Supplementary information

Aqueous electrosynthesis of silver indium selenide nanocrystals and photothermal properties

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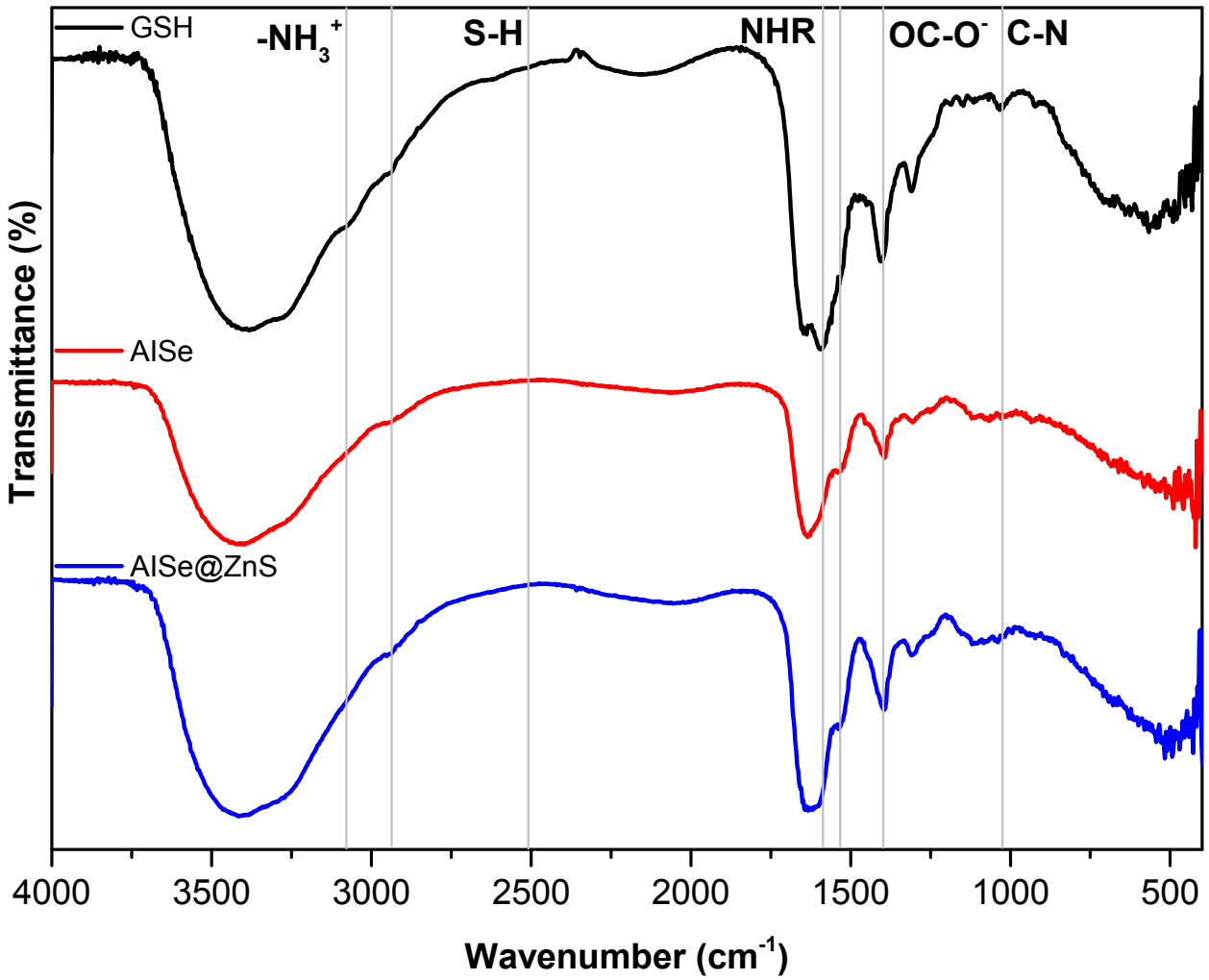

Figure S1. FTIR spectra of GSH, AISe and AISe/ZnS systems at pH 9.

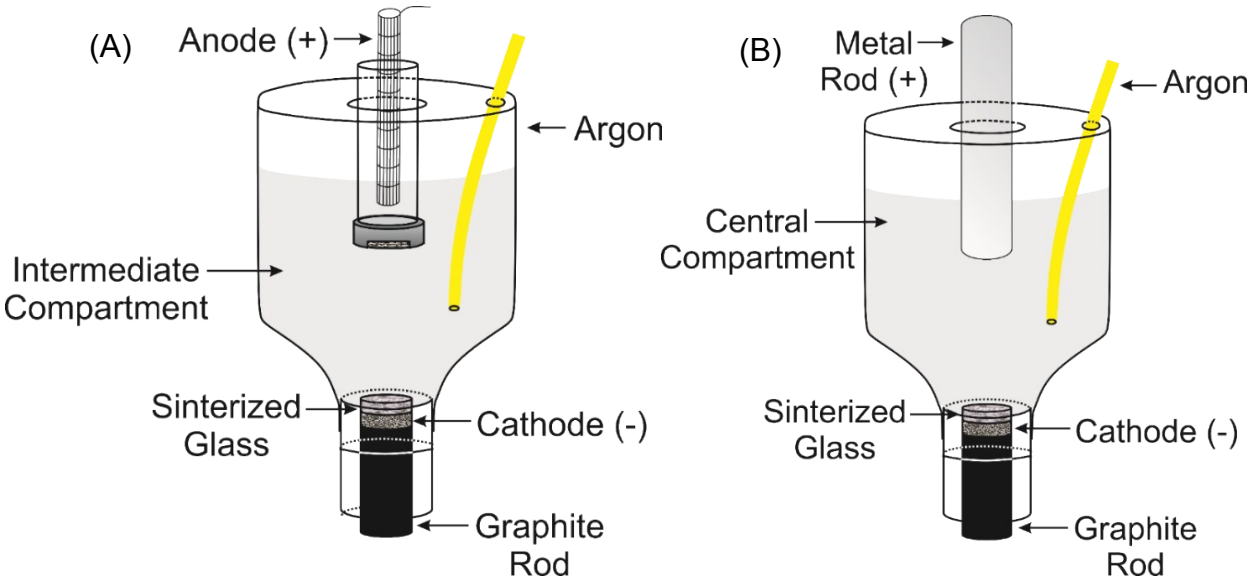

Figure S2: Electrochemical cell used for the synthesis of (A) AISe-GSH and (B) ZnS-GSH NCs (adapted from references 1 and 2).

Table S1. Emission wavelength and QY of AISe NCs synthesized by different methods.

entry	Strategy / Medium	Stabilizer	Chalcogen source	λ_{em}	QY (%)	Standard/λ_{exc}	Ref.
1	Hot injection (n-trioctylphosphine)	n-trioctylphosphine	Se^0	700 – 850 nm	< 5.0	-	[1]
2	Hot injection (1-octadecene)	Oleylamine/ dodecanethiol	Se^0	619 nm	6.9	R101/450 nm	[2]
3	Hot injection (aqueous)	L-GSH	$Se^0/NaBH_4$	642 nm	10.3	R6G / 382 nm	[3]
4	Hydrothermal (aqueous)	Gelatine/TGA	$Se^0/NaBH_4$	575 – 650 nm	< 5.0	Integrating sphere	[4]
5	**Electrochemical (aqueous)**	**L-GSH**	**Se^0**	**610 nm**	**18.6**	**R6G / 488 nm**	**This work**
6	**Electrochemical (aqueous)**	**L-GSH**	**Se^0/S^0**	**634 nm**	**16.0**	**R6G / 488 nm**	**This work**

References

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