

Supporting Information for:

Atom Economic Amidation with Acid Anhydrides using an Electrochemical Approach

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Table of Contents

Figures showing the electrochemical cell	2
Characterization information of synthesized compounds.....	3-5
Spectra of synthesized compounds	6-30
PMI calculation numbers	31
References	31-32

1. Picture of the electrochemical cell



Figure S1. Electrochemical H-Cell used to perform reactions described in the manuscript. Reticulated Vitreous Carbon electrodes were employed as cathode and anode.

2. Picture of the electrochemical reaction driven by 9-V battery.



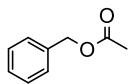
Figure S2. Electrochemical H-Cell used to perform reactions described in the manuscript. Reticulated Vitreous Carbon electrodes were employed as cathode and anode. Electrolysis is carried out by connecting the electrodes to a commercially available battery.

3. List of ^1H NMR and ^{13}C NMR peaks of products

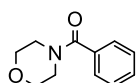
Multiplicity was indicated as follows: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet).



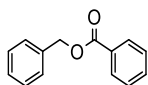
colorless oil, 72 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 2.03 (s, 3H), 3.39 (t, 2H, $J = 4.88$ Hz), 3.54 (t, 2H, $J = 4.86$ Hz), 3.61 (q, 4H, $J = 5.17$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 21.1, 41.7, 46.7, 66.6, 66.8, 169.3.



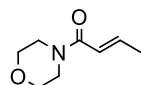
colorless oil, 77 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 2.102 (s, 3H,), 5.103 (s, 2H), 7.355 (m, 5H). ^{13}C NMR (CDCl_3 , 100 MHz) δ 21, 66, 128.2, 128.5



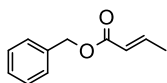
colorless oil, 100 mg, ^1H NMR (D-acetone, 400 MHz) δ 3.63 (broad singlet, 8H), 7.43 (m, 5H). ^{13}C NMR (D-acetone, 100 MHz) δ 66.5, 127.2, 128.3, 129.4, 136.2, 169.4.



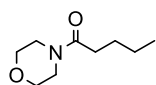
colorless oil, 114 mg, ^1H NMR (D-acetone, 400 MHz) δ 5.38 (s, 2H), 7.39 (3H, $J = 6.09$ Hz), 7.52 (t, 4H, $J = 7.60$ Hz), 7.63 (m, 1H, $J = 7.40$ Hz), 8.06 (m, 2H, $J = 7.12$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 66.2, 128.1, 128.5, 128.6, 129.4, 130.3, 133.1, 136.6, 165.7.



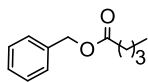
colorless oil, 79 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 1.87 (d, 3H, $J = 2.85$ Hz), 3.66 (broad multiplet, 8H, $J = 7.60$ Hz), 6.21 (d, 1H, $J = 5.55$ Hz), 6.88 (m, 1H, $J = 7.11$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 18.3, 42.2, 46.1, 66.8, 121, 142.2, 165.7.



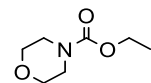
colorless oil, 83 mg, ^1H NMR (D-acetone, 400 MHz) δ 1.97 (dd, 3H, $J = 2.88$ Hz), 5.25 (s, 2H), 6.02 (dd, 1H, $J = 5.76$ Hz), 7.12 (m, 1H, $J = 5.62$ Hz), 7.48 (m, 5H, $J = 4.48$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 17.7, 40.1, 66.1, 117.9, 122.6, 128.6, 129.1, 137.3, 146.0, 166.4



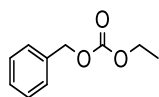
yellow oil, 86 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 0.88 (t, 3H, $J = 7.34$ Hz), 1.28 (m, 2H, $J = 5.62$ Hz), 1.55 (m, 2H, $J = 5.10$ Hz), 2.24 (t, 2H, $J = 7.70$ Hz), 3.42 (m, 2H, $J = 5.00$ Hz), 3.56 (m, 2H, $J = 5.16$ Hz), 3.61 (m, 4H, $J = 4.12$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 13.8, 22.5, 27.3, 32.8, 41.8, 46.0, 66.6, 66.9, 171.9



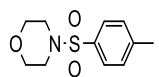
white solid, 106 mg, ^1H NMR (D-acetone, 400 MHz) δ 1.00 (t, 3H, $J = 7.36$ Hz), 1.44 (m, 2H, $J = 5.62$ Hz), 1.66 (m, 2H, $J = 4.30$ Hz), 2.45 (t, 2H, $J = 7.48$ Hz), 5.19 (s, 2H), 7.47 (bs, 5H). ^{13}C NMR (D-acetone, 100 MHz) δ 13.6, 22.5, 27.4, 34.1, 66.1, 117.9, 128.5, 129.1, 137.3, 173.8



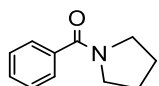
yellow oil, 87 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 1.26 (t, 3H, $J = 7.12$ Hz), 3.46 (m, 4H, $J = 4.88$ Hz), 3.64 (m, 4H, $J = 4.46$ Hz), 4.13 (q, 2H, , $J = 7.12$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 14.6, 20.3, 61.5, 66.6, 126.9, 127, 128, 155.6



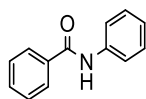
white solid, 92 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 1.30 (t, 3H, $J = 7.14$ Hz), 4.20 (q, 2H, $J = 7.15$ Hz), 5.15 (s, 2H), 7.36 (m, 5H, $J = 2.28$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 14.3, 64.1, 69.4, 128.3, 128.6, 135.3



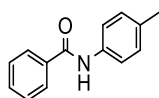
yellow solid, 137 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 2.45 (s, 3H), 2.91 (m, 4H, $J = 4.72$ Hz), 3.67 (m, 4H, $J = 4.74$ Hz), 7.47 (d, 2H, $J = 8.04$ Hz), 7.66 (d, 2H, $J = 8.24$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 12.9, 20.6, 46.2, 58.4, 65.7, 126.9, 129.5, 132.4, 143.9



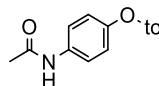
brown oil, 109 mg, ^1H NMR (D-acetone, 400 MHz) δ 1.89 (m, 4H, $J = 7.93$ Hz), 3.44 (t, 2H), 3.54 (t, 2H), 7.44 (m, 3H), 7.54 (m, 2H, $J = 7.72$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 24.1, 26.2, 45.9, 49.1, 127.1, 128.1, 129.5, 137.8, 168.6



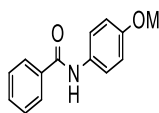
white solid, 110 mg, ^1H NMR (D-acetone, 400 MHz) δ 7.10 (t, 1H, $J = 7.4$ Hz), 7.35 (t, 2H, $J = 7.98$ Hz), 7.51 (m, 3H, $J = 6.76$ Hz), 7.56 (d, 2H, $J = 7.6$ Hz), 7.98 (m, 2H, $J = 4.3$ Hz), 9.52 (s, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 120.1, 123.6, 127.4, 128.4, 131.4, 139.5



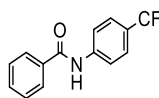
orange solid, 122 mg, ^1H NMR (D-acetone, 400 MHz) δ 2.29 (s, 3H), 7.15 (d, 2H, $J = 8.24$ Hz), 7.52 (m, 3H, $J = 8.8$ Hz), 7.72 (d, 2H, $J = 8.4$ Hz), 7.98 (d, 2H, $J = 7.28$ Hz), 9.46 (s, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 19.9, 120.2, 127.4, 128.4, 129.1, 131.3, 132.9, 135.5, 136.9, 165.2



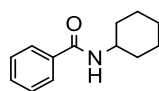
yellowish white solid, 125 mg, ^1H NMR (D-acetone, 400 MHz) δ 2.08 (s, 3H), 2.23 (s, 3H), 6.86 (m, 3H, $J = 4.12$ Hz), 7.06 (t, 1H, $J = 7.96$ Hz), 7.17 (t, 1H, $J = 7.72$ Hz), 7.28 (d, 1H, $J = 4.06$ Hz), 7.64 (d, 2H, $J = 9.0$ Hz), 9.18 (s, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 15.4, 23.3, 117.7, 118.8, 120.6, 121, 123.7, 127.2, 129.2, 131.4, 134.8, 153.3, 155.1, 167.7



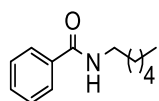
white solid, 126 mg, ^1H NMR (D-acetone, 400 MHz) δ 3.66 (s, 3H), 6.78 (d, 2H, $J = 3.02$ Hz), 7.41 (m, 3H, $J = 7.21$ Hz), 7.62 (d, 2H, $J = 9.04$ Hz), 7.86 (d, 2H, $J = 7.11$ Hz), 9.28 (s, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 54.75, 113.7, 121.7, 127.3, 128.3, 131.3, 132.6, 135.6, 156.2, 165.0



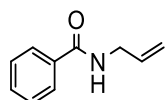
white solid, 123 mg, ^1H NMR (D-acetone, 400 MHz) δ 7.53 (t, 2H, $J = 7.44$ Hz), 7.60 (d, 1H, $J = 7.34$ Hz), 7.72 (d, 2H, $J = 8.64$ Hz), 8.02 (d, 2H, $J = 7.32$ Hz), 8.10 (d, 2H, $J = 8.40$ Hz), 9.86 (s, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 119.8, 119.9, 125.8, 127.6, 128.5, 131.8, 134.9, 142.9, 165.9



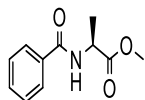
white solid, 120 mg, ^1H NMR (CDCl_3 , 400 MHz) δ 1.14 (m, 3H, $J = 8.42$ Hz), 1.31 (quartet of triplets, 2H, $J = 12.43$ Hz), 1.56 (doublet of triplets, 1H, $J = 12.96$ Hz), 1.66 (doublet of triplets, 2H, $J = 13.63$ Hz), 1.93 (doublet of doublets, 2H, $J = 12.30$ Hz), 3.88 (m, 1H, $J = 4.05$ Hz), 6.15 (broad doublet, 1H, $J = 6.48$ Hz), 7.31 (tt, 2H, $J = 6.70$ Hz), 7.38 (tt, 1H, $J = 6.69$ Hz), 7.68 (m, 2H, $J = 7.05$ Hz). ^{13}C NMR (CDCl_3 , 100 MHz) δ 24.9, 25.6, 30.9, 33.2, 77.1, 126.9, 128.4, 131.2, 135.1, 166.7



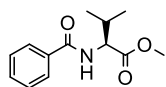
colorless oil, 118 mg, ^1H NMR (D-acetone, 400 MHz) δ 0.89 (tr, 3H, $J = 6.85$ Hz), 1.33 (m, 6H, $J = 6.63$ Hz), 1.62 (quint, 2H, $J = 7.26$ Hz), 3.40 (quart, 2H, $J = 7.34$ Hz), 7.44 (m, 2H, $J = 7.34$ Hz), 7.52 (m, 1H, $J = 7.39$ Hz), 7.84 (s, 1H), 7.92 (m, 2H, $J = 7.08$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 13.5, 22.4, 26.6, 31.5, 39.6, 127.1, 128.2, 130.9, 135.3, 166.4



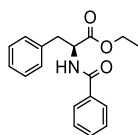
brown oil, 92 mg, ^1H NMR (D-acetone, 400 MHz) δ 4.02 (tt, 2H, $J = 5.62$ Hz), 5.21 (dq, 2H, $J = 17.19$ Hz), 5.95 (dq, 1H, $J = 27.46$ Hz), 7.46 (t, 2H, $J = 7.36$ Hz), 7.52 (t, 1H, $J = 7.27$ Hz), 7.94 (d, 2H, $J = 7.07$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 41.9, 114.9, 127.2, 128.5, 129.6, 131.2, 132.8, 134.9, 135.3, 166.6



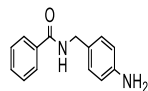
yellow oil, 114 mg, ^1H NMR (D-acetone, 400 MHz) δ 1.49 (d, 3H, $J = 7.32$ Hz), 3.7 (s, 3H), 4.68 (quint, 1H, $J = 7.33$ Hz), 7.46 (t, 2H, $J = 7.45$ Hz), 7.53 (m, 1H, $J = 7.33$ Hz), 7.95 (d, 2H, $J = 7.31$ Hz), 8.06 (m, 1H, $J = 4.22$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 16.7, 48.5, 51.5, 127.3, 128.3, 131.4, 134.3, 166.5, 173.1



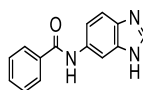
white solid, 116 mg, ^1H NMR (D-acetone, 400 MHz) δ 1.03 (dd, 6H, $J = 5.83$ Hz), 2.26 (m, 1H, $J = 6.77$ Hz), 3.73 (s, 3H), 4.60 (dd, 1H, $J = 4.96$ Hz), 7.46 (t, 2H, $J = 7.44$ Hz), 7.55 (t, 1H, $J = 7.34$ Hz), 7.73 (s, 1H), 7.93 (dd, 2H, $J = 7.16$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 17.9, 18.7, 30.6, 51.2, 58.2, 127.4, 128.3, 131.3, 134.53, 166.9, 172.2.



white solid, ^1H NMR (D-acetone, 400 MHz) δ 1.18 (t, 3H, $J = 7.12$ Hz), 3.21 (m, 1H, $J = 9.31$ Hz), 4.13 (quart, 2H, $J = 4.22$ Hz), 4.88 (m, 1H, $J = 4.22$ Hz), 7.2 (m, 1H, $J = 7.08$ Hz), 7.3 (m, 4H, $J = 7.15$ Hz), 7.42 (m, 2H, $J = 7.46$ Hz), 7.50 (m, 1H, $J = 7.41$ Hz), 7.83 (m, 2H, $J = 7.04$ Hz), 7.88 (m, 1H). ^{13}C NMR (D-acetone, 100 MHz) δ 13.6, 37.2, 54.4, 60.7, 126.6, 127.3, 128.1, 128.3, 131.4, 134.4, 137.5, 166.6, 171.6



white solid, 120 mg, ^1H NMR (D-acetone, 400 MHz) δ 4.08 (s, 2H), 4.37 (d, 2H, $J = 5.96$ Hz), 6.57 (d, 2H, $J = 8.40$ Hz), 7.06 (d, 2H, $J = 8.44$ Hz), 7.35 (s, 1H), 7.43 (m, 2H, $J = 7.36$ Hz), 7.49 (m, 1H, $J = 7.32$ Hz), 7.77 (m, 2H, $J = 7.08$ Hz). ^{13}C NMR (D-acetone, 100 MHz) δ 43.3, 114.9, 117.9, 127.5, 128.9, 129.3, 131.7, 135.4, 147.6, 167.2



white solid, 137 mg, ^1H NMR (D-acetone, 400 MHz) δ 7.52 (m, 2H, $J = 6.38$ Hz), δ 7.61 (m, 2H, $J = 7.2$ Hz), 7.73 (dt, 1H, $J = 8.93$ Hz), 8.03 (d, 2H, $J = 7.24$ Hz), 8.40 (m, 1H, $J = 2.44$ Hz), 9.57 (s, 1H). ^{13}C NMR (D-acetone, 400 MHz) δ 109.9, 110.9, 121.2, 123.9, 127.4, 128.4, 131.3, 132.6, 133.7, 135.6, 165.3

4. Spectra of synthesized compounds

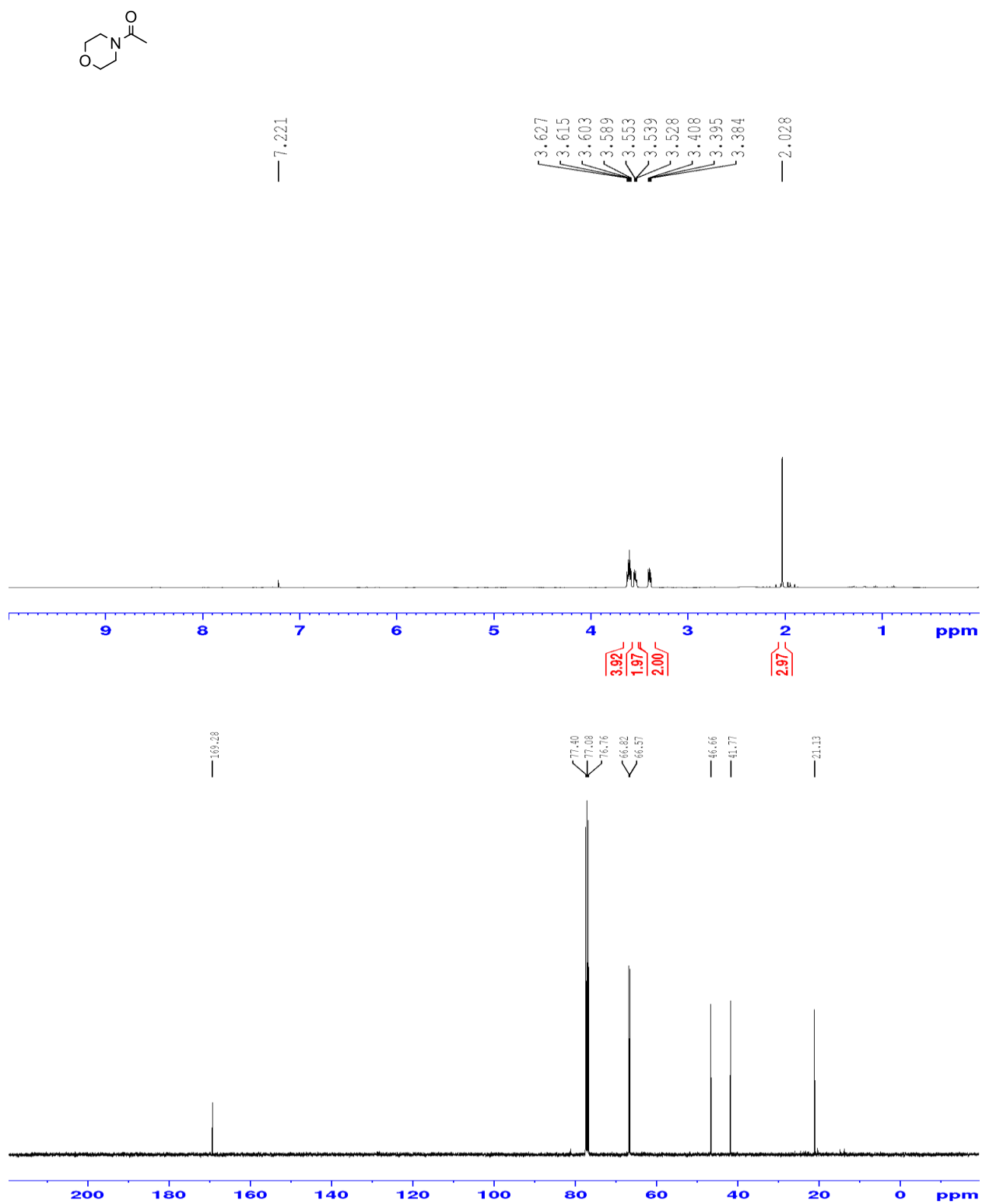


Figure S3 ¹H & ¹³C spectra of product 1a in CDCl₃. The spectra matched reference¹⁸

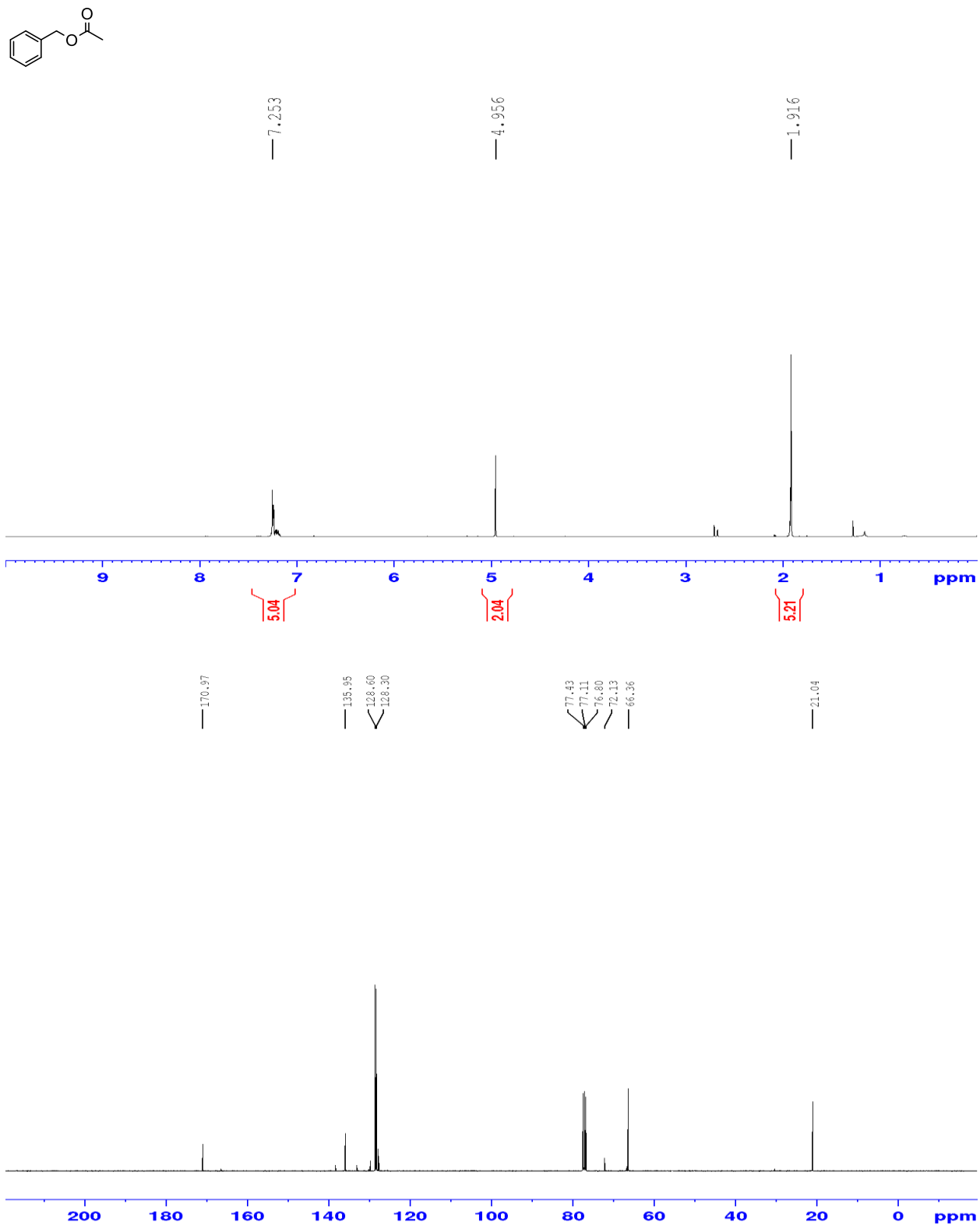


Figure S4 ¹H (D₂O) and ¹³C (CDCl₃) spectra of product 1b. The peak at 2.84 ppm is water from the acetone solvent. The spectra matched reference⁵

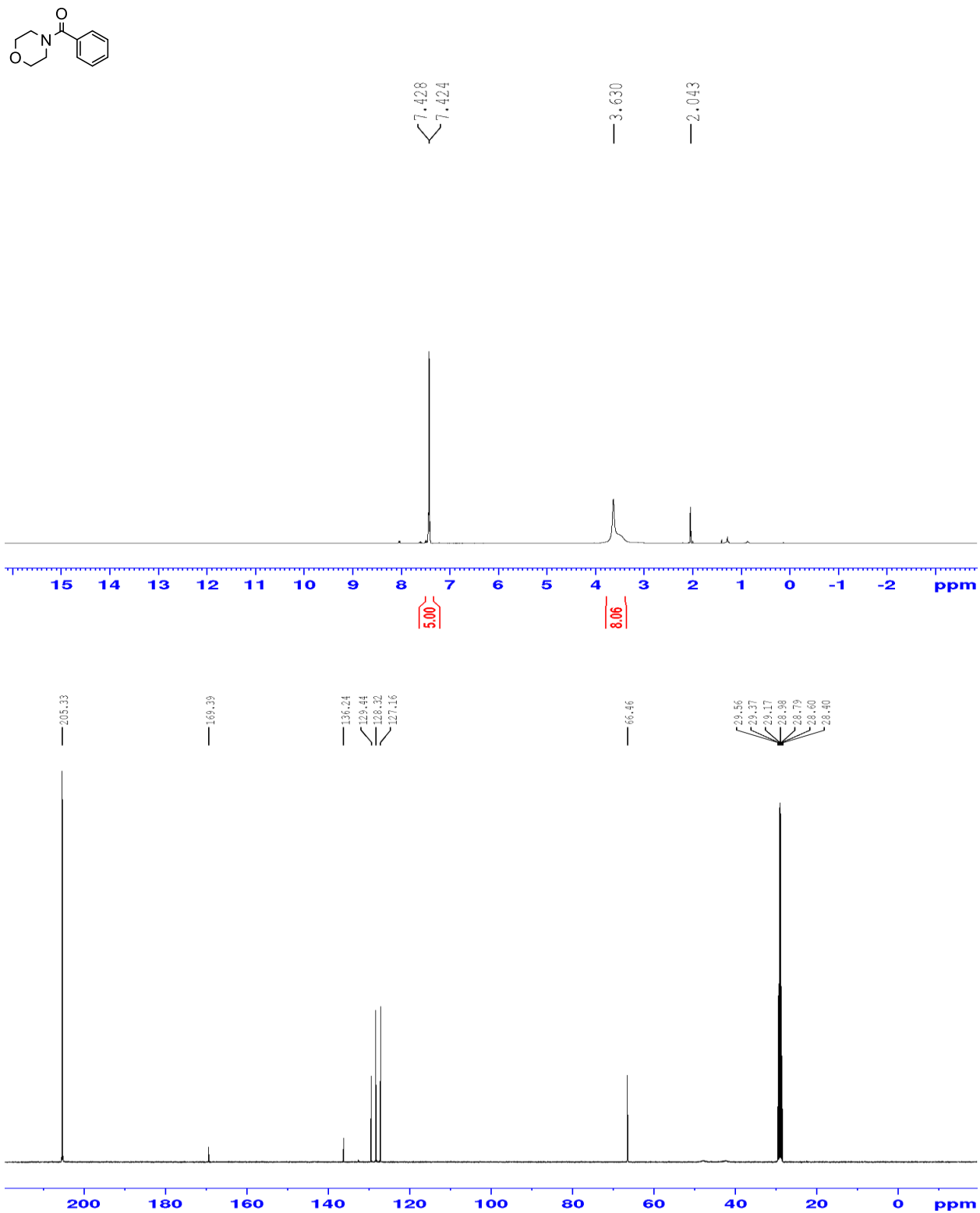


Figure S5 ¹H & ¹³C spectra of product 2a in D-acetone. The spectra matched reference¹²

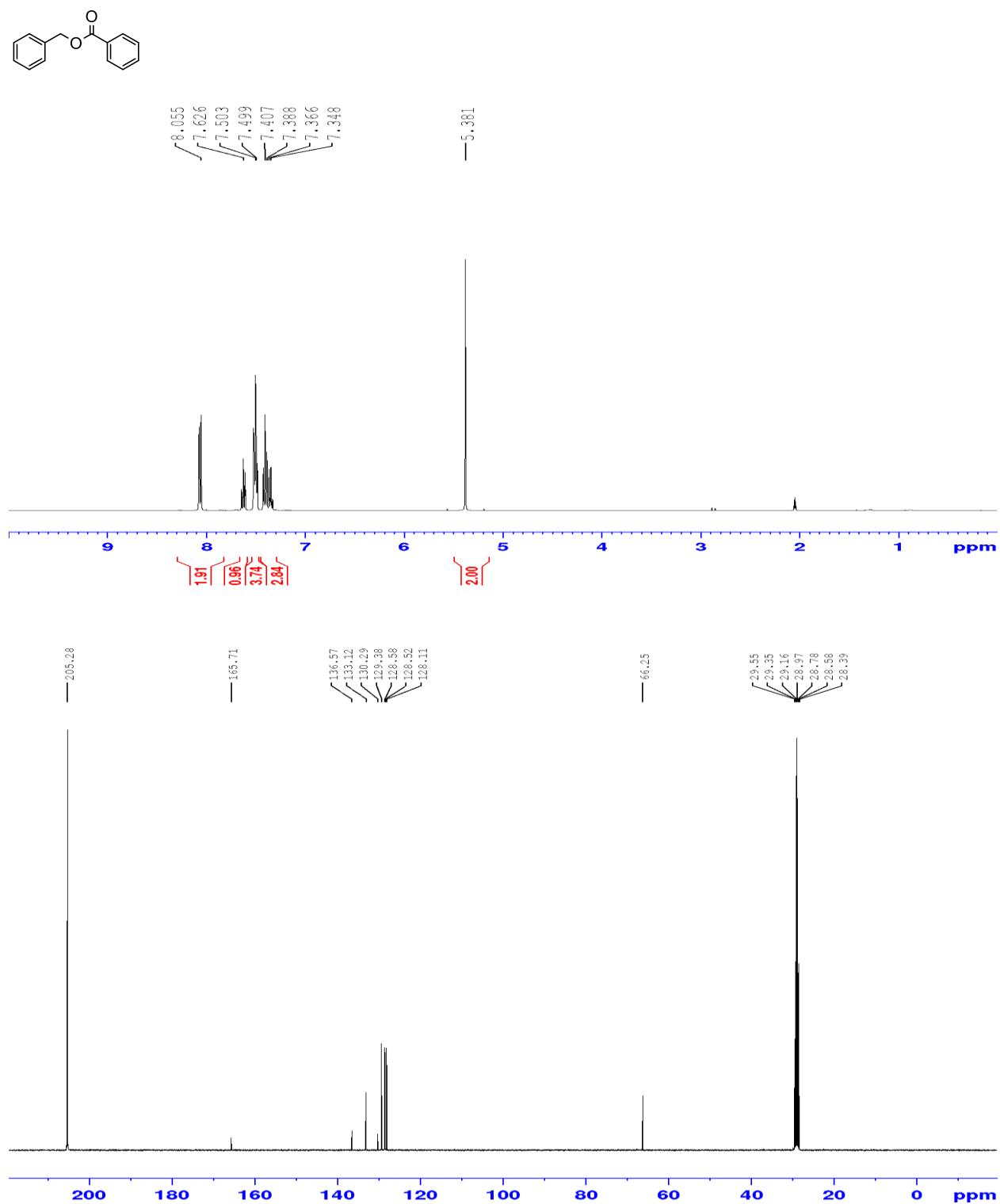


Figure S6 ¹H & ¹³C spectra of product 2b in D-acetone. The spectra matched reference¹⁶

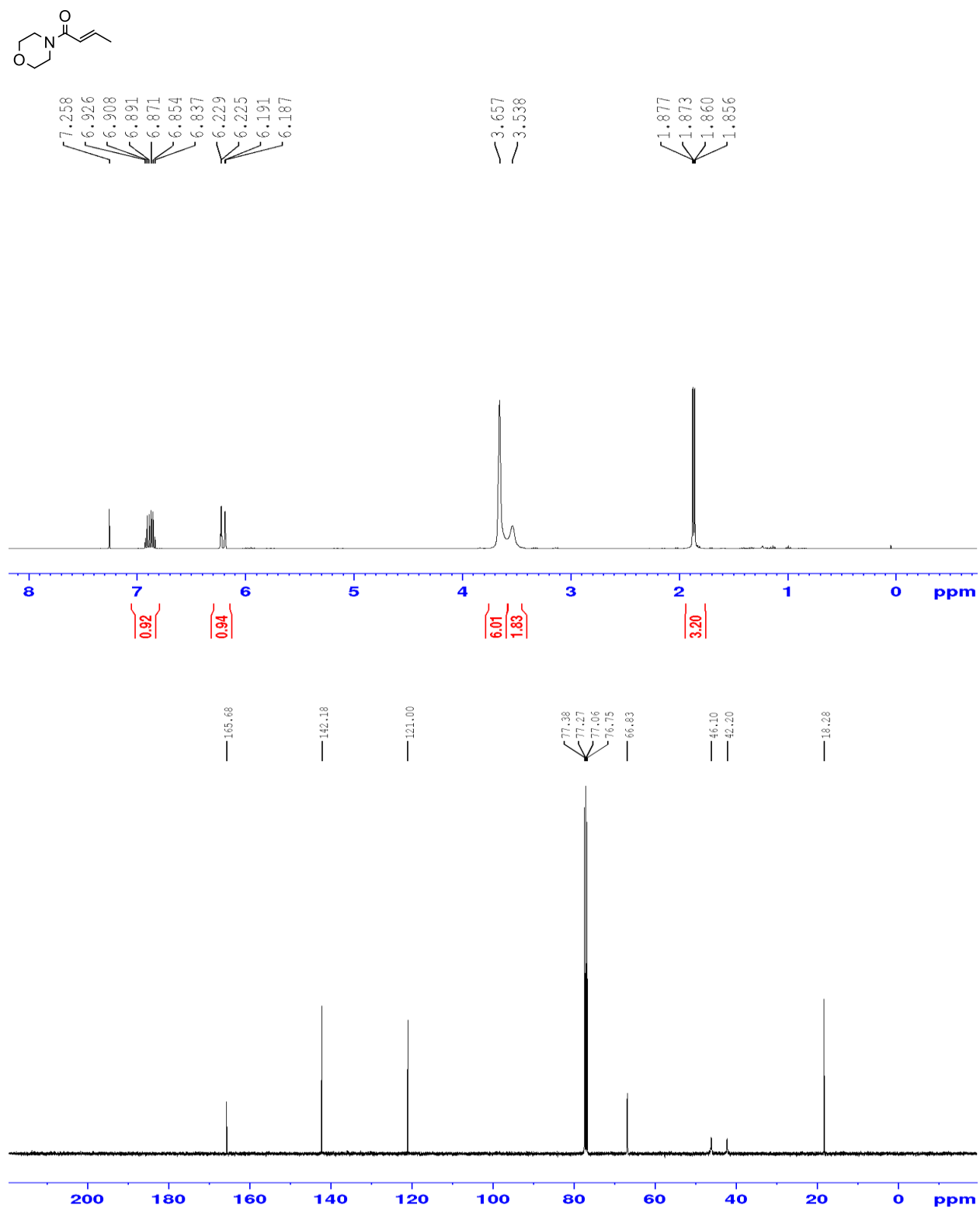


Figure S7 ¹H & ¹³C spectra of product 3a in CDCl₃. The spectra matched reference²¹

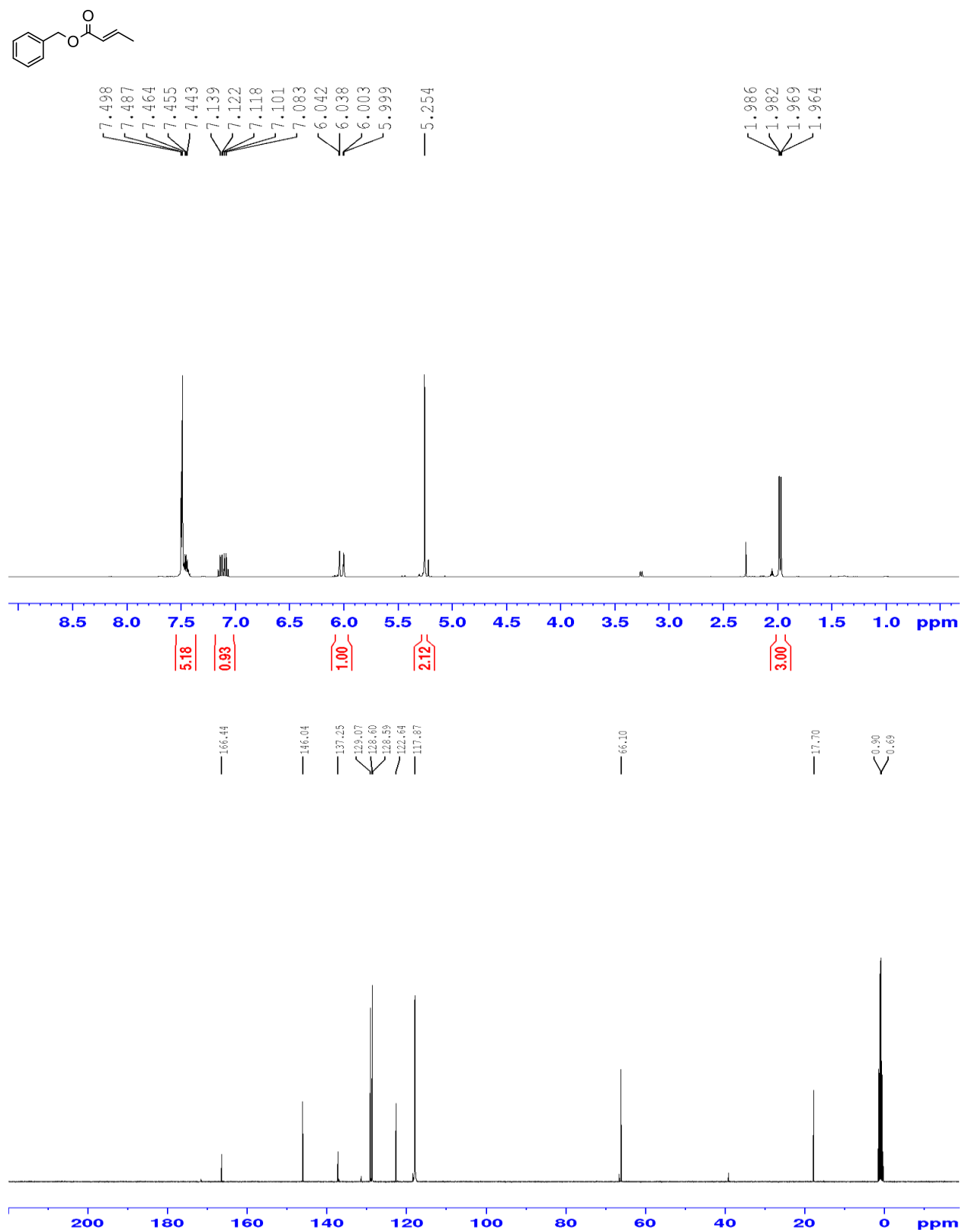


Figure S8 ¹H & ¹³C spectra of product 3b in in D-acetone. The spectra matched reference¹⁷

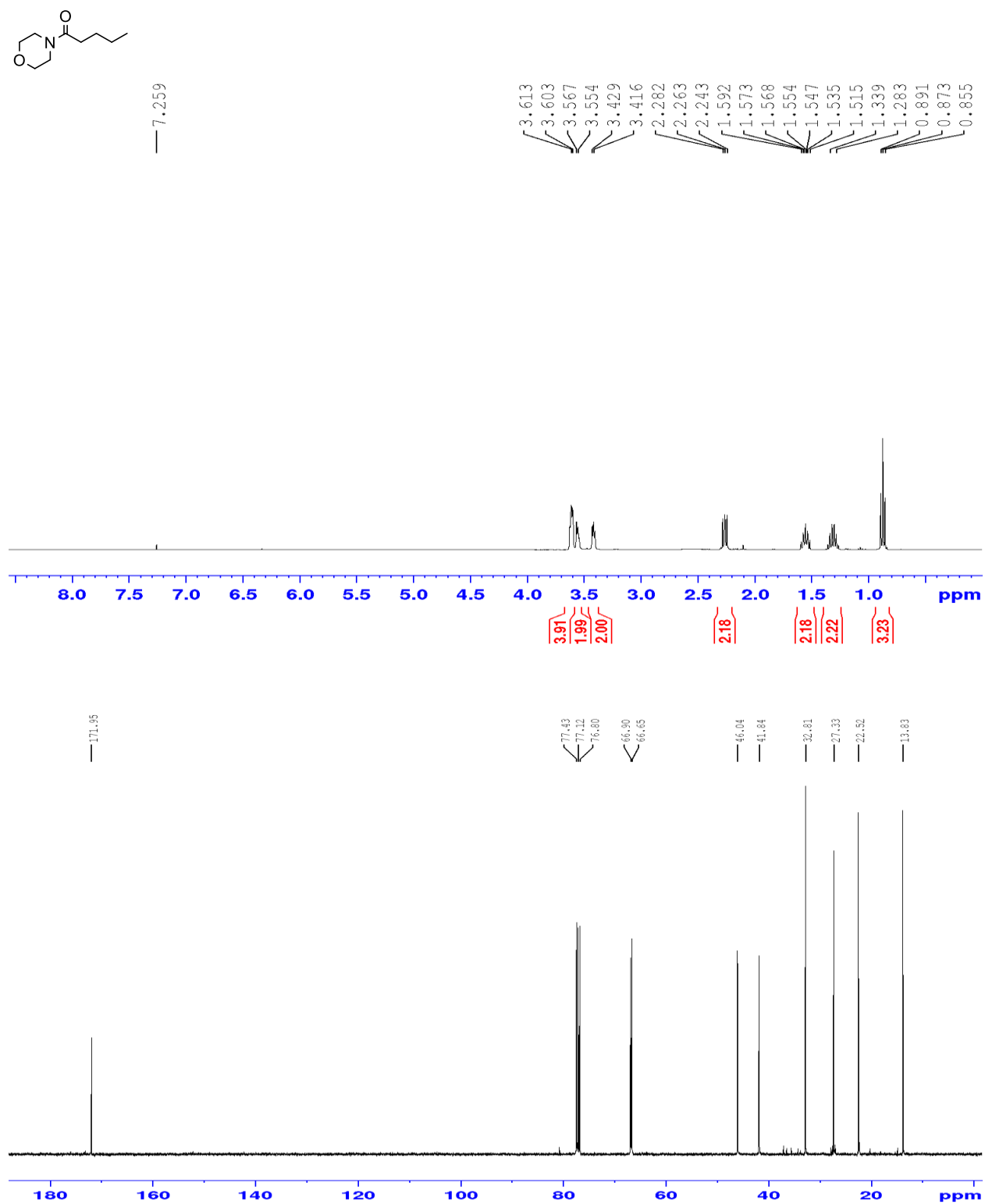


Figure S9 ¹H & ¹³C spectra of product 4a in CDCl₃. The spectra matched reference¹¹

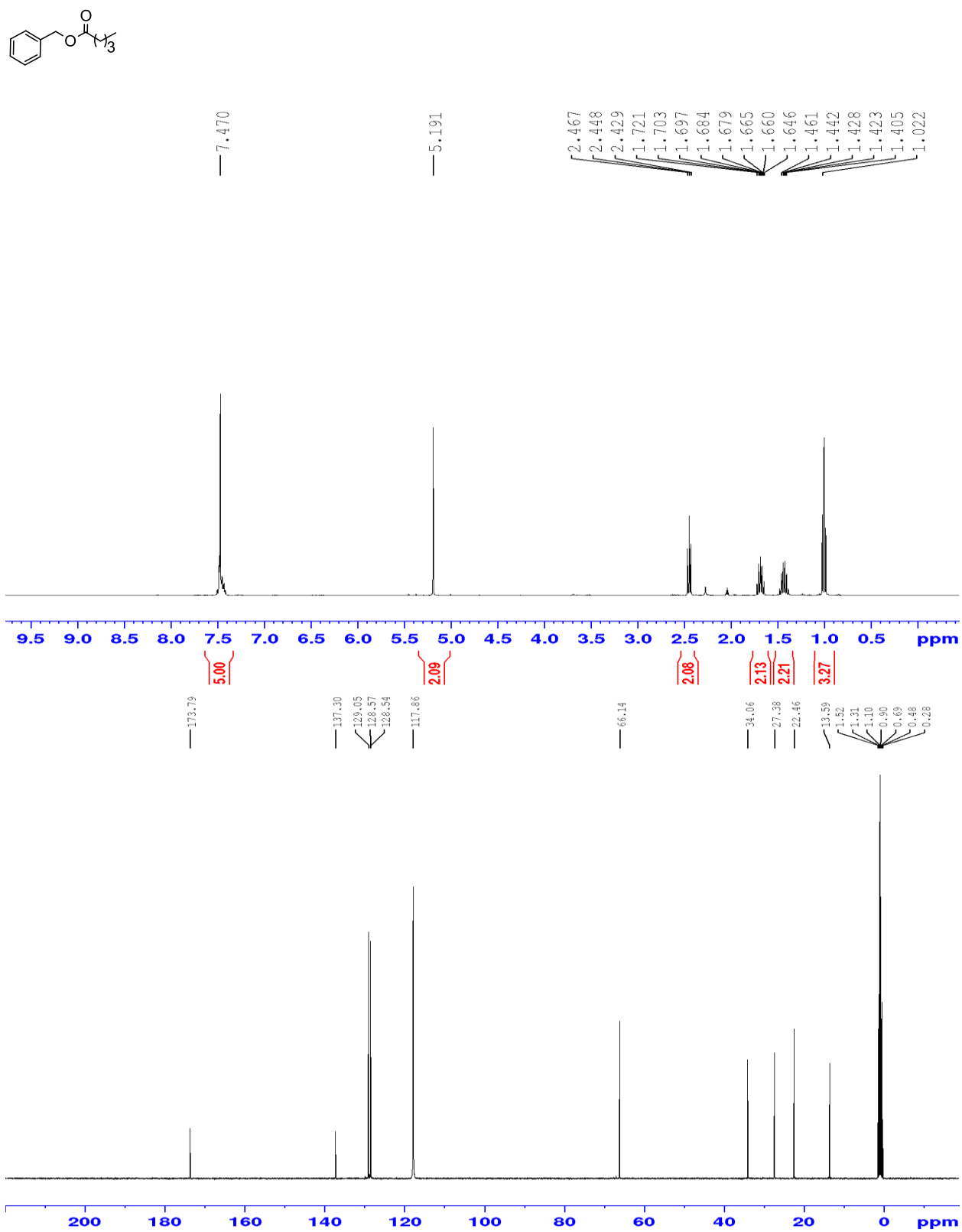


Figure S10 ^1H & ^{13}C spectra of product 4b in in D_2O . The spectra matched reference⁸

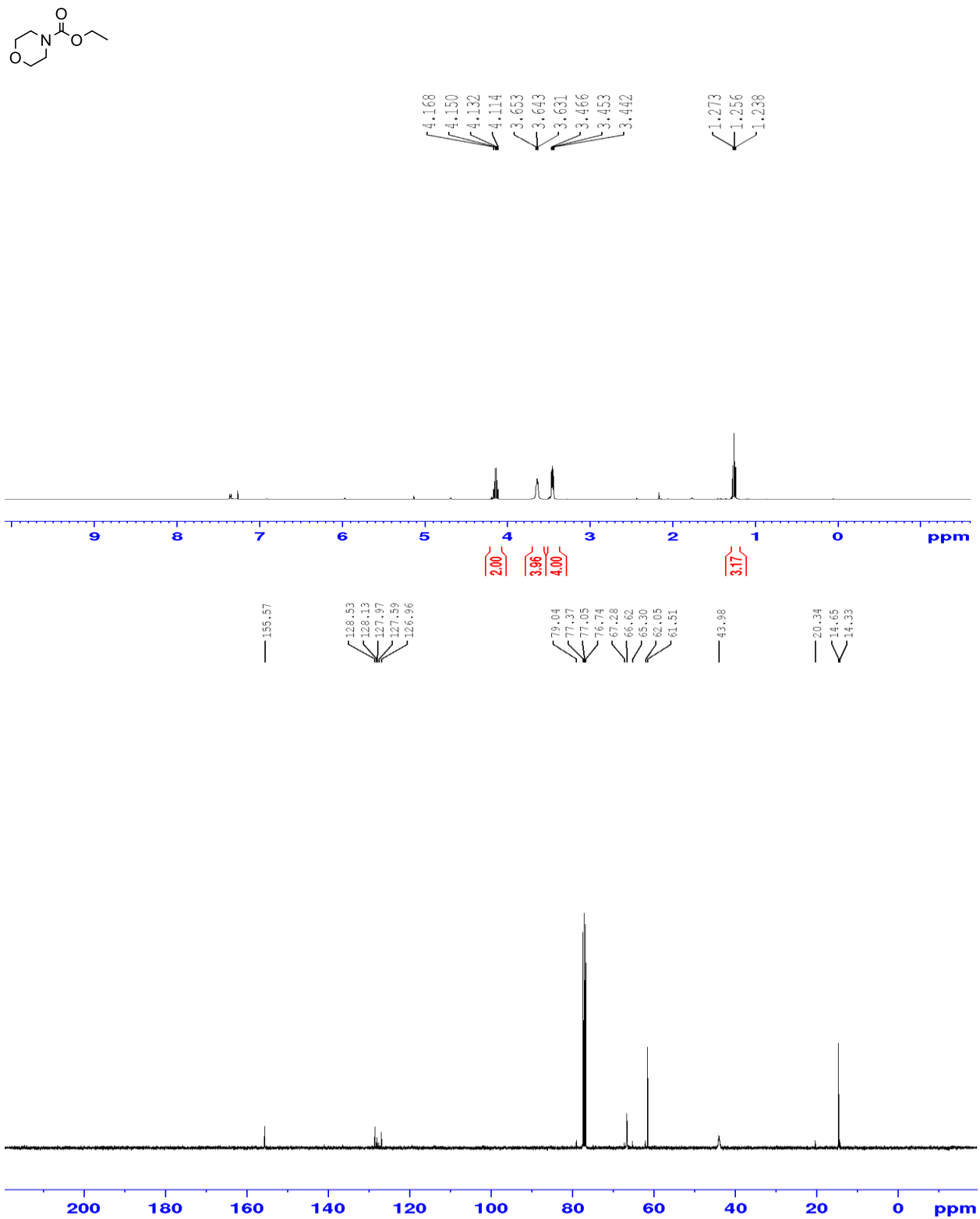


Figure S11 ¹H & ¹³C spectra of product 5a in CDCl₃. The spectra matched reference⁴

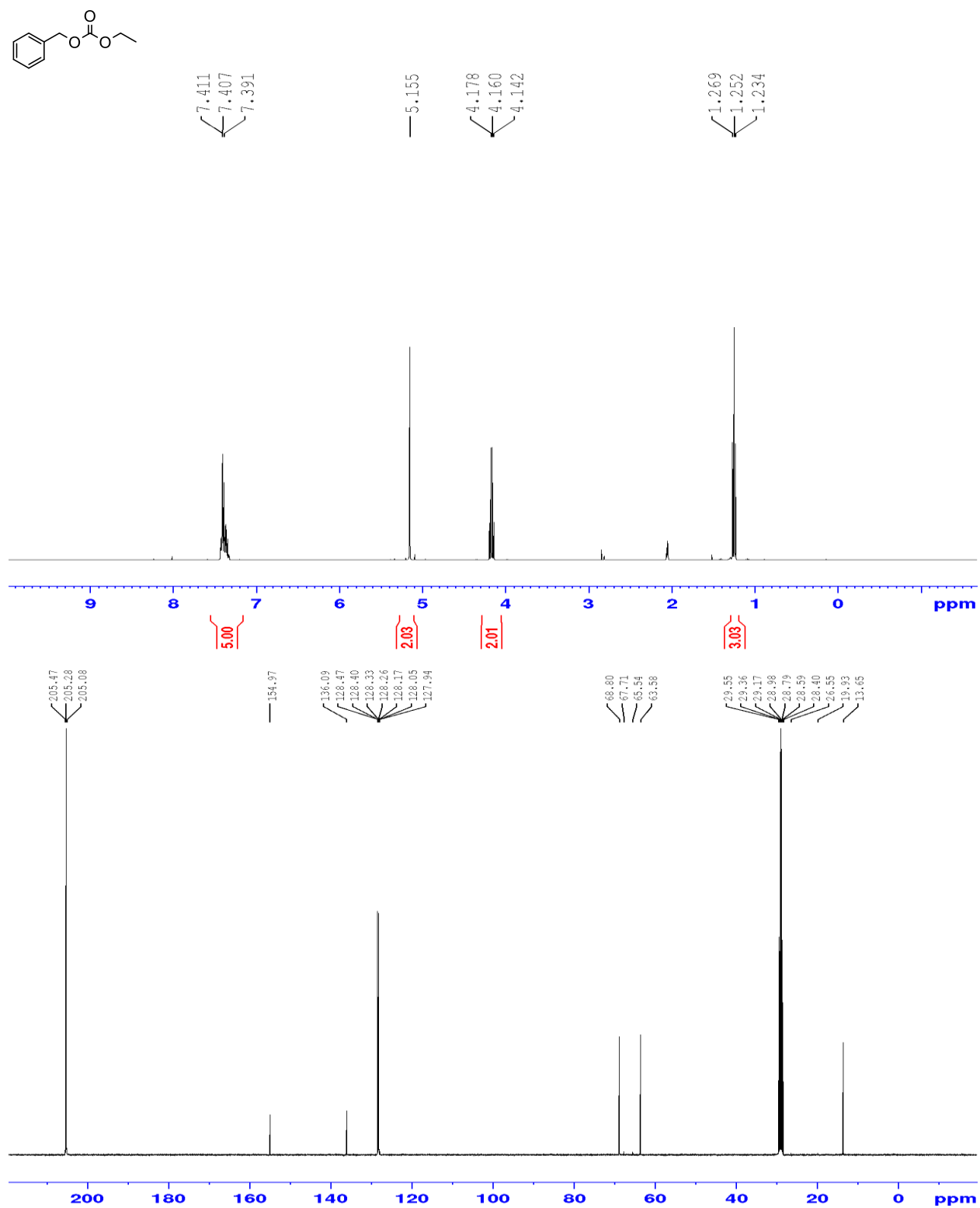


Figure S12 ¹H & ¹³C spectra of product 5b in D₂O. The spectra matched reference⁷

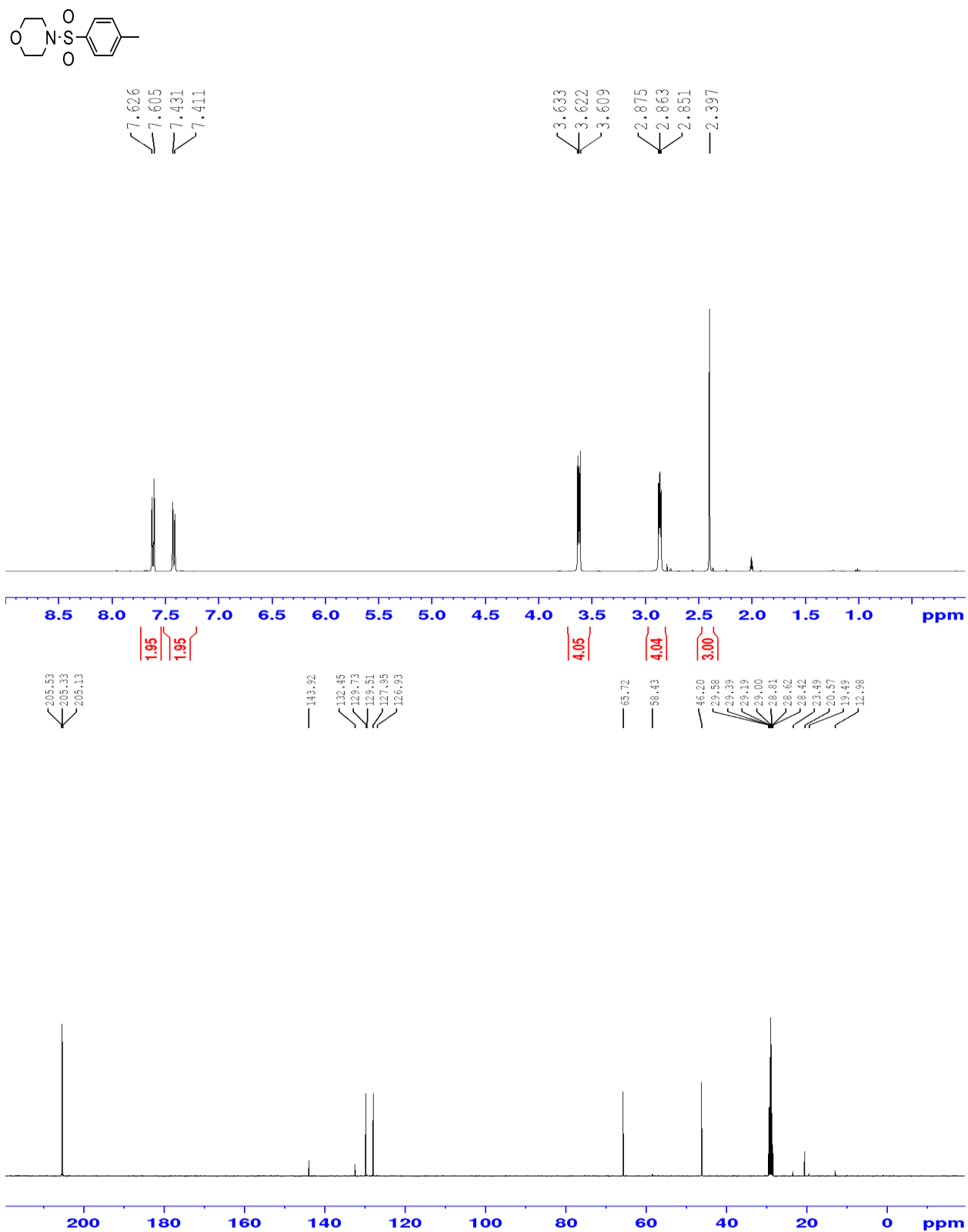


Figure S13 ^1H & ^{13}C spectra of product 6a in D-acetone. The spectra matched reference¹³

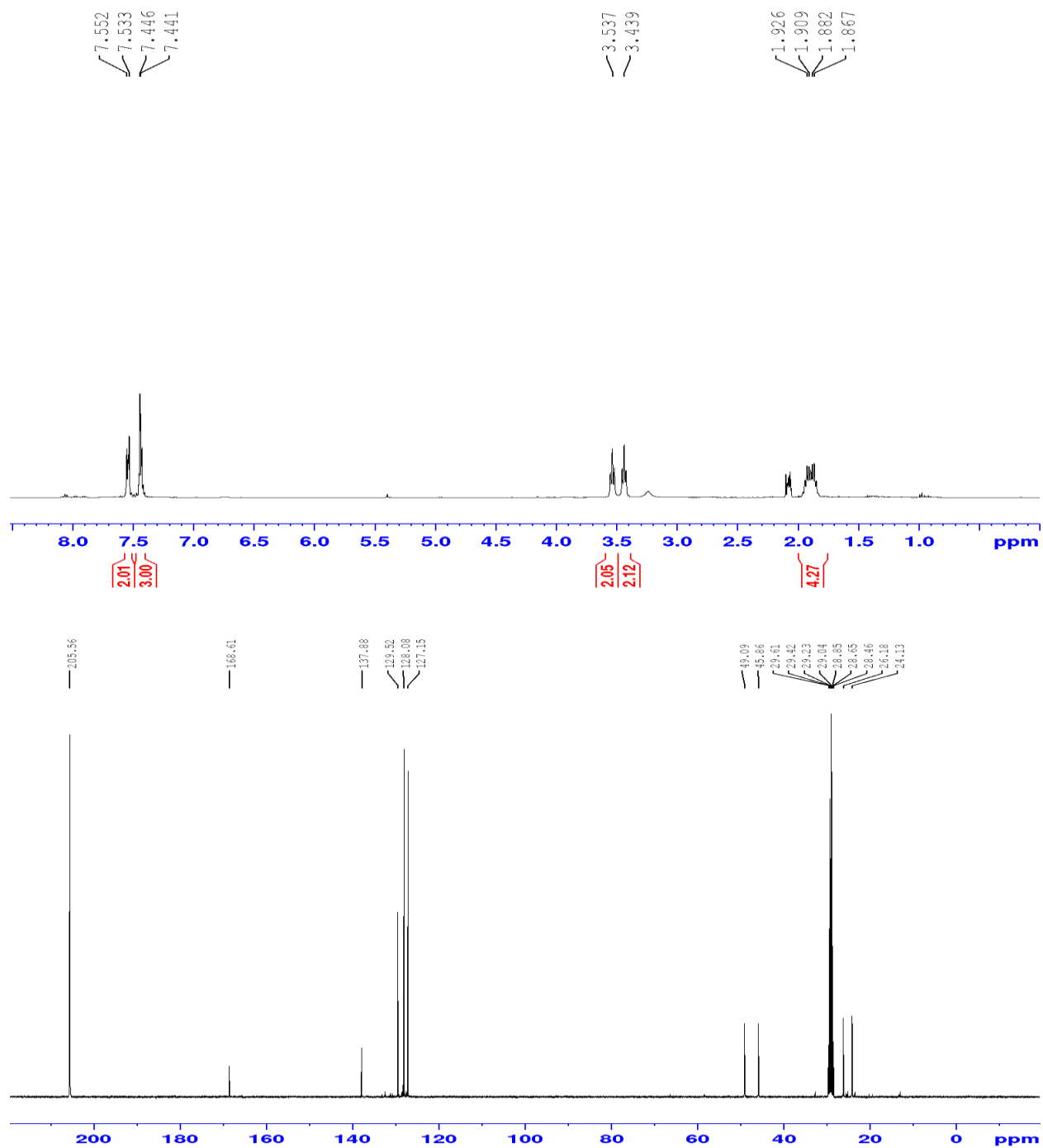
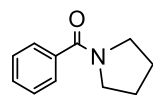


Figure S14 ¹H & ¹³C spectra of product 7 in D-acetone. The spectra matched reference¹²

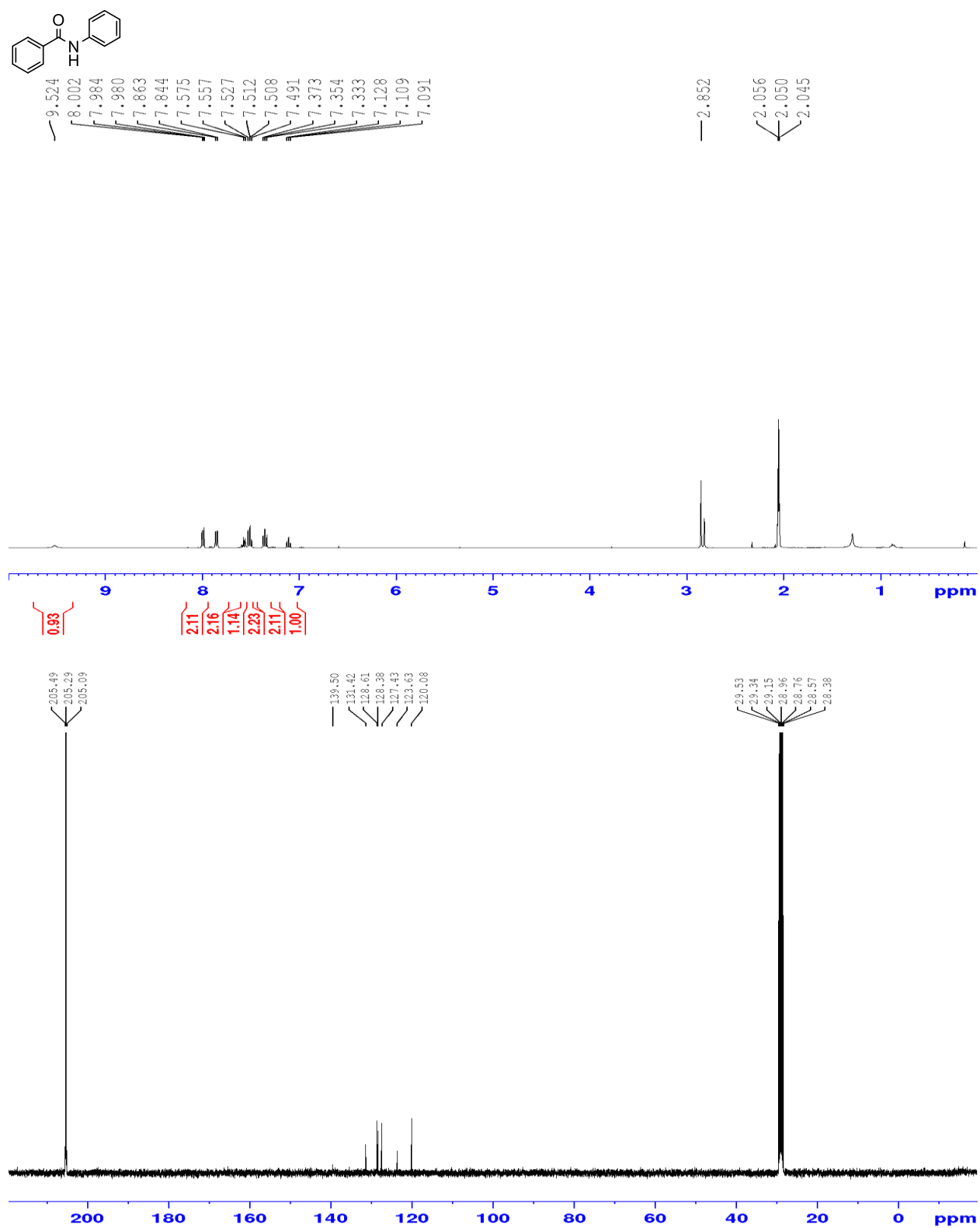


Figure S15 ¹H & ¹³C spectra of product 8 in D-acetone. The spectra matched reference¹⁵

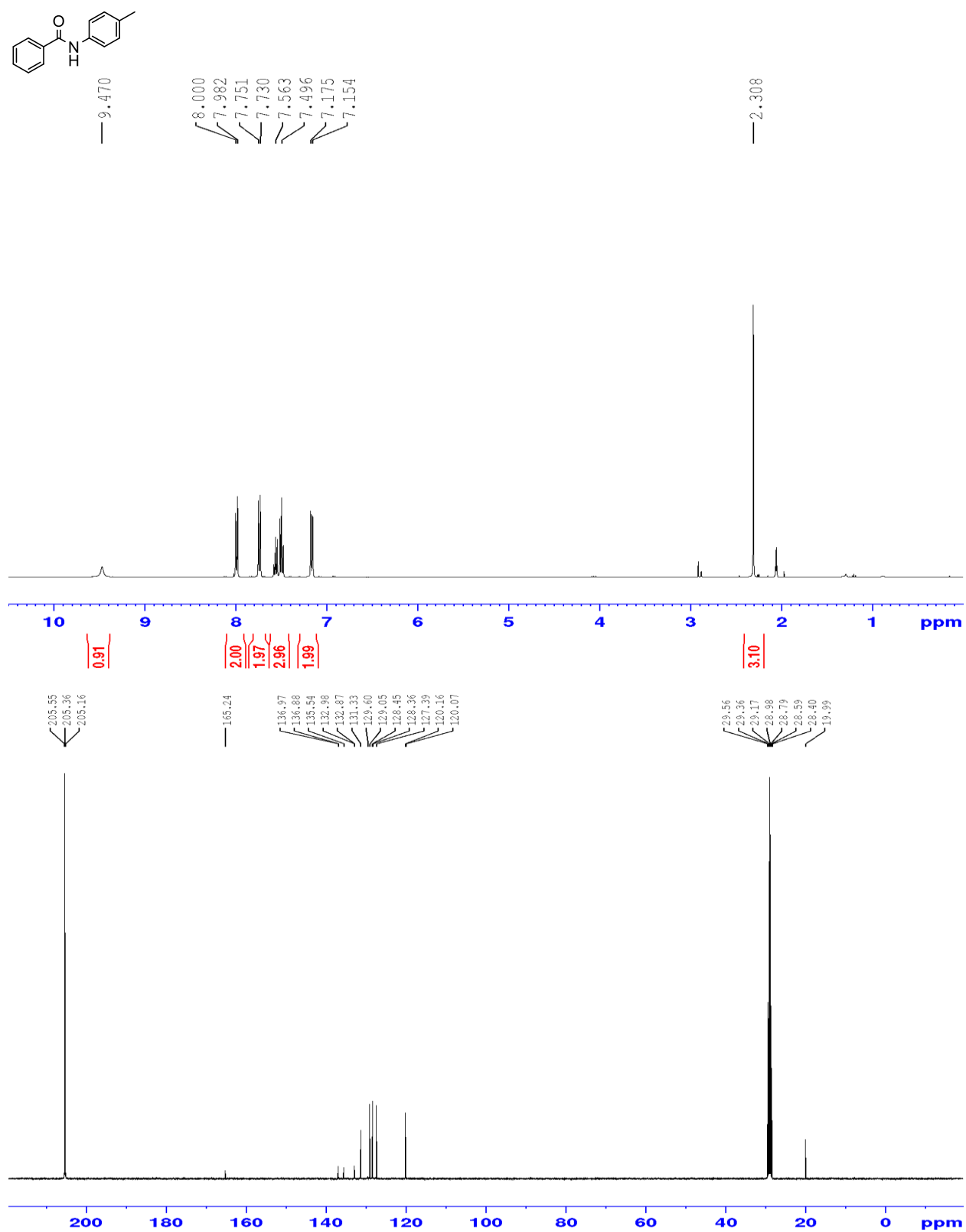


Figure S16 ^1H & ^{13}C spectra of product 9 in D-acetone. The spectra matched reference⁶

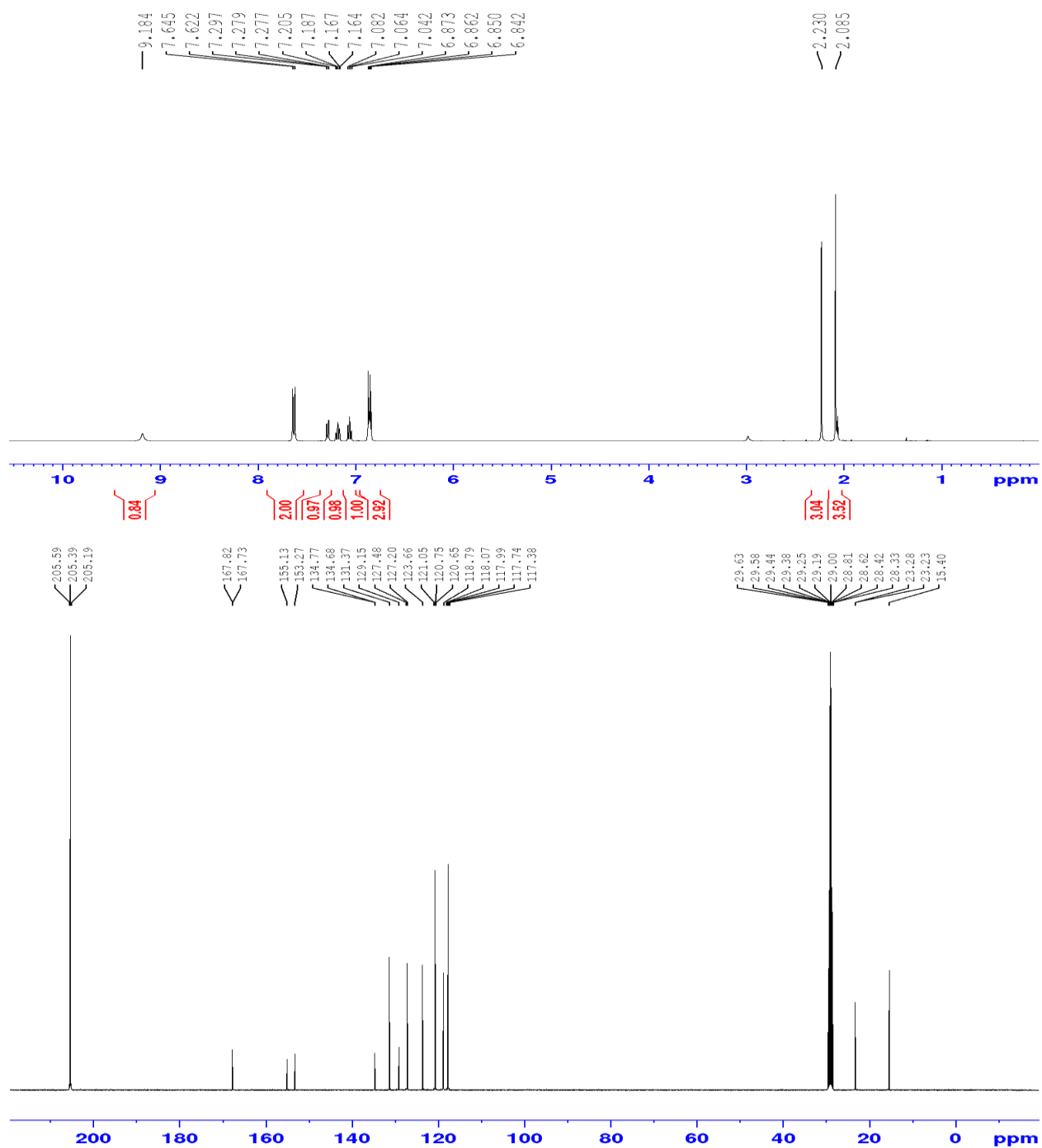
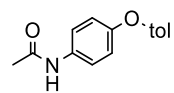


Figure S17 ¹H & ¹³C spectra of product 10 in D-acetone. The spectra matched reference²

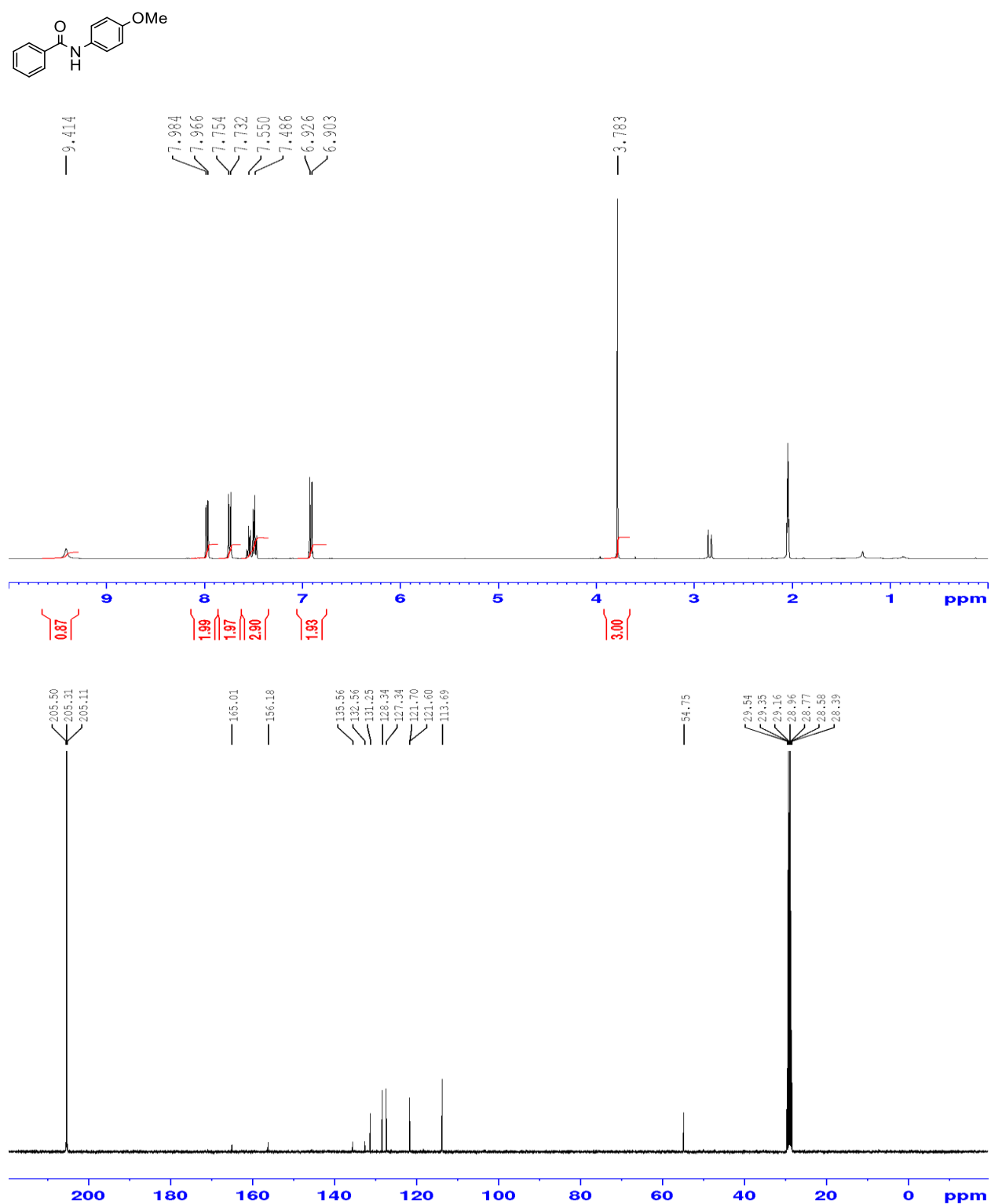


Figure S18 ¹H & ¹³C spectra of product 11 in D-acetone. The spectra matched reference²

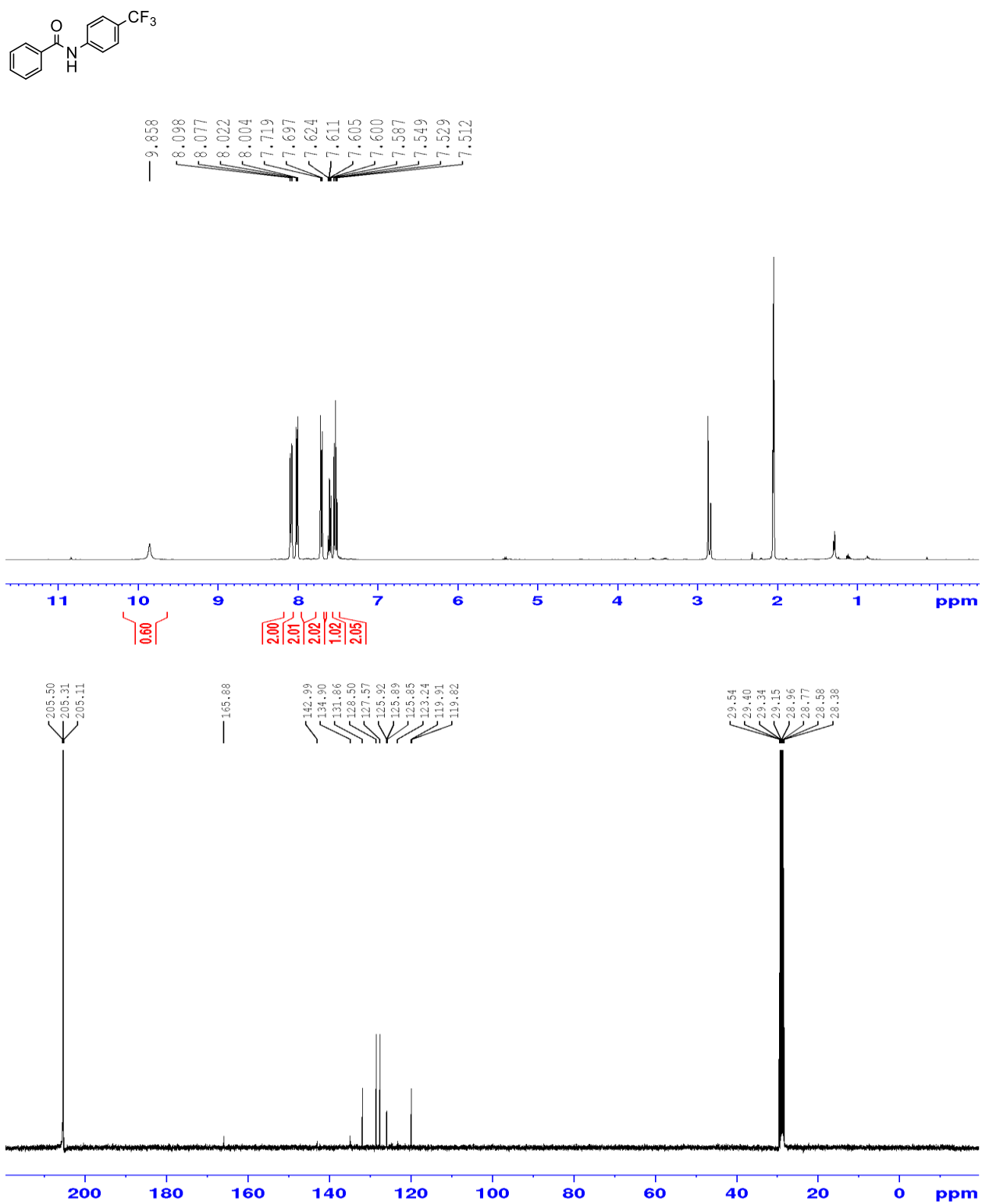


Figure S19 ¹H & ¹³C spectra of product 12 in D-acetone. The spectra matched reference¹⁶

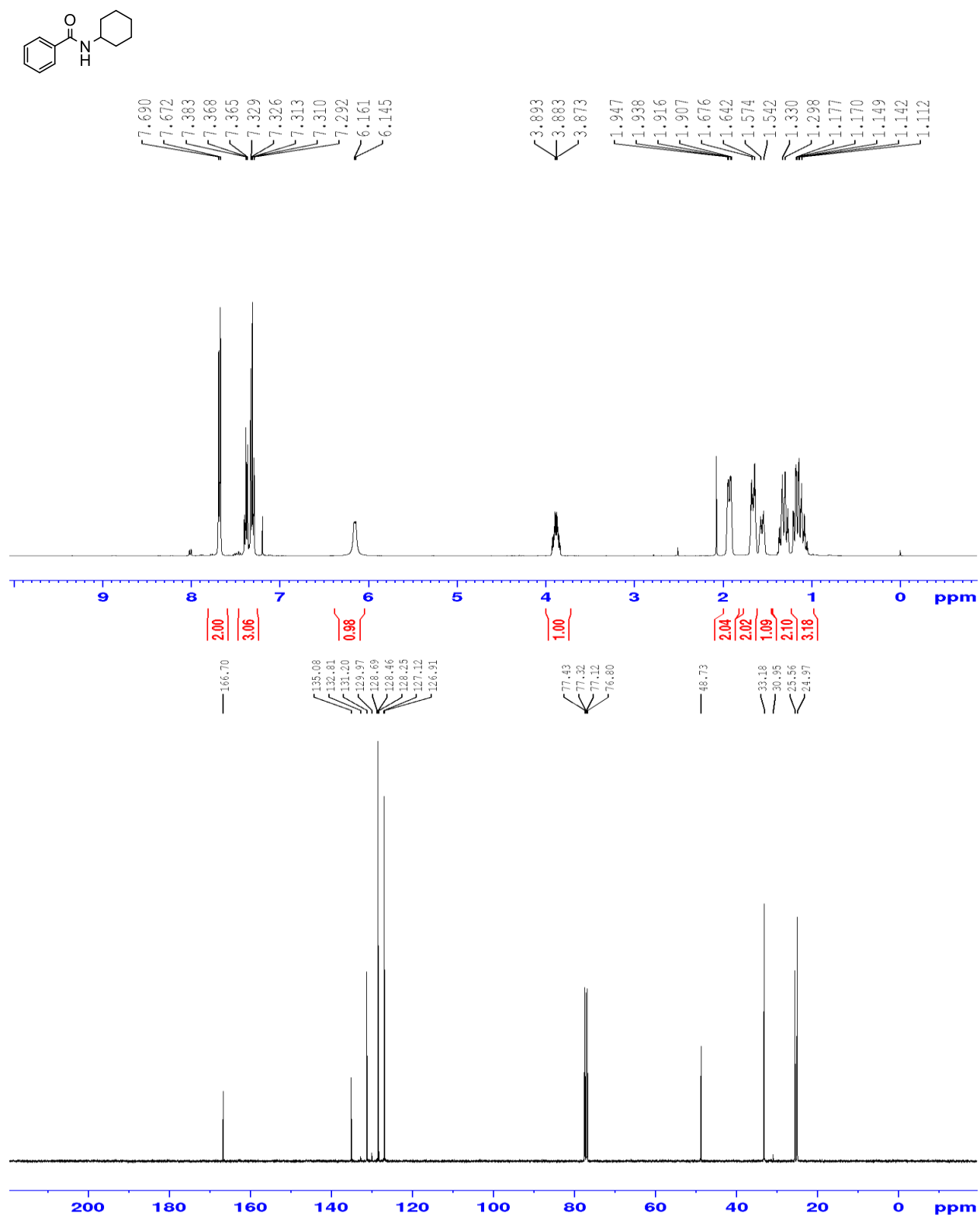


Figure S20 ¹H & ¹³C spectra of product 13 in CDCl₃. The spectra matched reference³

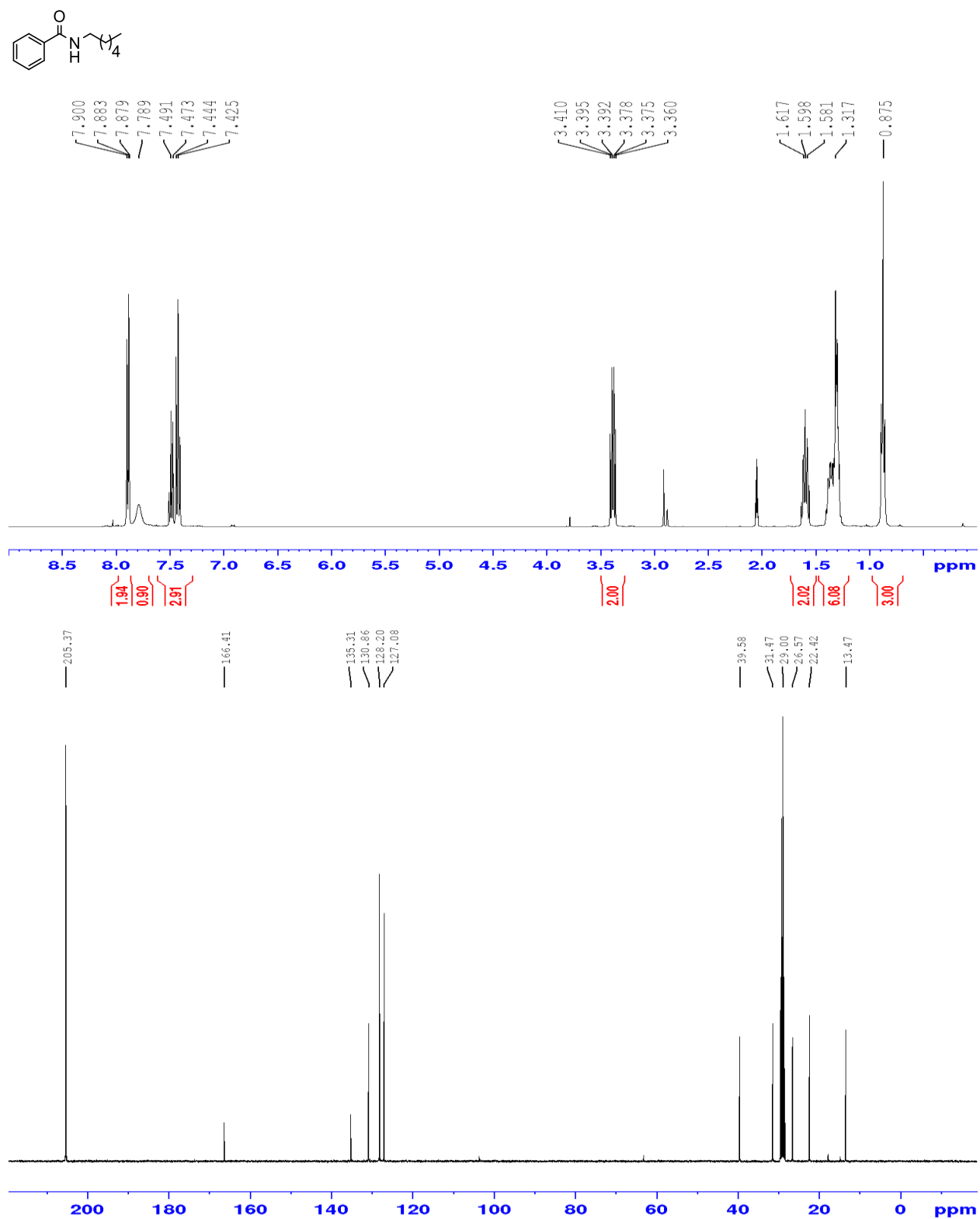


Figure S21 ¹H & ¹³C spectra of product 14 in D-acetone. The spectra matched reference¹⁰

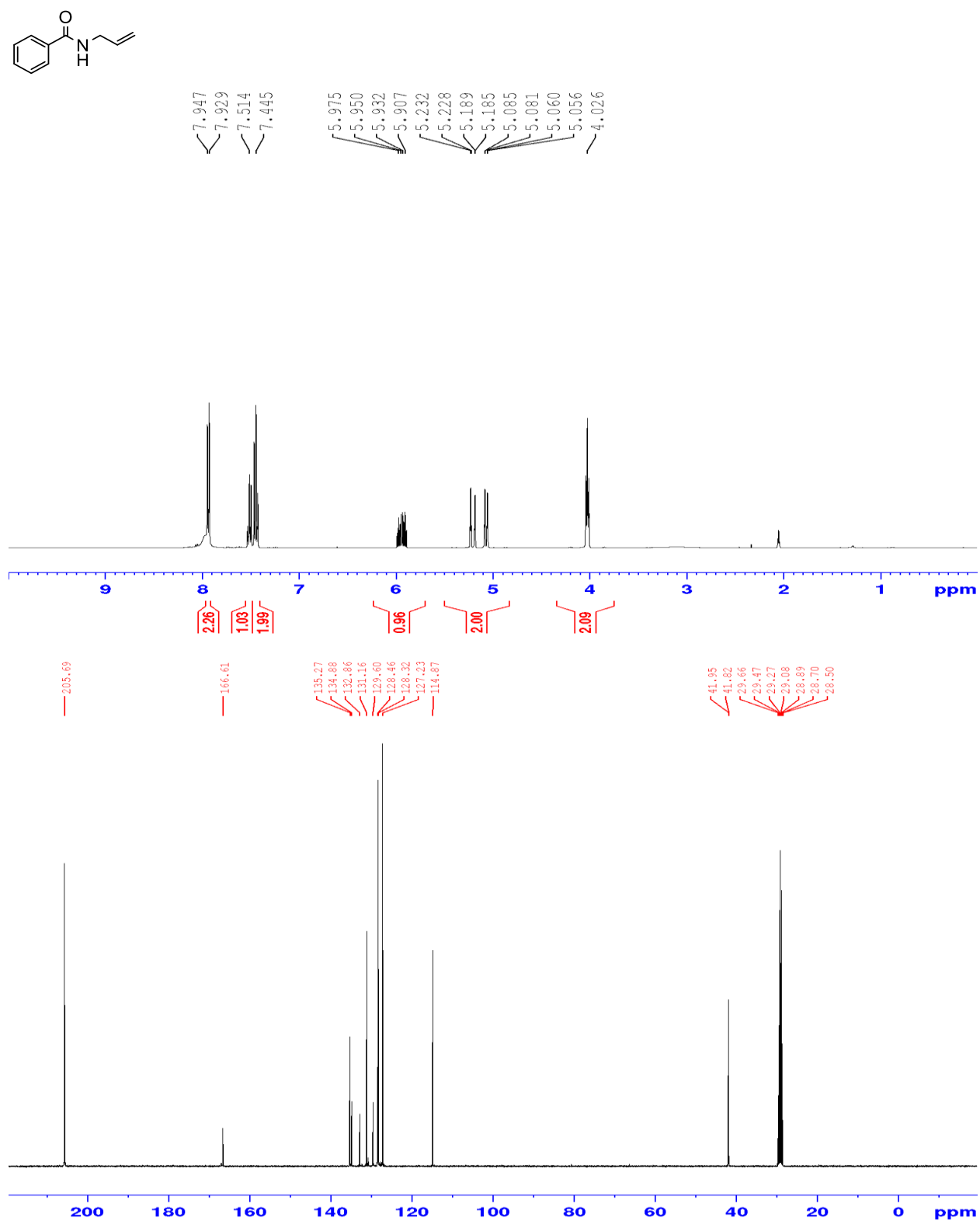


Figure S22 ¹H & ¹³C spectra of product 16 in D₂O. The spectra matched reference¹⁹

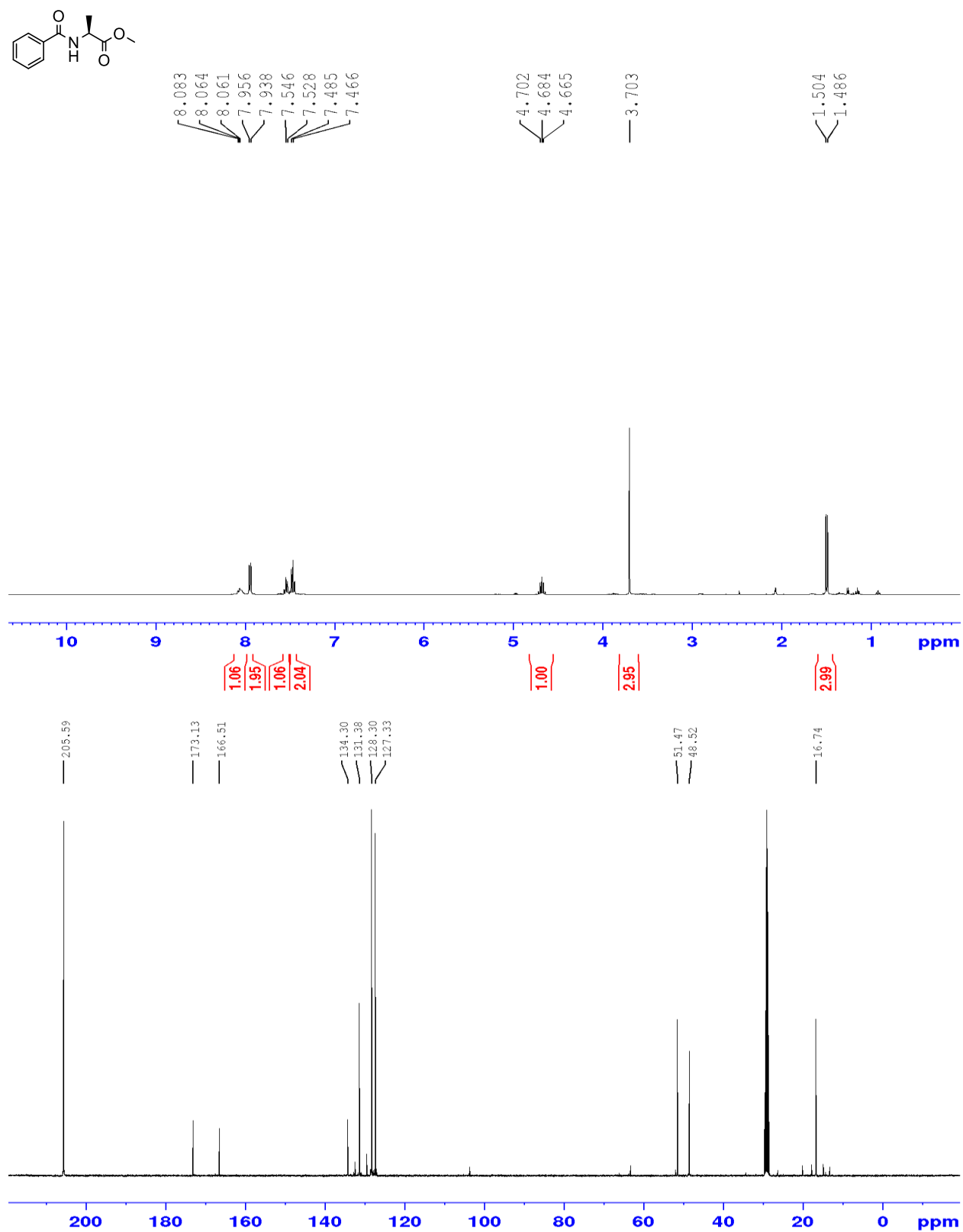


Figure S23 ¹H & ¹³C spectra of product 17 in D₂O. The spectra matched reference⁹

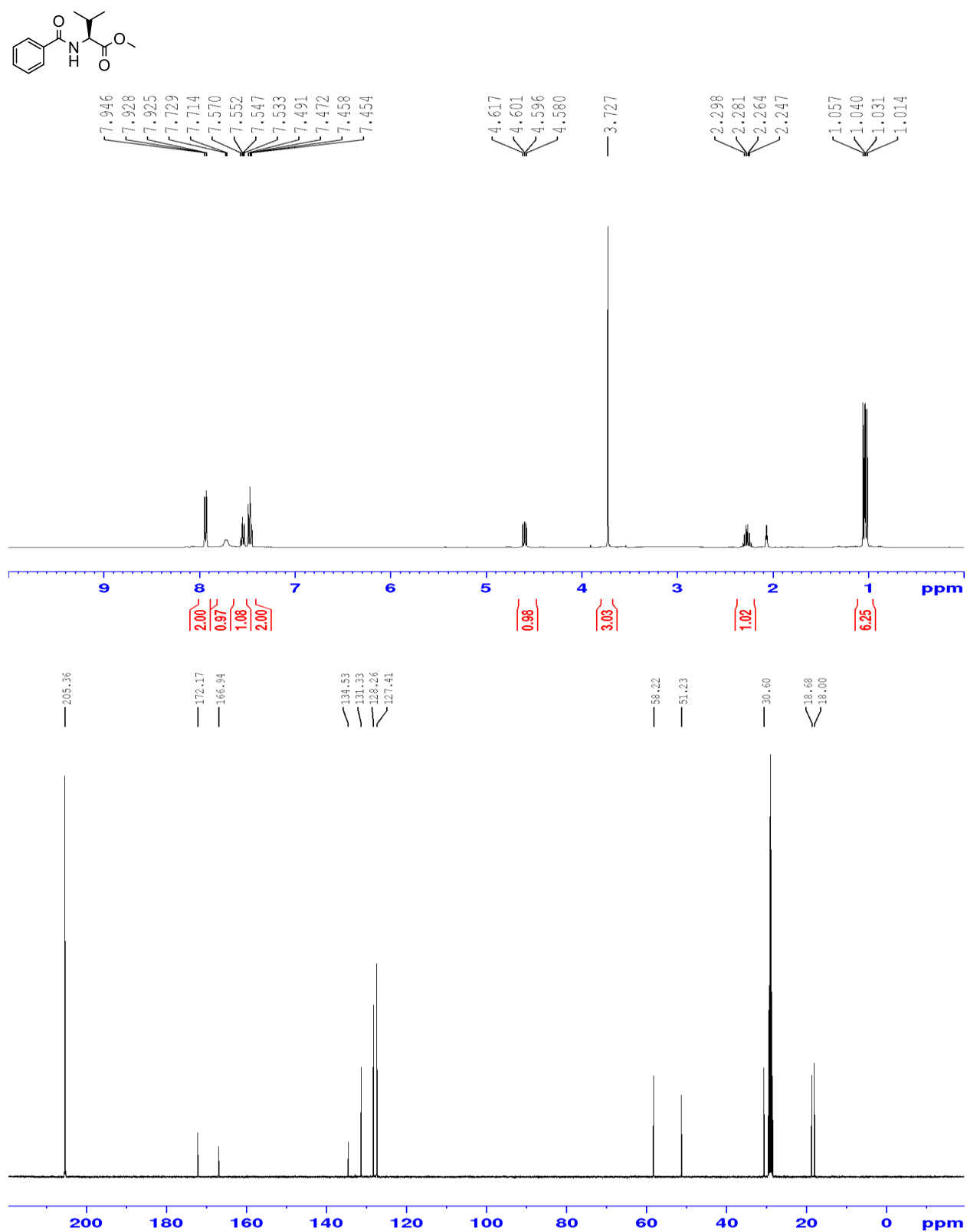


Figure S24 ¹H & ¹³C spectra of product 18 in D-acetone. The spectra matched reference¹⁴

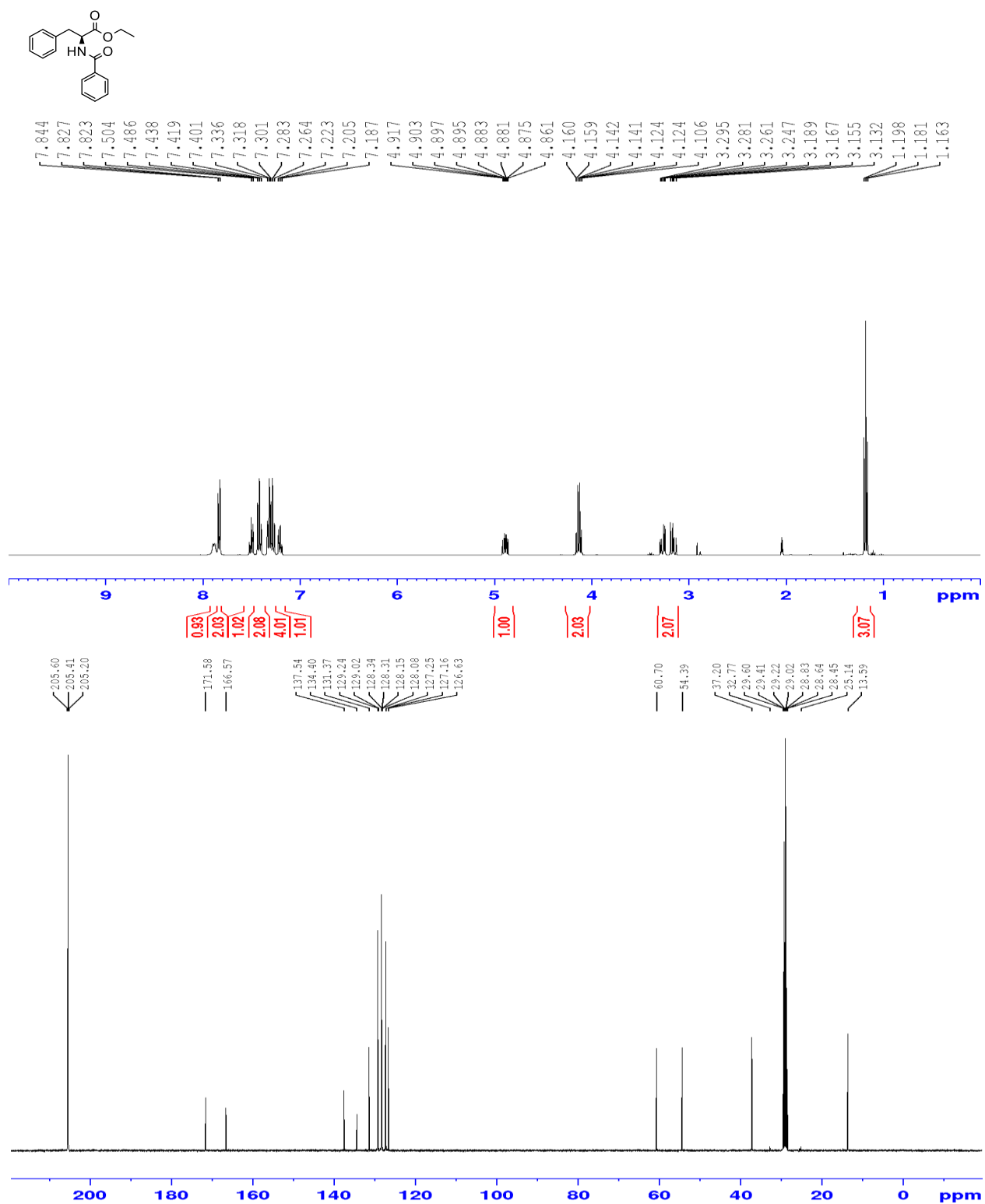


Figure S25 ^1H & ^{13}C spectra of product 19 in D_2O . The spectra matched reference¹⁴

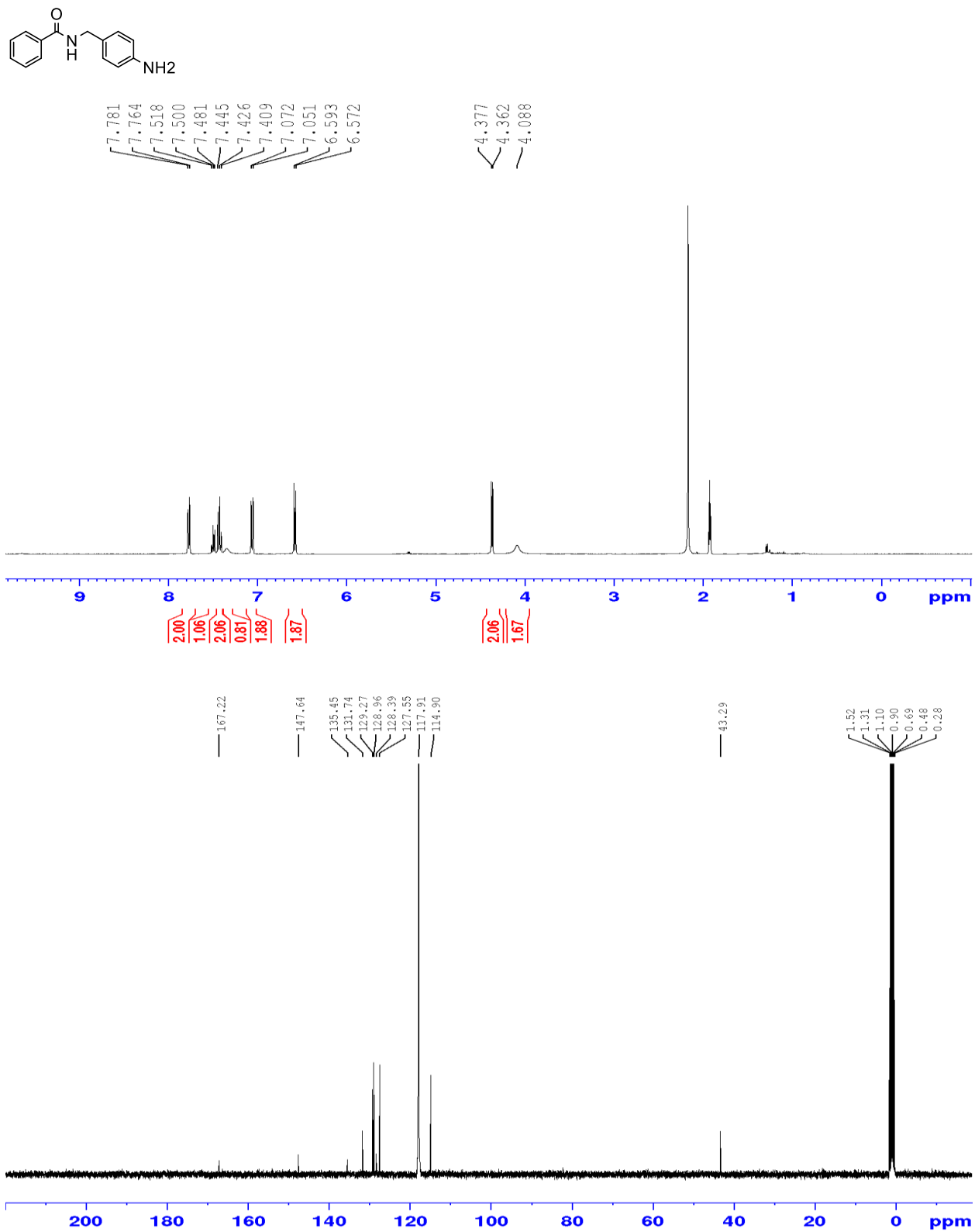


Figure S26 ¹H & ¹³C spectra of product 20 in D-acetone. The spectra matched reference¹⁴

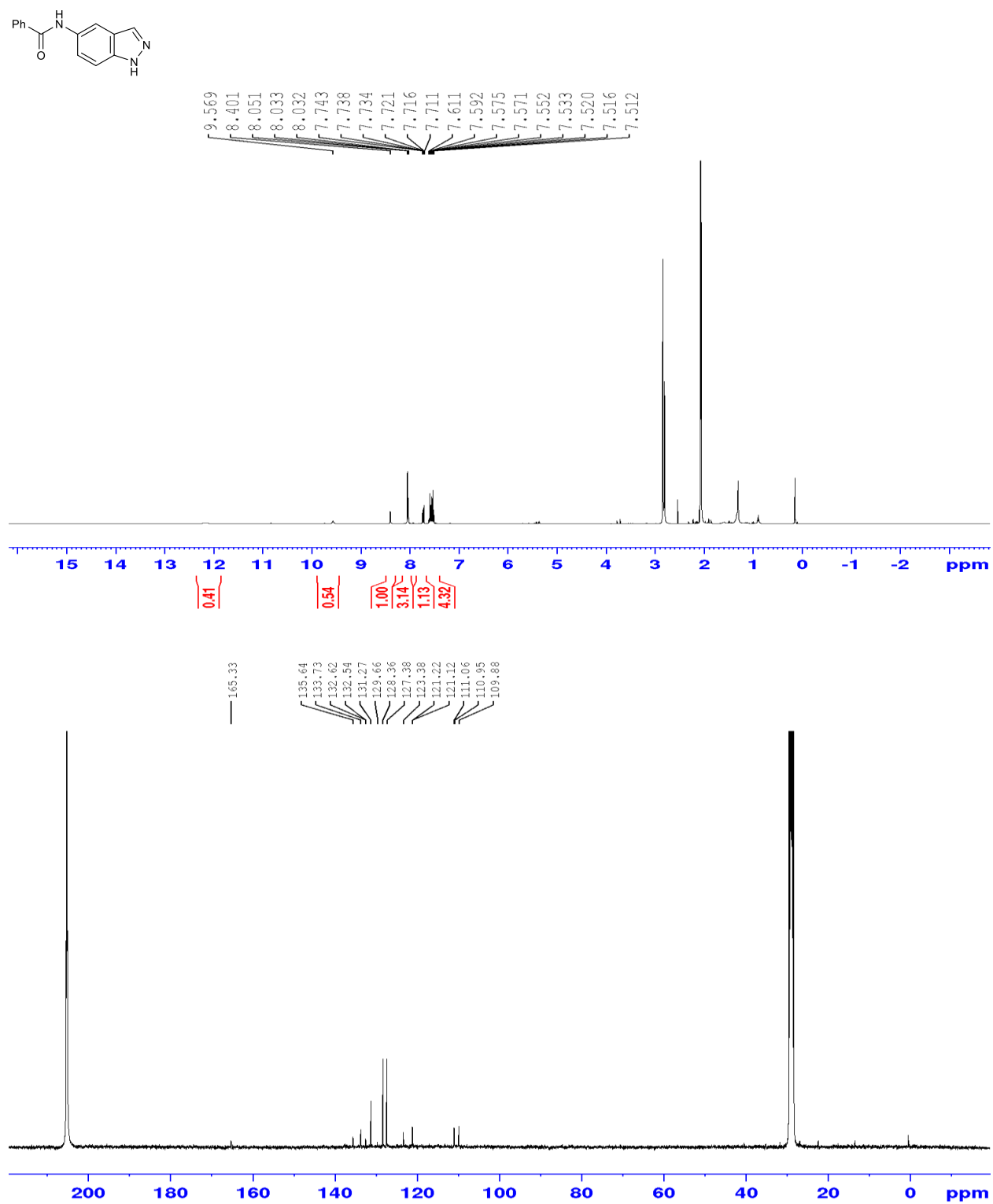


Figure S27 ¹H & ¹³C spectra of product 20. The spectra matched reference¹

PMI calculation for pyrrolidine amidation

A separate experiment was setup in an H cell using smaller volumes of solvent and electrolyte quantity. The volume of acetonitrile used for both cathode and anode was 3.5 mL. Each compartment contained 50 mM electrolyte. The cathode compartment contained 0.25 mmol of pyrrolidine and anode compartment contained 0.25 mmol ferrocene. After electrolysis is complete, an equivalent amount of benzoic anhydride was added to permit nucleophilic addition. It was followed by addition of an equivalent of benzyl bromide for nucleophilic substitution.

Mass of pyrrolidine = 0.018 g

Mass of benzoic anhydride = 0.056 g

Mass of benzyl bromide = 0.059 g

Mass of solvent = 5.5 g

Mass of electrolyte = 0.134 g

Mass of Ferrocene = 0.046 g

Mass of the amide = 0.096 g

Mass of the ester = 0.097 g

PMI = total mass/mass of products = 5.813/0.193 = 30.12

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