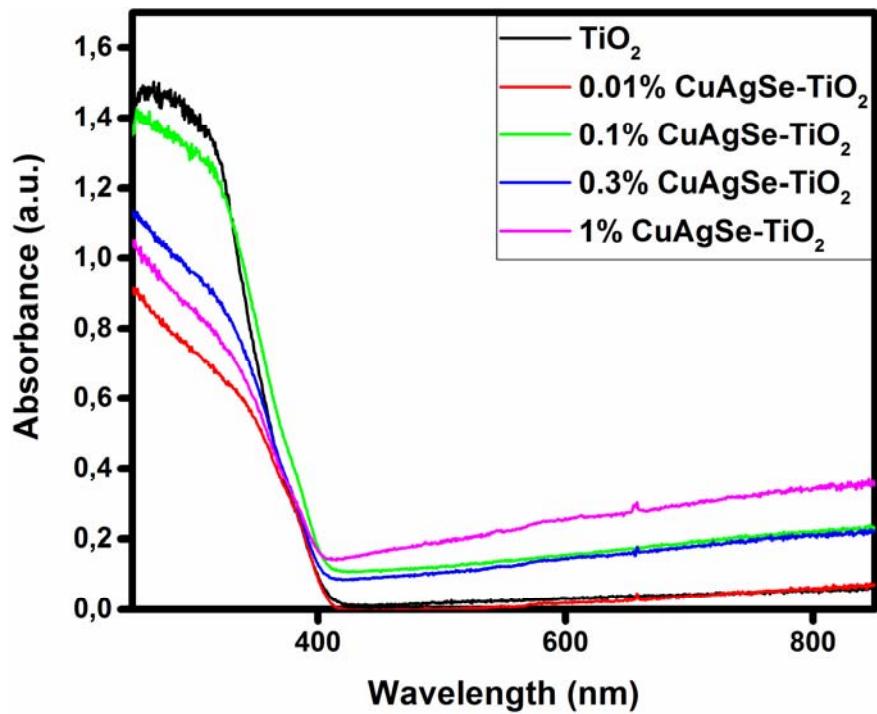
**Figure S2.** XRD patterns of n mol% CuAgSe-TiO<sub>2</sub> nanocomposites obtained from Cu<sub>2-x</sub>Se, Ag(TFA), 'Bu<sub>2</sub>Se and TiO<sub>2</sub> when stirred at room temperature for 6h. (a) 0.01% CuAgSe-TiO<sub>2</sub>, (b) 0.1% CuAgSe-TiO<sub>2</sub>, (c) 0.3% CuAgSe-TiO<sub>2</sub>, (d) 1% CuAgSe-TiO<sub>2</sub>, and (e) 10% CuAgSe-TiO<sub>2</sub>.



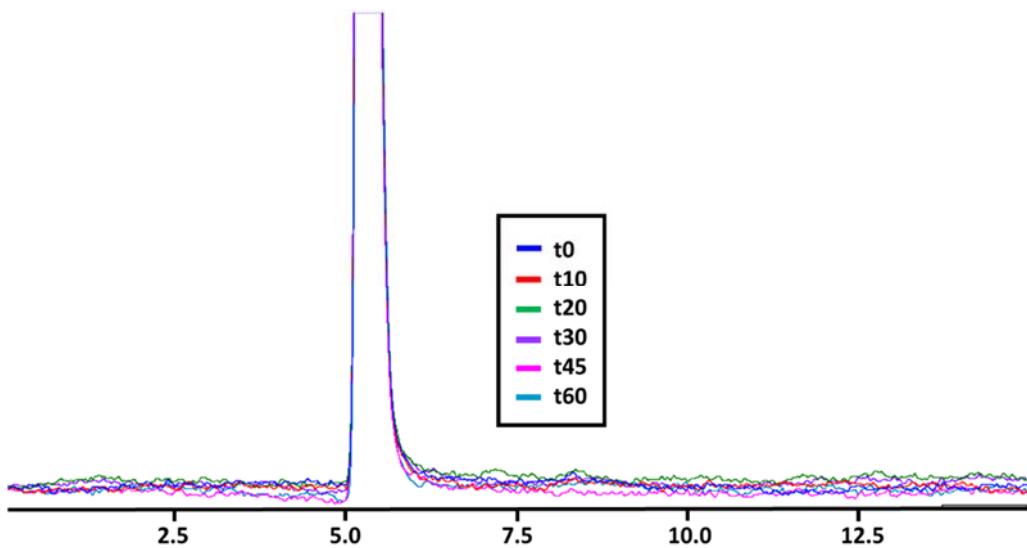
**Figure S3.** TEM image (a) and associated EDX analysis (b) of 10% CuAgSe-TiO<sub>2</sub> nanocomposites.



**Figure S4.** The experimental setup for FA photodegradation under UV light.



**Figure S5.** Absorbance spectra for n% CuAgSe-TiO<sub>2</sub> nanocomposites.



**Figure S6.** HPLC result after stability test using 0.3% CuAgSe-TiO<sub>2</sub> under acidic conditions (HNO<sub>3</sub>).

**Table S1.** N<sub>2</sub> adsorption and desorption isotherm and related data of TiO<sub>2</sub> and CuAgSe-TiO<sub>2</sub> nanocomposites.

Sample Name	Surface area (m <sup>2</sup> /g)	Total pore volume (cm <sup>3</sup> /g)	Av. Pore diameter (nm)
TiO <sub>2</sub>	58.9	0.23	15.6
0.01% CuAgSe-TiO <sub>2</sub>	53.7	0.52	38.9
0.1% CuAgSe-TiO <sub>2</sub>	52.4	0.49	37.5
0.3% CuAgSe-TiO <sub>2</sub>	51.2	0.41	32
1% CuAgSe-TiO <sub>2</sub>	54.9	0.24	17.6