

## Supporting Information

### Cobalt-catalyzed Synthesis of *N*-containing Heterocycles via Cyclization of Ortho-substituted Anilines with CO<sub>2</sub>/H<sub>2</sub>

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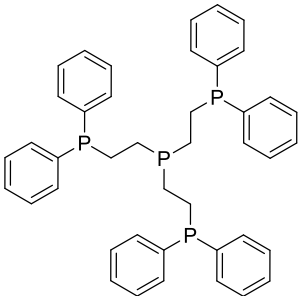
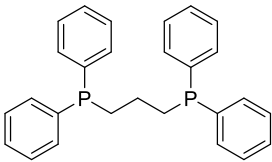
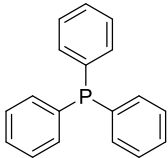
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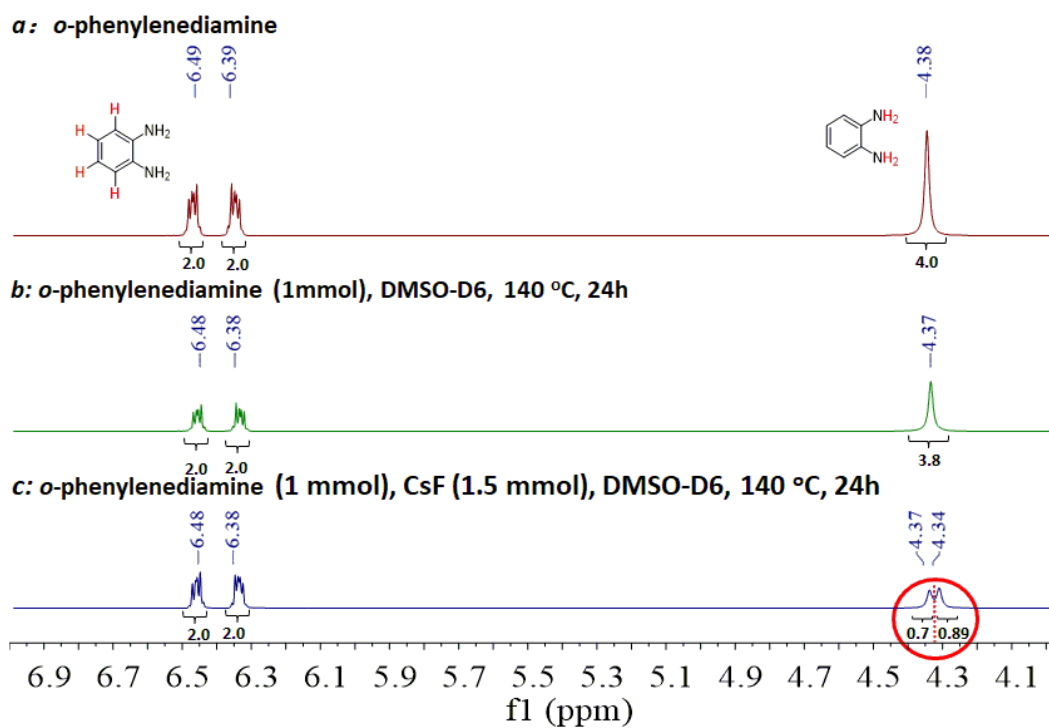
## Effects of ligands on the cyclization reaction

Table S1 Effects of ligands on the cyclization reaction <sup>a</sup>

| Entry | Ligand                                                                             | Yield (%) |
|-------|------------------------------------------------------------------------------------|-----------|
| 1     |   | 2         |
| 2     |   | <1        |
| 3     |  | 3         |

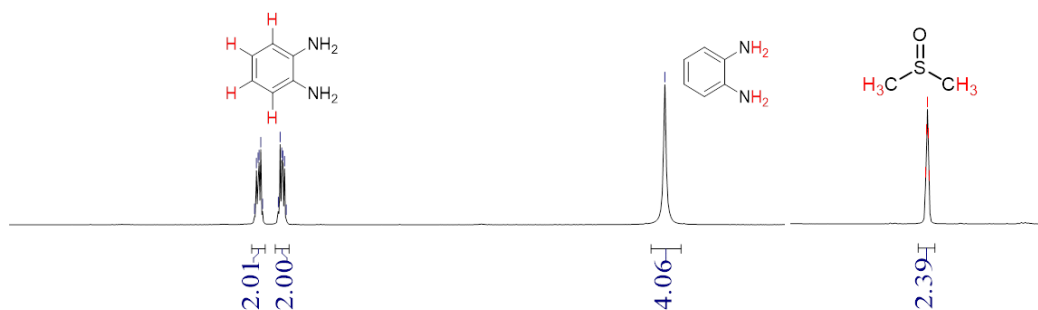
Reaction conditions: <sup>a</sup> *o*-phenylenediamine (1.0 mmol), CoF<sub>2</sub> (10 mol %), ligand (5 mmol %), CsF (1.5 equiv.), EtOH (3 mL), CO<sub>2</sub> pressure of 3 MPa, total pressure of 6 MPa, 140 °C, 24 h. The yield was determined by <sup>1</sup>H NMR, calibrated using 1,1,2,2-tetrachloroethane as the internal standard.

CsF activates ortho-substituted anilines via hydrogen bond interaction

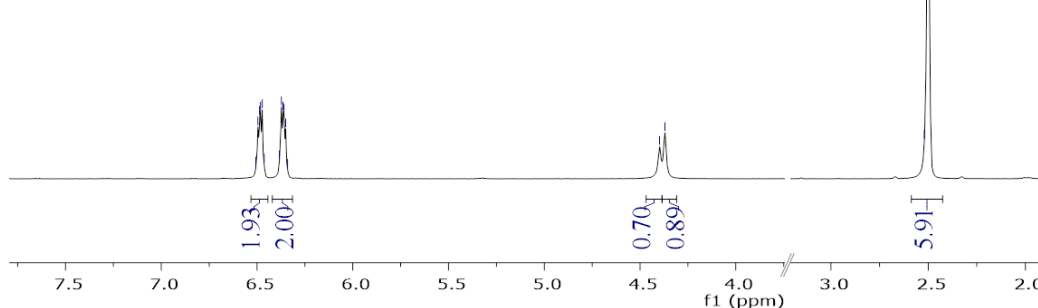


**Figure S1.**  $^1\text{H}$  NMR spectra: a) *o*-phenylenediamine; b) *o*-phenylenediamine (1.0 mmol), DMSO- $\text{d}_6$  (2.0 mL), 140 °C, 24h; c) *o*-phenylenediamine (1.0 mmol), CsF (1.5 equiv.), DMSO- $\text{d}_6$  (2.0 mL), 140 °C, 24h.

**a:** *o*-phenylenediamine (1mmol), DMSO-d<sub>6</sub>, 140 °C, 24h

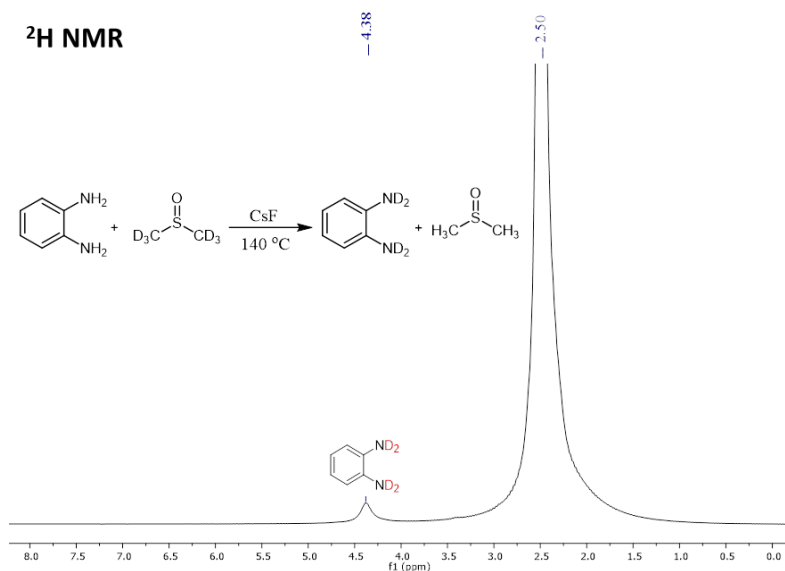


**b:** *o*-phenylenediamine (1 mmol), CsF (1.5 equiv.), DMSO-d<sub>6</sub>, 140 °C, 24h

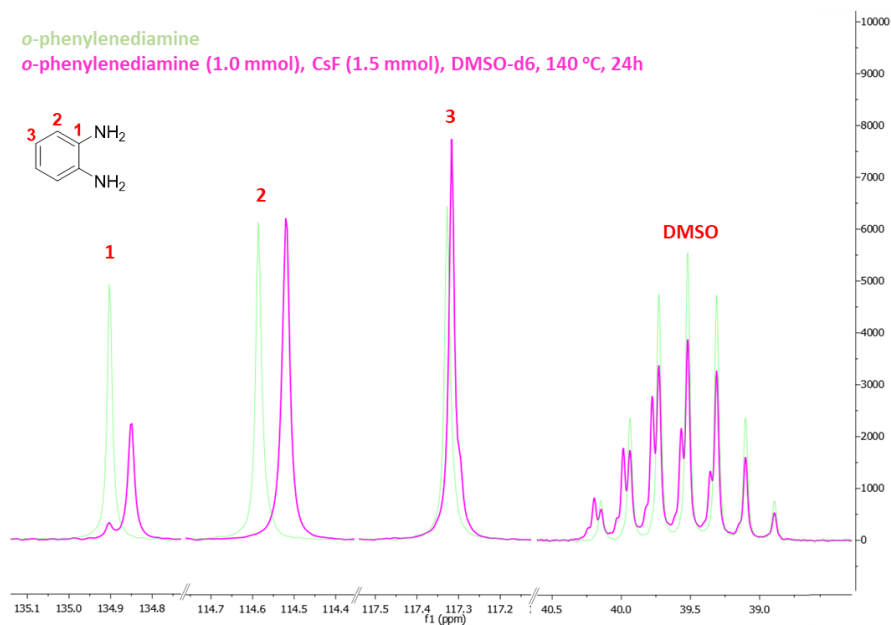


**Figure S2** <sup>1</sup>H NMR spectra: a) *o*-phenylenediamine (1.0 mmol), DMSO-d<sub>6</sub> (2.0 mL), 140 °C, 24h; b) *o*-phenylenediamine (1.0 mmol), CsF (1.5 equiv), DMSO-d<sub>6</sub> (2.0 mL), 140 °C, 24h.

Analysis: Counting the H number of in *o*-phenylenediamine from the above <sup>1</sup>H NMR spectra **a** and **b** (Figure S2), it was found that the H number of -NH<sub>2</sub> in *o*-phenylenediamine was decreasing from 4.06 to 1.59 (relative to the hydrogen atoms of benzene ring), while the H number of C-H in DMSO increased from 2.39 to 5.91 (the H number of DMSO was calibrated using the aromatic hydrogen of *o*-phenylenediamine as internal standard). At the same time, the -ND<sub>2</sub> (4.38 ppm) was observed in the <sup>2</sup>H NMR spectrum of the mixture of *o*-phenylenediamine and CsF (Figure S3). It is the evidence of H/D exchange between the -NH<sub>2</sub> of *o*-phenylenediamine with -CD<sub>3</sub> of DMSO-d<sub>6</sub> in the presence of CsF. It suggests that CsF can activate -NH<sub>2</sub> of *o*-phenylenediamine.

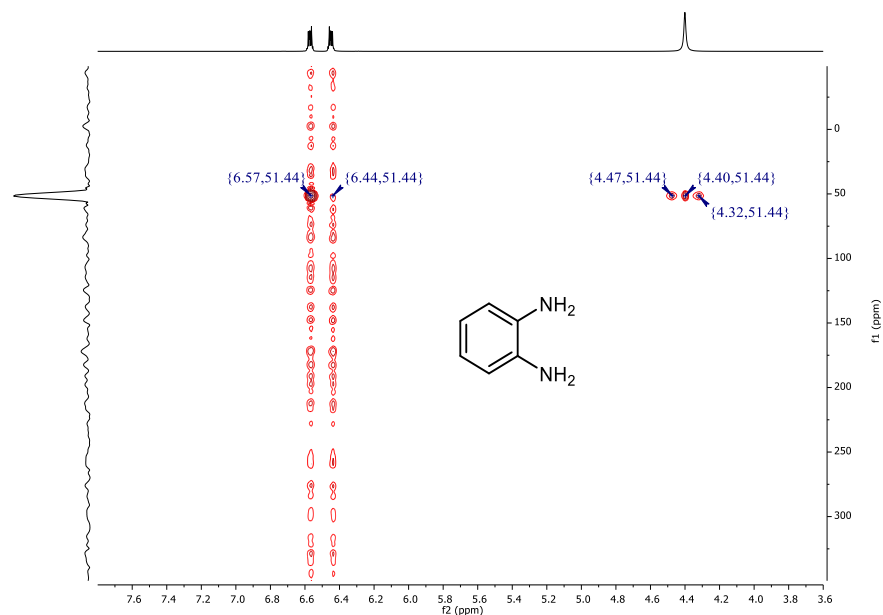


**Figure S3**  $^2\text{H}$  NMR spectrum of the mixture: *o*-phenylenediamine (1.0 mmol), CsF (1.5 equiv), DMSO- $d_6$  (2.0 mL), 140  $^\circ\text{C}$ , 24h.

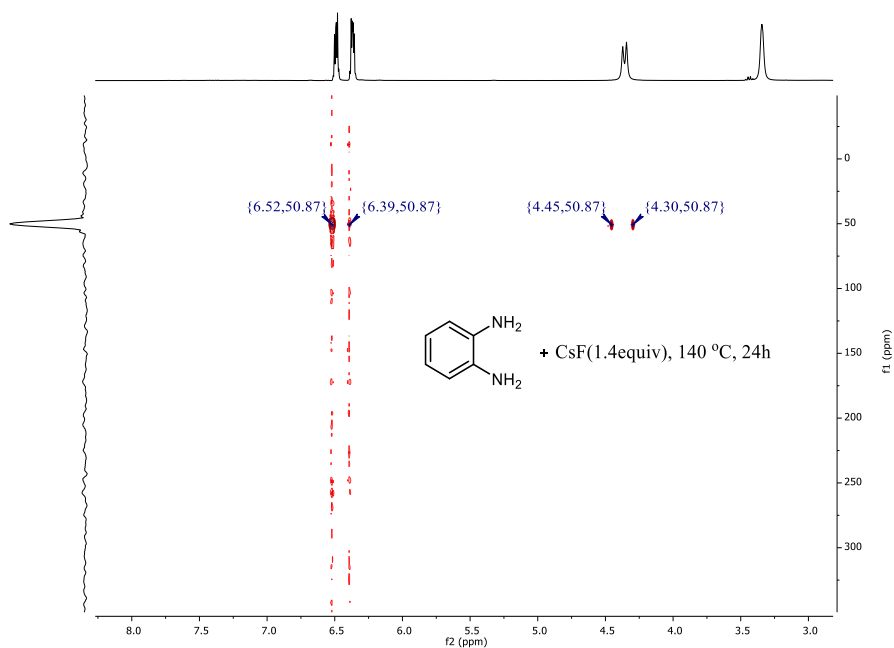


**Figure S4**  $^{13}\text{C}$  NMR spectra of the mixture: *o*-phenylenediamine (1.0 mmol), DMSO- $d_6$  (2.0 mL), 140  $^\circ\text{C}$ , 24h, (Green line); *o*-phenylenediamine (1.0 mmol) with CsF (1.5 equiv) in DMSO- $d_6$  (2.0 mL), 140  $^\circ\text{C}$ , 24h (Red line).

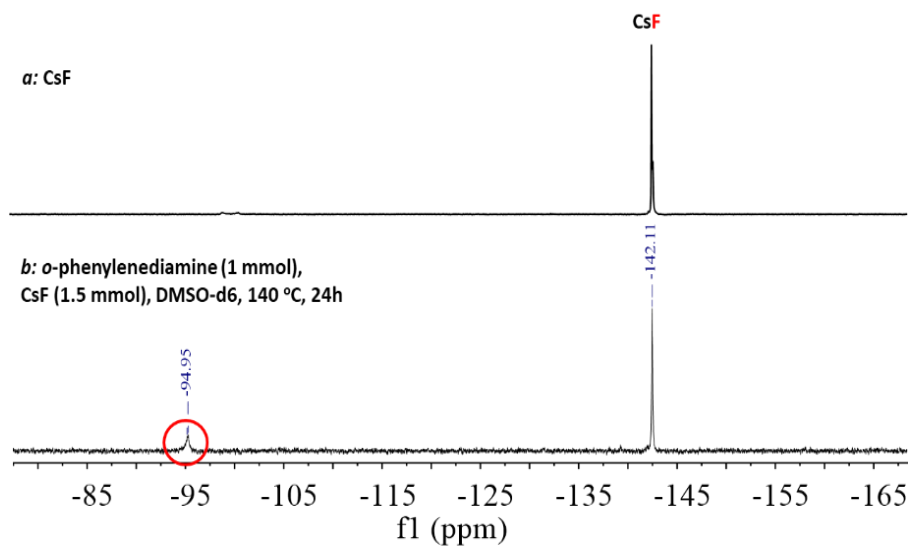
Analysis: Comparing the  $^{13}\text{C}$  signals of the mixtures of *o*-phenylenediamine, DMSO- $d_6$  with or without CsF (Figure S4), it was found that the chemical shift of *o*-phenylenediamine shifted upfield from 134.90 (C1), 117.33 (C2), 114.59 ppm (C3) to 134.85 (C1), 117.32 (C2), 114.52 ppm (C3). Carbon atoms of *o*-phenylenediamine become more electron-rich when mixed with CsF, suggesting that the carbon atom of *o*-phenylenediamine was activated by CsF to some extent.



**Figure S5**  $^1\text{H}$ - $^{15}\text{N}$  HMBC spectrum of *o*-phenylenediamine (DMSO- $\text{d}_6$ , 500 MHz) recorded with an evolution delay  $\tau$  of 62.5 ms for the optimal detection of long-range  $J_{\text{H-N}}$  of 8 Hz.  $^{15}\text{N}$  NMR (51 MHz, DMSO):  $\delta$  51.44, 51.44, 51.44, 51.44, 51.44.  $^1\text{H}$  NMR (500 MHz, DMSO):  $\delta$  6.57, 6.44, 4.47, 4.40, 4.32.

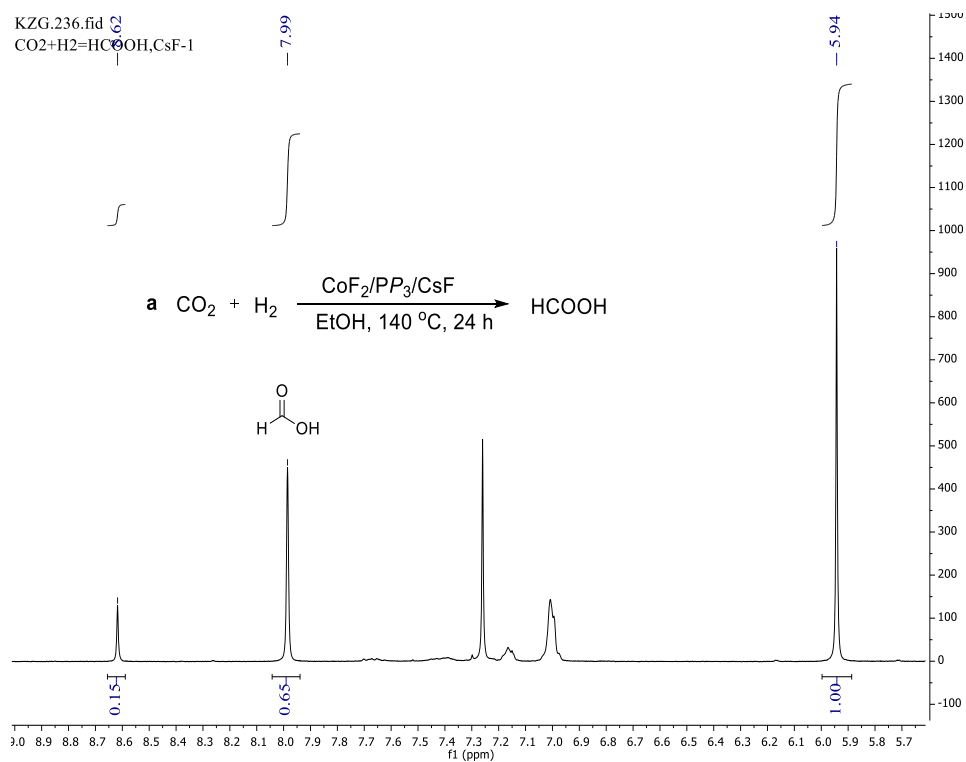


**Figure S6**  $^1\text{H}$ - $^{15}\text{N}$  HMBC spectrum of the mixture: *o*-phenylenediamine (1.0 mmol), CsF (1.5 equiv), DMSO- $\text{d}_6$  (2.0 mL), 140 °C, 24h (DMSO- $\text{d}_6$ , 500 MHz) recorded with an evolution delay  $\tau$  of 62.5 ms for the optimal detection of long-range  $J_{\text{H-N}}$  of 8 Hz.  $^{15}\text{N}$  NMR (51 MHz, DMSO):  $\delta$  50.87, 50.87, 50.87, 50.87.  $^1\text{H}$  NMR (500 MHz, DMSO):  $\delta$  6.52, 6.39, 4.45, 4.30.

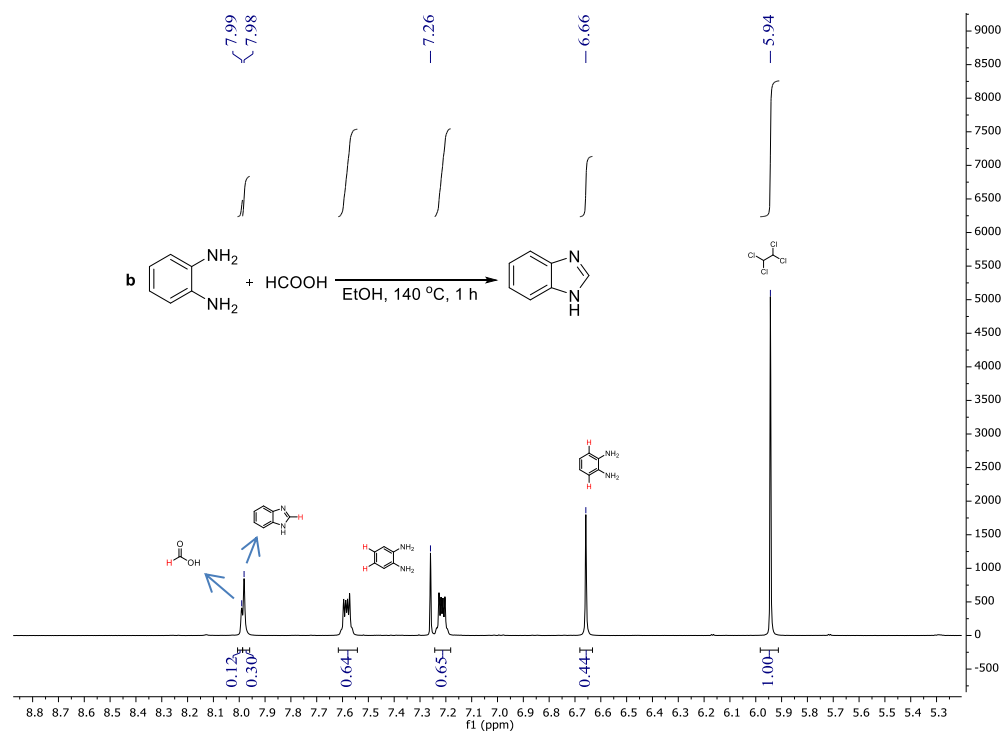


**Figure S7.**  $^{19}\text{F}$  NMR spectra: a) CsF; b) **1a** (1.0 mmol), CsF (1.5 equiv.), DMSO-d<sub>6</sub> (2.0 mL), 140 °C, 24h.

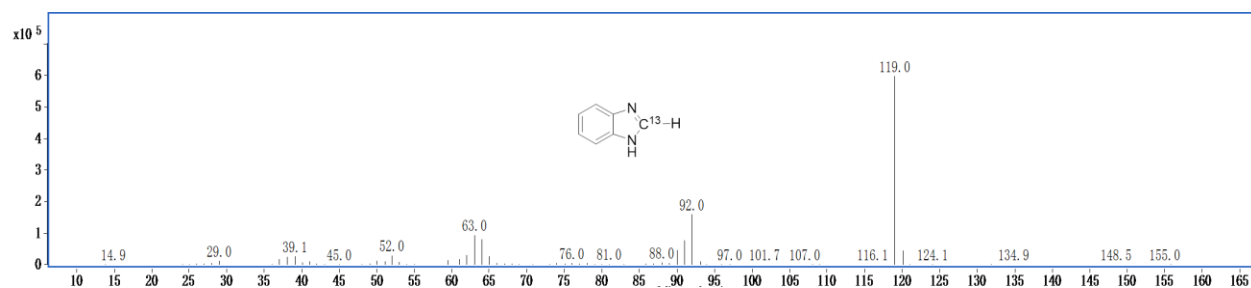
## Control experiments



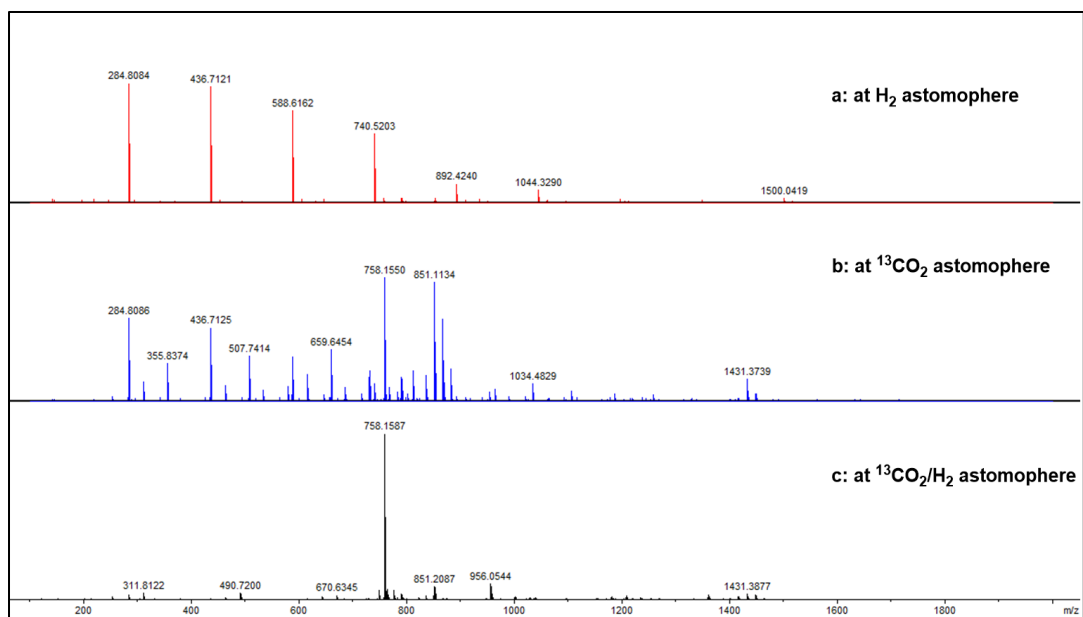
**Figure S8** <sup>1</sup>H NMR spectrum of the products of Scheme 2a. The yield was determined by <sup>1</sup>H NMR, calibrated using 1,1,2,2-tetrachloroethane as the internal standard.



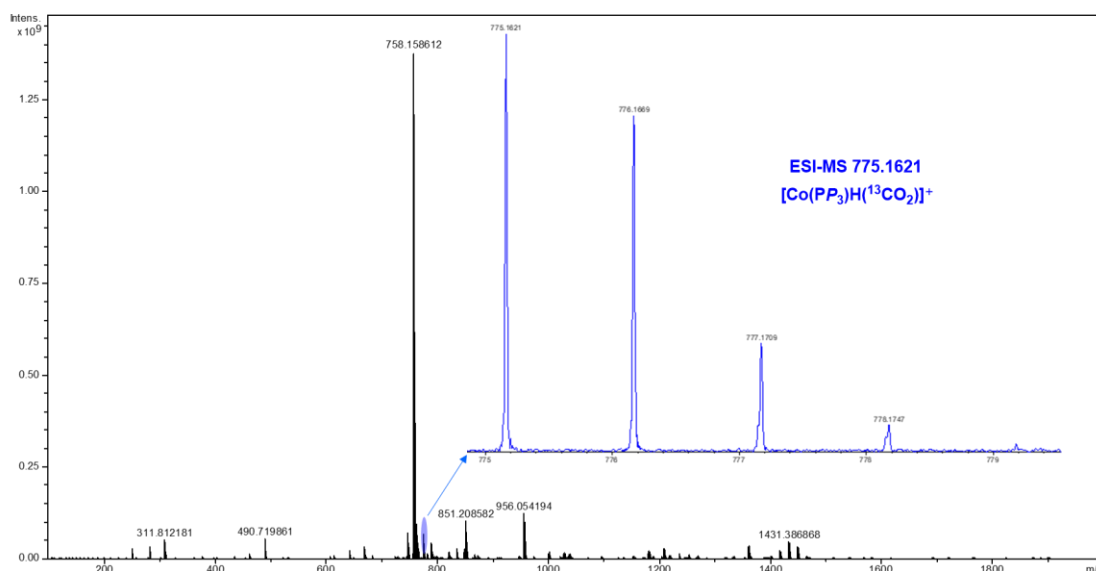
**Figure S9**  $^1\text{H}$  NMR spectrum of the products of Scheme 2b. The yield was determined by  $^1\text{H}$  NMR, calibrated using 1,1,2,2-tetrachloroethane as the internal standard.



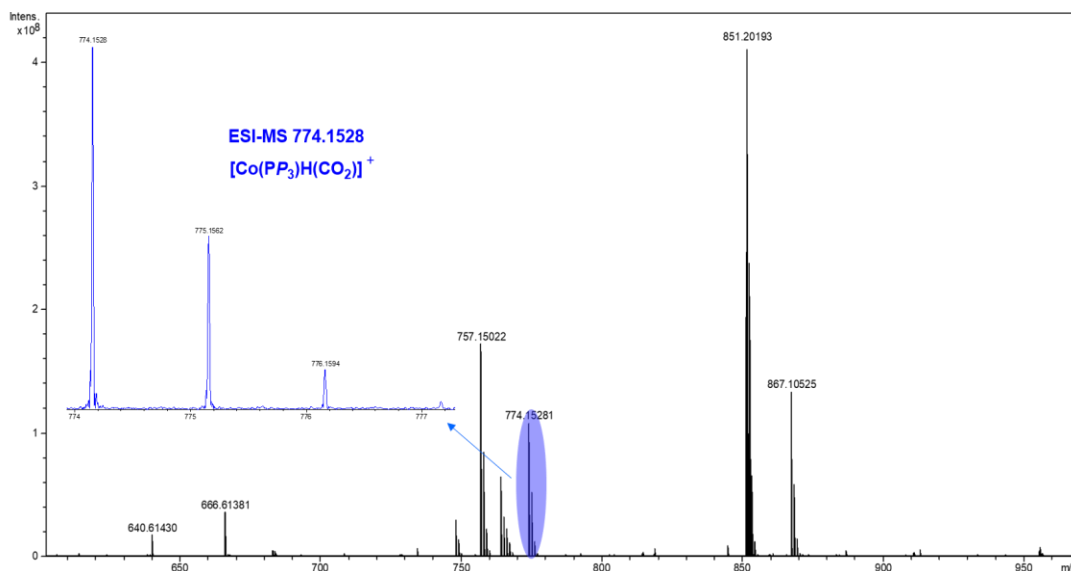
**Figure S10** MS spectrum of the cyclization reaction product in the  $^{13}\text{CO}_2$  isotope labelling experiment, Scheme 2c.



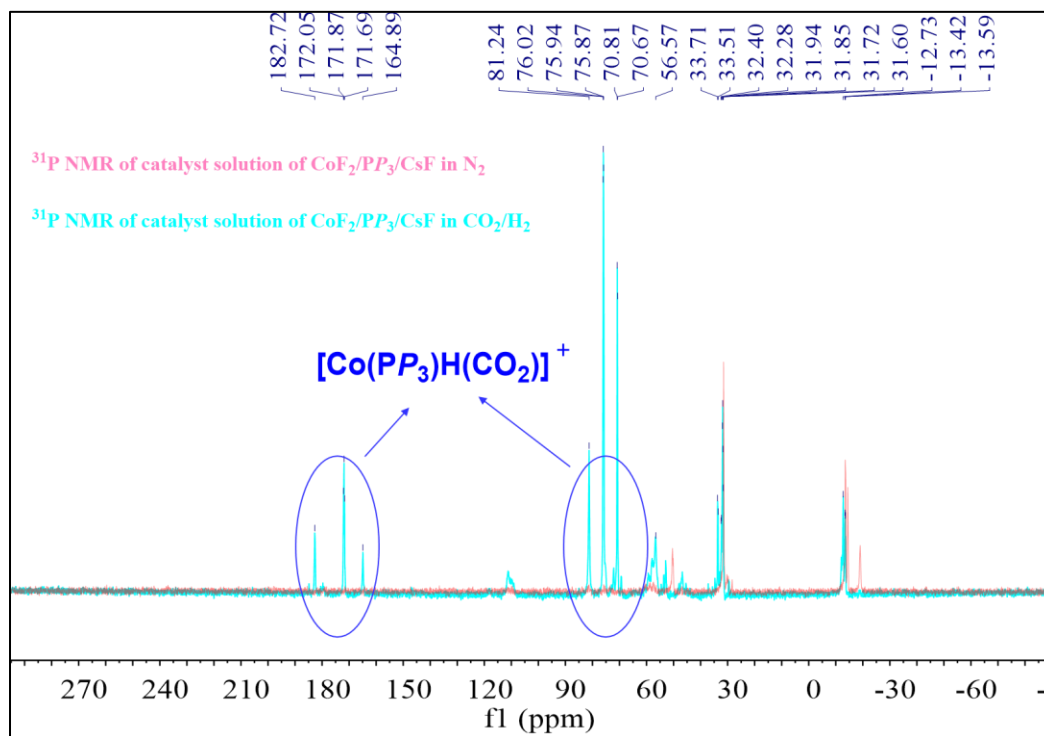
**Figure S11** ESI-MS spectra of the reaction mixture at different atmosphere. a, H<sub>2</sub> (3 MPa); b, <sup>13</sup>CO<sub>2</sub> (0.8 MPa); c, <sup>13</sup>CO<sub>2</sub> (0.8 MPa) and H<sub>2</sub> (3 MPa). Reaction conditions: CoF<sub>2</sub> (0.1 mmol), PP<sub>3</sub> (0.05 mmol), CsF (1.5 mmol), *o*-phenylenediamine (1.0 mmol), EtOH-D<sub>6</sub> (2 mL), 140 °C, 1.5h.



**Figure S12** HR-ESI-MS spectrum of the mixture of **1a** (1.0 mmol), CoF<sub>2</sub> (0.1 mmol), PP<sub>3</sub> (0.05 mmol), CsF (1.5 mmol), EtOH-D<sub>6</sub> (1.0 mL), <sup>13</sup>CO<sub>2</sub> (0.8 MPa), H<sub>2</sub> (3.0 MPa), 140 °C, 1.5 h.



**Figure S13** ESI-MS spectrum of the mixture of **1a** (1.0 mmol),  $\text{CoF}_2$  (0.1 mmol),  $\text{PP}_3$  (0.05 mmol),  $\text{CsF}$  (1.5 mmol),  $\text{EtOH}$  (2.0 mL),  $\text{CO}_2$  (3 MPa),  $\text{H}_2$  (3.0 MPa), 140 °C, 1.5 h.



**Figure S14**  $^{31}\text{P}$  NMR spectra of the catalyst solvent of **1a** (1.0 mmol),  $\text{CoF}_2$  (0.1 mmol),  $\text{PP}_3$  (0.05 mmol),  $\text{CsF}$  (1.5 mmol),  $\text{EtOH-d}_6$  (2.0 mL), at different atmosphere, 140 °C, 1.5 h.

## <sup>1</sup>H and <sup>13</sup>C NMR and HR-MS (ESI) data of isolated N-containing heterocycles

Benzimidazole (**1b**): White solid, yield=94%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.11 (s, 1H), 7.76-7.60 (m, 2H), 7.31 (m, 2H), 6.59 (s, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 140.53, 137.73, 123.14, 115.70, 77.48, 77.16, 76.84.

6-methyl-1H-benzo[d]imidazole (**2b**): White solid, yield=93%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 10.52 (s, 1H), 8.17 (s, 1H), 7.61 (d, 1H), 7.50 (s, 1H), 7.15 (d, 1H), 2.50 (s, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 140.86, 137.72, 136.50, 132.74, 124.42, 115.46, 114.93, 77.48, 77.16, 76.84, 21.68. HR-MS (ESI) calculated for C<sub>8</sub>H<sub>8</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 133.0761, found: 133.0760.

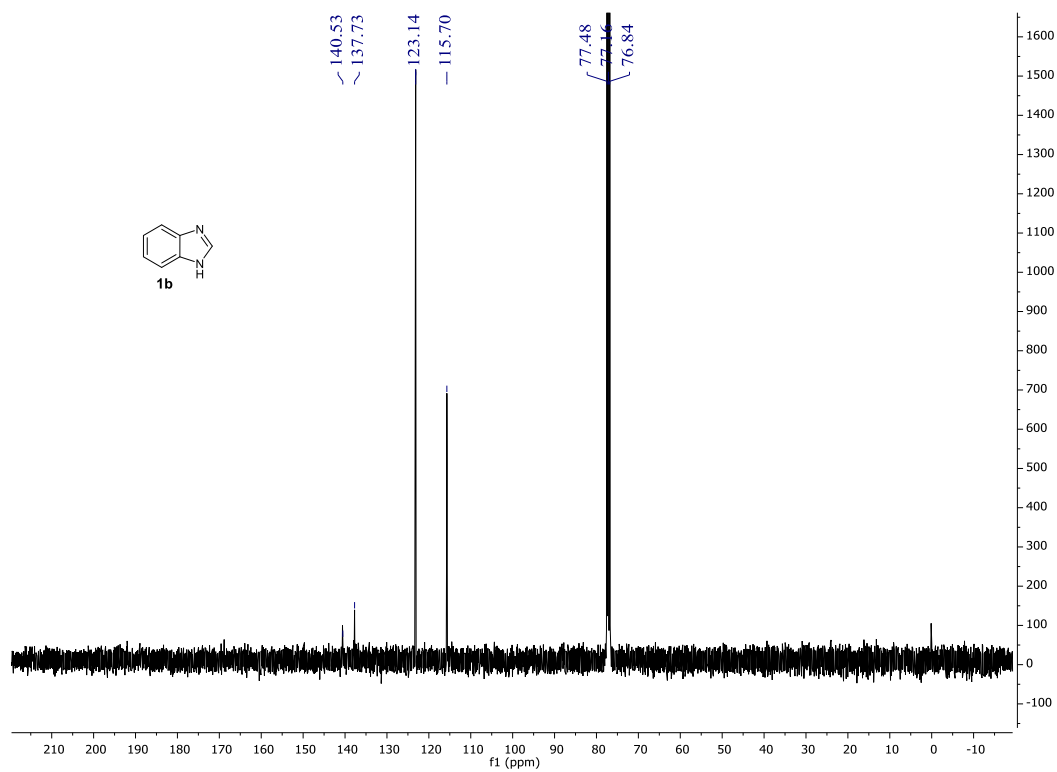
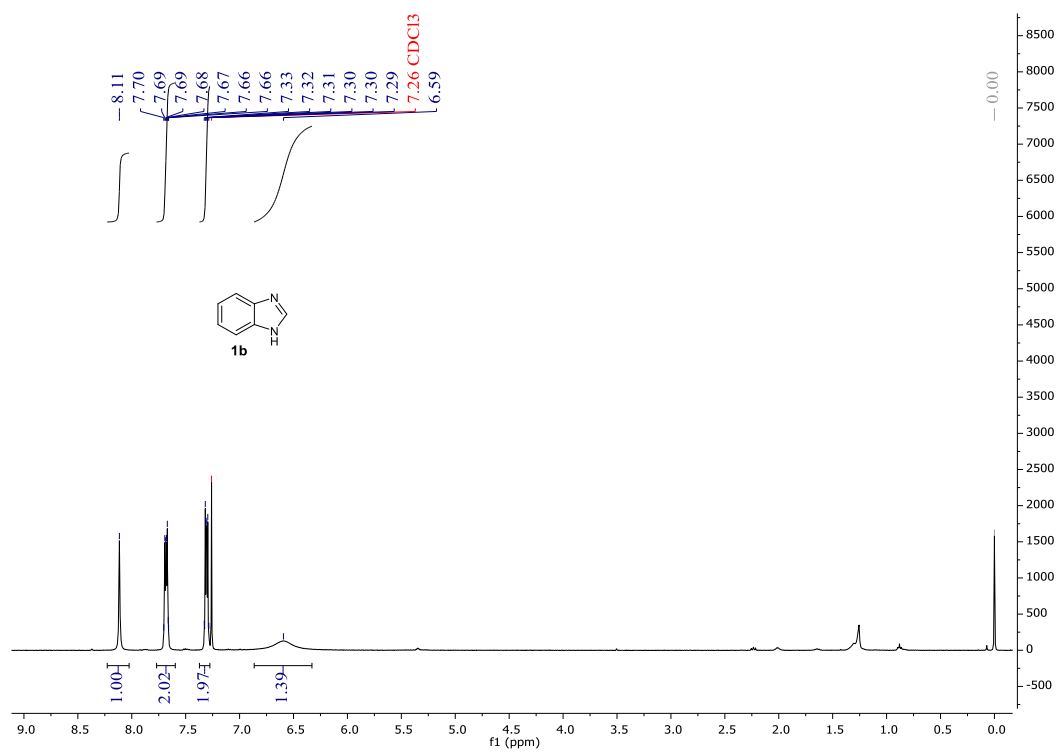
6-methoxy-1H-benzo[d]imidazole (**5b**): White solid, yield=85%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 9.99 (s, 1H), 8.06 (s, 1H), 7.56 (d, 1H), 7.11 (d, 1H), 6.94 (d, 1H), 3.83 (s, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 156.51, 140.18, 137.55, 132.82, 116.33, 112.48, 97.48, 77.16, 76.84, 76.52, 55.65. HR-MS (ESI) calculated for C<sub>8</sub>H<sub>8</sub>N<sub>2</sub>O [M+H]<sup>+</sup>: 149.0710, found: 149.0709.

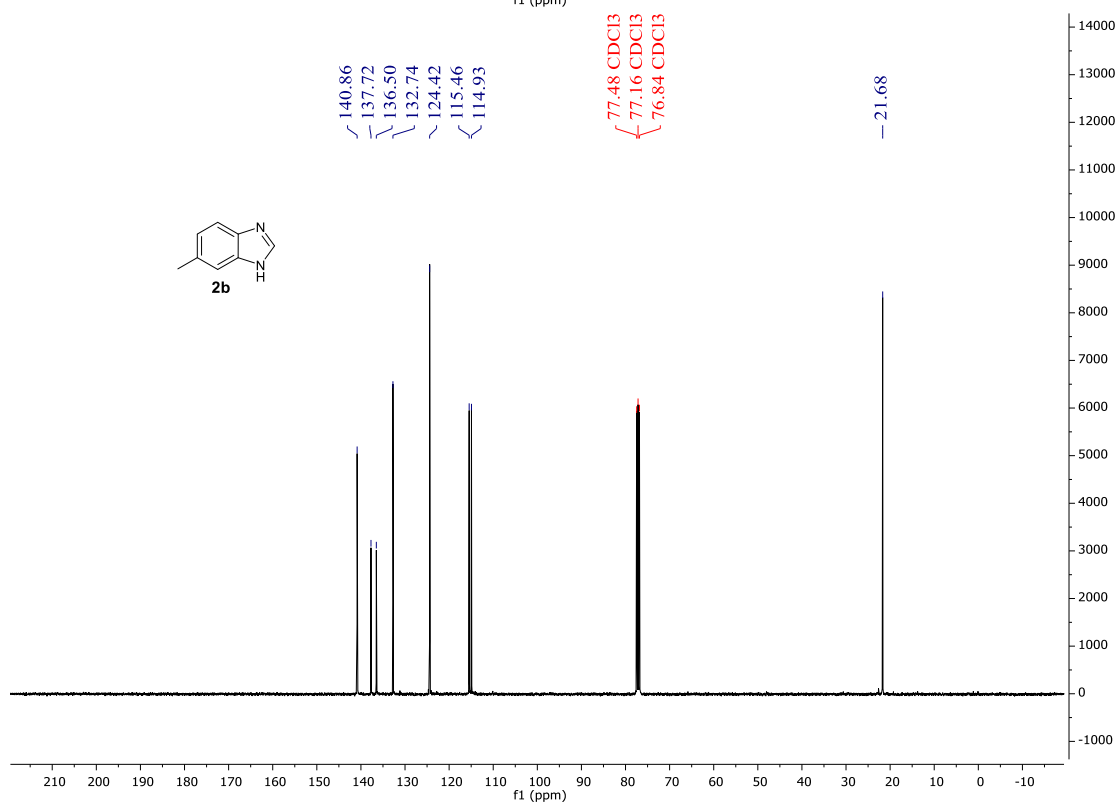
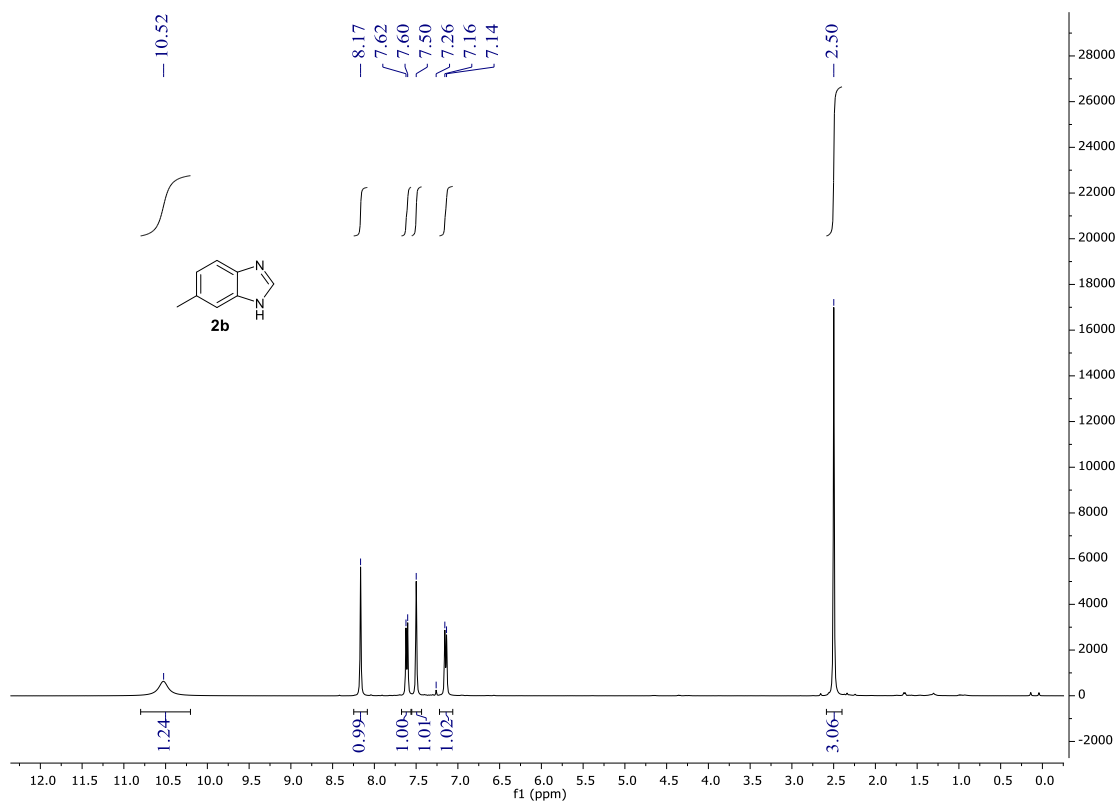
6-chloro-1H-benzo[d]imidazole (**7b**): Yellow solid, yield=86%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 10.79 (s, 1H), 8.24 (s, 1H), 7.66 (s, 1H), 7.58 (d, 1H), 7.26 (d, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 138.69, 136.62, 128.75, 123.72, 116.40, 115.47, 77.48, 77.16, 76.84. HR-MS (ESI) calculated for C<sub>7</sub>H<sub>5</sub>ClN<sub>2</sub> [M+H]<sup>+</sup>: 153.0215, found: 153.0214.

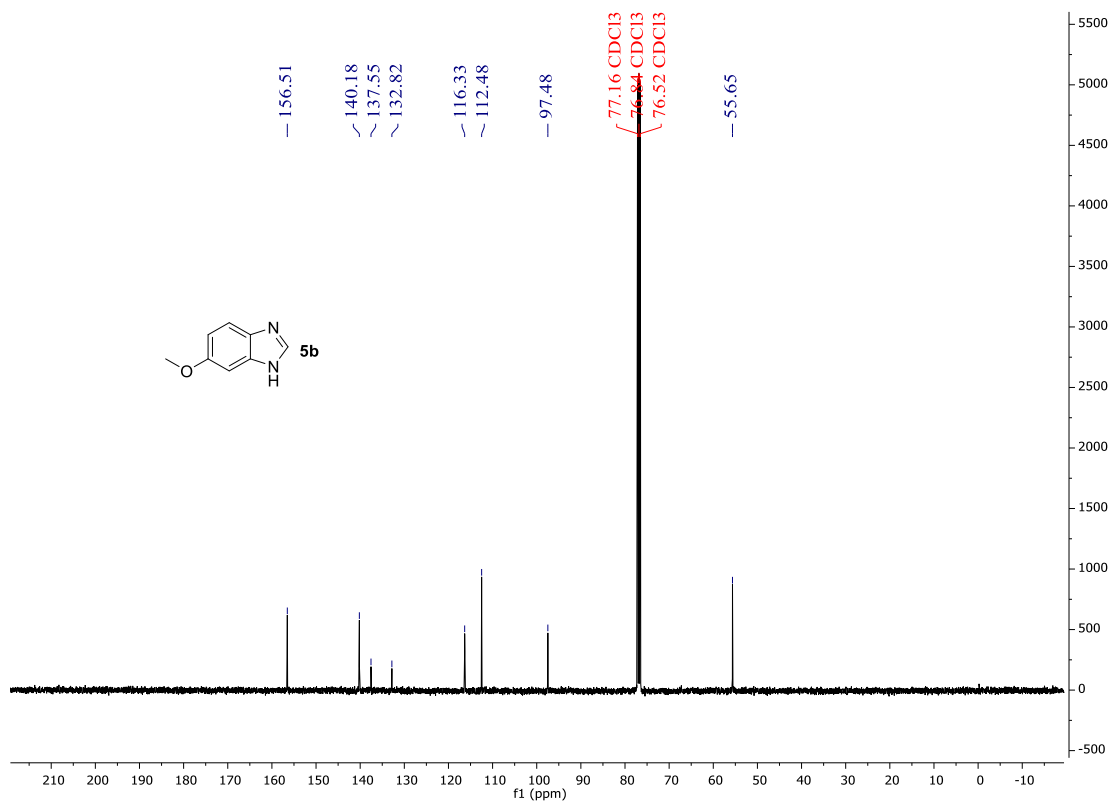
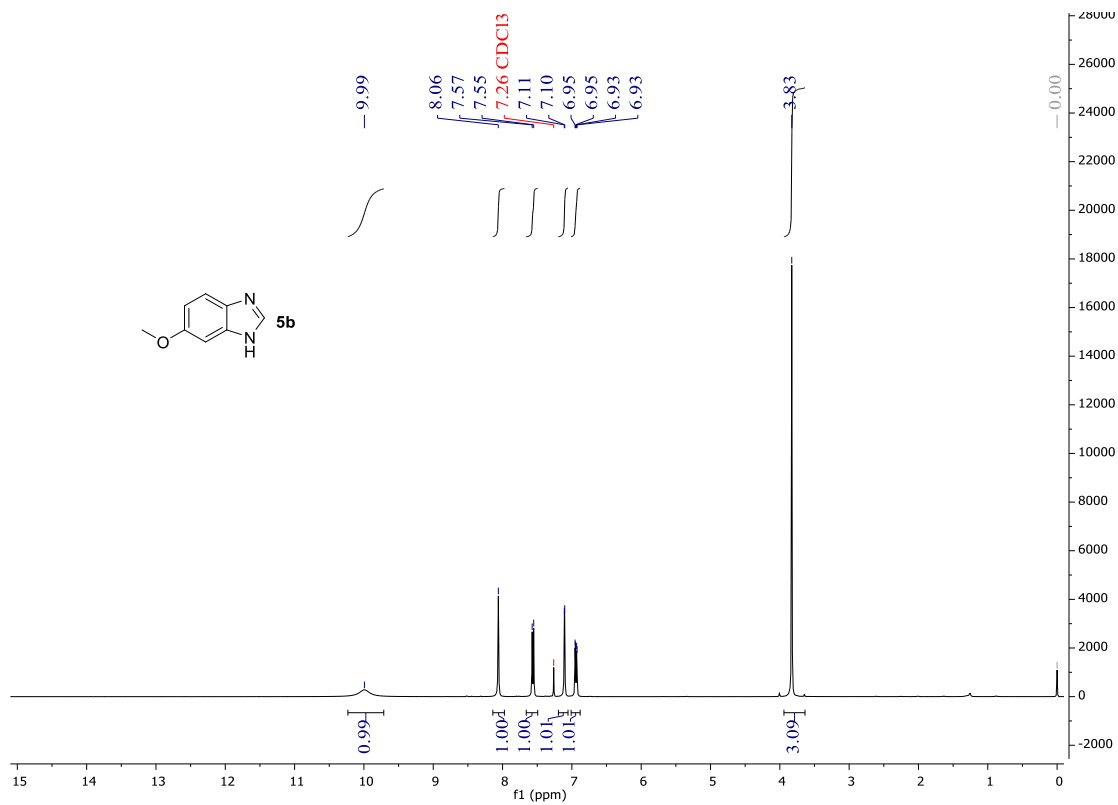
Benzothiazole (**18b**): Yellow solid, yield=92%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.96 (s, 1H), 8.13 (d, 1H), 7.92 (d, 1H), 7.49 (t, 1H), 7.41 (t, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 153.92, 153.24, 133.71, 126.17, 125.54, 123.63, 121.89, 77.48, 77.16, 76.84. HR-MS (ESI) calculated for C<sub>7</sub>H<sub>5</sub>NS [M+H]<sup>+</sup>: 136.0217, found: 136.0215.

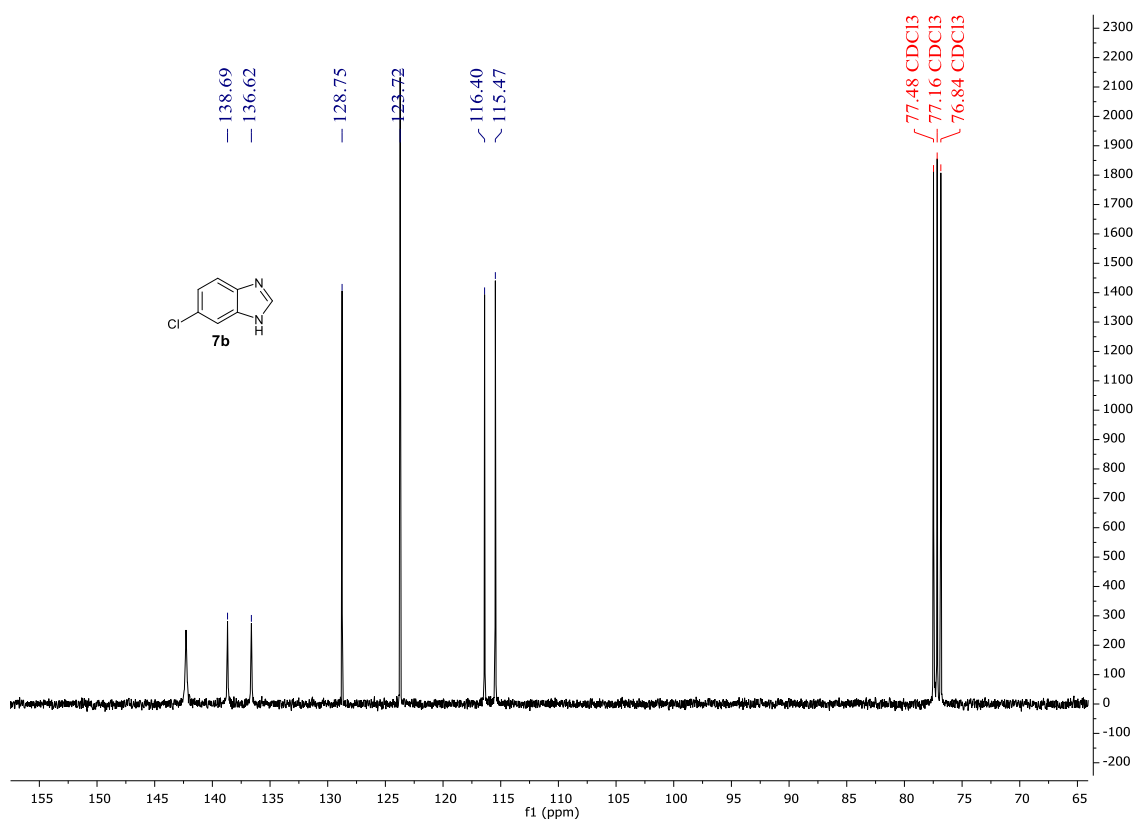
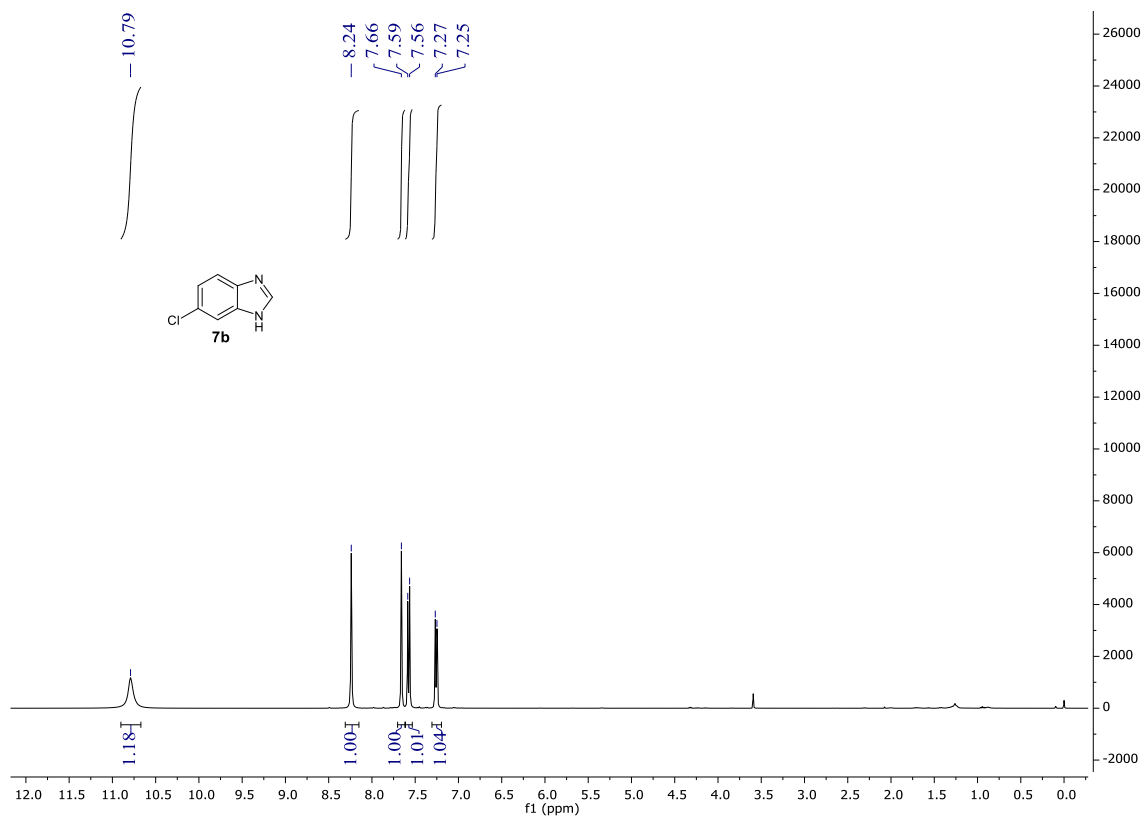
Benzoxazole (**19b**): Yellow solid, yield=70%, <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.08 (s, 1H), 7.78 (m, 1H), 7.60-7.50 (m, 1H), 7.42-7.30 (m, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 152.57, 150.02, 140.10, 125.65, 124.64, 120.66, 111.01, 77.48, 77.16, 76.84. HR-MS (ESI) calculated for C<sub>7</sub>H<sub>5</sub>NO [M+H]<sup>+</sup>: 120.0446, found: 120.0444.

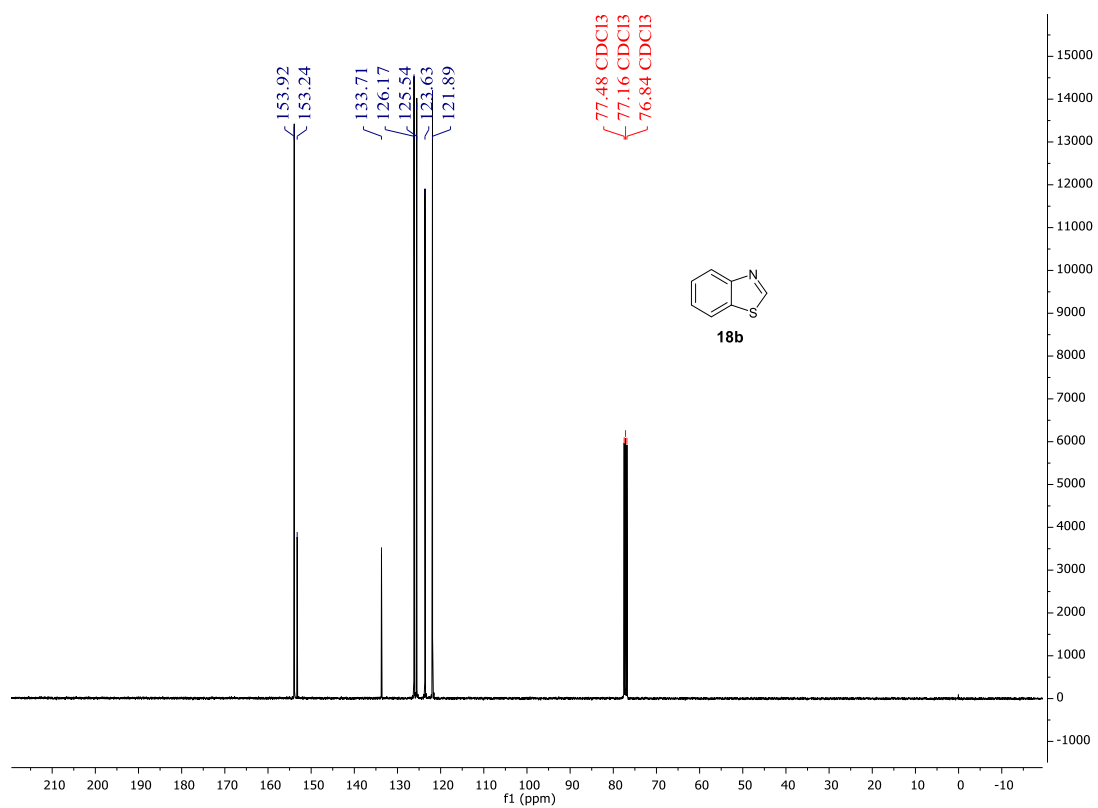
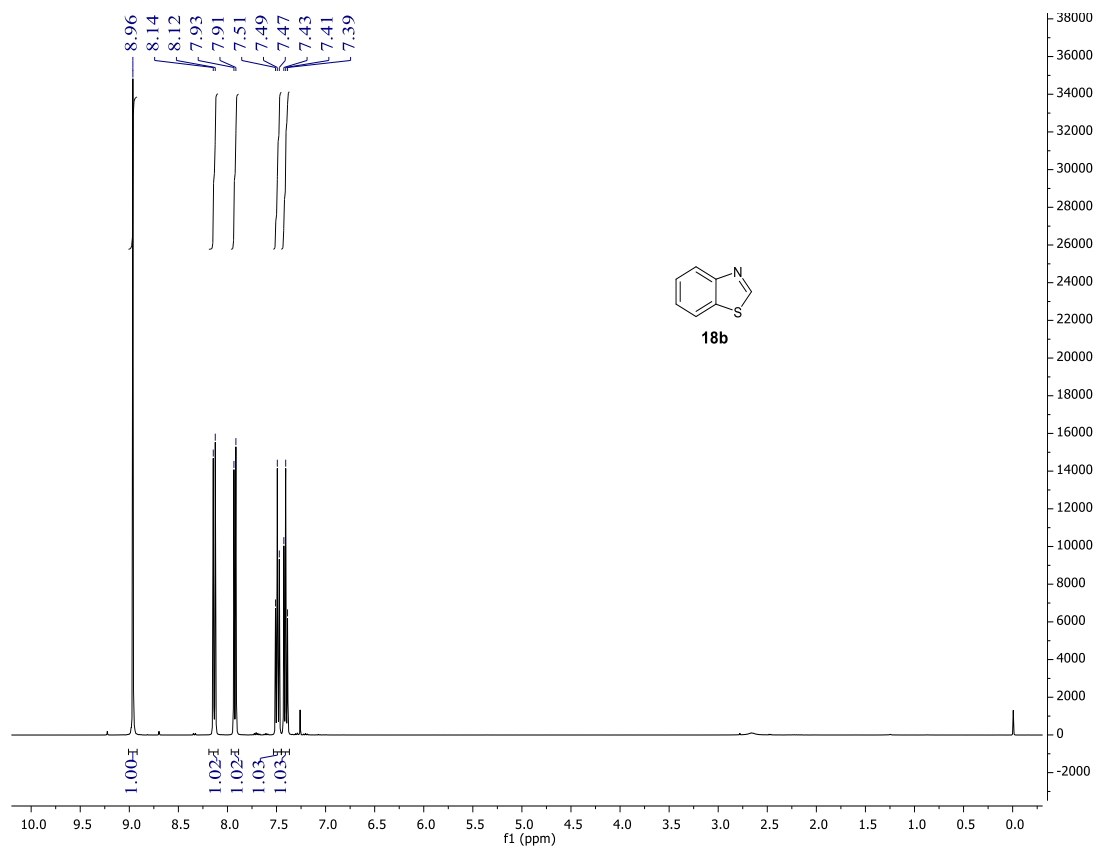
# <sup>1</sup>H and <sup>13</sup>C NMR spectra of the products

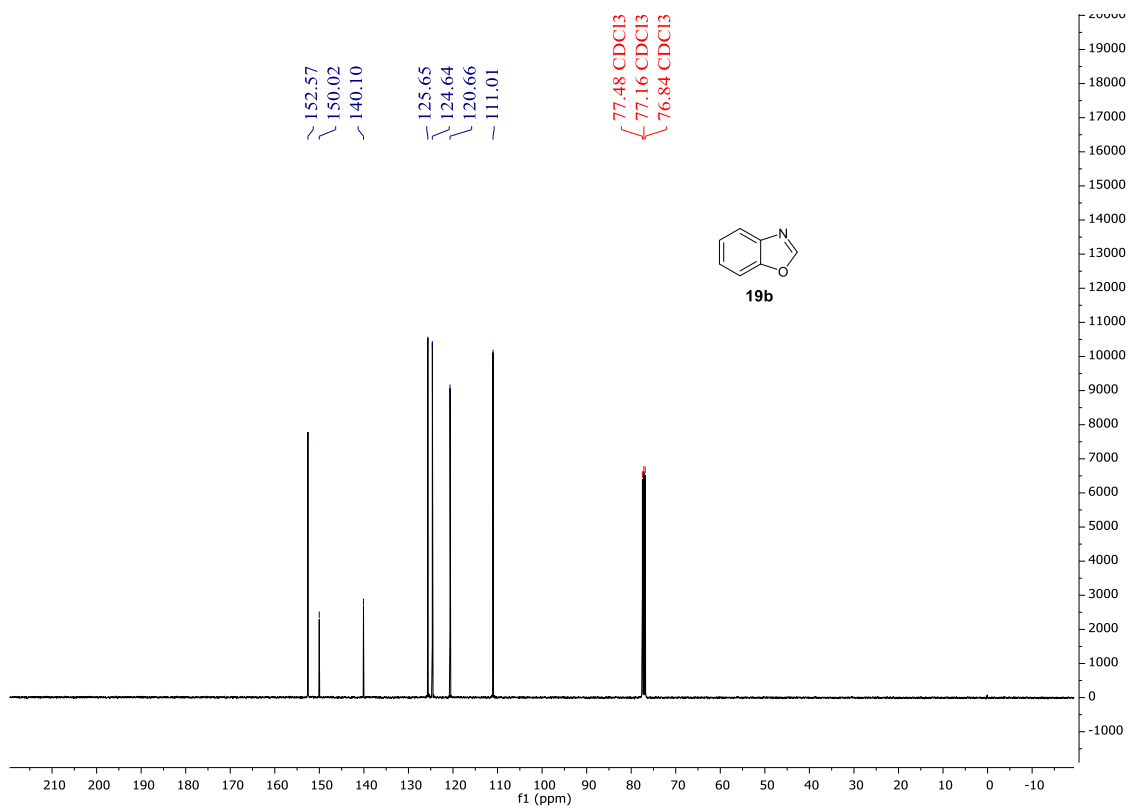
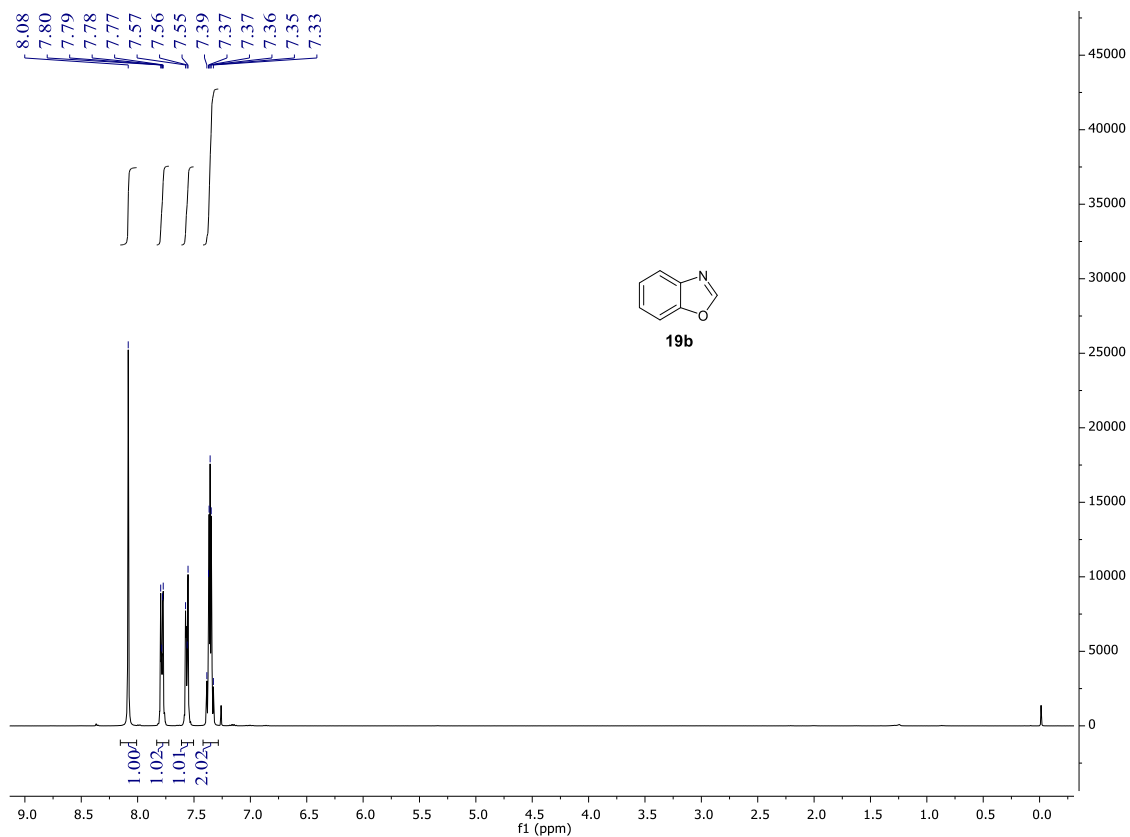












## References

- [1] Yu, B.; Zhang, H.; Zhao, Y.; Chen, S.; Xu, J.; Huang, C.; Liu, Z., Cyclization of *o*-phenylenediamines by CO<sub>2</sub> in the presence of H<sub>2</sub> for the synthesis of benzimidazoles. *Green Chem.* **2013**, *15*, 95-99.
- [2] Hao, L.; Zhao, Y.; Yu, B.; Zhang, H.; Xu, H.; Liu, Z., Au catalyzed synthesis of benzimidazoles from 2-nitroanilines and CO<sub>2</sub>/H<sub>2</sub>. *Green Chem.* **2014**, *16*, 3039-3044.