

Supporting information

Crystal Growth of Novel 3D Skeleton Uranyl Germanium Complexes: Influence of Synthetic Conditions to Crystal Structures

Haijian Li^{†,‡,*}, Philip Kegler[‡], Evgeny V. Alekseev^{‡,§,*}

[†]Science and Technology on Combustion and Explosion Laboratory, Xi'an Modern Chemistry Research Institute, Xi'an 710065, China

[‡]Institute of Energy and Climate Research (IEK-6), Forschungszentrum Jülich GmbH, 52428 Jülich, Germany

[§]Institut für Kristallographie, RWTH Aachen University, 52066 Aachen, Germany

*contact E-Mail: h.j.li@outlook.com, e.alekseev@fz-juelich.de

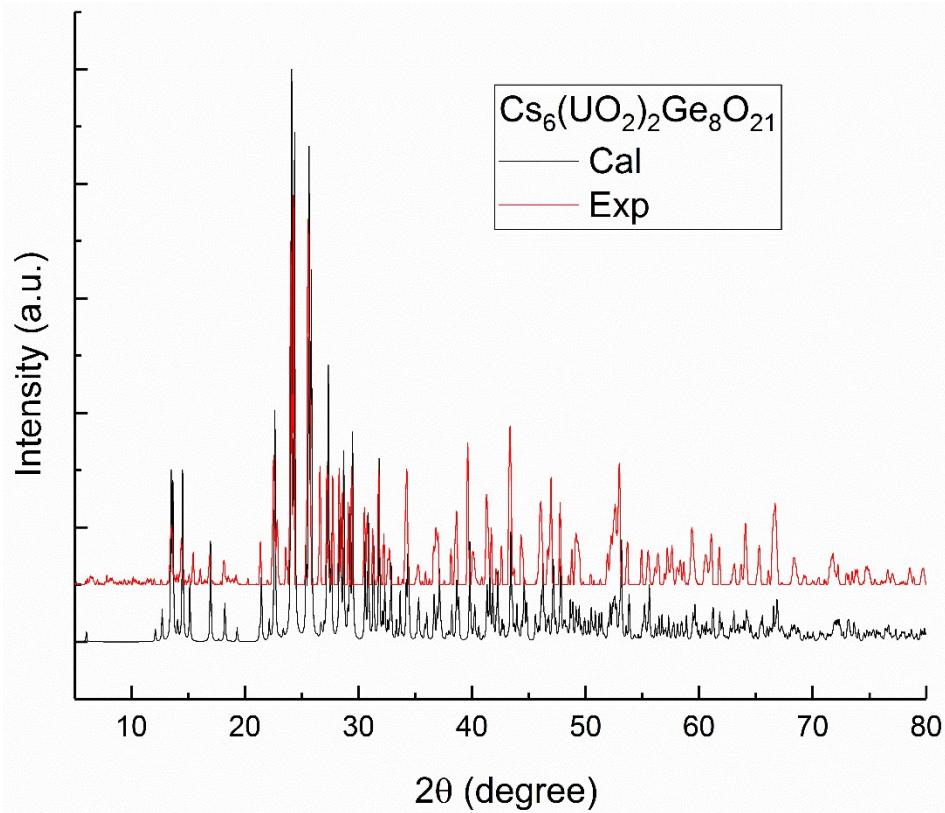


Figure S1. The theoretical (black line) and experimental (red line) powder diffraction patterns of $\text{Cs}_6(\text{UO}_2)_2\text{Ge}_8\text{O}_{21}$.

The theoretical diffraction patterns were calculated using *Mercury* software (version 3.8).¹



Figure S2(a). EDS analysis for $\text{K}_8\text{BrF}(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2$.

Table S1 (a). Atom ratio of $\text{K}_8\text{BrF}(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2$. (U is keep as 3)

	U	Ge	K	Br	F
Point1	3	4.29	7.71	1.03	0.56
Point2	3	4.13	6.20	1.16	0.82
Average	3	4.21	7.00	1.10	0.69

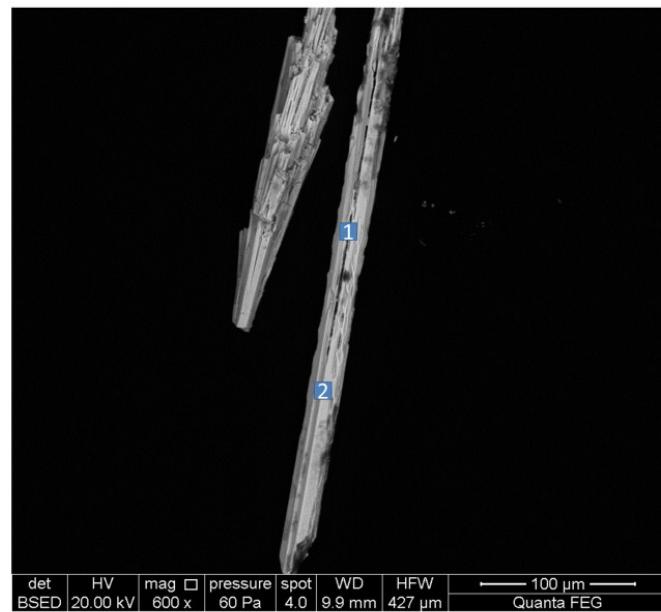


Figure S2(b). EDS analysis for $\text{Rb}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 0.5\text{H}_2\text{O}$.

Table S1 (b). Atom ratio of $\text{Rb}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 0.5\text{H}_2\text{O}$. (U is keep as 3)

	U	Ge	Rb
Point1	3	4.43	8.56
Point2	3	4.67	6.16
Average	3	4.55	7.36

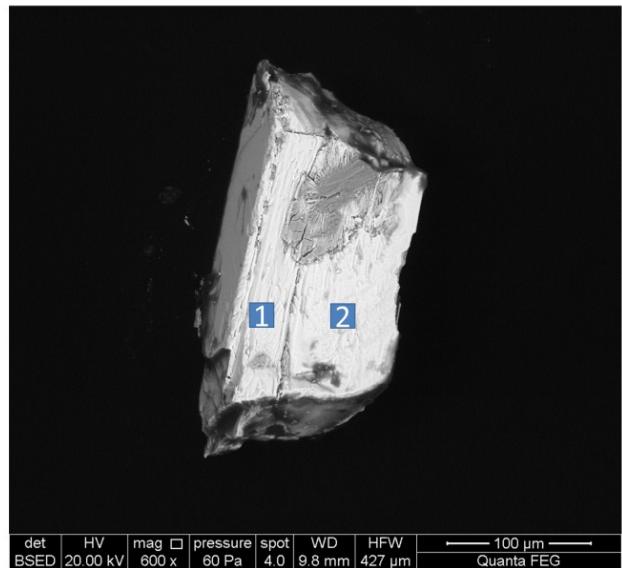


Figure S2(c). EDS analysis for $\text{Cs}_6(\text{UO}_2)_2\text{Ge}_8\text{O}_{21}$.

Table S1(c). Atom ratio of $\text{Cs}_6(\text{UO}_2)_2\text{Ge}_8\text{O}_{21}$. (U is keep as 2)

	U	Ge	Cs
Point1	2	9.42	6.92
Point2	2	8.72	6.30
Average	2	9.07	6.61

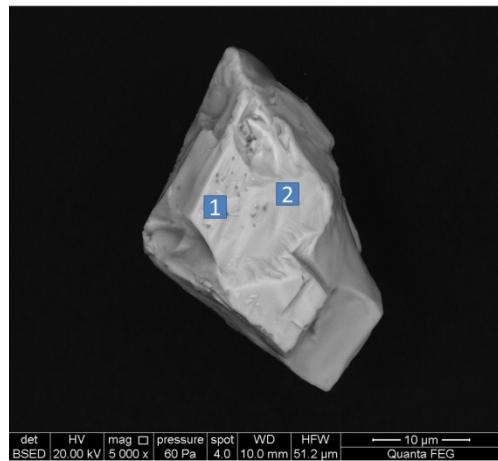


Figure S2(d). EDS analysis for $\text{Rb}_2(\text{UO}_2)_3(\text{GeO}_4)_2$.

Table S1(d). Atom ratio of $\text{Rb}_2(\text{UO}_2)_3(\text{GeO}_4)_2$. (U is keep as 3)

	U	Ge	Rb
Point1	3	2.28	2.45
Point2	3	2.28	2.35
Average	3	2.28	2.40

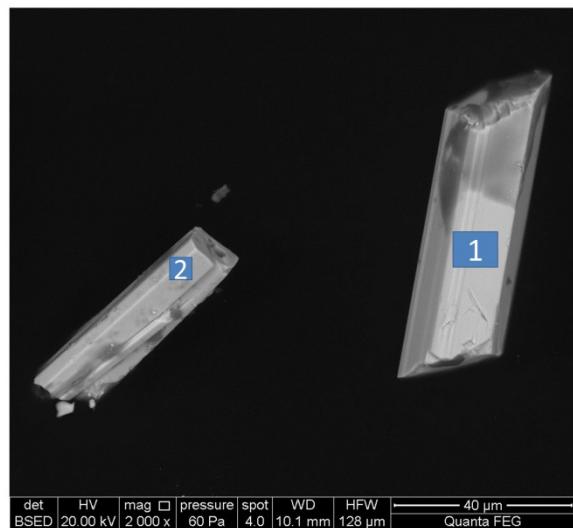


Figure S2(e). EDS analysis for $\text{Cs}_2(\text{UO}_2)_3(\text{GeO}_4)_2$.

Table S1(e). Atom ratio of $\text{Cs}_2(\text{UO}_2)_3(\text{GeO}_4)_2$. (U is keep as 3)

	U	Ge	Cs
Point1	3	2.44	2.24
Point2	3	2.87	2.29
Average	3	2.66	2.27

Table S2. Results of Bond valence calculation for $K_8BrF(UO_2)_3(Ge_2O_7)_2$, $Rb_6(UO_2)_3(Ge_2O_7)_2 \cdot 0.5H_2O$, $Cs_6(UO_2)_2Ge_8O_{21}$, and $A^+2(UO_2)_3(GeO_4)_2$ ($A^+ = Rb^+, Cs^+$), respectively. Note that for an atom, the corresponding valence V is obtained through the following formula:

$$V = \sum_i \exp[-[(R_i - d_i)/b]]$$

Here the bond valence parameter R_i and constant b of hexavalent U are taken from Burn.² The bond valence parameter R_i and constant b of other elements are provided by Brese and O'Keeffe.^{3,4}

(a) BVS calculation for $K_8BrF(UO_2)_3(Ge_2O_7)_2$.

	U(1)	U(2)	U(3)	Ge(1)	Ge(2)	K(1)	K(2)	K(3)	K(4)	Σ
O(1)	$2.2294^{2\downarrow}$					2.7261		2.942		2.039
O(2)		$2.2141^{2\downarrow}$			1.8070	2.6787	2.9075			2.089
O(3)		$1.8072^{2\downarrow}$				2.6929	2.9909			2.011
O(4)				1.7210	1.7376		2.8103 3.3141			2.128
O(5)			$2.2299^{2\downarrow}$	1.7051	1.6595			3.1104	3.0509	1.966
O(6)	$2.2671^{2\downarrow}$							3.0127 3.0135	2.9647	1.930
O(7)	$1.8389^{2\downarrow}$					2.7403			2.5753	1.986
O(8)			$1.8171^{2\downarrow}$		1.7109		2.7401	2.8104	3.0206	2.001
O(9)			$2.2608^{2\downarrow}$				3.3881		3.0170	2.044
O(10)		$2.21^{2\downarrow}$		1.7337 1.7342			3.1614		2.8043	1.980
Br(1)						3.1254 3.2347	3.0746 3.4651	3.2255 3.4430		1.039
F(1)						2.5579 2.5669	2.9525	3.2909	2.8581	0.625
Σ	5.651	6.039	5.797	4.276	4.256	1.759	1.221	0.892	0.926	

(b) BVS calculation for $\text{Rb}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 0.5\text{H}_2\text{O}$.

	U(1)	U(2)	U(3)	Ge(1)	Ge(2)	Rb(1)	Rb(2)	Rb(3)	Rb(4)	Σ
O(1)	$2.2331^{\times 2\downarrow}$			1.7071			3.1296		2.9922	2.038
O(2)				1.7576	1.7579	2.9729 3.4736	3.3052			2.193
O(3)		$2.2402^{\times 2\downarrow}$		1.7242			3.0554	3.1178		1.959
O(4)		$2.2334^{\times 2\downarrow}$			1.7373	3.0381			2.941	1.998
O(5)		$2.2471^{\times 2\downarrow}$			1.6742	3.2207		3.0893		2.070
O(6)		$1.8058^{\times 2\downarrow}$				2.9086	2.9473	3.0936		2.033
O(7)		$2.1967^{\times 2\downarrow}$		1.7468		3.2567		2.8746		1.999
O(8)	$2.2502^{\times 2\downarrow}$				1.7297		3.0570 3.1024			1.934
O(9)	$1.7953^{\times 2\downarrow}$						3.6024	2.7782	2.7422	2.177
O(10)		$1.7967^{\times 2\downarrow}$				2.9515		2.9945	2.7326	2.199
OW1						3.1866	3.2841			0.313
Σ	5.953	6.091	5.875	4.161	4.274	0.864	0.741	0.891	0.854	

(c) BVS calculation for $\text{Cs}_6(\text{UO}_2)_2\text{Ge}_8\text{O}_{21}$.

	U(1)	U(2)	Ge(1)	Ge(2)	Ge(3)	Ge(4)	Cs(1)	Cs(2)	Cs(3)	Cs(4)	Σ
O(1)			2.0108		2.2023	2.1940	3.4874	3.2398			1.248
O(2)		$2.2629^{\times 2\downarrow}$		1.7196			3.1425 3.1684		2.9257 3.2772		2.349
O(3)				1.7590	1.7661		3.3961	3.5577		3.1337	2.184
O(4)				1.7522		1.7662	3.0610	3.3311		3.4540	2.261
O(5)	1.8243 $^{\times 2\downarrow}$						3.2311	3.1485 3.6696			1.821
O(6)			1.8134	1.7370			3.4495	3.0971			2.089
O(7)	2.2169 $^{\times 2\downarrow}$				1.6916		3.1818	3.2929			2.093
O(8)			1.7596		1.7668		3.0804	3.6288			2.124
O(9)	2.2278 $^{\times 2\downarrow}$					1.7159	3.3469	3.1665 3.5602			2.041
O(10)		1.8239 $^{\times 2\downarrow}$				3.1333	3.1287		3.1377 3.6933		2.004
O(11)		2.2327 $^{\times 2\downarrow}$	1.7365			3.2407 3.4139			3.1746		2.022
O(12)			1.7493			1.7122	3.0627	3.7105			2.303
O(13)					1.7944	1.7944	3.4031	3.1174			1.985
Σ	5.876	5.742	3.8352	4.0694	3.949	4.026	0.894	1.175	0.821	0.859	

(d) BVS calculation for $\text{Rb}_2(\text{UO}_2)_3(\text{GeO}_4)_2$.

	U(1)	U(2)	U(3)	Ge(1)	Rb(1)	Σ
O(1)			1.7918		2.7862	2.126
O(2)	2.4659		2.5942	2.0275		1.708
O(3)	$2.4045^{\times 2\downarrow}$	$2.3348^{\times 2\downarrow}$		1.7629	3.0859	2.244
O(4)			1.7973		2.8864	1.993
O(5)	1.8149				2.8946	1.929
O(6)			2.3159 2.3160 $2.3212^{\times 2\downarrow}$	1.7612	3.4165	2.167
O(7)		1.8107			3.0977 3.5654	1.848
O(8)		2.3508		1.9024		1.861
O(9)	$2.2762^{\times 2\downarrow}$	$2.3650^{\times 2\downarrow}$		1.7699		2.100
O(10)	1.8040				2.9415	1.920
O(11)		1.7993			2.9673	1.914
Σ	5.836	5.916	5.913	3.997	1.206	

(e) BVS calculation for $\text{Cs}_2(\text{UO}_2)_3(\text{GeO}_4)_2$.

	U(1)	U(2)	U(3)	Ge(1)	Cs(1)	Σ
O(1)			1.7970		2.9001	2.165
O(2)	2.4735		2.5665	2.0410		1.687
O(3)	$2.4001 \times 2\downarrow$	2.3349 2.3350		1.7679	3.2875	2.096
O(4)			1.8049		2.9929	2.020
O(5)	1.8130				3.0611	1.983
O(6)			$2.3169 \times 2\downarrow$ $2.3294 \times 2\downarrow$	1.7600	3.6579	2.139
O(7)		1.8122			3.2845 3.3933	1.909
O(8)		2.3655		1.8941		1.875
O(9)	$2.2835 \times 2\downarrow$	$2.3812 \times 2\downarrow$		1.7716		2.070
O(10)	1.7996				3.0313	1.995
O(11)		1.7928			3.0646	1.984
Σ	5.840	5.883	5.847	3.981	1.318	

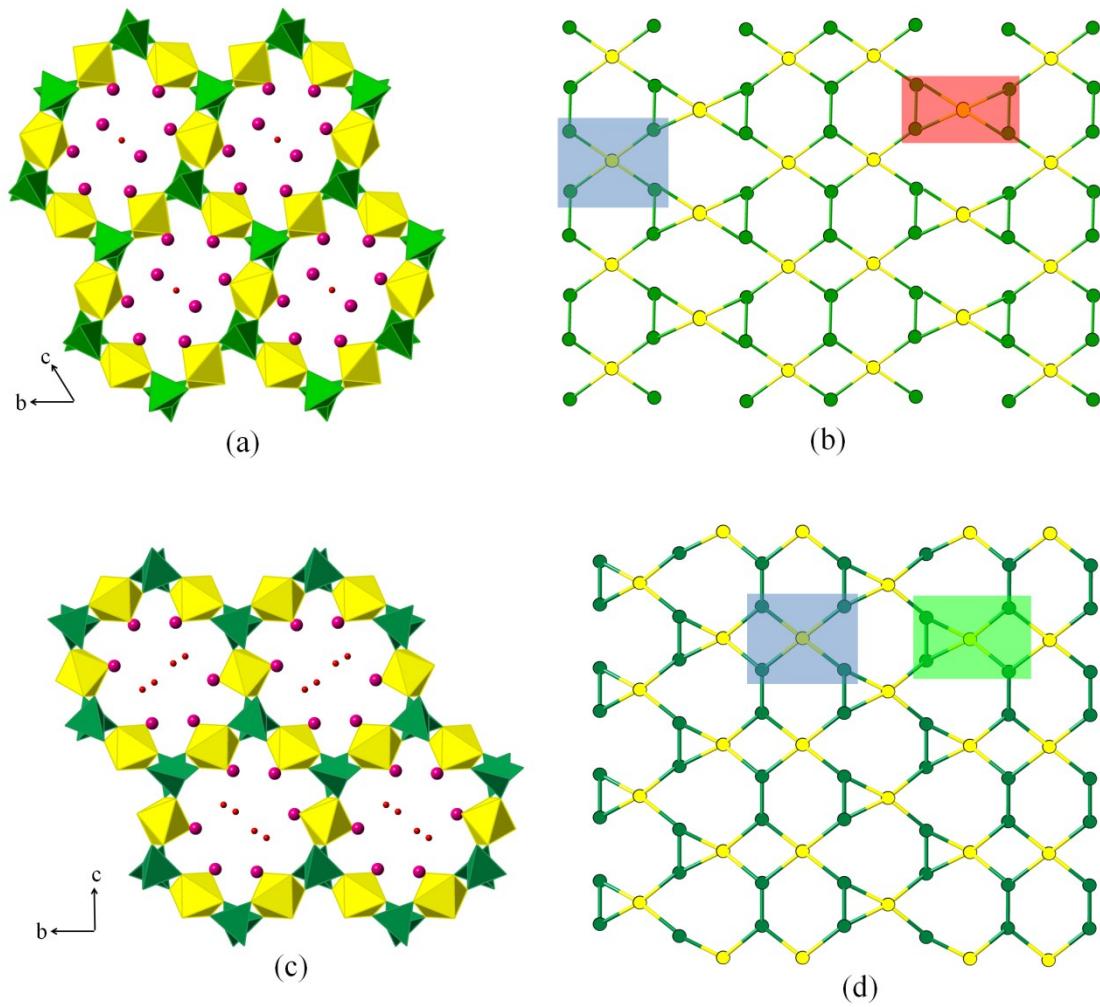


Figure S3. (a) Structure of $\text{Rb}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 0.5\text{H}_2\text{O}$ projected along a axis. (b) The unfolding wall of 12-membered channels constructed by $[\text{UGe}_4]$ pentamers (**A** type) and (**A2** type) in the structure of $\text{Rb}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 0.5\text{H}_2\text{O}$. (c) Structure of $\text{Cs}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 4\text{H}_2\text{O}^5$ projected along a axis. (d) The unfolding wall of 12-membered channels consisting of $[\text{UGe}_4]$ pentamers (**A** type) and (**A1** type) in the structure of $\text{Cs}_6(\text{UO}_2)_3(\text{Ge}_2\text{O}_7)_2 \cdot 4\text{H}_2\text{O}$.

Table S3 U=O bond lengths in uranyl $(\text{UO}_2)^{2+}$ ions for $A^+2(\text{UO}_2)_3(\text{GeO}_4)_2$ ($A^+ = \text{Rb, Cs}$) compounds and corresponding to the stretching frequencies of ν_1 and ν_3 vibrational modes.

Compounds	U=O bonds	Bond lengths (Å)	Calculated (cm ⁻¹)				Experimental (cm ⁻¹)	
			ν_1	ν_3	Average of ν_1	Average of ν_3	ν_1	ν_3
$\text{Cs}_2(\text{UO}_2)_3(\text{GeO}_4)_2$	U(1)=O(10)	1.798(5)	812	881	809	878	721, 753, 781, 813, 0, 909,	830, 836, 851. 87
	U(1)=O(5)	1.815(5)	796	860				
	U(2)=O(11)	1.792(5)	818	890				
	U(2)=O(7)	1.810(5)	800	865				
	U(3)=O(1)	1.793(5)	818	889				
	U(3)=O(4)	1.799(5)	811	880				
$\text{Rb}_2(\text{UO}_2)_3(\text{GeO}_4)_2$	U(1)=O(10)	1.804(5)	806	873	807	874	729, 757, 783, 818,	837, 854, 867, 908,
	U(1)=O(5)	1.816(5)	795	858				
	U(2)=O(11)	1.802(5)	808	876				
	U(2)=O(7)	1.812(5)	799	863				
	U(3)=O(1)	1.789(6)	821	893				
	U(3)=O(4)	1.798(6)	812	881				

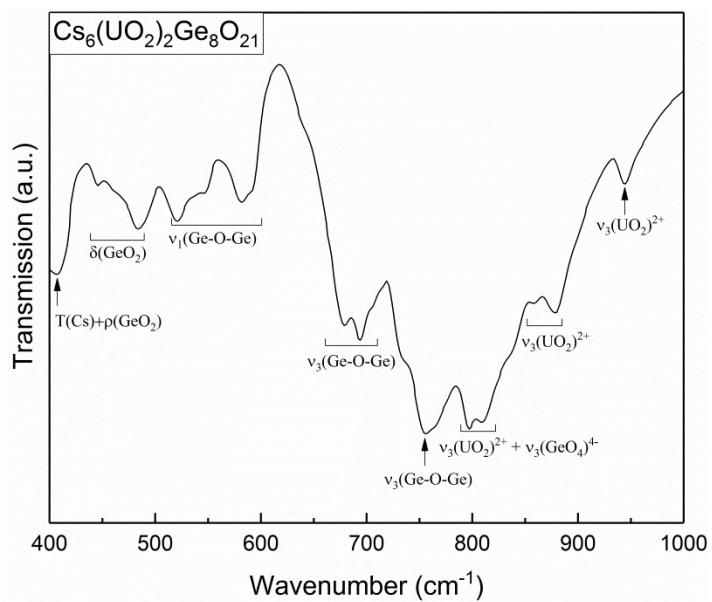


Figure S4. Infrared spectra of $\text{Cs}_6(\text{UO}_2)_2\text{Ge}_8\text{O}_{21}$ in the $400 - 1000 \text{ cm}^{-1}$ region.

References

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