

## Supporting Information

# Layer-by-layer membrane modification allows scandium recovery by nanofiltration

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## ***Elemental analysis***

The concentrations of Fe and Mg were analyzed in triplicate using Inductively Coupled Plasma Optical Emission Spectroscopy (ICP-OES) (Spectroblue SOP, Spectro Analytical Instruments, Kleve, Germany) using standard conditions (power of 1,400 W, coolant flow: 13 L/min, auxiliary flow: 1 L/min, nebulizer flow: 0.75 L/min). All other elements were measured using triple quadrupole Inductively Coupled Plasma Mass Spectrometry (qqq-ICP-MS) on an Agilent 8800 series machine (Agilent, Basel, Switzerland) using general-purpose operational settings. Quantification was performed via multi-element standards (Sigma- Aldrich). Rh was used as internal standard to account for matrix effects. Chloride concentrations were analyzed using ion chromatography (IC). The IC consisted of a Dionex 2100 system, equipped with an online eluent generator, a self-regenerating suppressor, a guard and analytical column (AG17-C and AS17-C, 2 mm) (all Dionex, Olten, Switzerland). Chloride was separated from other ions using a hydroxide gradient and quantified by conductivity detection.

Tab 1. Elemental concentration in the acid waste solution

Element	Concentration [mg/L]	Removal at pH 1.5 precipitation [ %]
Sc	76 - 90	~20
Ni	22 - 84	0
U	20 - 23	~40
Th	102 - 133	~75
Ti	3 906 - 4 228	100
Na	447 - 1 182	-
Al	6 124 - 7 722	0
Ca	667 - 839	-
Fe	39 374 - 45 160	0
K	401 - 631	-
Cl	153 150 - 180 000	-

Tab 2. Elemental concentration in the retentate after pH adjustment, dilution and filtration (60% permeate recovery)

Element	Concentration [mg/L]
Sc	23
U	5
Th	93
Fe	8 070

### *Filtration set-up*

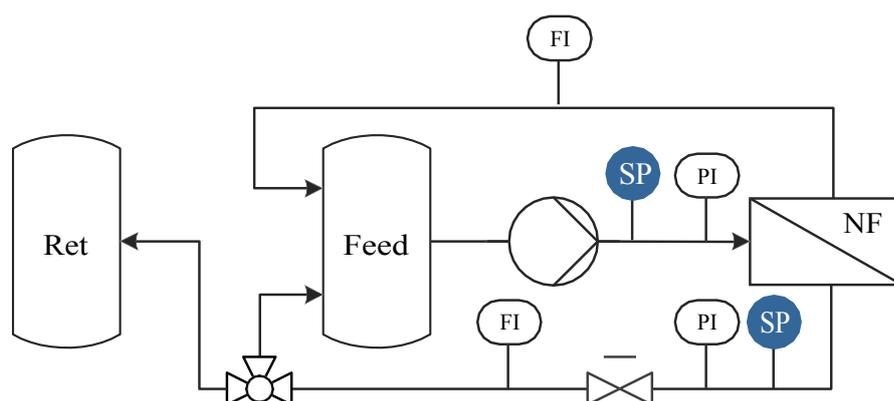


Figure 1: Flow chart of the custom made filtration unit used for the experiments with flowmeters (FI), pressure meters (PI) and nanofiltration membrane (NF)

### *Stability towards higher HCl concentrations*

Magnesium retention was used as an indicator of membrane acid stability. For this, 0.5 mM Mg was dissolved in deionized water. Mg retention was determined in cross-flow mode (5 bars of transmembrane pressure, TMP) using 0.5 mM Mg solution. The flow was 160 mL/min, resulting in a cross-flow velocity of 2.65 m/s and a Reynolds number  $> 2,300$  (thus a turbulent flow). HCl (32 wt%; Roth, Switzerland) was diluted in deionized water to 1 M, 2 M and 3 M, respectively, and Mg (as  $\text{Mg}_2\text{SO}_4$  heptahydrate,  $\geq 99\%$ , Sigma-Aldrich) added to a final concentration of 0.5 mM.

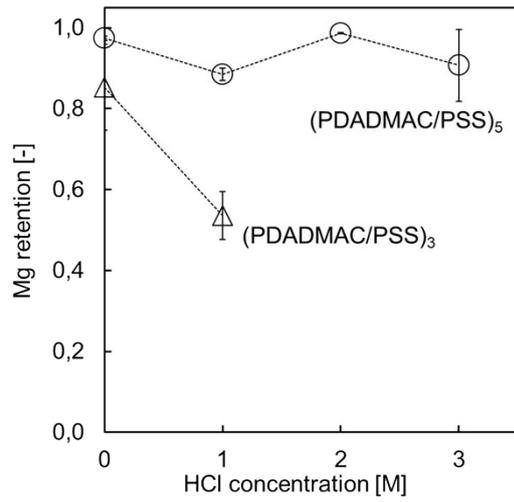


Figure 2: Acid stability in terms of Mg retention as a function of HCl concentration using 3 or 5 bi-layers systems. Coating was done at a higher (1 M NaCl) ionic strength .