# A Capped Trigonal Prismatic Cobalt(II) Complex as a Structure 

## Archetype for Single-Ion Magnets

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## Supporting Information

## Experimental Section

## Materials and General Characterizations

All manipulations were carried out using standard Schlenk techniques under a nitrogen atmosphere. All chemicals were purchased from commercial sources and used without further purification. The ligand [2,6-bis[bis(2pyridylmethyl)amino]methyl]pyridine (BPA-TPA) was prepared as described previously. ${ }^{\text {S1 }}$ Elemental analyses were performed on an Elementar Vario ELIII elemental analyzer. The powder XRD patterns were recorded at room temperature on a Bruker D8 Advance X-ray diffractometer.

## Synthesis of $\left[\mathbf{C o}{ }^{\text {II }}(\mathbf{B P A}-\mathrm{TPA}]\left(\mathrm{BF}_{4}\right)_{2}\right.$ (1-Co)

To the solution of $\mathrm{CoCl}_{2}(1.0 \mathrm{mmol}, 0.130 \mathrm{~g})$ in methanol ( 5 mL ) was added the solution of $\mathrm{AgBF}_{4}(2.0 \mathrm{mmol}, 0.390 \mathrm{~g})$ in methanol $(10 \mathrm{~mL})$. The insoluble silver chloride was immediately generated and then removed by filtration. The BPA-TPA ligand was added to the filtrate, and diethyl ether $(20 \mathrm{~mL})$ was allowed to slowly diffuse into the solution. After 2-4 days red crystals were obtained in 56 \% yield. Anal. Calc. for $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{~B}_{2} \mathrm{CoF}_{8} \mathrm{~N}_{7}$ : C, 50.72; H, 4.26; N, 13.36. Found: C, 50.65 ; H, 4.20; N, 13.42.

## Synthesis of $\left[\mathrm{Fe}^{\mathrm{II}}(\mathbf{B P A}-\mathrm{TPA}]\left(\mathrm{ClO}_{4}\right)_{2}(2-\mathrm{Fe})\right.$

Preparation of 2-Fe was performed according to the method of Hazell et al. ${ }^{\mathrm{S} 2}$ $\mathrm{Fe}\left(\mathrm{ClO}_{4}\right)_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$ was added to the solution of BPA-TPA ligand $(0.5 \mathrm{mmol}, 0.25 \mathrm{~g})$ in methanol ( 10 mL ), and the solution was stirred under dinitrogen for 10 min . The solution was allowed to stand in a closed vessel and pink crystals of formed overnight. The yield is $68 \%$. Anal. Calc. for $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{Cl}_{2} \mathrm{FeN}_{7} \mathrm{O}_{8}$ : C, 49.23; H, 4.13; N, 12.96. Found: C, 49.29; H, 4.06; N, 13.04.

## X-ray Structure Determination

Single crystal X-ray diffraction data for both complexes were collected using a Bruker APEX DUO diffractometer with a CCD area detector (Mo Ka radiation, $\lambda=$ $0.71073 \AA$ ) at $296 \mathrm{~K} .{ }^{\text {S3 }}$ The APEXII program was used for collecting frames of data, determining lattice parameters. Data were integrated through the SAINT. Absorption
corrections were applied using SADABS. ${ }^{\mathrm{S4}}$ The structures were solved using SHELXS-97 and subsequently completed by Fourier recycling using SHELXL 97 program. ${ }^{55}$

## Magnetic Measurements

Magnetic measurements of both complexes were performed with a Quantum Design SQUID VSM magnetometer. Direct current susceptibility data were collected between 1.8 and 300 K using an applied field of 2.0 kOe . Magnetization data were recorded from 0 to 7 T at $1.8,3$, and 5 K . The temperature and frequency dependent alternating-current susceptibility measurements under different applied static magnetic fields were carried out using an oscillating ac field of 2.0 Oe and ac frequencies ranging from 1 to 1000 Hz . The magnetic susceptibilities data were corrected for the sample holder as well as for diamagnetism of the constituent atoms (estimated using Pascal's constants).

## Computational details

Complete-active-space self-consistent field (CASSCF) calculations on complexes 1-Co and 2-Fe on the basis of single-crystal X-ray determined geometries have been carried out with MOLCAS $8.4^{\mathrm{S} 6}$ program package. The basis sets for all atoms are atomic natural orbitals from the MOLCAS ANO-RCC library: ANO-RCCVTZP for $\mathrm{Co}^{\mathrm{II}}$ or $\mathrm{Fe}^{\mathrm{II}}$ ion; VTZ for close N ; VDZ for distant atoms. The calculations employed the second order Douglas-Kroll-Hess Hamiltonian, where scalar relativistic contractions were taken into account in the basis set. And the spin-orbit couplings were handled separately in the restricted active space state interaction (RASSI-SO) procedure. For $\mathbf{1 - C o}$, the active electrons in 10 active spaces considering the $3 d$ double shell effect ( $5+5^{\prime}$ ) include all seven $3 d$ electrons (CAS(7 in $5+5^{\prime}$ ) for $\mathrm{Co}^{\mathrm{II}}$ ), and the mixed spin-free states are 50 (all from 10 quadruplets and all from 40 doublets for $\mathrm{Co}^{\mathrm{II}}$ ). And for 2-Fe, active electrons in 5 active spaces include all $d$ electrons (CAS (6 in 5) for $\mathrm{Fe}^{\mathrm{II}}$ ). We have mixed the maximum number of spin-free state which was possible with our hardware (all from 5 quintuplets, all from 45 triplets and all from 50 singlets for $\mathrm{Fe}^{\mathrm{II}}$. In the CASSCF calculation SINGLE_ANISO ${ }^{\text {S7 }}$ program was used to obtain the spin-free-orbit energy levels,
zero-field splitting (ZFS) parameters $D(E)\left(\mathrm{cm}^{-1}\right), \boldsymbol{g}$ tensors, magnetic axes, et al., based on the above CASSCF/RASSI-SO calculations.

Table S1 Summary of crystal data and refinement for 1-Co and 2-Fe.

|  | 1-Co | 2-Fe |
| :---: | :---: | :---: |
| Molecular formula | $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{~B}_{2} \mathrm{CoF}_{8} \mathrm{~N}_{7}$ | $\mathrm{C}_{31} \mathrm{H}_{31} \mathrm{Cl}_{2} \mathrm{FeN}_{7} \mathrm{O}_{8}$ |
| CCDC no | 1968362 | 1968363 |
| Formula weight | 734.18 | 756.38 |
| Temperature / K | 296(2) | 296(2) |
| crystal system | Triclinic | Triclinic |
| Space group | $P-1$ | $P-1$ |
| $a / \AA$ | 11.2070(3) | 11.5246(8) |
| $b / \AA$ | 12.5604(4) | 12.2303(9) |
| $c / \AA$ | 13.5908(4) | 14.0274(11) |
| $\beta /$ deg | 67.784(2) | 97.401(4) |
| $V / \AA^{3}$ | 1628.81(9) | 1632.4(2) |
| Z | 2 | 2 |
| $D_{\text {call }}, \mathrm{g} / \mathrm{cm}^{3}$ | 1.497 | 1.539 |
| $\mu / \mathrm{mm}^{-1}$ | 0.608 | 0.688 |
| $F(000)$ | 750 | 780 |
| Goodness-of-fit on $F^{2}$ | 1.018 | 1.059 |
| Final R indices $[I>2 \sigma(I)]$ | $\mathrm{R}_{1}=0.0428,$ | $\mathrm{R}_{1}=0.0553$, |
|  | $\mathrm{wR}_{2}=0.1012$ | $\mathrm{wR}_{2}=0.1601$ |
| $R$ indices (all data) | $\mathrm{R}_{1}=0.0630,$ | $\mathrm{R}_{1}=0.0852$ |
|  | $\mathrm{wR} 2=0.1144$ | $\mathrm{wR} 2=0.1985$ |

[^0]Table S2 Selected bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ for 1-Co and 2-Fe.

|  | 1-Co | 2-Fe |  |
| :--- | :---: | :--- | :---: |
| $\mathrm{Co}(1)-\mathrm{N}(1)$ | $2.2861(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(1)$ | $2.329(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(3)$ | $2.1123(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(2)$ | $2.209(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(4)$ | $2.2391(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(4)$ | $2.272(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(5)$ | $2.3327(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(5)$ | $2.271(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(6)$ | $2.1766(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(6)$ | $2.359(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(7)$ | $2.2910(19)$ | $\mathrm{Fe}(1)-\mathrm{N}(7)$ | $2.255(3)$ |
| $\mathrm{Co}(1)-\mathrm{N}(2)$ | $2.550(2)$ | $\mathrm{Fe}(1)-\mathrm{N}(3)$ | $2.554(3)$ |
|  |  |  |  |
| $\mathrm{N}(2)-\mathrm{Co}(1)-\mathrm{N}(3)$ | $74.02(7)$ | $\mathrm{N}(2)-\mathrm{Fe}(1)-\mathrm{N}(3)$ | $73.93(11)$ |
| $\mathrm{N}(4)-\mathrm{Co}(1)-\mathrm{N}(5)$ | $80.76(7)$ | $\mathrm{N}(4)-\mathrm{Fe}(1)-\mathrm{N}(5)$ | $85.39(11)$ |
| $\mathrm{N}(6)-\mathrm{Co}(1)-\mathrm{N}(7)$ | $73.34(7)$ | $\mathrm{N}(6)-\mathrm{Fe}(1)-\mathrm{N}(7)$ | $72.39(11)$ |

Table S3 The continuous shape measure (CSM) analyses of seven-coordinate geometries for 1-Co and 2-Fe by SHAPE software ${ }^{\mathrm{S8}}$

| CSM | 1-Co | 2-Fe |
| :---: | :---: | :---: |
| Heptagon | 33.749 | 33.315 |
| Hexagonal pyramid | 16.010 | 16.942 |
| Pentagonal bipyramid | 4.952 | 4.158 |
| Capped octahedron | 3.275 | 4.002 |
| Capped trigonal prism | 2.715 | 2.994 |



Figure S1 The coordination model of the central Co(II) ion for 1-Co.


Figure S2 The coordination model of the central $\mathrm{Fe}(\mathrm{II})$ ion for 2-Fe.


Figure S3. Crystal packing of 1-Co along the a axis. The H atoms are omitted for clarity.


Figure S4. Crystal packing of 2-Fe along the a axis. The H atoms are omitted for clarity.


Figure S5 The power X-ray diffraction patterns of 1-Co at room temperature.


Figure S6 The power X-ray diffraction patterns of 2-Fe at room temperature.

Table S4 The continuous shape measure (CSM) analyses of six-coordinate geometries for 1-Co and 2-Fe by SHAPE software ${ }^{\text {S8 }}$

| CSM | 1-Co | $\mathbf{2 - F e}$ |
| :---: | :---: | :---: |
| Hexagon | 27.163 | 28.542 |
| Pentagonal pyramid | 13.385 | 13.207 |
| Octahedron | 11.716 | 10.197 |
| Trigonal prism | 3.914 | 4.240 |



Figure $\mathbf{S 7}$ The $M$ versus $H / T$ curves at different temperatures for 1-Co.


Figure S8 The $M$ versus $H / T$ curves at different temperatures for 2-Fe.

Table S5 Zero field splitting patameters obtained fitting the experimental data using the PHI program for 1-Co and 2-Fe.

|  | 1-Co | 2-Fe |
| :---: | :---: | :---: |
|  | Fitting of the $\chi_{\mathrm{M}} T$ versus $T$ |  |
| $D, \mathrm{~cm}^{-1}$ | 21.4 | -15.7 |
| $\|E\|, \mathrm{cm}^{-1}$ | 1.00 | 3.27 |
| $g$ | 2.21 | 2.16 |
|  | Fitting of the $M$ versus $H$ at 1.8, 3 and 5 K |  |
| $D, \mathrm{~cm}^{-1}$ | 13.7 | -10.6 |
| $\|E\|, \mathrm{cm}^{-1}$ | 0.01 | 2.30 |
| $g$ | 2.25 | 2.21 |
|  | ab initio calculations |  |
| $D, \mathrm{~cm}^{-1}$ | 17.1 | -10.0 |
| $\|E\|, \mathrm{cm}^{-1}$ | -2.7 | 3.1 |
| $g_{x}, g_{y}, g_{z}$ | 2.349, 2.282, 2.148 | 2.002, 2.106, 2.247 |



Figure S9 Frequency dependence of the out-of-phase ( $\chi_{\mathrm{M}}{ }^{\prime \prime}$ ) ac susceptibility at 1.8 K under the applied de fields from 0 to 3.0 KOe for $\mathbf{1 - C o}$. The solid lines are foreye guide.


Figure S10 Frequency dependence of the out-of-phase ( $\chi_{\mathrm{M}}{ }^{\prime \prime}$ ) ac susceptibility at 1.8 K under the applied dc fields from 0 to 1.0 KOe for 2-Fe. The solid lines are for eye guide.

Table S6 The parameters obtained by fitting Cole-Cole plot for 1-Co.

| $\mathrm{T} / \mathrm{K}$ | $\chi_{\mathrm{S}}$ | $\chi_{\mathrm{T}}$ | $\tau$ | $a$ |
| :---: | :---: | :---: | :---: | :---: |
| 1.8 | 0.132 | 0.866 | 0.03804 | 0.19 |
| 1.9 | 0.127 | 0.817 | 0.03304 | 0.18 |
| 2.0 | 0.123 | 0.769 | 0.02843 | 0.17 |
| 2.2 | 0.117 | 0.698 | 0.02135 | 0.15 |
| 2.4 | 0.111 | 0.635 | 0.01298 | 0.11 |
| 2.6 | 0.107 | 0.583 | 0.00607 | 0.06 |
| 2.8 | 0.104 | 0.545 | 0.00242 | 0.03 |
| 3.0 | 0.108 | 0.508 | 0.00094 | 0.01 |
| 3.4 | 0.147 | 0.455 | 0.00017 | 0.01 |



Figure S11 Relaxation time of the magnetization $\ln (\tau)$ vs $T^{-1}$ plot under the 1.0 KOe applied field for 1-Co.

Table S7 The reported seven-coordinate Co(II)-SIMs

| Complex | $D(E) / \mathrm{cm}^{-1}$ | $\boldsymbol{U}_{\text {eff }} / \mathbf{K}$ | Ref. |
| :---: | :---: | :---: | :---: |
| [Co(BPA-TPA] $\left(\mathrm{BF}_{4}\right)_{2}{ }^{\text {a }}$ | 13.7 (0.01) ${ }^{\text {d }}$ | 39.7 | this work |
| $\left[\mathrm{Co}\left(\mathrm{H}_{2} \mathrm{dapb}\right)\left(\mathrm{H}_{2} \mathrm{O}\right)\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)^{\mathrm{b}}$ <br> ( H 2 dapb $=2,6$-diacetylpyridine bis(benzoyl hydrazine) | 32.4 (0) ${ }^{\text {d }}$ | 81.2 | S9a |
| $\begin{aligned} & {\left[\mathrm{CoL}_{\mathrm{N} 5}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right] \mathrm{Cl}_{2} \cdot 4 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}} \\ & \left(\mathrm{~L}_{\mathrm{N} 5}=2,13\right. \text {-dimethyl-3,6,9,12- tetraaza-1(2,6)-pyridina- } \\ & \text { cyclotridecaphane-2,12-diene }) \end{aligned}$ | 24.6 (0.014) ${ }^{\text {d }}$ | 29.8 | S9a |
| $\begin{aligned} & {\left[\mathrm{Co}(\mathrm{dapb})(\mathrm{im})_{2}\right] \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}} \\ & (\mathrm{im}=\text { imidazole }) \end{aligned}$ | $24.8(0.0016)^{\text {d }}$ | 62.3 | S9a |
| $\left[\mathrm{Co}(\mathrm{DAPBH})\left(\mathrm{NO}_{3}\right)\left(\mathrm{H}_{2} \mathrm{O}\right)\right]\left(\mathrm{NO}_{3}\right)^{\mathrm{b}}$ <br> DAPBH=2,6-diacetylpyridinebis(2'-pyridylhydrazone) | $30^{\text {e }}$ | 50 | S9b |
| $\left[\mathrm{Co}(4 \text {-tert-butylpyridine })_{3}\left(\mathrm{NO}_{3}\right)_{2}\right]^{\mathrm{b}}$ | $35.8(0.07)^{\text {e }}$ | 25.5 | S9c |
| $\left[\mathrm{Co} \text { (isoquinoline) } 3_{3}\left(\mathrm{NO}_{3}\right)_{2}\right]^{\mathrm{b}}$ | 35.7 (1.81) ${ }^{\text {e }}$ | 15.8 | S9c |
| $\begin{aligned} & {[\mathrm{CoL}]\left(\mathrm{ClO}_{4}\right)_{2}{ }^{\mathrm{b}}} \\ & (\mathrm{~L}=3,12-\mathrm{Bis}(2-\text { methylpyridine)-3,12,18-triaza-6,9- } \\ & \text { dioxabicyclo[12.3.1]octadeca-1,14,16-triene) } \end{aligned}$ | $34^{\text {d }}$ | 24.3 | S9d |
| $\left[\mathrm{Co}(\mathrm{tdmmb})\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right]\left[\mathrm{BF}_{4}\right]_{2}{ }^{\text {b }}$ | $35.1(-2.7)^{\text {f }}$ | 42.2 | S9e |
| $\left[\mathrm{Co}(\mathrm{tdmmb})(\mathrm{CN})_{2}\right] \cdot 2 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | $37.7(-3.4)^{\mathrm{f}}$ | 48.9 | S9e |
| $\left[\mathrm{Co}(\mathrm{tdmmb})(\mathrm{NCS})_{2}\right]^{\mathrm{b}}$ | $34.5(-3.6)^{\mathrm{f}}$ | 49.2 | S9e |
| $\begin{aligned} & {\left[\mathrm{Co}(\mathrm{tdmmb})(\mathrm{SPh})_{2}\right]^{\mathrm{b}}} \\ & (\mathrm{tdmmb}=1,3,10,12 \text {-tetramethyl-1,2,11,12-tetraaza}[[3] \\ & (2,6) \text {-pyridino }[3](2,9)-1,10 \text {-phenanthrolinophane-2,10- } \\ & \text { diene) } \end{aligned}$ | $39.7(-6.8){ }^{\text {f }}$ | 54.7 | S9e |


| $\left[\mathrm{Co}\left(\mathrm{H}_{2} \mathrm{daps}\right)(\mathrm{MeOH})_{2}\right]^{\text {b }}$ | 43.1 (3.3) ${ }^{\text {d }}$ | 33.5 | S9f |
| :---: | :---: | :---: | :---: |
| $\left[\mathrm{Co}\left(\mathrm{H}_{4}\right.\right.$ daps $\left.)(\mathrm{NCS})(\mathrm{MeOH})\right] \cdot(\mathrm{ClO} 4) \cdot(\mathrm{MeOH})^{\mathrm{b}}$ | $41.5(1.5)^{\text {d }}$ | 28.4 | S9f |
| $\left[\mathrm{Co}\left(\mathrm{H}_{4} \mathrm{daps}\right)(\mathrm{NCS})_{2}\right] \cdot(\mathrm{MeOH})_{2}{ }^{\text {b }}$ | 38.8 (2.1) ${ }^{\text {d }}$ | 23.6 | S9f |
| $\left(\mathrm{Ph}_{4} \mathrm{P}\right)_{2}\left[\mathrm{Co}\left(\mathrm{NO}_{3}\right)_{4}\right]$ | $10.9(1.56)^{\text {e }}$ | 12 | S9g |
| $\left(\mathrm{MePh}_{3} \mathrm{P}\right)_{2}\left[\mathrm{Co}\left(\mathrm{NO}_{3}\right)_{4}\right]$ | $12.74(2.20)^{\text {e }}$ | 20 | S9g |
| $\left[\mathrm{Co}(2,2 \text { '-bipyridine })\left(\mathrm{NO}_{3}\right)_{2}\left(\mathrm{CH}_{3} \mathrm{CN}\right)\right]^{\text {b }}$ | $32.9(0.2)^{\text {e }}$ | 56.7 | S9h |
| $\left[\mathrm{Co}(1,10 \text {-phenanthroline })\left(\mathrm{NO}_{3}\right)_{2}\left(\mathrm{CH}_{3} \mathrm{CN}\right)\right]^{\text {b }}$ | $31.4(0.1)^{\text {e }}$ | 46.0 | S9h |
| [Co(8-carboxymethoxy-2- <br> carboxylicquinoline) $\left.\left.\left(\mathrm{H}_{2} \mathrm{O}\right)_{3}\right] \cdot \mathrm{H}_{2} \mathrm{O}\right]^{\text {b }}$ | 70.4 (8.89) ${ }^{\text {d }}$ | 5.3 | S9i |
| $\begin{aligned} & {\left[\mathrm{Co}(\text { pypzbeyz })\left(\mathrm{NO}_{3}\right)_{2}\right]^{\mathrm{b}}} \\ & \text { pypzbeyz=N-((6-(1H-pyrazol-1-yl)pyridin-2- } \\ & \text { yl)methylene)benzohydrazide } \end{aligned}$ | $29.9(0.31)^{\text {d }}$ | 49.8 | S9j |
| $\begin{aligned} & {[\mathrm{CoL}] \cdot \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}} \\ & \mathrm{H}_{2} \mathrm{~L}=3,12,18 \text {-triaza-6,9-dioxabicyclo[12.3.1] octadeca- } \\ & \text { 1,14,16-triene-3,12-diacetic acid) } \end{aligned}$ | 29.1 (0) ${ }^{\text {d }}$ |  | S9k |
| $\left[\mathrm{Co}(\mathrm{L}) \mathrm{Cl}_{2}\right] \cdot 2 \mathrm{CH}_{3} \mathrm{OH}^{\mathrm{b}}$ | $38^{\text {d }}$ |  | S91 |
| $\left[\mathrm{Co}(\mathrm{L}) \mathrm{Br}_{2}\right]^{\mathrm{b}}$ | $41^{\text {d }}$ | 4.2 | S91 |
| $\begin{aligned} & {\left[\mathrm{Co}(\mathrm{~L}) \mathrm{I}_{2}\right]^{\mathrm{b}}} \\ & \mathrm{~L}=3,12,18 \text {-triaza-6,9-dioxabicyclo[12.3.1]octadeca- } \\ & 1(18), 14,16 \text {-triene } \end{aligned}$ | $35^{\text {d }}$ | 4.5 | S91 |
| $\left[\mathrm{Co}\left(\mathrm{H}_{4} \mathrm{~L}\right)(\mathrm{DMF})\left(\mathrm{H}_{2} \mathrm{O}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot(\mathrm{DMF})$ | 35.92 (1.42) ${ }^{\text {d }}$ | 25 | S9m |
| $\left[\mathrm{Co}\left(\mathrm{H}_{4} \mathrm{~L}\right)(\mathrm{MeOH})\left(\mathrm{H}_{2} \mathrm{O}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot(\mathrm{MeOH})$ | 37.23 (0.93) ${ }^{\text {d }}$ | 15 | S9m |
| $\begin{aligned} & {\left[\mathrm{Co}\left(\mathrm{H}_{4} \mathrm{~L}\right)(\mathrm{DEF})\left(\mathrm{H}_{2} \mathrm{O}\right)\right]\left(\mathrm{NO}_{3}\right)_{2}} \\ & \mathrm{H}_{4} \mathrm{~L}=2,2^{\prime} \text {-(pyridine-2,6-diylbis(ethan-1-yl-1- } \\ & \text { ylidene))bis(N-phenylhydrazinecarboxamide) } \end{aligned}$ | 43.76 (0.84) ${ }^{\text {d }}$ | 4 | S9m |
| $\left[\mathrm{Co}\left(\mathrm{L}-\mathrm{N}_{3} \mathrm{O}_{2}\right)(\mathrm{MeCN})_{2}\right]\left[\mathrm{BPh}_{4}\right]_{2}$ | $36.9(0.2)^{\text {d }}$ | 32 | S9n |
| $\left[\mathrm{CoL}^{5-\mathrm{ONHtBu}}\right] \mathrm{Cl}_{2}{ }^{\text {c }}$ | $9.2(0.02)^{\text {d }}$ | 13.2 | S9o |

 of dc magnetic data; ${ }^{\mathrm{e}}$ Values obtained from the electron paramagnetic resonance; ${ }^{\mathrm{f}}$ Values obtained from calculations.

Table S8 Calculated spin-free energies $\left(\mathrm{cm}^{-1}\right)$ of the lowest ten terms of complexes 1-Co $(S=3 / 2)$ and 2-Fe ( $S=2$ ) using CASSCF/RASSI-SO with MOLCAS 8.4

|  | 1-Co | 2-Fe |
| :---: | :---: | :---: |
|  | $E / \mathrm{cm}^{-1}$ | $E / \mathrm{cm}^{-1}$ |
| 1 | 0.0 | 0.0 |
| 2 | 3918.1 | 591.6 |
| 3 | 4867.9 | 4625.7 |
| 4 | 5209.4 | 5233.3 |
| 5 | 7106.6 | 6492.7 |
| 6 | 8387.7 | 19202.6 |
| 7 | 10053.2 | 20522.6 |
| 8 | 21125.0 | 21174.1 |
| 9 | 21875.8 | 22046.4 |
| 10 | 25197.0 | 22560.7 |

Table S9 Calculated weights of the five most important spin-orbit-free states for the lowest two spin-orbit states of $\mathbf{1 - C o}$ and $\mathbf{2 - F e}$ using CASSCF/RASSI-SO with MOLCAS 8.4.

|  | Spin-orbit <br> states | $\begin{aligned} & \text { Energy } \\ & \left(\mathrm{cm}^{-1}\right) \end{aligned}$ | Spin-free states, Spin, Weights |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1-Co | 1 | 0.0 | $\begin{gathered} 1,1.5,0.97 \\ 64 \end{gathered}$ | $\begin{gathered} 2,1.5,0.01 \\ 40 \end{gathered}$ | $\begin{gathered} 3,1.5,0.00 \\ 64 \end{gathered}$ | $\begin{gathered} \hline 4,1.5,0.00 \\ 14 \end{gathered}$ | $\begin{gathered} 17,0.5,0.0 \\ 005 \end{gathered}$ |
|  | 2 | 35.4 | $\begin{gathered} \hline 1,1.5,0.98 \\ 37 \end{gathered}$ | $\begin{gathered} 4,1.5,0.00 \\ 56 \end{gathered}$ | $\begin{gathered} 3,1.5,0.00 \\ 51 \end{gathered}$ | $\begin{gathered} 2,1.5,0.00 \\ 44 \end{gathered}$ | $\begin{gathered} 19,0.5,0.0 \\ 003 \end{gathered}$ |
| 2-Fe | 1 | 0.0 | $\begin{gathered} 1,2.0,0.94 \\ 02 \end{gathered}$ | $\begin{gathered} 2,2.0,0.05 \\ 48 \end{gathered}$ | $4,2.0,0.00$ <br> 14 | $5,2.0,0.00$ <br> 14 | $\begin{gathered} 3,2.0,0.00 \\ 12 \end{gathered}$ |
|  |  | 2.7 | $\begin{gathered} 1,2.0,0.93 \\ 64 \end{gathered}$ | $\begin{gathered} 2,2.0,0.05 \\ 95 \end{gathered}$ | $\begin{gathered} 5,2.0,0.00 \\ 15 \end{gathered}$ | $\begin{gathered} 3,2.0,0.00 \\ 11 \end{gathered}$ | $\begin{gathered} 4,2.0,0.00 \\ 07 \end{gathered}$ |
|  | 2 | 22.9 | $\begin{gathered} 1,2.0,0.97 \\ 58 \end{gathered}$ | $\begin{gathered} 2,2.0,0.01 \\ 84 \end{gathered}$ | $\begin{gathered} 4,2.0,0.00 \\ 33 \end{gathered}$ | $\begin{gathered} 3,2.0,0.00 \\ 08 \end{gathered}$ | $\begin{gathered} 5,2.0,0.00 \\ 08 \end{gathered}$ |
|  | 3 | 41.7 | $\begin{gathered} 1,2.0,0.97 \\ 37 \end{gathered}$ | $\begin{gathered} 2,2.0,0.02 \\ 32 \end{gathered}$ | $\begin{gathered} 4,2.0,0.00 \\ 11 \end{gathered}$ | $\begin{gathered} 5,2.0,0.00 \\ 10 \end{gathered}$ | $\begin{gathered} 3,2.0,0.00 \\ 03 \end{gathered}$ |
|  |  | 45.9 | $\begin{gathered} 1,2.0,0.98 \\ 81 \end{gathered}$ | $\begin{gathered} 2,2.0,0.00 \\ 81 \end{gathered}$ | $\begin{gathered} 4,2.0,0.00 \\ 20 \end{gathered}$ | $\begin{gathered} 5,2.0,0.00 \\ 08 \end{gathered}$ | $\begin{gathered} 3,2.0,0.00 \\ 03 \end{gathered}$ |



Figure S12 Calculated (red solid line) data of magnetic susceptibilities of 1-Co.


Figure S13 Calculated (red solid line) data of magnetic susceptibilities of 2-Fe.

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[^0]:    ${ }^{\mathrm{a}} \mathrm{wR}_{2}=\left[\Sigma\left[w\left(\mathrm{~F}_{\mathrm{o}}^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}\right] / \Sigma\left[w\left(\mathrm{Fo}^{2}\right)^{2}\right]\right]^{1 / 2}, \mathrm{R}_{1}=\Sigma| | \mathrm{F}_{\mathrm{o}}\left|-\left|\mathrm{F}_{\mathrm{c}}\right|\right| / \Sigma\left|\mathrm{F}_{\mathrm{o}}\right|$.

