Characterization of heavy metal binding properties of periplasmic metal uptake protein CLas-ZnuA2

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Supplementary figures and tables

Table S1: Metal coordination distances in Mn-specific available structures of Cluster A-I

proteins. The values indicated by * represents the higher than ideal coordination distance.

| Protein | Metal ion | Metal coordinating residues | | | | | | Observed geometry | Ref. |
|------------|--------------|-----------------------------|--------|--------|-------|--------|-------|---------------------|------|
| | | His39 | His106 | Glu172 | | Asp247 | | - | |
| | | NE2 | NE2 | OE1 | OE2 | OD1 | OD2 | - | |
| CLas-ZnuA2 | Cd2+ | 2.24 | 2.23 | 2.23 | 2.20 | 2.25 | 2.23 | Octahedral | |
| CLas-ZnuA2 | Mn2+ | 2.26 | 2.23 | 2.31* | 2.19 | 2.22 | 2.18 | Square pyramidal | [1] |
| | Zn2+ | 2.08 | 2.07 | 2.51* | 2.00 | 2.04 | 2.03 | Square pyramidal | [2] |
| PsaA | Mn2+ | 2.11 | 2.15 | 2.08 | 2.39* | 2.43* | 2.14 | Tetrahedral | [3] |
| | Zn2+ | 2.00 | 2.01 | 2.05 | 2.60* | 2.03 | 2.84* | Tetrahedral | [4] |
| Sau-MntC | Mn2+ | 2.05 | 2.07 | 2.28 | 2.77* | 2.32* | 2.22 | Tetrahedral | [5] |

| | Zn2+ | 2.05 | 2.10 | 2.56* | 2.07 | 2.05 | 2.54* | Tetrahedral | [6] |
|----------|------|------|------|-------|-------|-------|-------|-------------|-----|
| MtsA | Fe | 2.09 | 2.02 | 2.12 | 2.44* | 2.79* | 1.99 | Tetrahedral | [7] |
| Syn-MntC | Mn2+ | 2.17 | 1.87 | 2.28 | 2.42* | 2.86* | 1.93 | Tetrahedral | [8] |

Table S2: Average values of RMSD, radius of gyration, SASA and intra-H bond of native CLas-**ZnuA2 native, CLas-ZnuA2** -H106C Mutant and CLas-ZnuA2 -D247C Mutant.

| S.No. | Compound | RMSD (Å) | Gyration (Å) | SASA (Å ²) | Intra-hydrogen number |
|-------|-------------------------|-------------|-----------------|------------------------|--------------------------|
| 1 | CLas-ZnuA2- Native | 0.96 | 19.38 | 12433.2 | 265.2 |
| 2 | CLas-ZnuA2-H106C Mutant | 1.25 | 19.31 | 12286.3 | 262.1 |
| 3 | CLas-ZnuA2-D247C Mutant | 1.38 | 19.48 | 12681.3 | 260.8 |

 Table S3: Kinetic parameters of CLas-ZnuA2 interactions with different divalent metal ions:

association rates, dissociation rates and equilibrium constants are given.

| Analyte | k _{a1} | k _{d1} | K _{a2} | k _{d2} | K _D |
|------------------|------------------------|-----------------|------------------------|--------------------|------------------------|
| (divalent | $(M^{-1}s^{-1})$ | (s^{-1}) | (s ⁻¹) | (s ⁻¹) | (M) |
| metal) | | | | | |
| Ba ²⁺ | 8.2 x 10 ⁻⁵ | 50.54 | 0.003511 | 0.02283 | 5.3 x 10 ⁻⁵ |
| Cd^{2+} | 1.8 x 10 ⁸ | 1571 | 6.6 x 10 ⁻⁴ | 0.003931 | 7.1 x 10 ⁻⁶ |
| Co ²⁺ | $1.2 \ge 10^4$ | 0.6472 | 0.001923 | 0.009397 | 4.3 x 10 ⁻⁵ |
| Hg^{2+} | 1.1 x 10 ⁴ | 0.1420 | 0.02584 | 0.01579 | 4.8 X 10 ⁻⁶ |

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Figure S1. Circular diachroism studies (A), (B) and (C) Shows thermal denaturation of wildtype, mutant H106C and D247C monitored by CD spectroscopy at 222 nm in presence of Cd^{2+} and Ba^{2+} respectively.