

Supplementary Information

Prospects for finding Junge variability-lifetime relationships for micropollutants in the Danube river.

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Monitoring stations in the Danube river as used for STREAM-EU

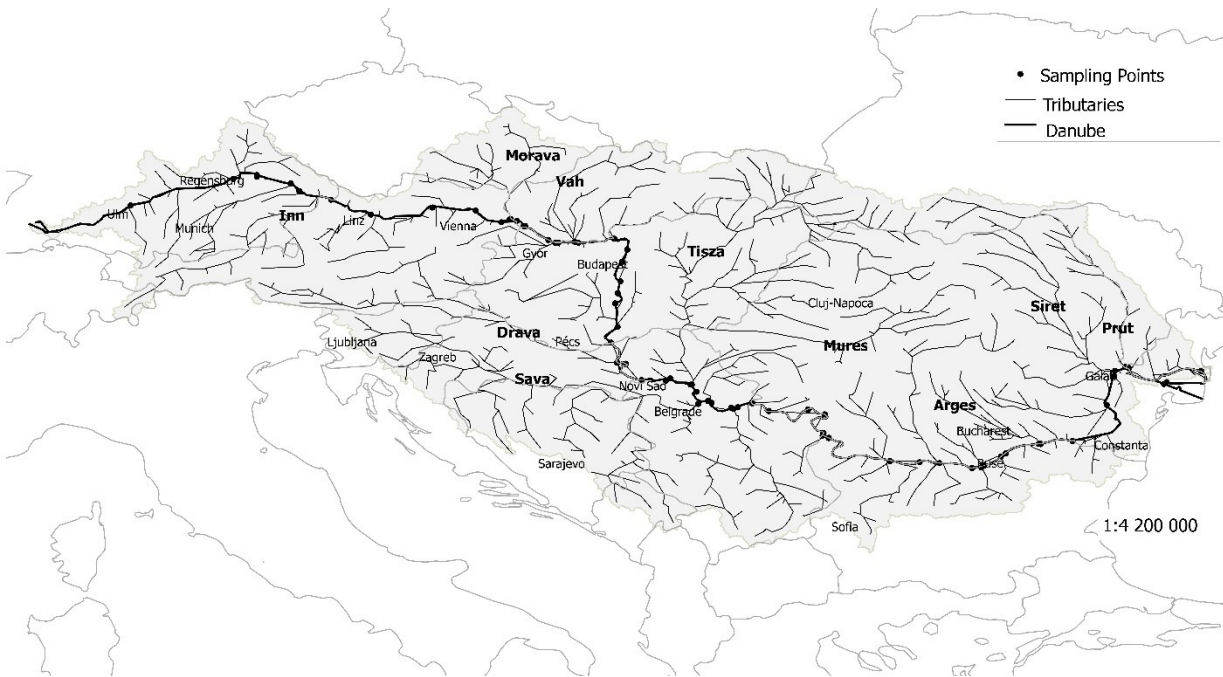
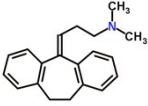
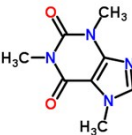
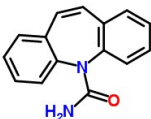
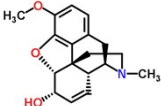
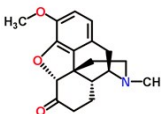
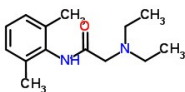
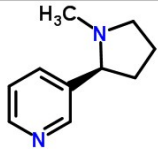
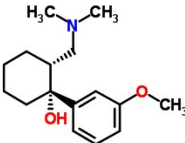
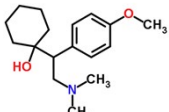


Figure S1. Map used in Lindim, et. al.¹ displaying the 68 sampling points along the Danube river that were modeled in STREAM-EU^{1, 2}. The daily concentrations of 4 hypothetical micropollutants were modeled in these stations for the year 2013 and used to calculate spatial and temporal Junge variability-lifetime relationships.

Properties of the monitored micropollutants that were used in the empirical Junge relationship

Table S1. Micropollutants monitored by Croatian Waters in the third Joint Danube Survey (JDS3-CW) and their relevant properties.

Compound	CAS	Chemical group and function	Structure ‡	log D_{ow} *	Main ionic form at pH 6.5-8.5*	log K_{AW} +
Amitriptyline	50-48-6	Pharmaceutical: Antidepressant		2.48	Cationic	-5.55
Caffeine	58-08-2	Stimulant		-0.55	Neutral	-8.83

Carbamazepine	298-46-4	Pharmaceutical: Anti-epileptic		2.77	Neutral	-8.35
Codeine	76-57-3	Pharmaceutical: Analgesic/ opioid		-0.45	Cationic	-11.51
Hydrocodone	125-29-1	Pharmaceutical: Analgesic		0.73	Cationic	-9.58
Lidocaine	137-58-6	Pharmaceutical: Anesthetic		2.33	Cationic	-8.27
Nicotine	54-11-5	Stimulant		-0.04	Cationic	-6.91
Tramadol	27203-92-5	Pharmaceutical: Analgesic/ opioid		0.62	Cationic	-9.20
Venlafaxine	93413-69-5	Pharmaceutical: Antidepressant		1.22	Cationic	-9.08

40 Note: ‡ obtained from ChemSpider. * Obtained from ChemAxon Chemicalize web-based software
41 (www.chemicalize.com). Distribution coefficient ($\log D_{OW}$) was obtained at pH 7.4, main ionic form was
42 deduced from the software-estimated strongest acidic or basic pK_a + Calculated from Henry's Law
43 constant at 25°C obtained from EPI Suite V4.11³

44

45 Method used to fit measured concentrations of JDS3-CW to log-normal
46 distributions and to estimate the measurements below LOQ.

47

48 The empirical relative standard deviation (σ/μ) for spatial variability was calculated for the nine
49 micropollutants measured in the JDS3 (Table S1). To avoid bias in the estimation of means and standard
50 deviations calculated only from measurements above LOQ, we imputed values below LOQ by fitting the
51 concentration measurements to log-normal distributions and extrapolating values in the lower tail (as
52 recommended for censored data ^{4,5}) and not from the parameters of the fitted log-normal distribution.

53 Concentrations in (ng/L) obtained for compounds measured in the JDS3-CW were first log-transformed and
54 then inverse-ranked from highest to smallest, meaning that if N concentrations were below LOQ, then the

55 highest concentration corresponding to rank 68 and the lowest to 68-N. An empirical cumulative probability
56 (Y) was calculated for each log-concentration with Equation S1.

57
$$\text{Eq. S1.} \quad Y = \text{rank} / 68$$

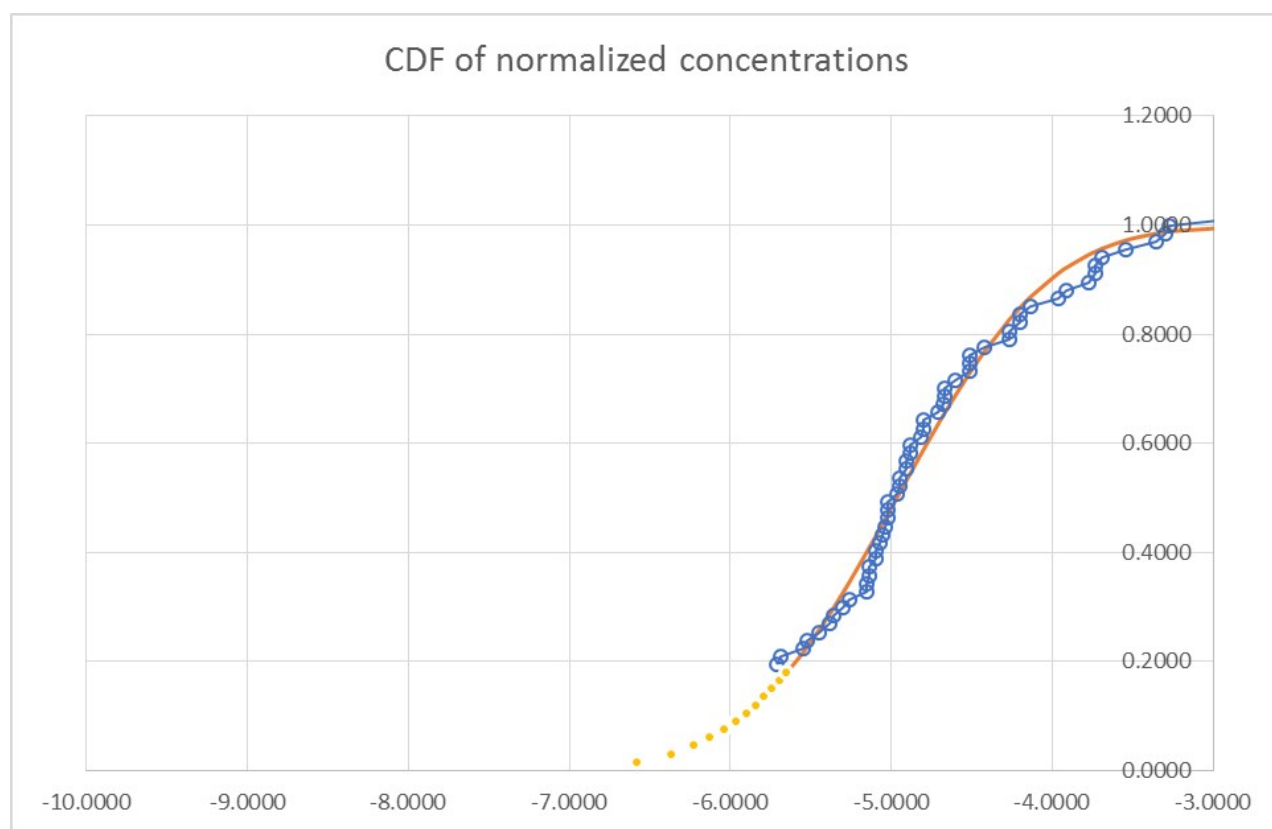
58 A modeled cumulative probability (Y') according to a normal distribution was calculated with initial
59 estimated values for the mean (μ) and standard deviation of Y (σ).

60 Based on a method described in Reference 3, the squared error between each Y and Y' was calculated and
61 summed. Using the “Solver” tool in Excel, the first estimates of $\mu_{\text{fit-lognorm}}$ and $\sigma_{\text{fit-lognorm}}$ were changed in
62 order to minimize the sum of squared errors.⁶

63 The N concentrations below LOQ were then calculated using the optimized $\mu_{\text{fit-lognorm}}$ and $\sigma_{\text{fit-lognorm}}$ and
64 combined with the original concentration measurements for the subsequent calculation of σ/μ intended for
65 the calibration of the empirical Junge relationship. The $\mu_{\text{fit-lognorm}}$ and $\sigma_{\text{fit-lognorm}}$ were not back-transformed
66 and used directly for the calculation of σ/μ , because this is known to cause bias in the estimation of central-
67 tendency and dispersion parameters of small skewed samples.^{4, 5}

68 Figure S2 shows the graphical visualization of the fitting.

69



70

71 Figure S2. Cumulative distribution function for the log-normal concentrations measured in the JDS3-CW (in blue),
72 the fitted normal distribution (in orange) and the imputed concentrations below LOQ (in yellow).

73

74 A single arithmetic mean, standard deviation and subsequently the relative standard deviation σ/μ , were
 75 calculated for each micropollutant using the concentrations measurements above the LOQ (in blue)
 76 combined with the imputed data below the LOQ (in yellow) (as recommended for censored data ^{4,5}) and
 77 not from the parameters of the fitted log-normal distribution (in orange).

78 Compounds where all concentrations were close to LOQ (i.e. sulfamethoxazole) were excluded for this
 79 part of analysis due to the low resolution and measurement error.

80

81 Evaluation of sorption of cationic compounds to soil and sediment (K_d calculation). 82

83 Seven of the compounds used in the empirical Junge relationship (shown in Tables S1 and S6) occur mainly
 84 as cationic species in freshwater. Sorption of cationic pharmaceuticals to soil and sediment is higher
 85 compared to neutral pharmaceuticals with similar K_{OW} , and thus sorption might be underestimated by a
 86 criteria based only on D_{OW} . Therefore sediment-water partition coefficients (K_d) were calculated according
 87 to Droge ⁷ to assess degree of sorption to sediment and dissolved matter from the distribution coefficient
 88 D_{OW} . The estimated K_d of the seven cationic compounds is lower than the K_d calculated with the sorption
 89 criteria we used (i.e., $\log D_{OW} < 4$, see SI), so therefore we decided these chemicals meet our criteria and
 90 can be included in this study.

91

92 The distribution coefficient (K_d) was calculated using two approaches:

- 93 1) Using a refined sorption model for organic cations to soil and clay (Eq. S4), by calculating D_{OC}
 94 (Eq. S2Eq. S2 ⁷ $\log D_{OC} = 1.53Vx + 0.32NA_i - 0.27$
 95) and K_{CEC} (Eq. S3) from the micropollutant molecule volume V_x and surface area NA_i ⁷
 96

97	<i>Eq. S2</i> ⁷	$\log D_{OC} = 1.53Vx + 0.32NA_i - 0.27$
98	<i>Eq. S3</i> ⁷	$\log K_{CEC,clay} = 1.22Vx + 0.22NA_i + 1.09$
99	<i>Eq. S4</i> ⁷	$K_d = K_{CEC,clay} \cdot (CEC_{soil} - 3.4 \cdot f_{OC}) + f_{OC} \cdot D_{OC}$

100

- 101 2) Using a calculated from a traditional linear regression between D_{OC} and D_{OW} (Eq. S5)⁸, and then
 102 using this D_{OC} for calculation of K_d as in Equation S4.
 103

104 *Eq. S5* ⁸ $D_{OC} = 10^{0.31 \log D_{OW} + 2.78}$

105 The average and maximum concentrations of dissolved organic carbon (DOC) measurements from JDS3-
 106 CW were taken into account for the calculations of K_d and since the exact composition of the DOC in the
 107 Danube river is unknown, the properties of standard Eurosoils ES-1 and ES-5⁷ were used.

108

109 Table S2. Average and maximum dissolved organic matter (DOM) parameters obtained from the JDS3-CW data.⁹

	DOM	
average	2.98	mg/L
maximum	5.50	mg/L

110

111 Table S3. Standard Eurosoil 1 and 5 (ES-1 and ES-5) characteristics used on the K_d calculations according to Reference 6. ES-1 is the standard soil
112 with highest content of clay and lowest f_{oc} , and ES5 the one with highest f_{oc} and lowest clay content.⁷

Soil type	f_{oc}		CEC_{soil}		CEC_{clay}	
ES-1	0.013	kg oc / kg solid dw	0.299	mmol/kg	0.2548	mol/kg
ES-5	0.0925	kg oc / kg solid dw	0.327	mmol/kg	0.0125	mol/kg

113

114 Table S4. Calculation of K_d for two different types of soil (ES-1 and ES-5) and average or maximum dissolved organic matter (DOM) measured in
115 the Danube during the JDS3. The K_d of the compounds in the empirical Junge relationship do not exceed the K_d of the theoretical D_{OW} limit chosen.

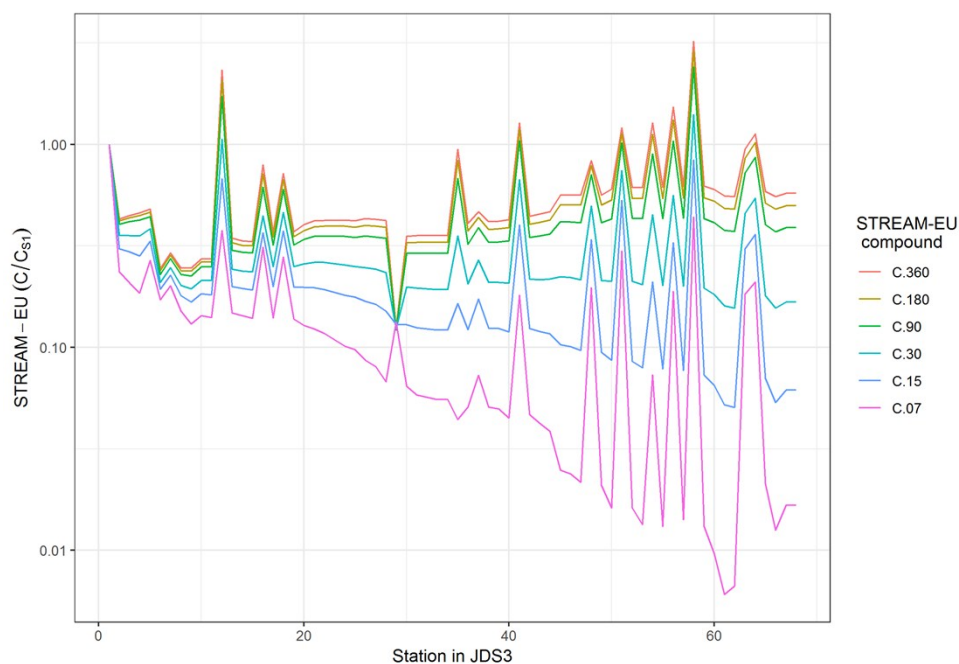
Compound	$\log D_{OW}$	ES-1, average DOM			ES-5, average DOM		ES-1, maximum DOM		ES-5, maximum DOM			
		$\log D_{OC}$	$\log D_{OC}$	$\log K_{CEC}$	$\log K_d$	$\log K_d$	$\log K_d$	$\log K_d$	$\log K_d$	$\log K_d$	$\log K_d$	$\log K_d$
		(Eq S5)	(Eq S2)	(Eq S3)	(Eq S5+S4)	(Eq S2+S4)	(Eq S5+S4)	(Eq S2+S4)	(Eq S5+S4)	(Eq S2+S4)	(Eq S5+S4)	(Eq S2+S4)
Amitriptyline	2.48	3.55	3.82	4.88	-1.23	-1.23	-2.42	-2.33	-0.97	-0.96	-2.16	-2.07
Caffeine	-0.55	2.61	1.45	3.46	-2.64	-2.66	-3.66	-3.94	-2.37	-2.39	-3.39	-3.67
Carbamazepine	2.77	3.64	2.50	4.30	-1.79	-1.82	-2.71	-3.08	-1.52	-1.55	-2.45	-2.82
Codeine	-0.45	2.64	3.63	4.72	-1.40	-1.39	-2.68	-2.50	-1.13	-1.12	-2.42	-2.24
Hydrocodone	0.73	3.01	3.63	4.72	-1.40	-1.39	-2.65	-2.50	-1.13	-1.12	-2.38	-2.24
Lidocaine	2.33	3.50	3.20	4.38	-1.72	-1.73	-2.75	-2.87	-1.45	-1.46	-2.48	-2.61
Nicotine	-0.04	2.77	2.15	3.54	-2.55	-2.57	-3.54	-3.77	-2.29	-2.30	-3.27	-3.51
Tramadol	0.62	2.97	3.77	4.84	-1.28	-1.27	-2.55	-2.38	-1.02	-1.00	-2.29	-2.11
Venlafaxine	1.22	3.16	3.68	4.77	-1.35	-1.34	-2.59	-2.45	-1.08	-1.07	-2.32	-2.19
Theoretical sorption limit	4	4.02	3.82	4.88	-1.22	-1.23	-2.24	-2.33	-0.95	-0.96	-1.98	-2.07

116

117

118 STREAM-EU concentration predictions for the JDS3 monitoring campaign

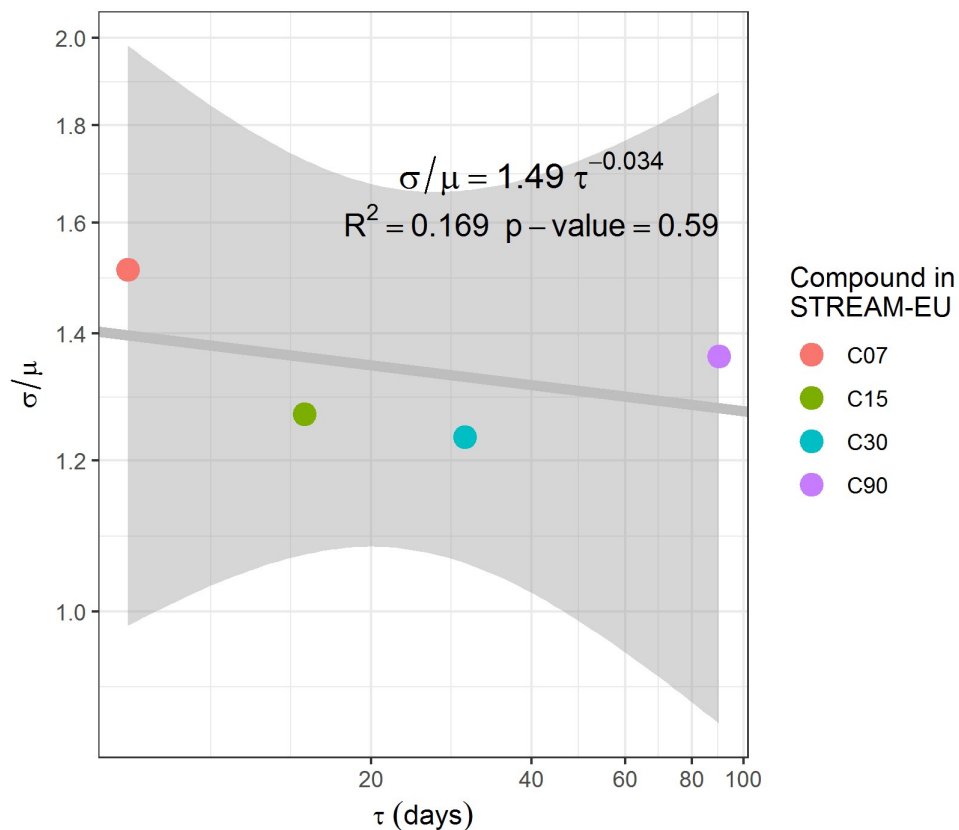
119



120

121 Figure S3. Normalized concentrations predicted for the corresponding date and station sampled during the
 122 third Joint Danube Survey (JDS3). The normalization was relative to the concentration of the compound in
 123 the first station (S1)

124 The normalized modeled concentrations ($C/C_{\text{station 1}}$) have more variation for compounds with shorter half-
 125 life (i.e. C07 with $\tau = 7$ days), than for longer half-lives (i.e. C360 with $\tau = 360$ days). Variation is similar
 126 for the three most persistent hypothetical compounds with half-lives of 90, 180 and 360 days.



127

128 Figure S4. Junge relationship of STREAM-EU concentrations predicted for dates and location of
 129 measurements performed in the Joint Danube Survey (JDS3). Only the four compounds with shortest half-
 130 lives (7, 15, 30 and 90 days) were used to derive this relationship.

131

132 Junge relationships from STREAM-EU synthetic data

133

134 Table S5. Summary of Junge relationships for four hypothetical chemicals with concentrations modeled by
 135 STREAM-EU. The parameter a represents the intercept, and b the slope of the Junge relationships. The
 136 temporal relationships were calculated for 67 monitoring stations along the Danube river, the spatial
 137 relationships were calculated daily for the year 2013.

	Temporal relationships Mean (5 th , 95 th percentiles)	Spatial relationship Mean (5 th , 95 th percentiles)
Total number	67	365
a	1.33 (0.62, 2.98)	1.32 (0.99, 1.89)
b	-0.327 (-0.651, -0.088)	-0.154 (-0.212, -0.103)
p-value (slope)	0.027 (0.002, 0.085)	0.079 (0.024, 0.181)
R²	0.95 (0.84, 0.99)	0.85 (0.67, 0.95)

138 Note: the 5th and 95th percentiles are shown instead of a confidence interval.

139

140 Relative standard deviation and half-lives of micropollutants

141

142 Table S6. Relative Standard Deviations (σ/μ) and half-lives (τ) of micropollutants used to generate an
143 empirical spatial Junge relationship. σ/μ were calculated from concentrations measured in the JDS3-CW,
144 imputing measurements below limits of quantification (LOQ). In cases where there were two or more half-
145 lives reported in the literature τ is the geometric mean, and the 95% confidence factor (Cf) is reported as
146 described in reference ¹⁰.

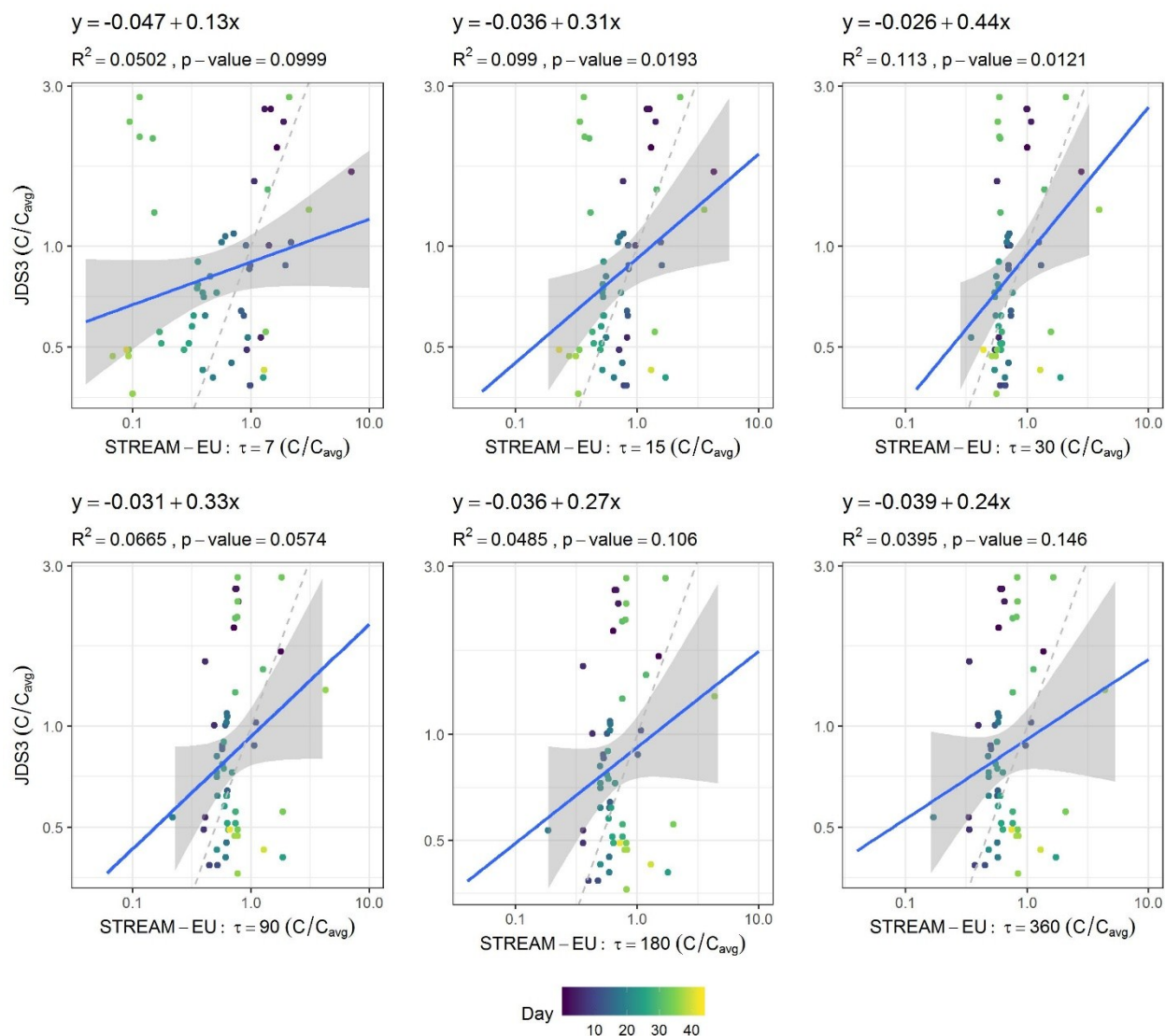
Micropollutant	Empirical σ/μ in the Danube river	Literature τ (days)	Cf in τ	References for τ
amitriptyline	0.8034	5.0	NA	11
caffeine	1.3448	5.7	3.39	11-19
carbamazepine	0.5461	181.8	2.37	12, 14-20
codeine	0.7877	34.6	1.03	12, 21
hydrocodone	0.7811	18.5	NA	12
lidocaine	1.0447	100.6	NA	22
nicotine	0.7773	3.2	NA	12
tramadol	1.0067	81.4	1.84	22, 23
venlafaxine	0.8945	44.7	3.43	22, 24

147

148 Several studies have experimental designs with replicates at different conditions (i.e. comparing different
149 study sites¹²) and report more than one half-life for each compound. If this was the case, then the compiled
150 half-lives were first averaged per compound and per study as the geometric mean. Second, all studies for
151 each compound were pooled in order to compute a broad geometric mean for each micropollutant, this
152 geometric mean is reported in the “literature τ ”.

153 Concentrations of micropollutants in the JDS3 vs predicted concentrations for
 154 hypothetical chemicals in STREAM-EU

Amitriptyline (5 days)

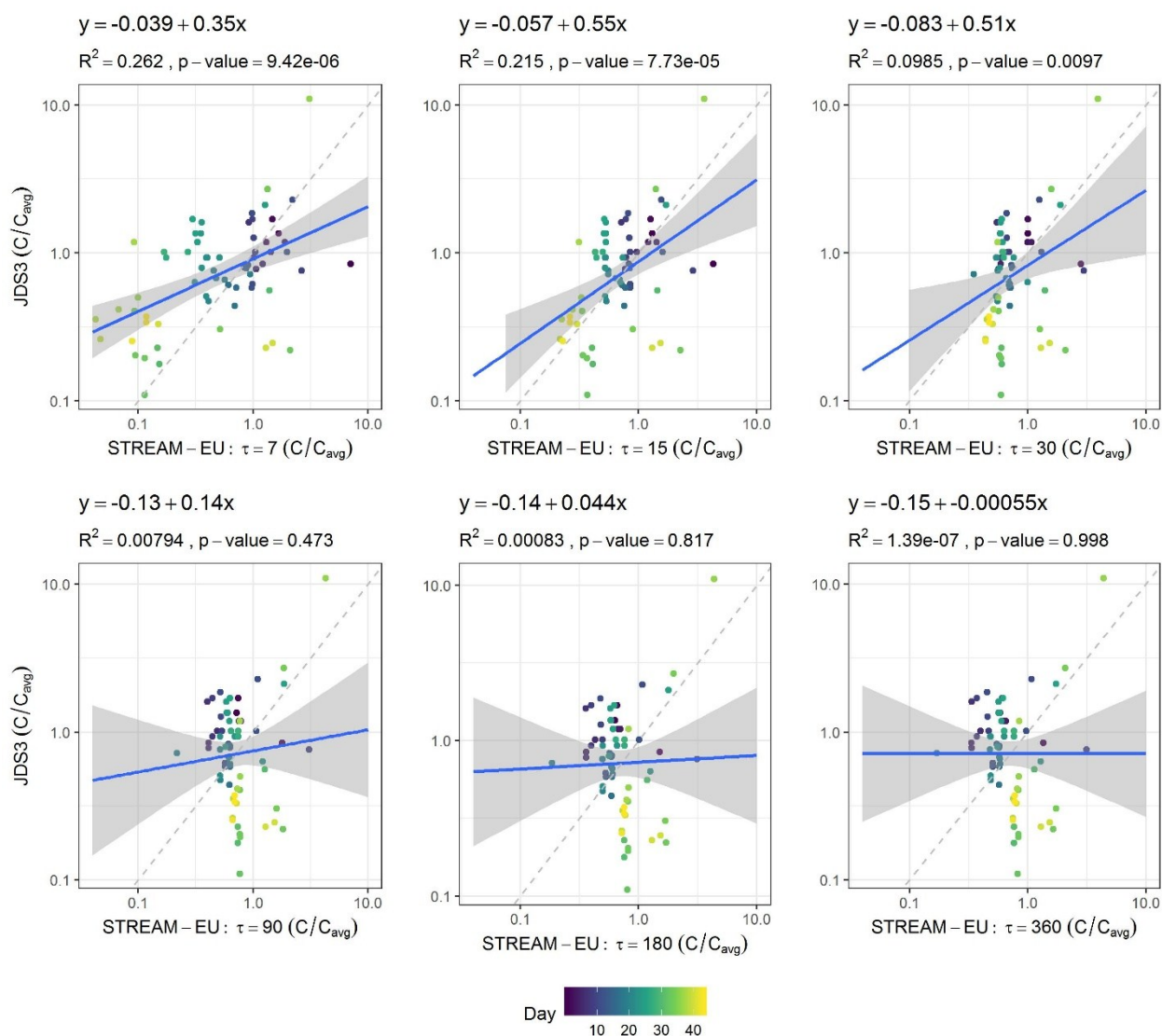


155

156 Figure S5. Normalized concentrations measured for amitriptyline in the JDS3 campaign compared to the
 157 predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation
 158 half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the
 159 JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the
 160 experimental half-life is shown in brackets.

161

Caffeine (5.7 days)

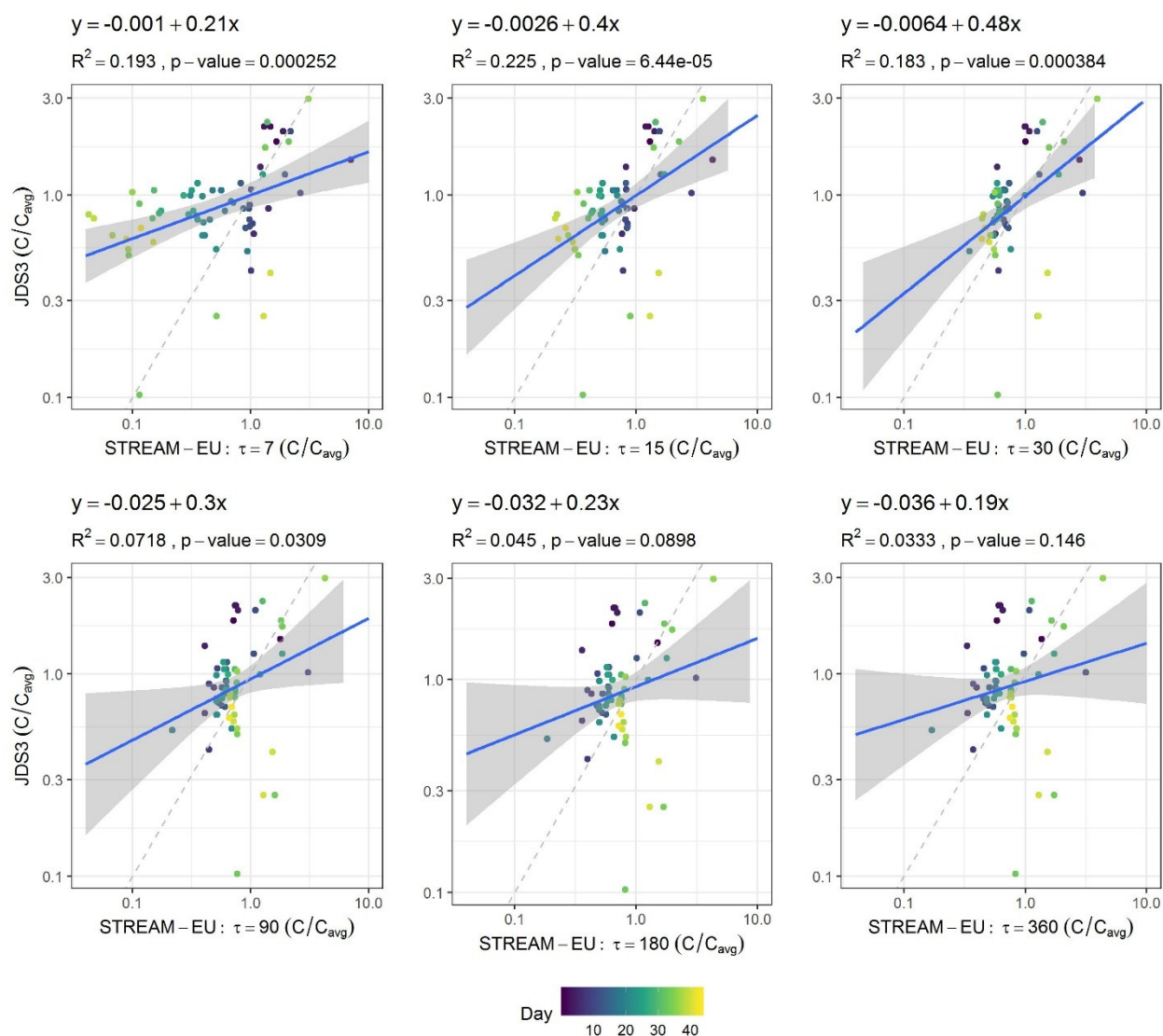


162

163 Figure S6. Normalized concentrations measured for caffeine in the JDS3 campaign compared to the
 164 predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation
 165 half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the
 166 JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the
 167 experimental half-life is shown in brackets.

168

Carbamazepine (182 days)



169

170 Figure S7. Normalized concentrations measured for carbamazepine in the JDS3 campaign compared to the
 171 predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation
 172 half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the
 173 JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the
 174 experimental half-life is shown in brackets.

175

Codeine (35 days)

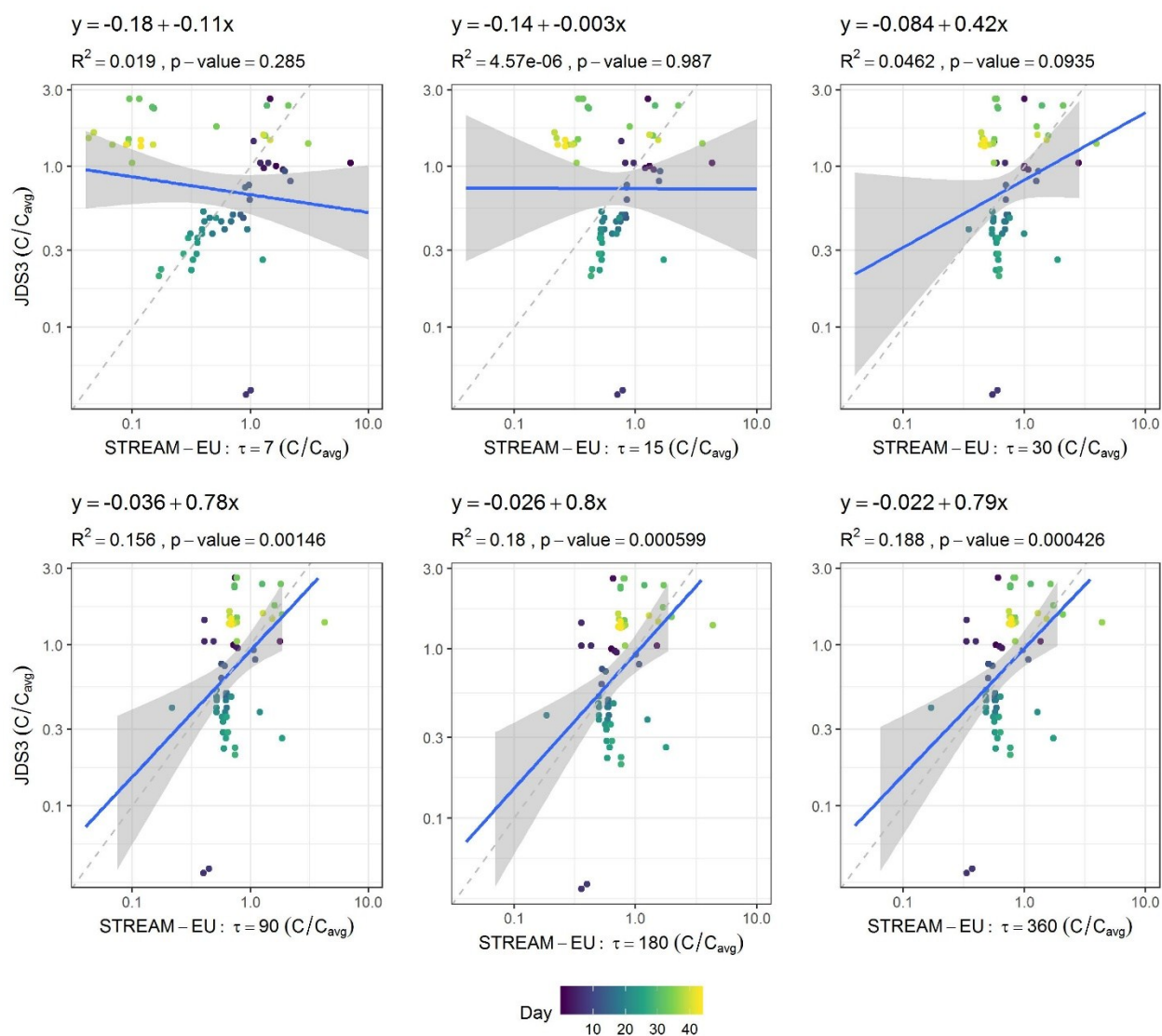


Figure S8. Normalized concentrations measured for codeine in the JDS3 campaign compared to the predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the experimental half-life is shown in brackets.

Hydrocodone (18 days)

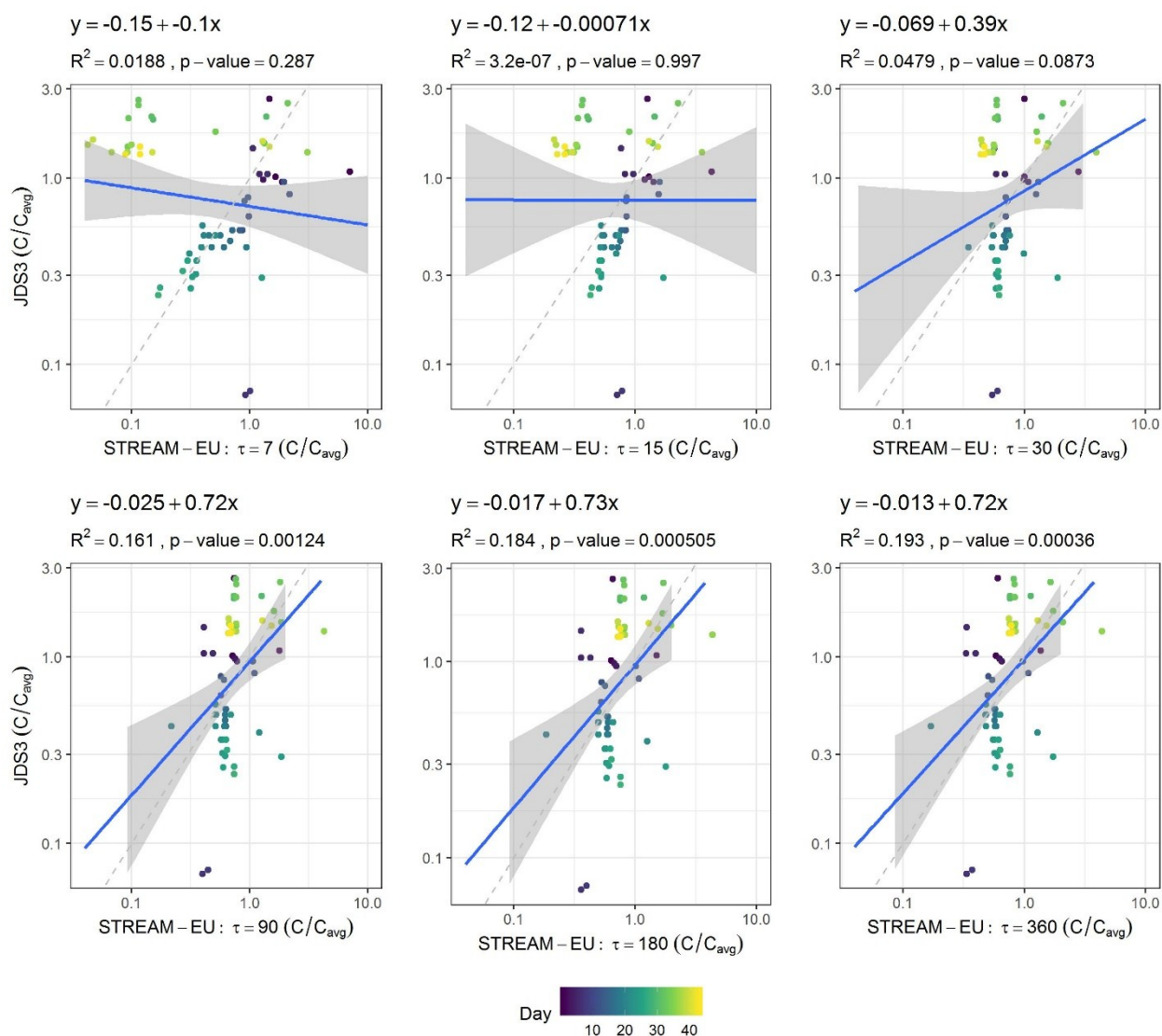
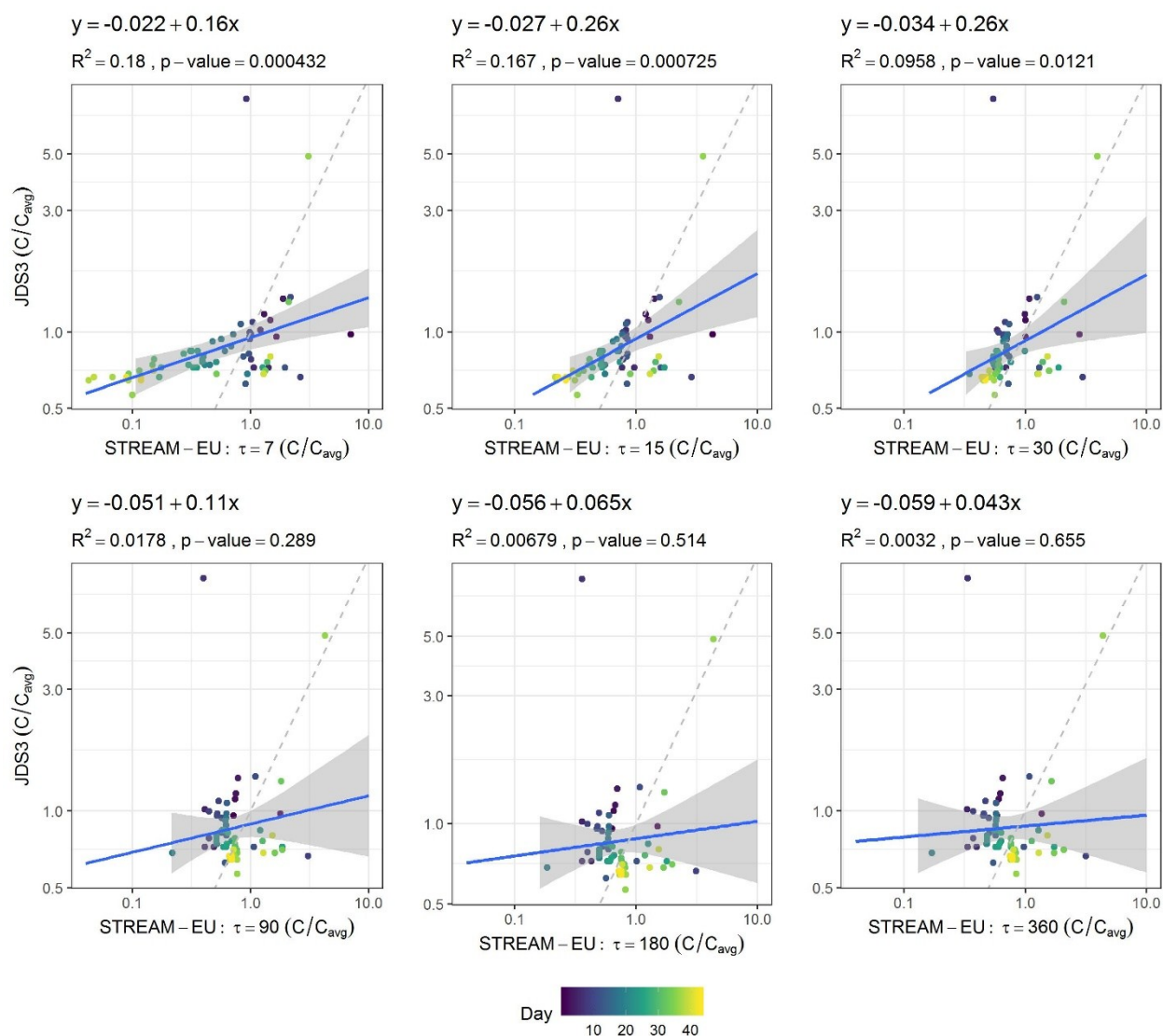


Figure S9. Normalized concentrations measured for hydrocodone in the JDS3 campaign compared to the predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the experimental half-life is shown in brackets.

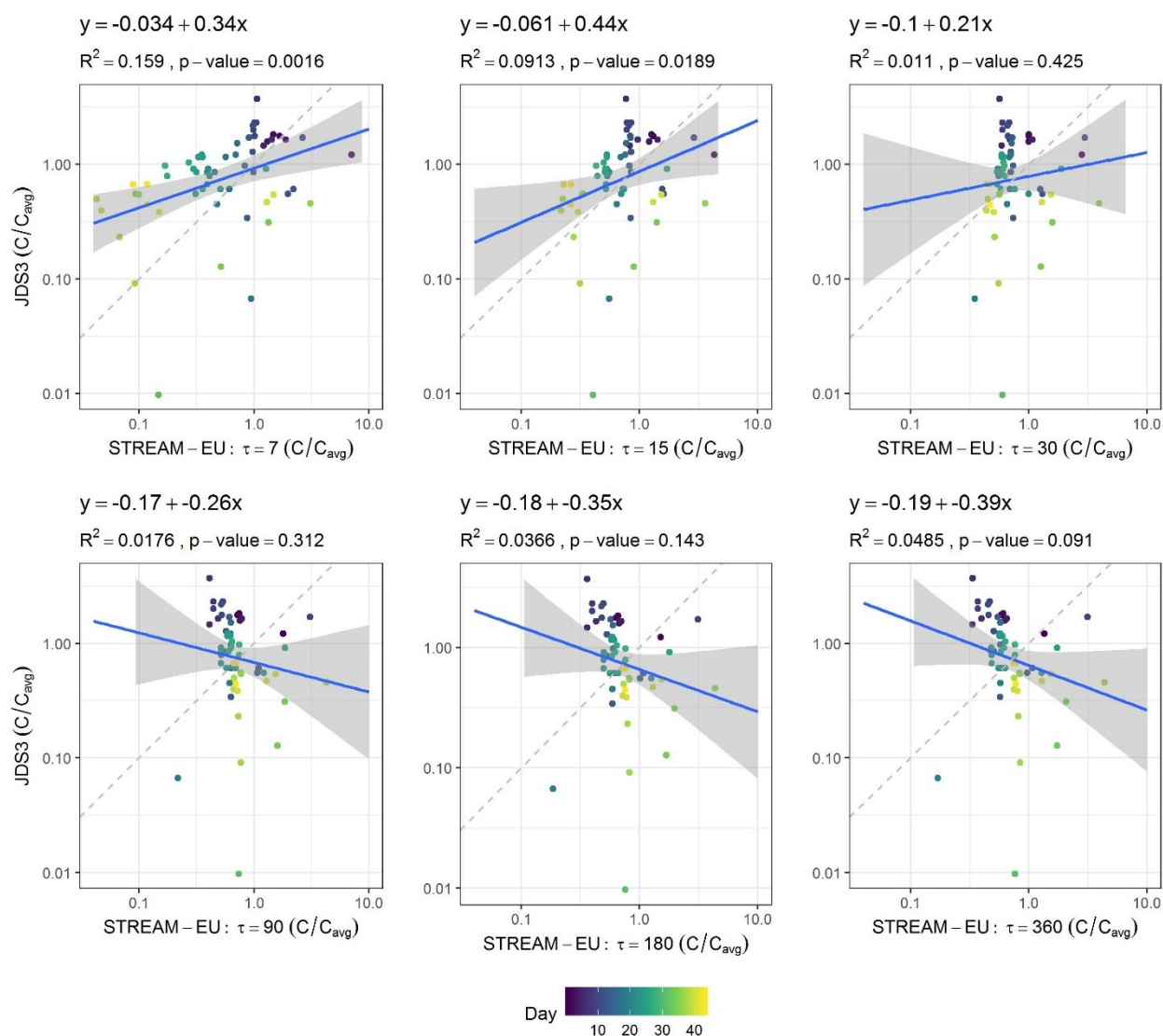
Lidocaine (101 days)



190

191 Figure S10. Normalized concentrations measured for lidocaine in the JDS3 campaign compared to the
 192 predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation
 193 half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the
 194 JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the
 195 experimental half-life is shown in brackets.

Nicotine (3.2 days)



196

197 Figure S11. Normalized concentrations measured for nicotine in the JDS3 campaign compared to the
 198 predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation
 199 half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the
 200 JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the
 201 experimental half-life is shown in brackets.

202

Tramadol (81 days)

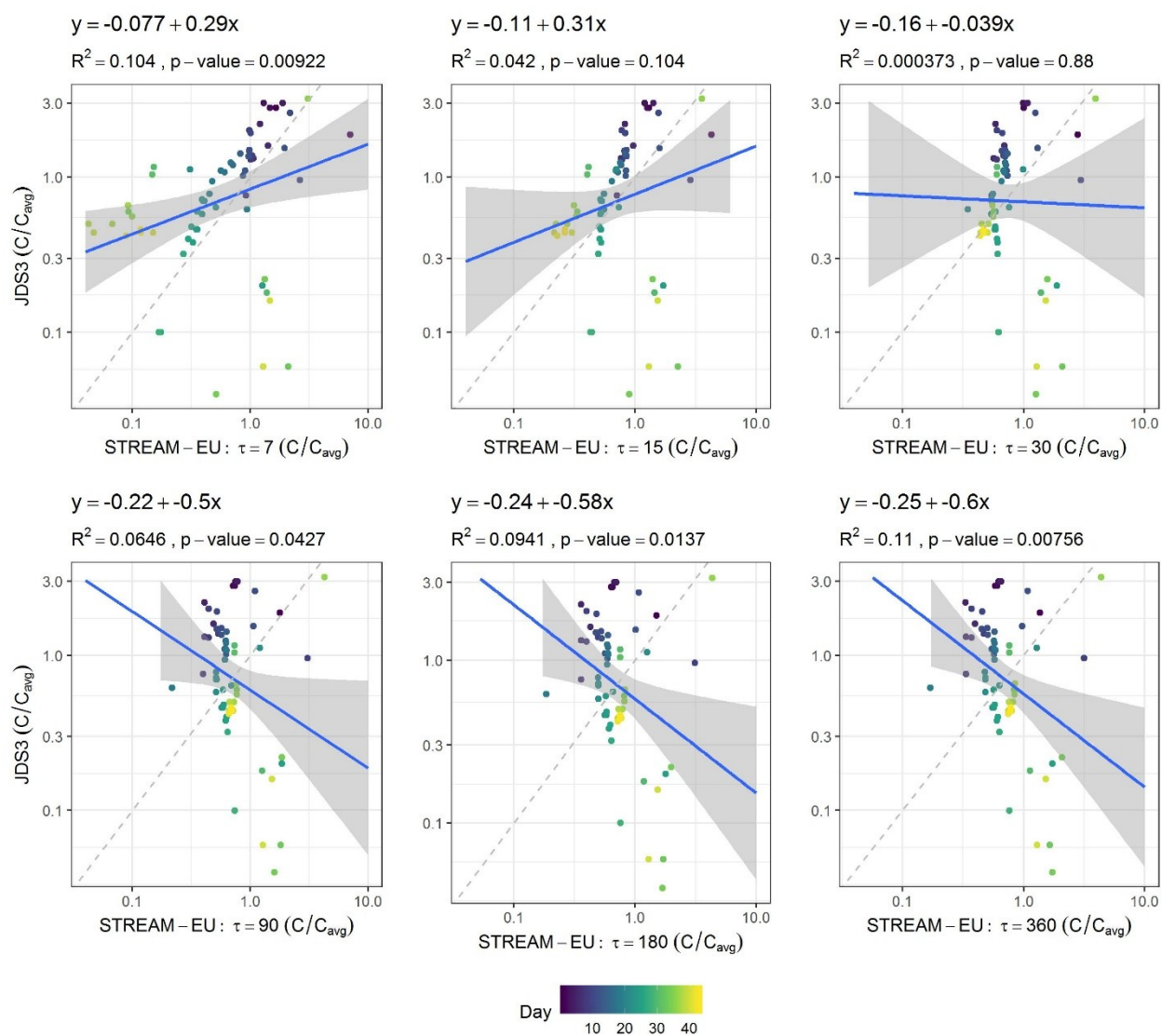
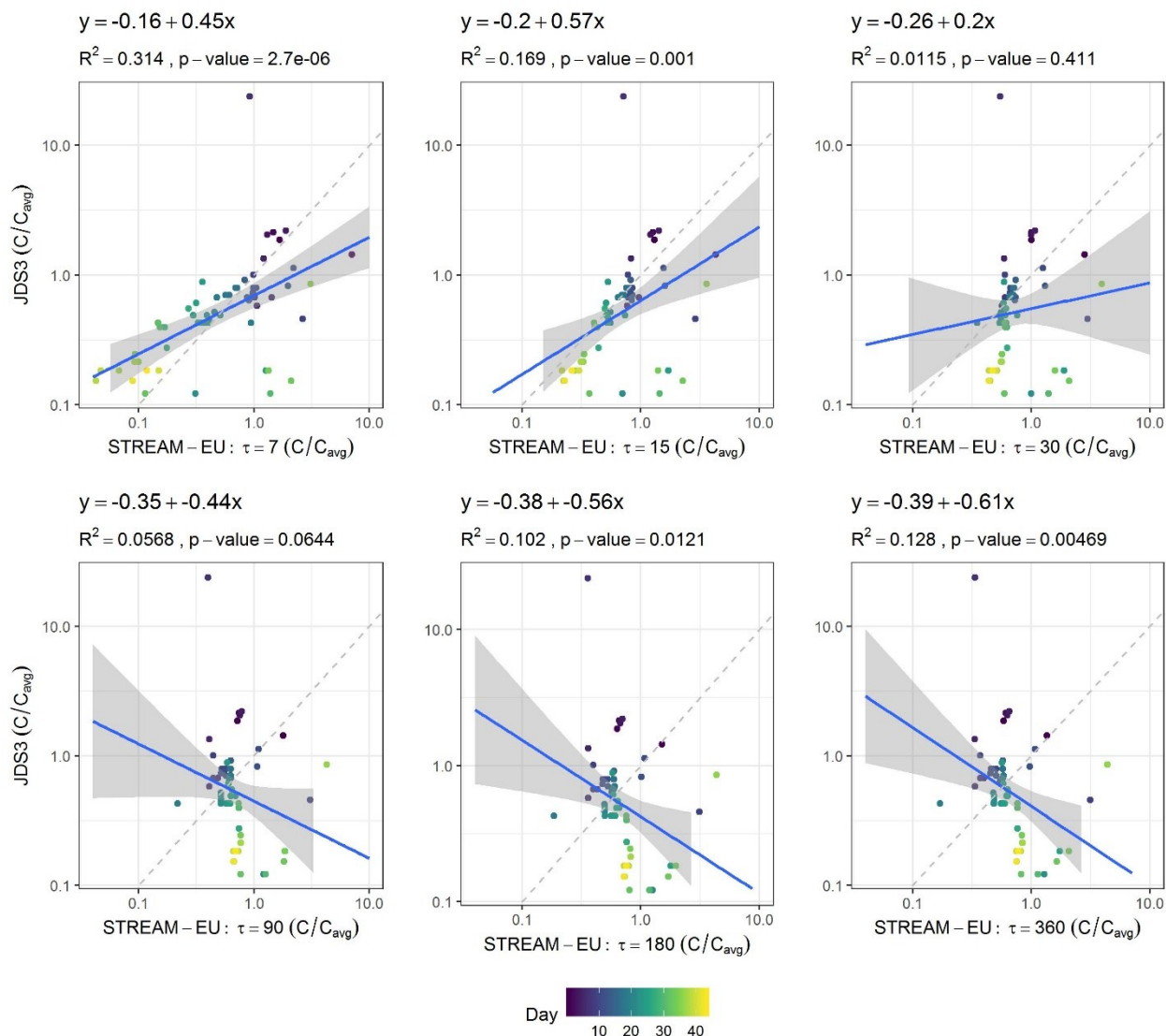


Figure S12. Normalized concentrations measured for tramadol in the JDS3 campaign compared to the predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the experimental half-life is shown in brackets.

209

Venlafaxine (45 days)



210

Figure S13. Normalized concentrations measured for venlafaxine in the JDS3 campaign compared to the predicted normalized concentrations in STREAM-EU of the six hypothetical chemicals with biodegradation half-lives of 7, 15, 30, 90, 180 and 360 days and for the corresponding date and station sampled during the JDS3. The normalization was relative to the average concentration of the compound. In the plot's title, the experimental half-life is shown in brackets.

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217 REFERENCES

- 218 1. C. Lindim, I. T. Cousins and J. vanGils, Estimating emissions of PFOS and PFOA to the Danube
219 River catchment and evaluating them using a catchment-scale chemical transport and fate model,
220 *Environmental Pollution*, 2015, **207**, 97-106.
- 221 2. C. Lindim, J. van Gils and I. T. Cousins, A large-scale model for simulating the fate & transport
222 of organic contaminants in river basins, *Chemosphere*, 2016, **144**, 803-810.
- 223 3. US-EPA., Estimation programs interface (EPI) Suite, *EPA's Office of Pollution Prevention Toxics*
224 *and Syracuse Research Corporation (SRC)*, 2012.
- 225 4. N. Shoari and J. S. Dube, Toward improved analysis of concentration data: Embracing nondetects,
226 *Environmental Toxicology and Chemistry*, 2018, **37**, 643-656.
- 227 5. D. R. Helsel, *Statistics for censored environmental data using Minitab and R*, John Wiley & Sons,
228 Inc., 2nd edn., 2012.
- 229 6. E. G. John, Simplified Curve Fitting using Spreadsheet Add-ins, *International Journal of*
230 *Engineering and Education*, 1998, **14**, 375-380.
- 231 7. S. T. Droge and K. U. Goss, Development and evaluation of a new sorption model for organic
232 cations in soil: contributions from organic matter and clay minerals, *Environmental Science &*
233 *Technology*, 2013, **47**, 14233-14241.
- 234 8. A. Franco, J. Struijs, T. Gouin and O. R. Price, Evolution of the sewage treatment plant model
235 SimpleTreat: applicability domain and data requirements, *Integrated Environmental Assessment*
236 *and Management*, 2013, **9**, 560-568.
- 237 9. ICPDR, International Commission for the Protection of the Danube River: Danube River Basin
238 Water Quality Database, <https://danubis.icpdr.org/user?destination=home>, (accessed 25/11/2015).
- 239 10. M. MacLeod, A. J. Fraser and D. Mackay, Evaluating and expressing the propagation of
240 uncertainty in chemical fate and bioaccumulation models, *Environmental Toxicology and*
241 *Chemistry*, 2002, **21**, 700-709.
- 242 11. Y. Aminot, L. Fuster, P. Pardon, K. Le Menach and H. Budzinski, Suspended solids moderate the
243 degradation and sorption of waste water-derived pharmaceuticals in estuarine waters, *Science of*
244 *The Total Environment*, 2018, **612**, 39-48.
- 245 12. M. J. Benotti and B. J. Brownawell, Microbial degradation of pharmaceuticals in estuarine and
246 coastal seawater, *Environmental Pollution*, 2009, **157**, 994-1002.
- 247 13. S. M. Blunt, J. D. Sackett, M. R. Rosen, M. J. Benotti, R. A. Trenholm, B. J. Vanderford, B. P.
248 Hedlund and D. P. Moser, Association between degradation of pharmaceuticals and endocrine-
249 disrupting compounds and microbial communities along a treated wastewater effluent gradient in
250 Lake Mead, *Science of The Total Environment*, 2018, **622-623**, 1640-1648.
- 251 14. J. L. Conkle, J. Gan and M. A. Anderson, Degradation and sorption of commonly detected PPCPs
252 in wetland sediments under aerobic and anaerobic conditions, *Journal of Soils and Sediments*,
253 2012, **12**, 1164-1173.
- 254 15. J. C. Durán-Álvarez, B. Prado, D. González, Y. Sánchez and B. Jiménez-Cisneros, Environmental
255 fate of naproxen, carbamazepine and triclosan in wastewater, surface water and wastewater
256 irrigated soil — Results of laboratory scale experiments, *Science of The Total Environment*, 2015,
257 **538**, 350-362.
- 258 16. M. W. Lam, C. J. Young, R. A. Brain, D. J. Johnson, M. A. Hanson, C. J. Wilson, S. M. Richards,
259 K. R. Solomon and S. A. Mabury, Aquatic persistence of eight pharmaceuticals in a microcosm
260 study, *Environmental Toxicology and Chemistry*, 2004, **23**, 1431-1440.
- 261 17. D. Löffler, J. Römbke, M. Meller and T. A. Ternes, Environmental fate of pharmaceuticals in
262 water/sediment systems, *Environmental Science & Technology*, 2005, **39**, 5209-5218.
- 263 18. C. Tixier, H. P. Singer, S. Oellers and S. R. Müller, Occurrence and fate of carbamazepine,
264 clofibric acid, diclofenac, ibuprofen, ketoprofen, and naproxen in surface waters, *Environmental*
265 *Science & Technology*, 2003, **37**, 1061-1068.

- 266 19. H. Yamamoto, Y. Nakamura, S. Moriguchi, Y. Nakamura, Y. Honda, I. Tamura, Y. Hirata, A.
267 Hayashi and J. Sekizawa, Persistence and partitioning of eight selected pharmaceuticals in the
268 aquatic environment: laboratory photolysis, biodegradation, and sorption experiments, *Water*
269 *Research*, 2009, **43**, 351-362.
- 270 20. H. Zou, M. Radke, A. Kierkegaard, M. MacLeod and M. S. McLachlan, Using chemical
271 benchmarking to determine the persistence of chemicals in a Swedish lake, *Environmental Science*
272 *& Technology*, 2015, **49**, 1646-1653.
- 273 21. A. Y.-C. Lin, Y.-C. Lin and W.-N. Lee, Prevalence and sunlight photolysis of controlled and
274 chemotherapeutic drugs in aqueous environments, *Environmental Pollution*, 2014, **187**, 170-181.
- 275 22. P. C. Rua-Gomez and W. Puttmann, Degradation of lidocaine, tramadol, venlafaxine and the
276 metabolites O-desmethyltramadol and O-desmethylvenlafaxine in surface waters, *Chemosphere*,
277 2013, **90**, 1952-1959.
- 278 23. Z. Li, A. Sobek and M. Radke, Flume experiments to investigate the environmental fate of
279 pharmaceuticals and their transformation products in streams, *Environmental Science &*
280 *Technology*, 2015, **49**, 6009-6017.
- 281 24. M. Rožman, V. Acuña and M. Petrović, Effects of chronic pollution and water flow intermittency
282 on stream biofilms biodegradation capacity, *Environmental Pollution*, 2018, **233**, 1131-1137.

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