

1 **Electronic Supplemental Information (ESI): Removal of estrogenic compounds via iron**
2 **electrocoagulation: removal mechanisms and impact of water quality**

3 Emily K. Maher¹, Kassidy N. O'Malley¹, Joe Heffron¹, Jingwan Huo², Yin Wang², Brooke K.
4 Mayer¹, Patrick J. McNamara^{1*}

5 ¹Department of Civil, Construction and Environmental Engineering, Marquette University,
6 Milwaukee, WI 53211, USA.

7 ²Department of Civil and Environmental Engineering, University of Wisconsin – Milwaukee,
8 Milwaukee, WI 53210, USA.

9 *Corresponding author

10 Patrick McNamara*: (414)288-2188, patrick.mcnamara@mu.edu

11 Department of Civil, Construction and Environmental Engineering
12 Marquette University
13 1637 W Wisconsin Ave
14 Milwaukee, WI 53233 USA.

15

16 **Table of Contents**

17 **S1: Physical-chemical properties of estrogenic compounds**

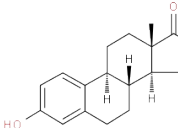
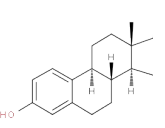
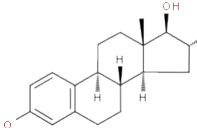
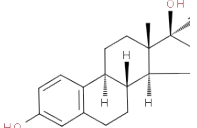
18 **S2: X-Ray Diffraction Patterns**

19 **S3: Post-hoc Statistical Analysis P-values**

20 **S3: Oxidation**

21 **S1: Physical-chemical properties of estrogenic compounds**

22 **Table S1.** Physical-chemical properties of estrogenic steroids

Property	Estrone (E1)	17β- Estradiol (E2)	Estriol (E3)	17α- Ethinylestrad iol (EE2)
Molecular Formula	C ₁₈ H ₂₂ O ₂	C ₁₈ H ₂₄ O ₂	C ₁₈ H ₂₄ O ₃	C ₂₀ H ₂₄ O ₂
CAS No.	53-16-7	50-28-2	50-27-1	57-63-6
Structure¹				
Type	Natural	Natural	Natural	Synthetic
Molecular Weight (g mol⁻¹)²	270.4	272.4	288.4	296.403
Log K_{ow}²	3.43	3.94	2.81	4.15
pKa³	10.5-10.7	10.3-10.8	10.3-10.8	10.4 ⁴
Water Solubility²	13	13	13	4.8
Henry's Law Constant (atm m³ mol⁻¹)⁵	3.8x10 ⁻¹⁰	3.64x10 ⁻¹¹	1.33x10 ⁻¹²	7.94x10 ⁻¹²
Vapor Pressure²	2.3x10 ⁻¹⁰	2.3x10 ⁻¹⁰	6.7x10 ⁻¹⁵	4.5x10 ⁻¹¹

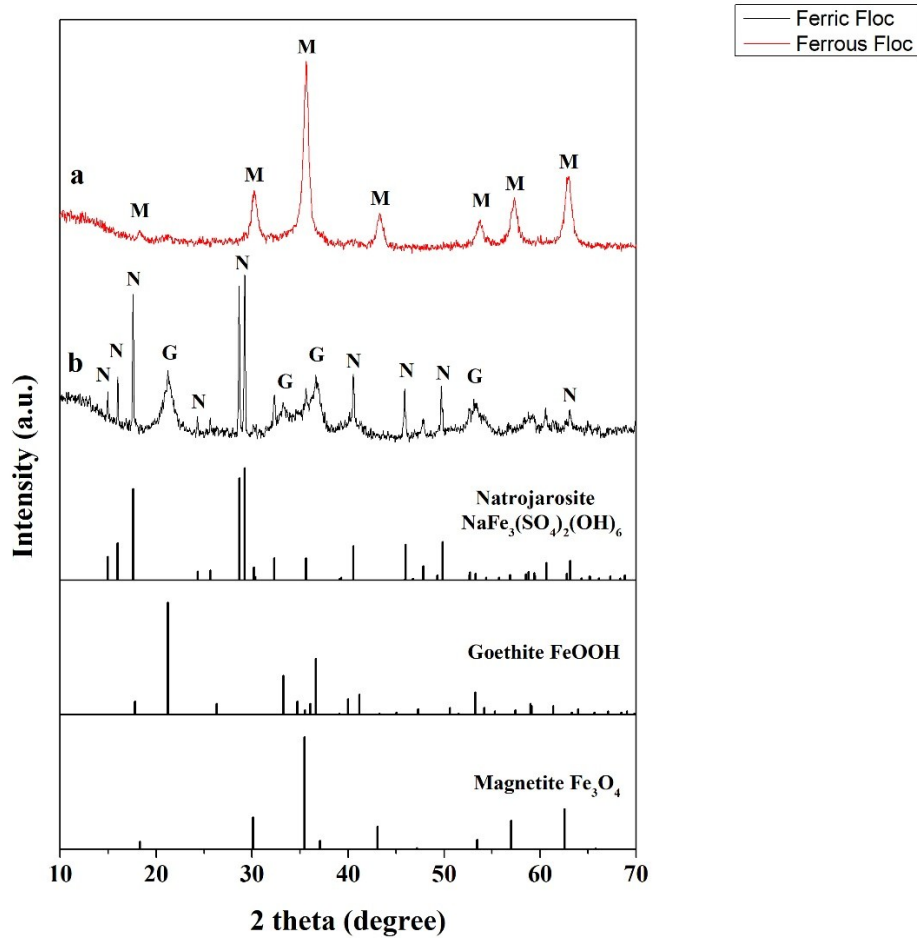
23

24

25

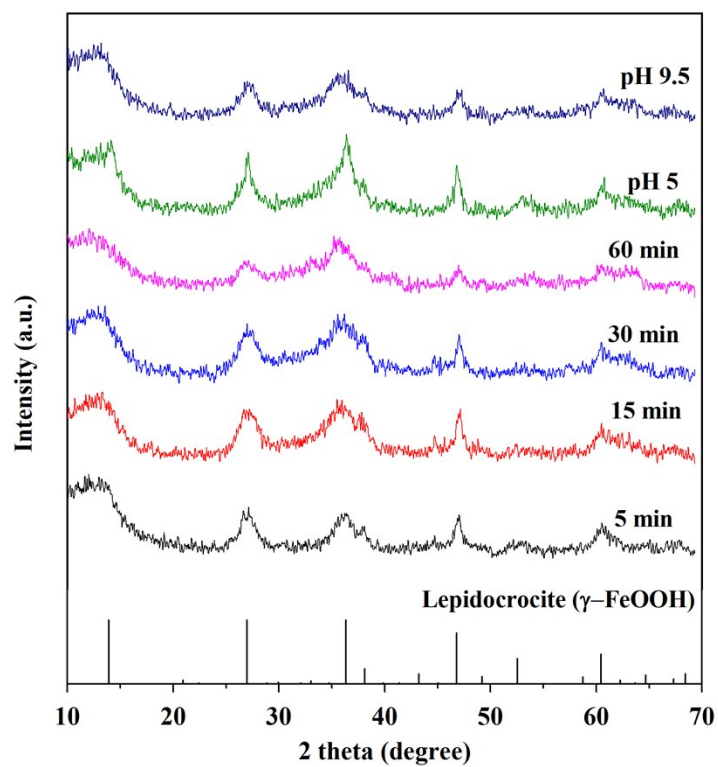
26 S2: X-Ray Diffraction Patterns

27 X-ray diffraction (XRD) patterns of the ferric and ferrous flocs were obtained using a Bruker D8
28 Discover A25 diffractometer with a copper $K\alpha$ radiation to determine their crystalline phases.
29 The XRD scans were recorded from 2θ of $10^\circ - 70^\circ$ using a step size of 0.02° and a count time of
30 0.4 s per step.
31



32

33 Figure S1. XRD patterns of iron flocs produced from conventional coagulation jar tests using (a)
34 FeSO_4 and (b) $\text{Fe}_2(\text{SO}_4)_3$. The reference patterns for goethite (G, 29-0713), magnetite (M, 65-
35 3107), and natrojarosite (N, 51-1567) are included for comparison.
36



37
38 Figure S2. XRD patterns of iron floc produced at different pH values and electrocoagulation
39 times. The reference pattern for lepidocrocite (01-0136) is included for comparison.

40 **S3: Post-hoc Statistical Analysis P-values**

41 **Table S2.** Post-hoc p-values with Fisher's LSD for Multiple Comparisons at Variable pH

Estradiol (E1)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
pH 5.5 vs. pH 7.0	-0.3738	-0.4897 to -0.2579	Yes	0.0002
pH 5.5 vs. pH 9.5	-0.7919	-0.9078 to -0.676	Yes	<0.0001
pH 7.0 vs. pH 9.5	-0.4181	-0.534 to -0.3022	Yes	0.0001
17β- Estradiol (E2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
pH 5.5 vs. pH 7.0	-0.571	-0.6584 to -0.4835	Yes	<0.0001
pH 5.5 vs. pH 9.5	-0.9274	-1.015 to -0.8399	Yes	<0.0001
pH 7.0 vs. pH 9.5	-0.3564	-0.4439 to -0.2689	Yes	<0.0001
Estriol (E3)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
pH 5.5 vs. pH 7.0	-0.424	-0.5426 to -0.3053	Yes	0.0001
pH 5.5 vs. pH 9.5	-0.7667	-0.8854 to -0.648	Yes	<0.0001
pH 7.0 vs. pH 9.5	-0.3427	-0.4614 to -0.2241	Yes	0.0004
17α-Ethynylestradiol (EE2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
pH 5.5 vs. pH 7.0	-0.1295	-0.1772 to -0.08182	Yes	0.0006
pH 5.5 vs. pH 9.5	0.2091	0.1614 to 0.2567	Yes	<0.0001
pH 7.0 vs. pH 9.5	0.3386	0.2909 to 0.3862	Yes	<0.0001

42

43 **Table S3.** Post-hoc p-values with Fisher's LSD for Multiple Comparisons at Variable Turbidities

Estradiol (E1)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
0.0 NTU vs. 2.0 NTU	0.1251	-0.02172 to 0.272	No	0.085
0.0 NTU vs. 60.0 NTU	-0.05162	-0.1985 to 0.09522	No	0.441
2.0 NTU vs. 60.0 NTU	-0.1768	-0.3127 to -0.0408	Yes	0.0171
17β- Estradiol (E2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
0.0 NTU vs. 2.0 NTU	0.321	0.1744 to 0.4676	Yes	0.001
0.0 NTU vs. 60.0 NTU	0.09973	-0.04685 to 0.2463	No	0.1553
2.0 NTU vs. 60.0 NTU	-0.2213	-0.357 to -0.08554	Yes	0.0055
17α-Ethynylestradiol (EE2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
0.0 NTU vs. 2.0 NTU	0.2854	0.1568 to 0.4141	Yes	0.0009
0.0 NTU vs. 60.0 NTU	0.2728	0.1441 to 0.4015	Yes	0.0012
2.0 NTU vs. 60.0 NTU	-0.01262	-0.1318 to 0.1065	No	0.8131

44

45 **Table S4.** Post-hoc p-values with Fisher's LSD for Multiple Comparisons at Variable DOC

46 Concentrations

17β- Estradiol (E2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
0.0 vs. 0.5 mg L ⁻¹	-0.117	-0.4257 to 0.1918	No	0.4078
0.0 vs. 15.0 mg L ⁻¹	0.3051	-0.00363 to 0.6139	No	0.0522
0.5 vs. 15.0 mg L ⁻¹	0.4221	0.1362 to 0.708	Yes	0.0093

47

48 **Table S5.** Post-hoc p-values for Fisher's LSD for Multiple Comparisons with EC vs Conventional
 49 Coagulation

Estradiol (E1)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC with Iron vs. CC - Ferric	0.9123	0.6593 to 1.165	Yes	<0.0001
EC with Iron vs. CC - Ferrous	0.6917	0.4387 to 0.9447	Yes	0.0001
CC - Ferric vs. CC - Ferrous	-0.2205	-0.4735 to 0.03248	No	0.0808
17 β - Estradiol (E2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC with Iron vs. CC - Ferric	0.9149	0.6642 to 1.166	Yes	<0.0001
EC with Iron vs. CC - Ferrous	0.6829	0.4322 to 0.9336	Yes	0.0001
CC - Ferric vs. CC - Ferrous	-0.232	-0.4827 to 0.01872	No	0.0662
Estriol (E3)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC with Iron vs. CC - Ferric	0.7493	0.5818 to 0.9167	Yes	<0.0001
EC with Iron vs. CC - Ferrous	0.6917	0.5242 to 0.8592	Yes	<0.0001
CC - Ferric vs. CC - Ferrous	-0.05757	-0.225 to 0.1099	No	0.4614
17 α -Ethinylestradiol (EE2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC with Iron vs. CC - Ferric	0.7493	0.5818 to 0.9167	Yes	<0.0001
EC with Iron vs. CC - Ferrous	0.6917	0.5242 to 0.8592	Yes	<0.0001
CC - Ferric vs. CC - Ferrous	-0.05757	-0.225 to 0.1099	No	0.4614

50

51 **Table S6.** Post-hoc p-values for Fisher's LSD for Multiple Comparisons for EC and Oxidant Scavenging
 52 Tests

Estradiol (E1)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC Only vs. EC with t-BuOH	0.1824	0.01885 to 0.346	Yes	0.0342
EC Only vs. EC with MeOH	0.3384	0.1749 to 0.502	Yes	0.0023
EC with t-BuOH vs. EC with MeOH	0.156	-0.00752 to 0.3196	No	0.0583
17 β - Estradiol (E2)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC Only vs. EC with t-BuOH	0.38	0.2492 to 0.5107	Yes	0.0004
EC Only vs. EC with MeOH	0.5553	0.4246 to 0.6861	Yes	<0.0001
EC with t-BuOH vs. EC with MeOH	0.1754	0.04461 to 0.3061	Yes	0.0168
Estriol (E3)				
Fisher's LSD	Mean Diff.	95.00% CI of diff.	Significant?	P Value
EC Only vs. EC with t-BuOH	0.225	0.04136 to 0.4086	Yes	0.0241
EC Only vs. EC with MeOH	0.487	0.3034 to 0.6706	Yes	0.0006
EC with t-BuOH vs. EC with MeOH	0.262	0.07836 to 0.4456	Yes	0.013

*EE2 was removed below the detection limit.

53

54

55 **S4: Ultrasonic Assisted Extraction**

56 **Table S7.**

		Estradiol (E1)	17 β - Estradiol (E2)	Estriol (E3)	17 α - Ethinylestradiol (EE2)
LOD ($\mu\text{g L}^{-1}$)		27	3	3	13
LOQ ($\mu\text{g L}^{-1}$)		107	14	14	51
Initial Concentration ($\mu\text{g L}^{-1}$)		460	440	469	452
Final Concentration ($\mu\text{g L}^{-1}$)		N.D.	<14	<14	<51
Percent Removal (%)		100	98	98	97
Spiked Sample* ($\mu\text{g L}^{-1}$)	1	130	80	87	65
	2	121	87	94	69
	3	138	96	101	77
Unspiked Sample ($\mu\text{g L}^{-1}$)	1	N.D.	<14	<14	N.D.
	2	N.D.	<14	<14	N.D.
	3	N.D.	<14	<14	N.D.
Average Percent Recovery (%)		137	116	126	108
Standard Deviation		19	9	8	7

*Spiked concentration were 68 $\mu\text{g L}^{-1}$ for E1, 69 $\mu\text{g L}^{-1}$ for E2 and E3, and 65 $\mu\text{g L}^{-1}$ for EE2.

N.D. = Non-detect

57

58

59

60

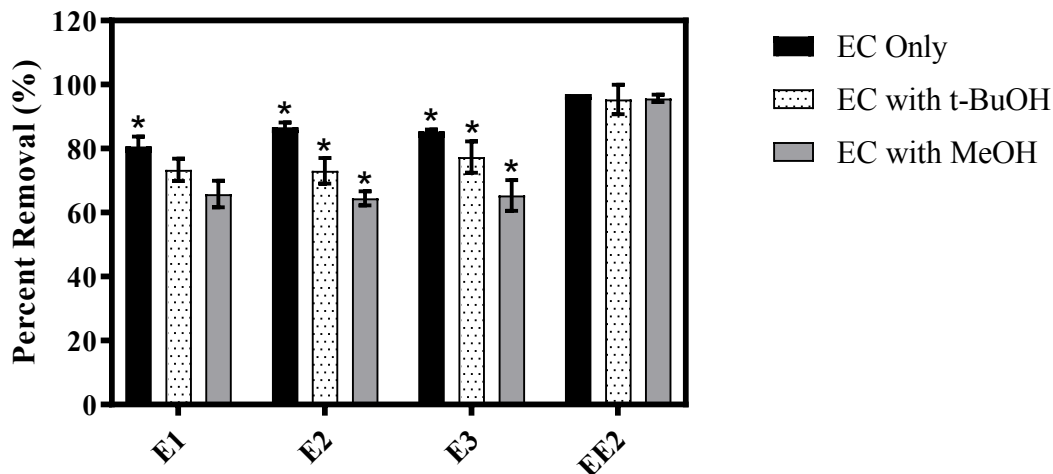
61

62

63

64

65 **S5: Oxidation**



66

67 Figure S3. Removal of E1, E2, E3, and EE2 for an EC test with iron electrodes (n=3) compared
 68 to a test with oxidant scavengers of 0.25 M t-BuOH (n=3) or 0.95 M MeOH (n=3). E1, E2, and
 69 E3 were significantly different for all tests (p-values < 0.007). There was no significant
 70 difference for EE2 (p-value =0.77). The error bars represent the standard error of the mean.
 71 Asterisks indicate significant difference between all other conditions within a single estrogenic
 72 compound. Bars indicate difference between specific conditions.

73

74 **Table S8.** EE2 dosed EC reactor results for transformation to E1, E2, and E3

Test	Time (min.)	E1		E2		E3		EE2	
		Area	S/N* Ratio	Area	S/N Ratio	Area	S/N Ratio	Area	S/N Ratio
1	0	N.P.	N.P.	N.P.	N.P.	N.P.	N.P.	290053	140.19
	5	4675	3.6	N.P.	N.P.	N.P.	N.P.	223636	123.24
	120	12701	3.98	N.P.	N.P.	N.P.	N.P.	N.P.	N.P.
2	0	N.P.	N.P.	N.P.	N.P.	N.P.	N.P.	286970	137.65
	5	9614	3.72	N.P.	N.P.	N.P.	N.P.	246019	119.68
	120	12719	4.89	N.P.	N.P.	N.P.	N.P.	N.P.	N.P.

N.P. = No peak

S/N Ratio = signal-to-noise ratio

*S/N Ratio \geq 3.0 sample is detectable; S/N Ratio \geq 10.0 sample is quantifiable.

Data indicate that E1 was formed from oxidation of EE2

75

76 **Table S9.** E2 dosed EC reactor results for transformation to E1, EE2, and E3

Test	Time (min.)	E1		E2		E3		EE2	
		Area	S/N* Ratio	Area	S/N Ratio	Area	S/N Ratio	Area	S/N Ratio
1	0	N.P.	N.P.	47574	118.9	N.P.	N.P.	N.P.	N.P.
	120	N.P.	N.P.	19042	25.62	1651	3.45	N.P.	N.P.
2	0	N.P.	N.P.	133863	318.1	N.P.	N.P.	N.P.	N.P.
	120	N.P.	N.P.	26517	5.1	8750	8.25	N.P.	N.P.

N.P. = No peak

S/N Ratio = signal-to-noise ratio

*S/N Ratio ≥ 3.0 sample is detectable; S/N Ratio ≥ 10.0 sample is quantifiable.

Data indicate that E3 was formed from oxidation of E2

77

78

79 **References**

- 80 1 ChemSpider | Search and share chemistry, <http://www.chemspider.com/>, (accessed 13
81 December 2018).
- 82 2 G. G. Ying, R. S. Kookana and Y. J. Ru, *Environ. Int.*, 2002, 28, 545–551.
- 83 3 T. A. Hanselman, D. A. Graetz and A. C. Wilkie, *Environ. Sci. Technol.*, 2003, 37, 5471–
84 5478.
- 85 4 M. I. Toral, F. Nacaratte, F. Nova-Ramírez and R. Otipka, *J. Chil. Chem. Soc.*, 2013, 58,
86 1779–1784.
- 87 5 C. Zhang, Y. Li, C. Wang and W. Cai, , DOI:10.1080/10643389.2015.1061881.
88