

Supporting Information

The solvent and zinc source dual-induced synthesis of a two dimensional zeolitic imidazolate framework with farfalle-shape and its crystal transformation to zeolitic imidazolate framework-8

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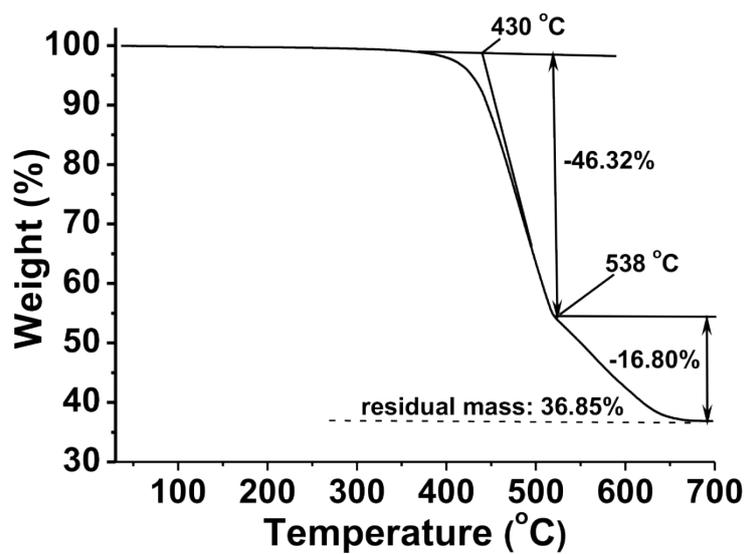


Fig. S1 TG analysis of ZIF-F after activation at 120 °C for 12 h.

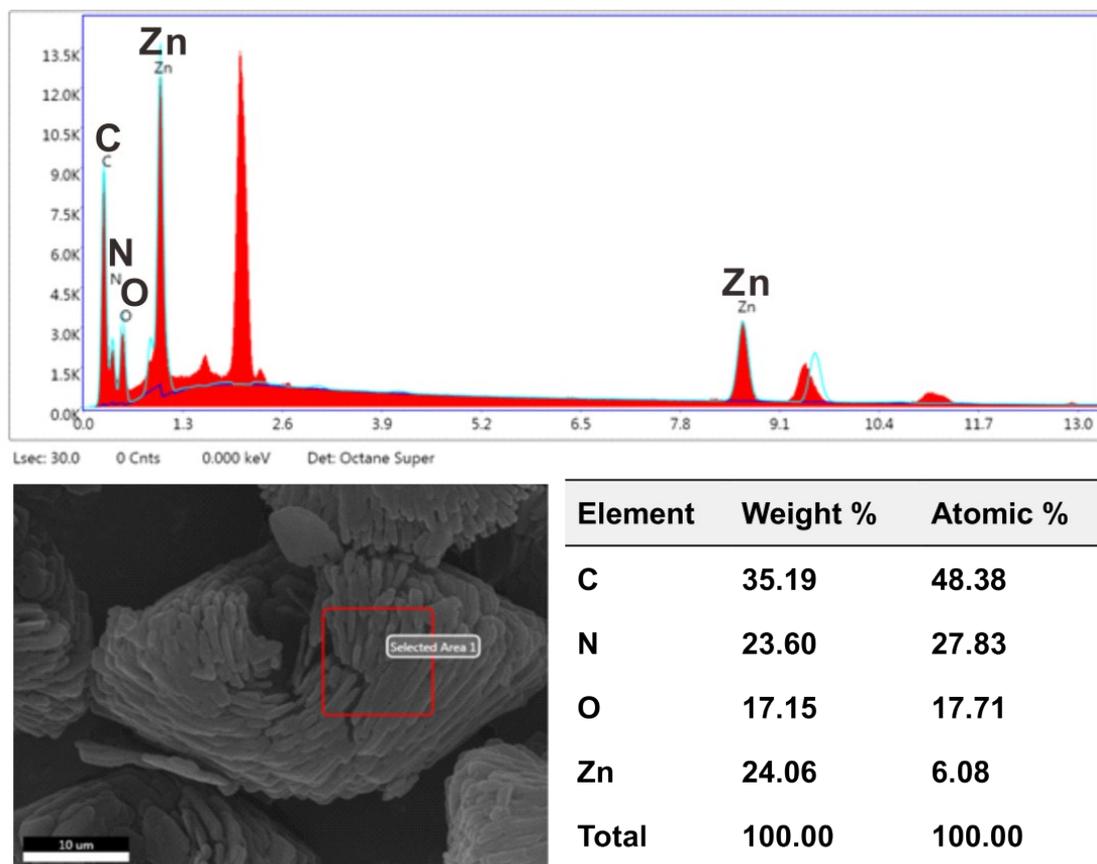


Fig. S2 EDX analysis of ZIF-F.

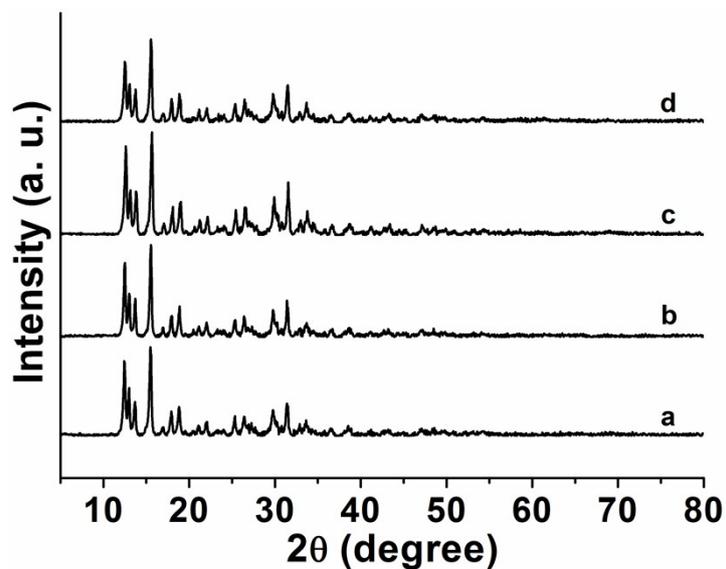


Fig. S3 XRD patterns of re-produced ZIF-F samples: (a, b, c) three repetitions with the normal system, (d) A fivefold expansion of the synthesis system.

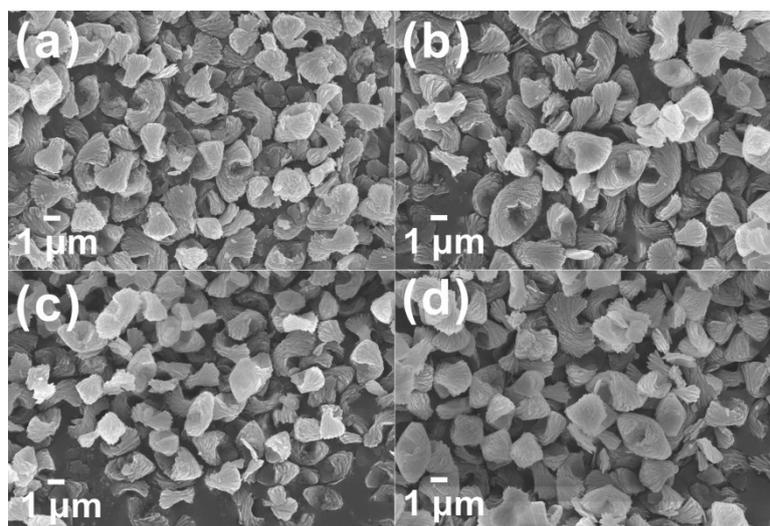


Fig. S4 SEM images of re-produced ZIF-F samples: (a, b, c) three repetitions with the normal system, (d) A fivefold expansion of the synthesis system.



Fig. S5 Photo pictures and SEM images of ZIF-F before and after thermal treatments under different temperatures for 12 h: (a, f) before heating, (b, g) 200 °C, (c, h) 300 °C,

(d, i) 350 °C, (e, j) 400 °C.

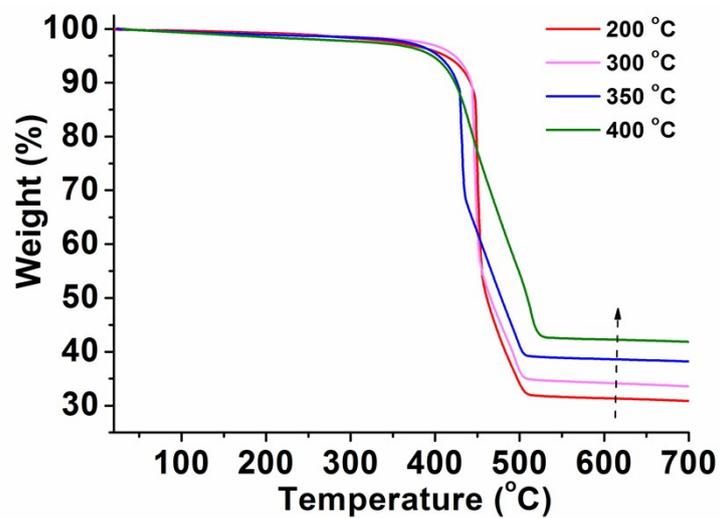


Fig. S6 TG analysis of the calcined samples after 200, 300, 350, 400 °C for 12 h , respectively.

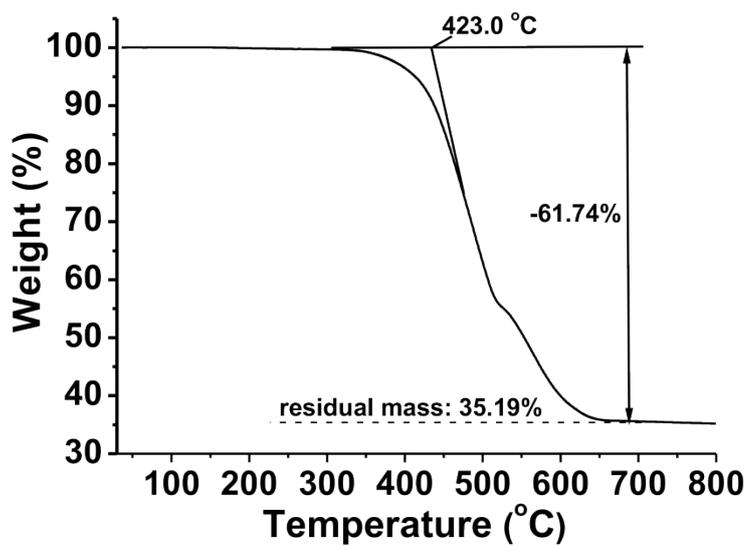


Fig. S7 TG analysis of the prepared ZIF-8.

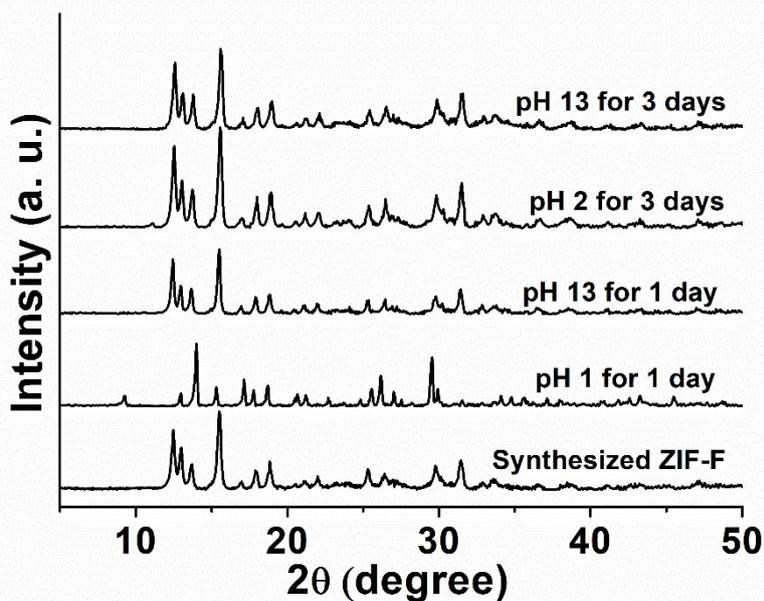


Fig. S8 XRD patterns of ZIF-F after immersion in acid and basic solutions.

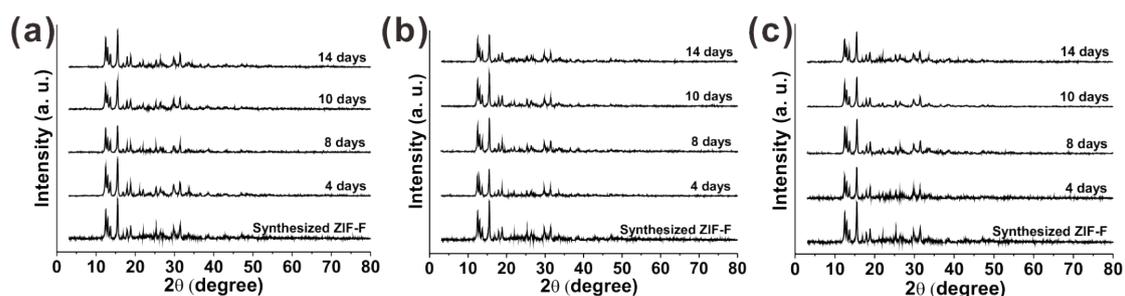


Fig. S9 XRD patterns of ZIF-F after immersion in organic solvents: (a) DMF, (b) DCM, (c) toluene.

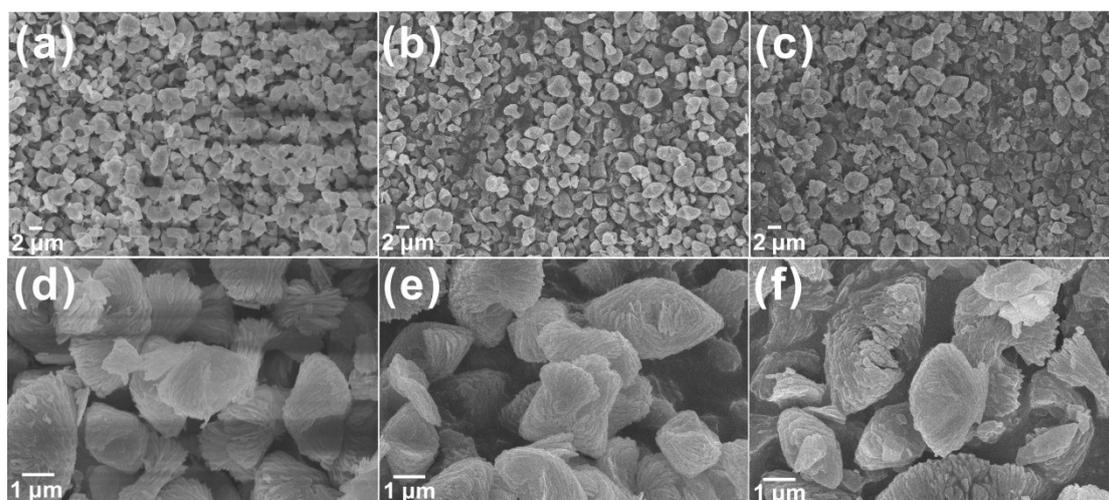


Fig. S10 SEM images of ZIF-F after immersion in organic solvents for 14 days: (a, d) DMF, (b, e) DCM, (c, f) toluene.

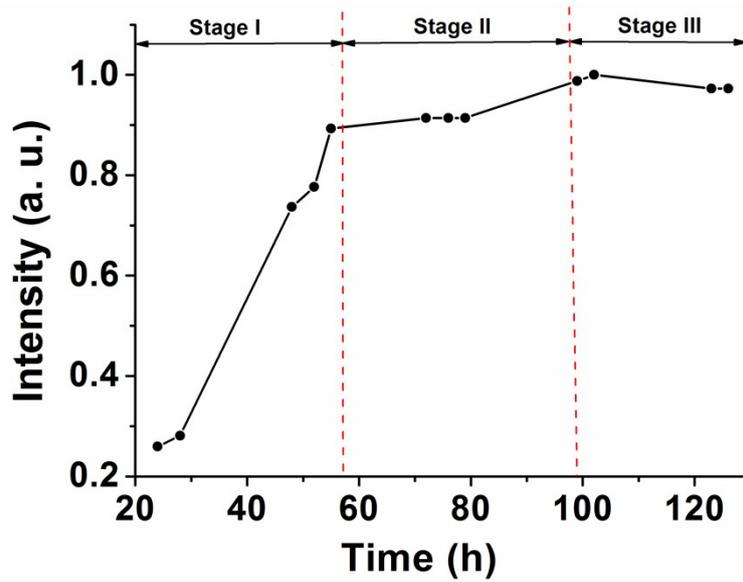


Fig. S11 Curve of extent of crystallization (α) as a function of time (t) obtained from time-resolved in situ ^1H NMR.

Avrami's Analysis

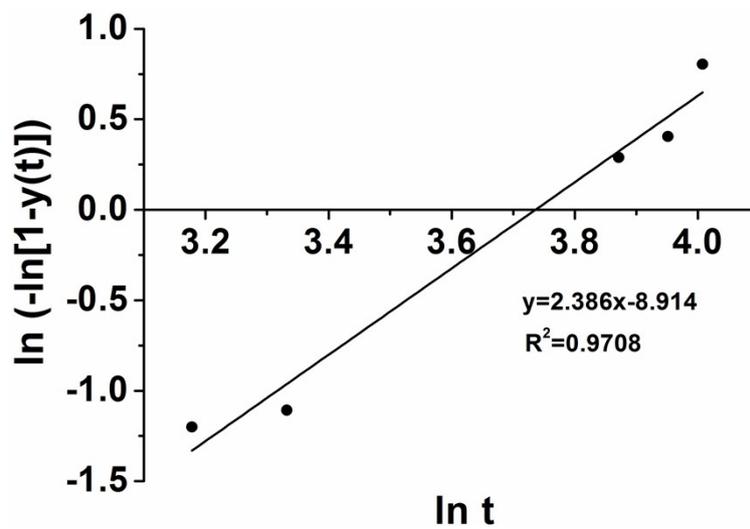


Fig. S12 The linear regression analysis of Avrami's classical model.

Avrami's Equation

$$1 - y(t) = e^{-kt^n} \quad (\text{S1})$$

Linear form of Avrami's Equation

$$\ln(-\ln[1 - y(t)]) = \ln k + n \ln t \quad (\text{S2})$$

Where t is the synthesis time, k is the scaling constant, n is Avrami's constant and $y(t)$ is the ZIF-8 relative crystallinity as a function of time.

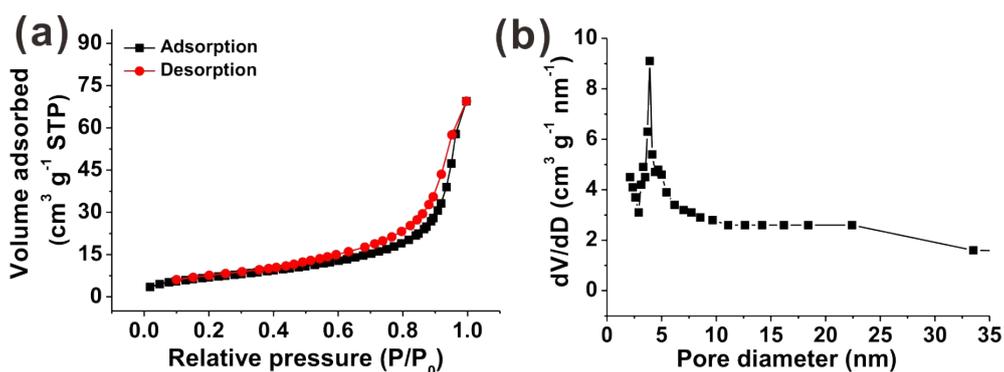


Fig. S13 (a) Nitrogen adsorption-desorption isotherm of ZIF-F at 77.3 K, (b) Pore diameter distribution of ZIF-F obtained by BJH method.

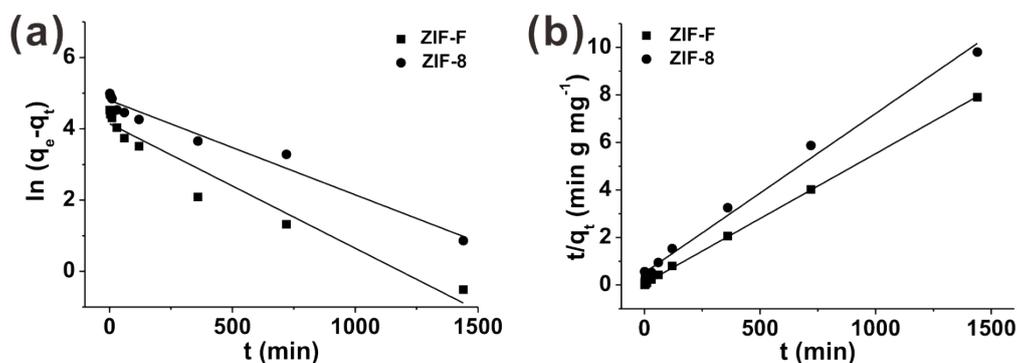


Fig. S14 The pseudo-first-order kinetic models (a) and pseudo-second-order kinetic models (b) of ZIF-F and ZIF-8.

Table S1 Kinetic parameters of ZIF-F and ZIF-8 for the CR adsorption

	pseudo-second-order kinetic model			pseudo-first-order kinetic model	
	$q_{e(cal)}$ (mg g ⁻¹)	k_2 (g mg ⁻¹ min ⁻¹)	R^2	k_1 (min ⁻¹)	R^2
ZIF-F	182.82	4.93×10^{-4}	0.9996	3.49×10^{-3}	0.9425
ZIF-8	149.25	8.69×10^{-5}	0.9908	2.65×10^{-3}	0.9687