Supporting Information

A new family of cation-disordered Zn(Cu)-Si-P compounds as high-performance anodes for next-generation Li-ion batteries

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 Table S1. Crystallographic parameters and reliability factors for the cation-disordered ZnSiP₂ sample from XRD refinement.

Compound	ZnSiP ₂		
Sp.Gr.	F-43m		
<i>a</i> , Å	5.34025		
<i>V</i> , Å ³	152.3		
2θ-interval, °	5-120		
Number of reflections	13		
Rwp, %	7.23		
Rp, %	8.9		
Rexp, %	2.67		
χ2	1.62		
RB, %	1.77		

Table S2. Fractional atomic coordinates and isotropic displacement parameters (Å2) of ZnSiP₂.

	х	У	Z	$B_{ m iso}$	Occ.
Р	0.25	0.25	0.25	1	1
Si	0	0	0	1	0.5
Zn	0	0	0	1	0.5



Figure S1. TEM image and the selected area electron diffraction (SAED) pattern of the cation-disordered ZnSiP₂ powder.



Figure S2. High-resolution TEM image of the cation-disordered ZnSiP₂.



Figure S3. Discharge/charge profiles of the cation-disordered ZnSiP₂, Zn/C, Si/C and P/C electrodes at a current density of 200 mA g⁻¹.



Figure S4. The initial anodic scan at different scan rates of the cation-disordered $ZnSiP_2$ anodes: black (Electrode 1), the initial anodic scan at a scan rate of 0.1 mV s⁻¹. Red (Electrode 2), the initial anodic scan at a scan rate of 0.02 mV s⁻¹; blue (Electrode 2), the 4th anodic scan at a scan rate of 0.02 mV s⁻¹.



Figure S5. The initial discharge profiles of the ZnSiP₂ anodes cycled at ultralow current density of 20, 40, 80, and 100 mA g⁻¹.



Figure S6. Initial three discharge/charge profiles of the cation-disordered ZnSiP₂ electrode at a current density of 200 mA g⁻¹



Figure S7. a) Schematic of the cation-disordered ZnSiP₂ crystal structure (red ball: P atom, gray white ball: Zn atom); b) the electronic structure based on the above model. c) the crystal structure model of the cation-ordered ZnSiP₂; d) the electronic structure of the cation-ordered ZnSiP₂.



Figure S8. Electrochemical impedance spectroscopy study of a cation-disordered ZnSiP₂ electrode and a cation-ordered ZnSiP₂ electrode: a) impedance spectrum of a cell with a cation-disordered ZnSiP₂ electrode and that with a cation-ordered ZnSiP₂ electrode acquired under identical conditions, and b) charge transfer resistance (R_{ct}) determined from the impedance spectra shown in part a).



Figure S9. Li-ion diffusion barrier energy and its corresponding diffusion paths calculated based on the simulated model as shown in Figure S5a.



Figure S10. Li-ion diffusion barrier energy and its corresponding diffusion paths calculated based on the simulated model as shown in Figure S5c.



Figure S11. The cells were executed with a constant pulse for 30 min and then rest for 10 h: a) the initial discharge/charge profile of the cation-disordered ZnSiP₂ and b) the initial discharge/charge profile of the cation-ordered ZnSiP₂. Li-ion diffusion coefficients of the cation-disordered ZnSiP₂ and cation-ordered ZnSiP₂ anodes during discharge (c) and charge (d) processes.



Figure S12. Discharge/charge profiles of the cation-disordered ZnSiP₂ used for ex-situ characterizations.



Figure S13. Schematic illustration of the voids suitable for Li-ion accommodation within the crystal structure of the cation-disordered Zn₂Si₂P₄ (one unit cell).



Figure S14. Ex-situ HRTEM and SAED patterns at the elected working potentials corresponding to Figure S9.



Figure S15. Schematic of the Li₂ZnSi crystal structure.



Figure S16. The ex-situ XRD (a) and HRTEM image (b) of the ZnSiP₂ electrode after 30 cycles.



Figure S17. The ex-situ XPS of the ZnSiP₂ electrode after 30 cycles: a) high-resolution survey of P 2p; b) high-resolution survey of Si 2p.



Figure S18. The ex-situ Raman of the ZnSiP₂ electrode after 30 cycles.



Figure S19. XRD pattern of the cation-disordered $ZnSiP_2/C$ composite.



Figure S20. FSEM images of the cation-disordered ZnSiP₂ carbon composite: a) low-magnitude, b) high-magnitude.



Figure S21. N₂ adsorption-desorption isotherms of a) the cation-disordered ZnSiP₂, and b) its carbon composite.



Figure S22. Cycle stability of the cation-disordered ZnSiP₂/C anode at the current density of 2000 mA g⁻¹.