

## Supporting Information

### Epitaxial patterned $\text{Bi}_2\text{FeCrO}_6$ nanoisland arrays with room temperature multiferroic properties

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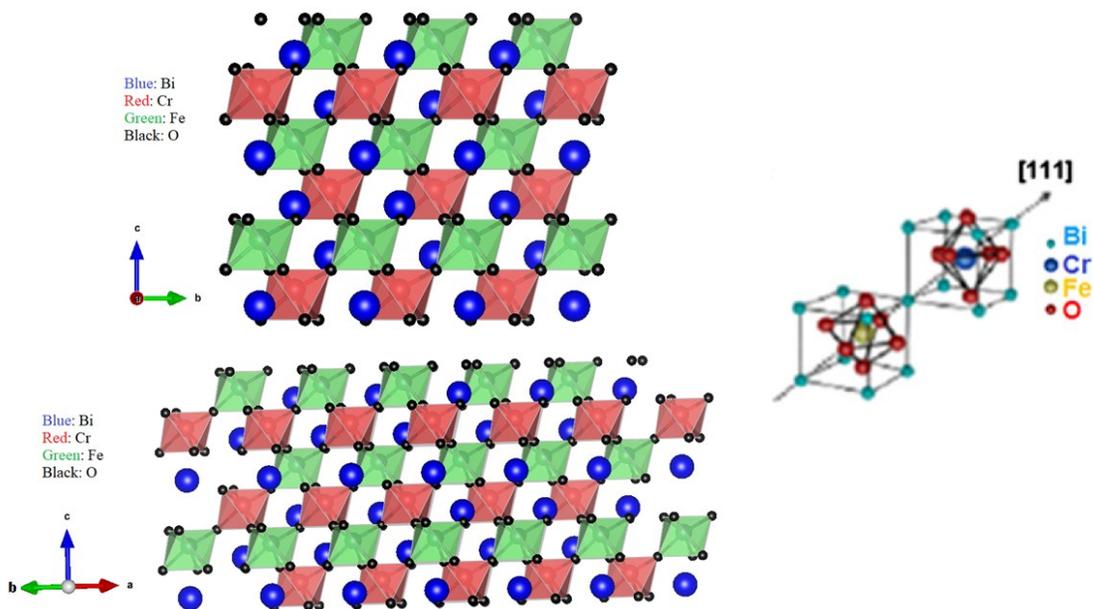
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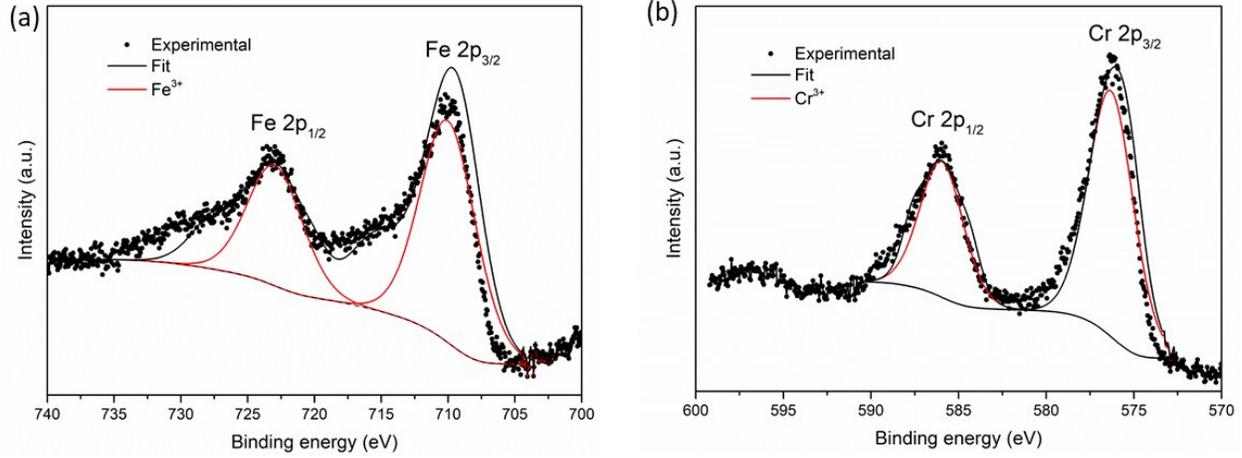
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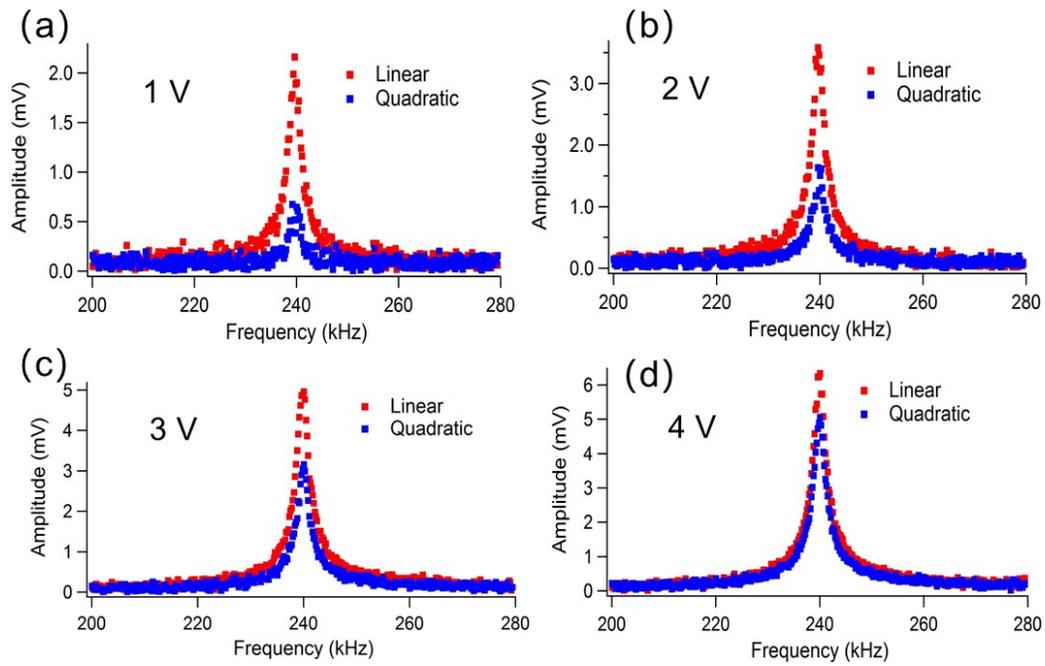


**Figure. S1.** Left: Crystal structure of the double perovskite  $\text{Bi}_2\text{FeCrO}_6$  from different perspectives. O and Bi atoms are denoted by the smallest ball with black and the biggest ball with blue, respectively. Fe and Cr octahedra are colored green and red, respectively. Right: double-perovskite structure involved in perovskite Bi-Fe-O and perovskite Bi-Cr-O unit cells, along Fr-Cr ions [111] direction. Adapted from ref. 1.



**Figure S2.** X-ray photoelectron spectroscopy (XPS) of (a) Fe 2p and (b) Cr 2p lines for BFCO patterned nanostructures on Nb-SrTiO<sub>3</sub> (100) substrate.

The Fe 2p XPS spectra reveal the 2p<sub>3/2</sub> and 2p<sub>1/2</sub> doublets arising from spin-orbit splitting (Fig. S2(a)). The binding energies for Fe2p<sub>3/2</sub> and Fe2p<sub>1/2</sub> peaks are about 710.1 and 723.2 eV, respectively. From the Fig. S2(a) we deduce that the oxidation state of Fe in our BFCO/NSTO nanostructures is Fe<sup>3+</sup>. The Cr 2p XPS spectra reveal the 2p<sub>3/2</sub> and 2p<sub>1/2</sub> doublets arising from spin-orbit splitting (Fig. S2(b)). The binding energies for Cr2p<sub>3/2</sub> and Cr2p<sub>1/2</sub> peaks are about 576.4 and 586.2 eV, respectively, implying that the oxidation state of the Cr ion on BFCO is Cr<sup>3+</sup>.<sup>2</sup>



**Figure S3.** The first and second harmonic PFM responses of BFCO nanoisland under (a) 1, (b) 2, (c) 3, and (d) 4 V AC excitation.

**Reference:**

1 R. Nechache, F. Rosei, *J. Solid Stat. Chem.*, 2012, **189**, 12.

2 R. Nechache, C. V. Cojocaru, C. Harnagea, C. Nauenheim, M. Nicklaus, A. Ruediger, F. Rosei and A. Pignolet, *Adv. Mater.*, 2011, **23**, 1724.