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Support Information

Bimental composites for photocatalytic reduction of CO2 to CO in the near-infrared region

by SPR effect

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Figure S1. XRD of different proportions of CuNi/C-550(a). SEM CuNi/C-300(b).



Figure S2. XRD of C(a). FTIR patterns of C(b). TEM of C(c,d).



 $\label{eq:sigma} Figure \ S3.SEM \ of \ Cu_{0.66}Ni_{0.33}BTC(a), \quad Cu_{0.75}Ni_{0.25}BTC(b), \ Cu_{0.33}Ni_{0.66}BTC(c), \ Cu_{0.25}Ni_{0.75}BTC(d)$



Figure S4 .SEM of CuNi/C-450



Figure S5. .SEM of CuNi/C-550



Figure S6.SEM of CuNi/C-550



Figure S7.SEM of CuNi/C-650



Figure S8 .SEM of CuNi/C-750



Figure S9. EDS of Cu/C-550.



Figure S10. EDS of Ni/C-550.



Figure S11. EDS of CuNi/C-450.



Figure S12. EDS of CuNi/C-650



Figure S13. EDS of CuNi/C-750



Figure S14. XPS of CuNi/C-450



Figure S16 .Sample adsorption and desorption capacity for N2 (a)Cu/C-550. (b)CuNi/C-450. (c)CuNi/C-550. (d)CuNi/C-



 $\label{eq:size} Figure \ S17 \ . Pore \ size \ distribution(a) CuNi/C-450. \ (b) \ CuNi/C-550. \ (c) \ CuNi/C-650. \ (d) \ Cu/C-550. \ (e) \ Ni/C-550. \ (f) \ Sample \ adsorption \ and \ desorption \ capacity \ for \ N_2$



Figure S18. Electrical properties: photocurrent response of the samples.

Test Conditions	CO precipitation rate	CH ₄ precipitation rate
	$(\mu mol \cdot g^{-1}h^{-1})$	$(\mu mol \cdot g^{-1}h^{-1})$
Without TEOA	0	0
Without CO ₂	0.012	0
Dark conditions	0	0
Without catalyst	0	0

Table S1. Gas precipitation under full light (without filter) under different conditions. (The catalyst added in all reactions is CuNi/C-550.)