

Supplementary Materials

Cone-like titanate immobilized on polyacrylonitrile nanofibers: hierarchical architecture for effective photocatalytic activity

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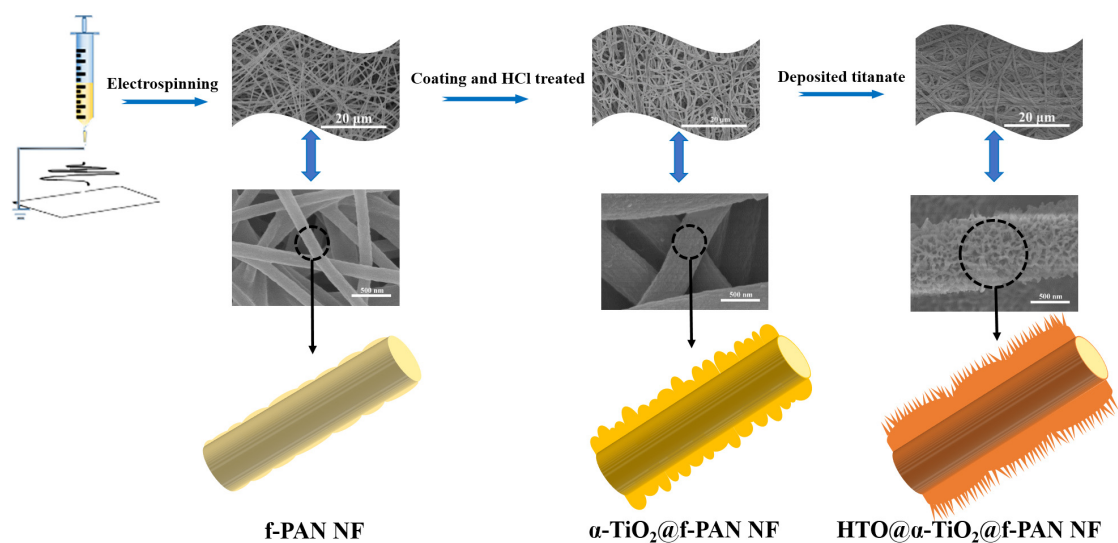


Figure S1 The scheme of fabrication process for the HTO@ α -TiO₂@f-PAN NF photocatalysts.

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The spectrum of LED light source (Fig. S2) in our experiment system was investigated by USB4000 optical fiber UV-VIS spectrometer, the region between 300 nm-400 nm is about 3.5 %, while that between 400 nm-888 nm is about 96.5 %.

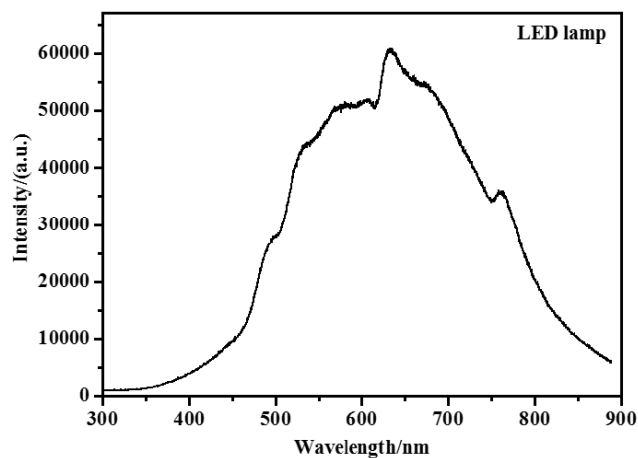


Figure S2 Relative spectral radiance of LED light source.

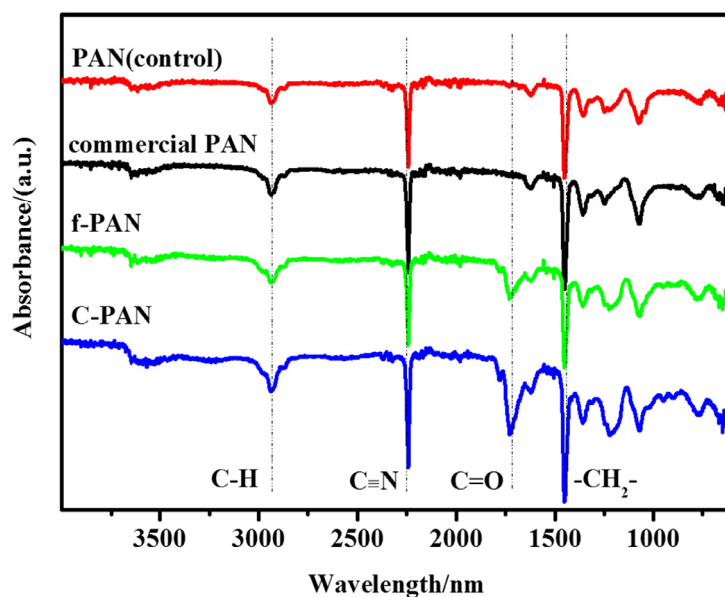


Figure S3 FTIR spectra of different samples.

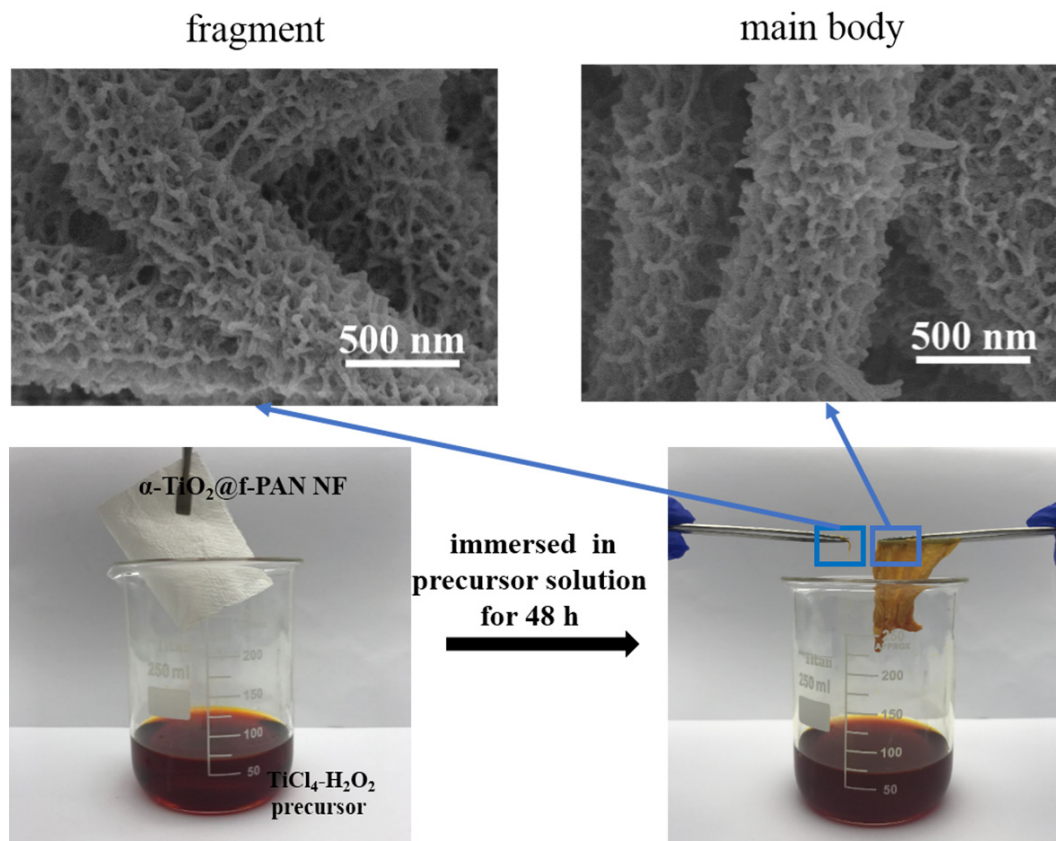


Figure S4 Photographs of the preparation process for 48-HTO@ α -TiO₂@f-PAN NF and SEM images of different part of 48-HTO@ α -TiO₂@f-PAN NF sample.

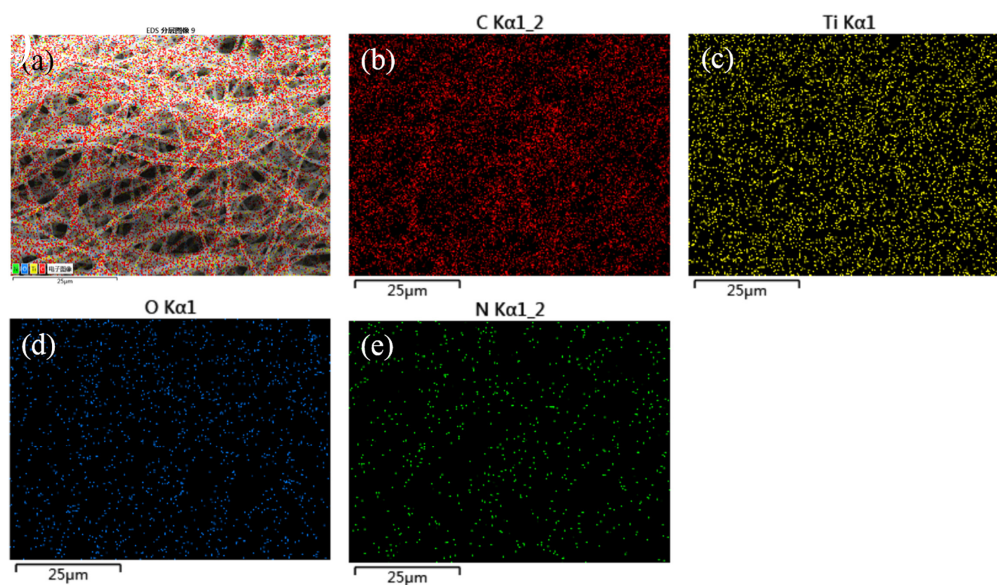


Figure S5 The SEM element mapping images of S-HTO@ α -TiO₂@f-PAN NF.

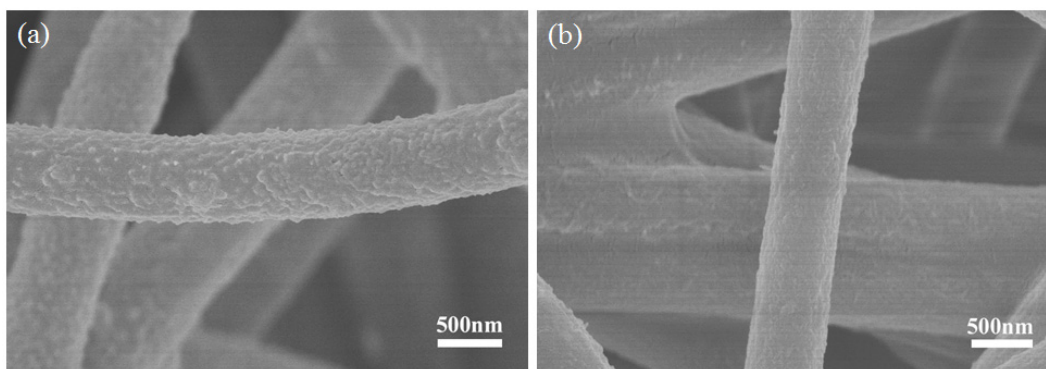


Figure S6 SEM images of (a) 36-HTO@ α -TiO₂@f-PAN NF prepared without cyanuric acid and (b) HTO@f-PAN NF prepared without α -TiO₂ layer.

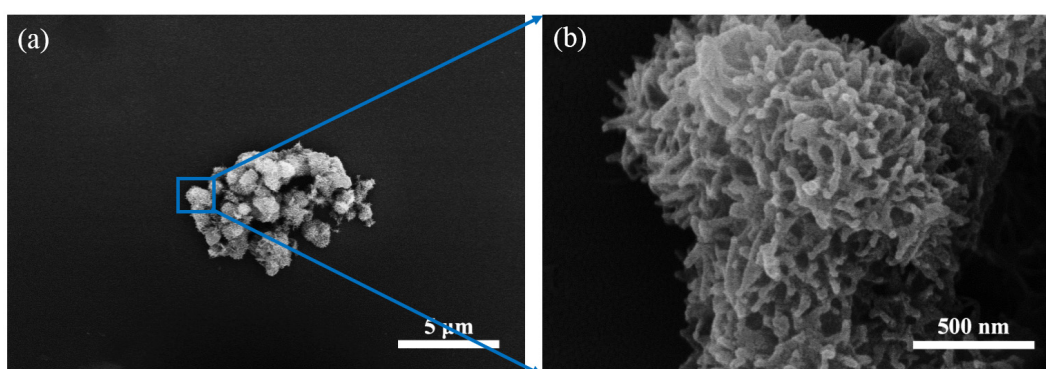


Figure S7 SEM images using different magnification of titanate powder deposited for 36h in the peroxide precursor with no fiber carriers.

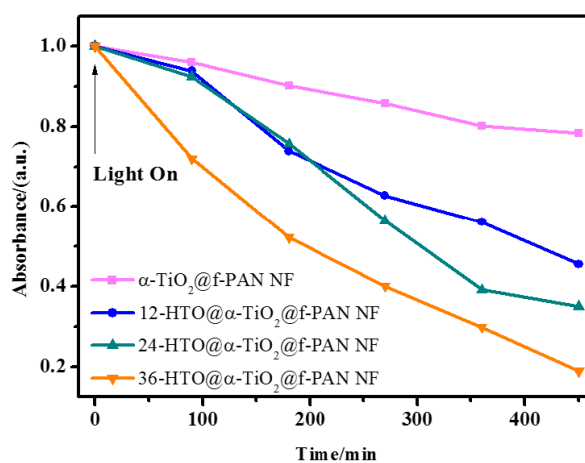


Figure S8 The absorbance ($\lambda=554$ nm) of RhB aqueous solution when using HTO@ α -TiO₂@f-PAN NF samples under LED light irradiation.

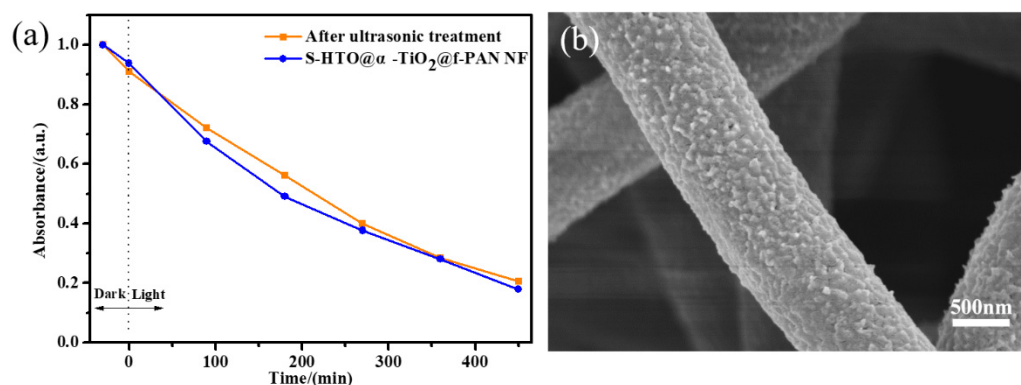


Figure S9 (a) The absorbance of RhB ($\lambda=554\text{nm}$) under LED over S-HTO@ α -TiO₂@f-PAN NF before and after ultrasonic treatment and (b) SEM image of HTO@ α -TiO₂@f-PAN NF after ultrasonic treatment.

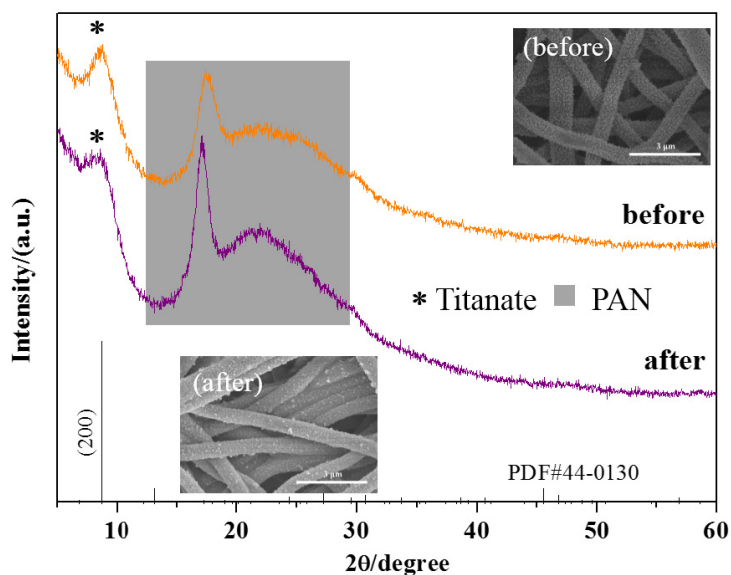


Figure S10 XRD patterns of S-HTO@ α -TiO₂@f-PAN NF after five times photocatalytic degradation of RhB aqueous solution and inset SEM images of S-HTO@ α -TiO₂@f-PAN NF before and after photocatalytic degradation of RhB.

The photocatalytic activity of tetracycline hydrochloride (TC-HCl), a colorless organic pollutant, was also tested here. In these typical experiments, 0.06 g S-HTO@ α -TiO₂@f-PAN NF composite photocatalyst was dispersed in the reactor containing 100 mL TC-HCl aqueous solution of different concentration (100mg/L, 50mg/L, 25mg/L). The removal rate of organic pollutant was determined by UV-vis spectrophotometer (TU-1900, China) at the characteristic wavelength of 357 nm. [S1-S3]

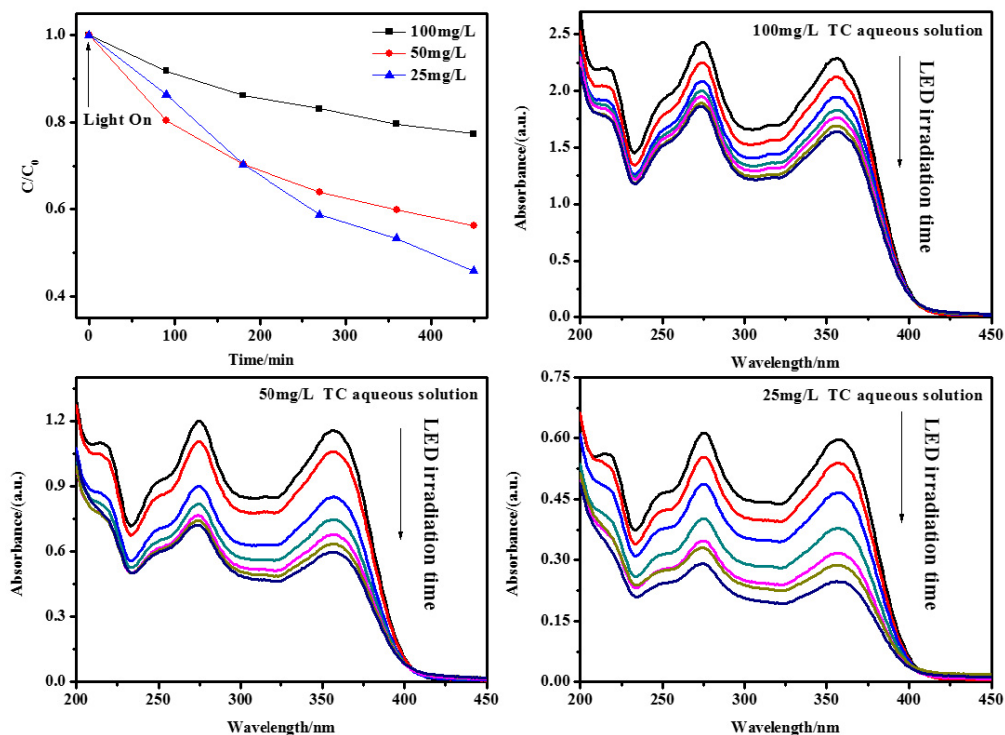


Figure S11 Concentration variation of TC-HCl aqueous solution under LED light irradiation in presence of S-HTO@ α -TiO₂@f-PAN NF composite photocatalyst.

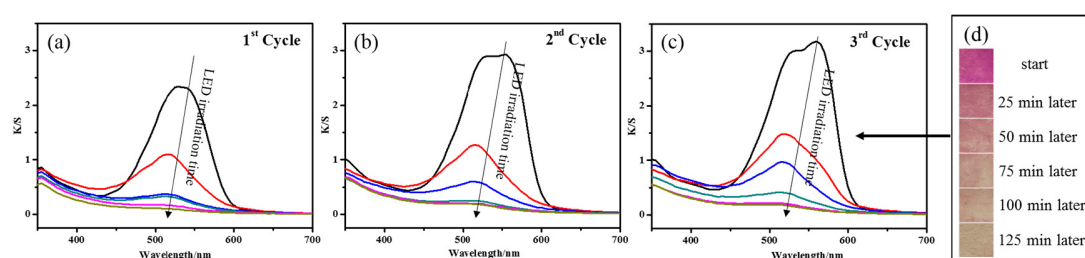


Figure S12 (a-c) The K/S curves of S-HTO@ α -TiO₂@f-PAN NF stained by RhB under LED irradiation for three cycles and (d) Photos of the S-HTO@ α -TiO₂@f-PAN NF stained by RhB in the third cycle.

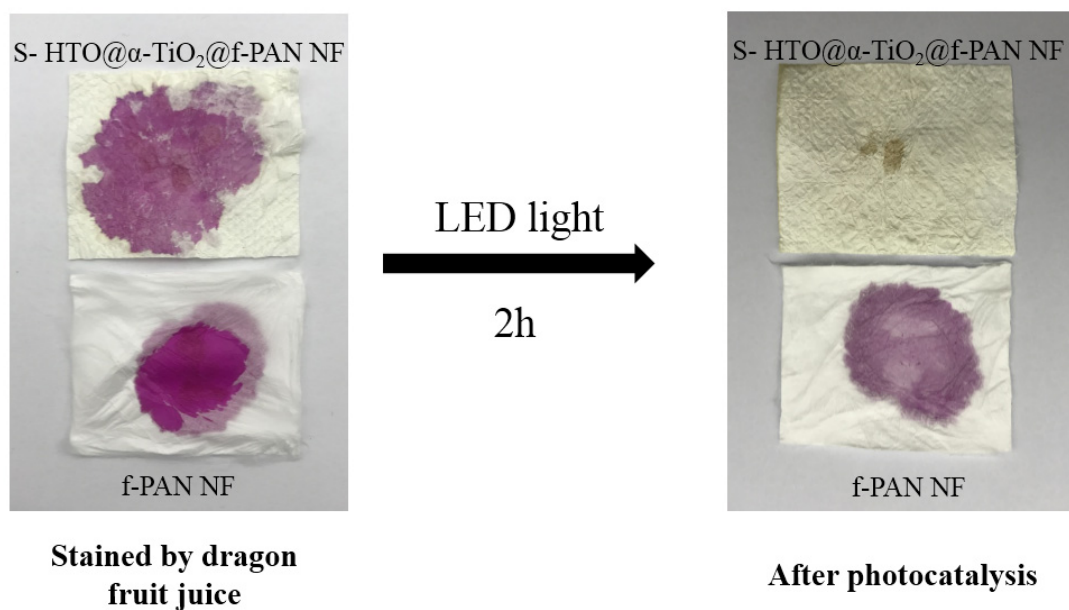


Figure S13 Photocatalytic degradation of dragon fruit juice stains.

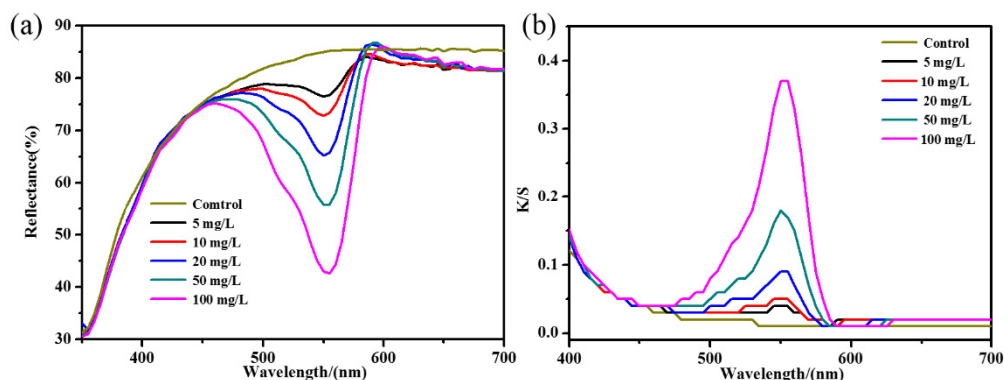


Figure S14 (a) Reflectance and (b) K/S curves of PAN fabric stained with different concentrations of RhB aqueous solution.

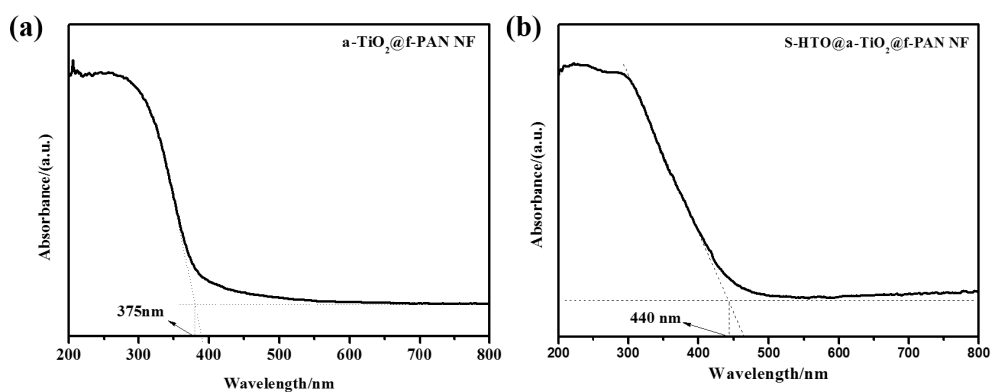


Figure S15 UV-vis-DRS of (a) α -TiO₂@f-PAN NF and (b) S-HTO@ α -TiO₂@f-PAN NF.

References

- [S1]. R. Guan, X. Yuan, Z. Wu, H. Wang, L. Jiang, J. Zhang, Y. Li, G. Zeng and D. Mo, *Chemical Engineering Journal*, 2018, 350, 573-584.
- [S2]. L. Rimoldi, D. Meroni, G. Cappelletti and S. Ardizzone, *Catalysis Today*, 2017, 281, 38-44.
- [S3]. S. Panneri, M. Thomas, P. Ganguly, B. N. Nair, A. P. Mohamed, K. G. K. Warriar and U. S. Hareesh, *Catalysis Science & Technology*, 2017, 7, 2118-2128.