Supporting Information

Experimental and geochemical simulation of nickel carbonate mineral precipitation by carbonate-laden ureolytic fungal culture supernatants

Qianwei Li*a,b,#, Daoqing Liu^{c,#}, Chunmao Chen*b, Zhiguo Shaoa, Huazhen Wangb,

Jicheng Liu^b, Qiangbin Zhang^b, Geoffrey Michael Gadd^d

Methods

Quantification of polysaccharides produced by fungi

The concentration of polysaccharide in fungal supernatant before and after mixture with $NiCl_2$ solution was measured using the Phenol-Sulfuric acid method, following the basic protocol of Dubois et al.¹.

Table S1. Proportion of polysaccharide removed after mixture of fungal growth supernatant and 50 mM NiCl₂.

Mixture ratio	Removal of polysaccharide (%)		
(V _s : V _m)	1% inoculation	10% inoculation	
5:0.5	5.4	8.0	
7.5:0.5	6.8	7.9	
10:0.5	7.4	8.2	

Table S2. Elemental composition of fungal growth supernatant and

minerals precipitated.

Sample	Elemental Composition (%)			
	С	Ν	Ni	0
Fungal supernatant (S)	52.97	11.31	-	35.72
S + Ni ²⁺	47.88	16.85	4.51	30.76

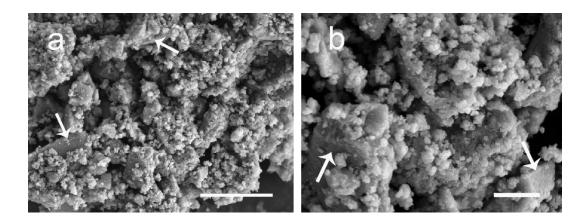


Fig. S1. Scanning electron microscopy and elemental analysis of minerals precipitated by the mixture of NiCl₂ and $(NH_4)_2CO_3$.

Minerals were formed by mixture of 0.5 ml 50 mM NiCl₂ and 10 ml AP1 medium containing (a) 33 mM, (b) 330 mM (NH₄)₂CO₃. Scale bars: (a) = 3 μ m, (b) = 600 nm.

Reference

1. M. Dubois, K. A. Gilles, J. K. Hamilton, P. t. Rebers and F. Smith, *Anal Chem*, 1956, **28**, 350-356.