

Supporting Information

Single sea urchin–MoO₃ nanostructure for surface enhanced Raman spectroscopy of dyes

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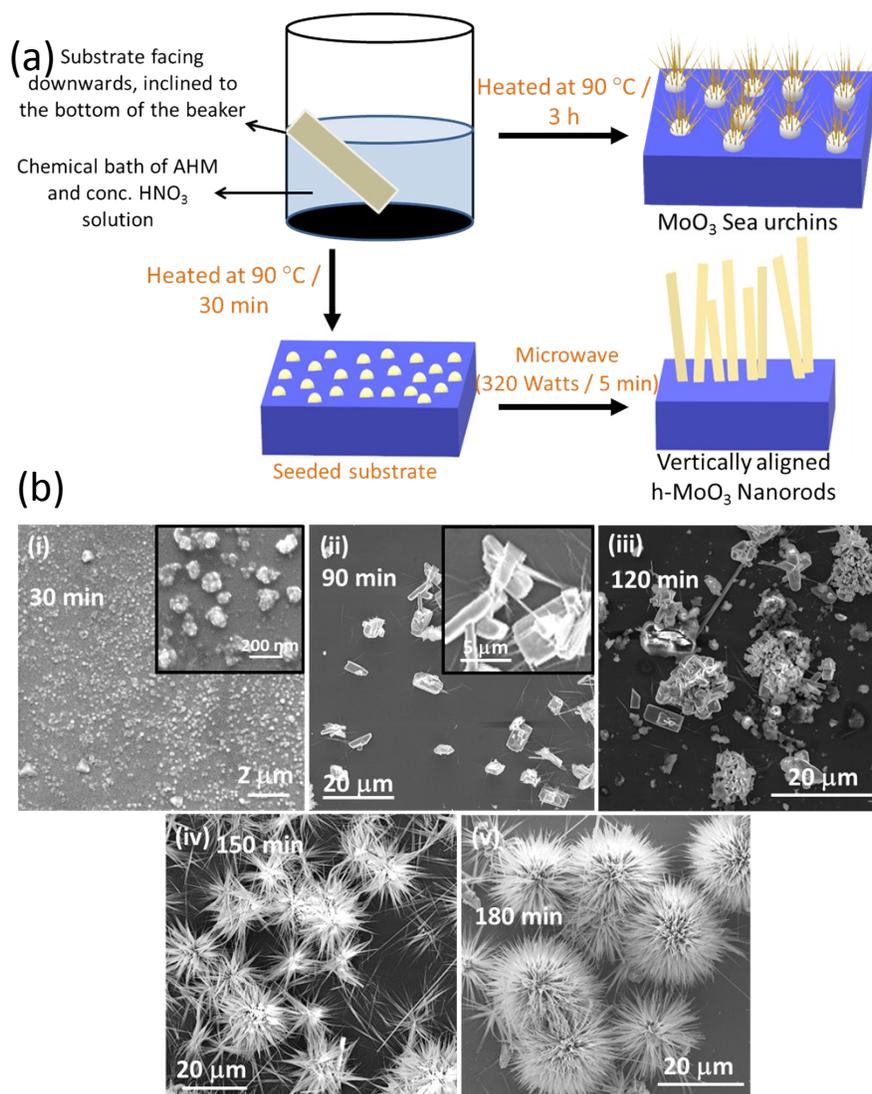


Figure S1. (a) Schematic of synthesis of MoO₃ sea urchins and h-MoO₃ nanorods employing chemical bath deposition and microwave methods, respectively. (b) FESEM images of MoO₃ sea urchin growth at various reaction stages from 30 to 180 minutes.

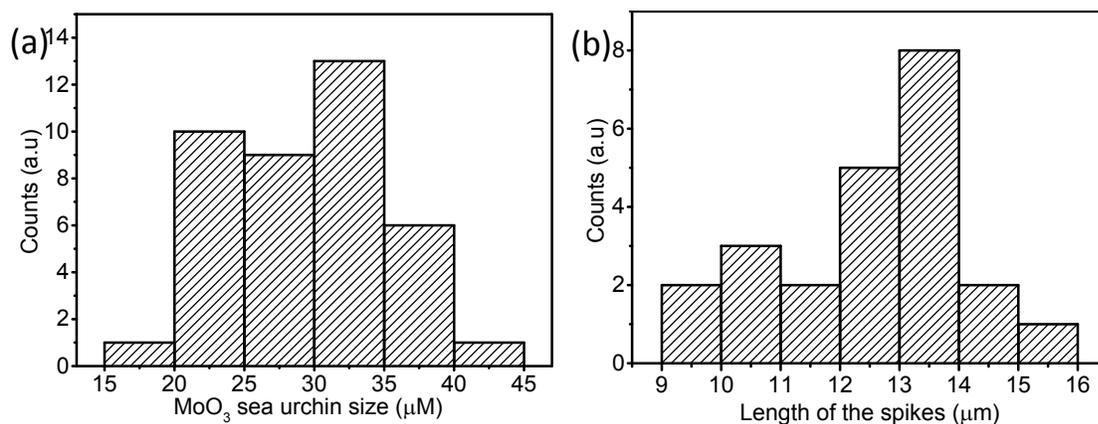


Figure S2. Histograms of MoO₃ sea urchins (a) Size distribution (from FESEM) and (b) length of the spikes (TEM images).

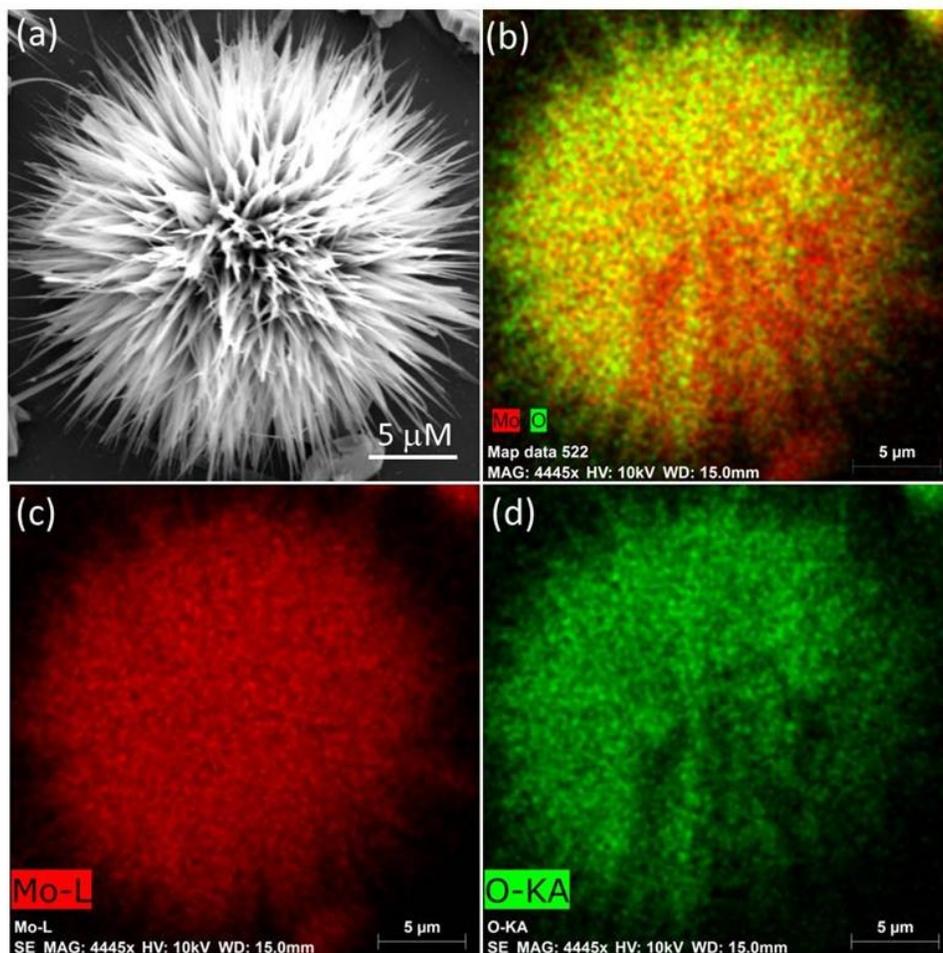


Figure S3. (a) FESEM image of MoO₃ sea urchin (b) combined EDS map of Mo L level and O K level (c) Corresponding EDS mapping of Mo L and (d) O K level.

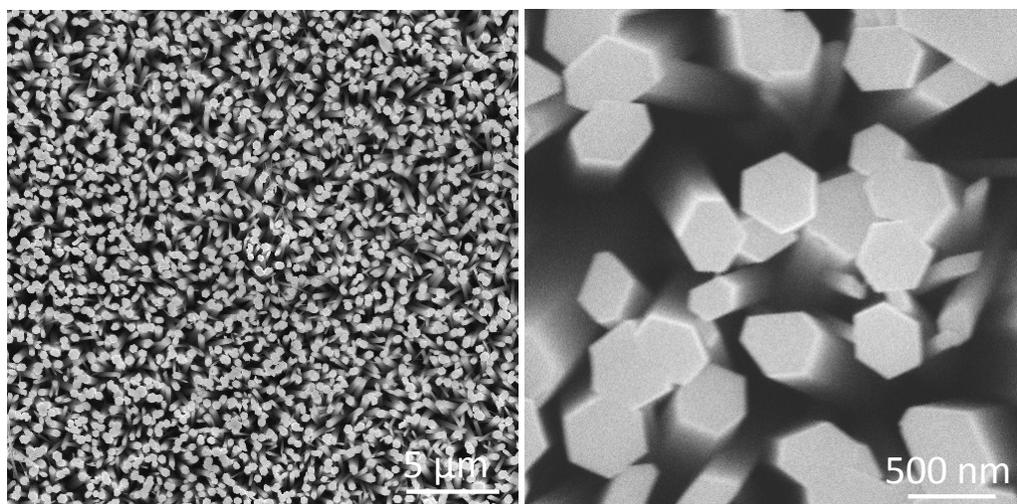


Figure S4. FESEM images of vertically aligned h-MoO₃ nanorods grown on Si substrate employing microwave method (a) low magnification image (b) high magnification image

Figure S4 displays the low and high magnified FESEM image of h-MoO₃ nanorods. From figure S4b, one can see that the NRs are 1D hexagonal rods. The h-MoO₃ nanorods are well oriented, uniform and grown vertically to the substrate with the diameter of ~400 nm and length of ~5.5 μm.

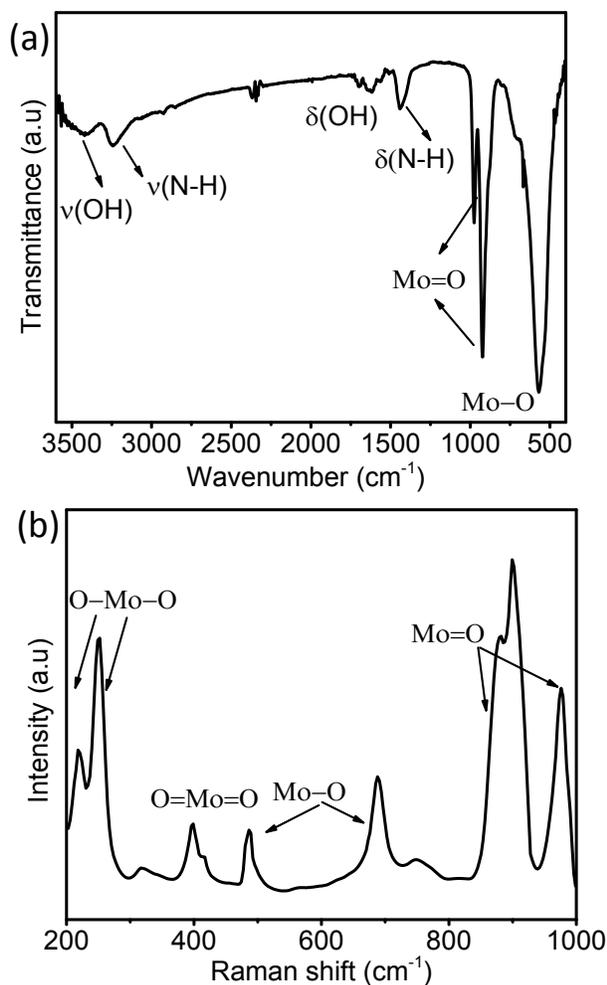


Figure S5. (a) FTIR spectra (b) Raman spectra of vertically aligned MoO₃ nanorods.

FTIR spectra of h-MoO₃ display three strong vibrational peaks in the range of 400-1000 cm⁻¹. The three strong vibrational peaks detected at 569 cm⁻¹, 923 cm⁻¹ and 992 cm⁻¹, associated with the stretching modes of Mo-O bonds, stretching and bending modes of Mo=O bonds. The broad peaks appeared at 3432 cm⁻¹ and 1630 cm⁻¹ due to stretching and bending vibrations of hydroxyl groups (-OH) present. The weak broad peaks at 3249 cm⁻¹ and 1440 cm⁻¹ correspond to the stretching and bending vibrations of N-H of NH₄⁺ groups. In the Raman spectra for h-MoO₃, the peaks at 877 and 977 cm⁻¹ correspond to the vibrational modes of O=Mo whereas the peaks at 247 and 218 cm⁻¹ arise due to vibrations of the O-Mo-O bonds. The lower intensity peaks at 484 and 687 cm⁻¹ are attributed to the O-Mo vibrations and the peak at 397 cm⁻¹ has been assigned to O=Mo=O vibrations.

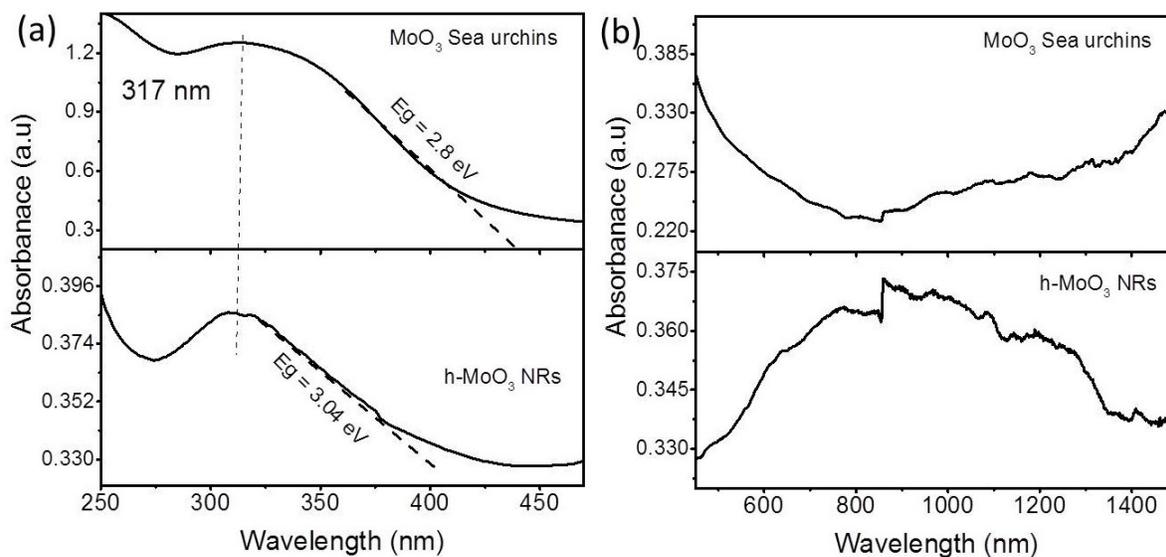


Figure S6. (a) UV-visible-NIR absorption spectra of as-synthesized MoO₃ nanostructures such as MoO₃ sea urchins and vertically aligned h-MoO₃ nanorods (b) Zoomed view of absorption spectra in (a).

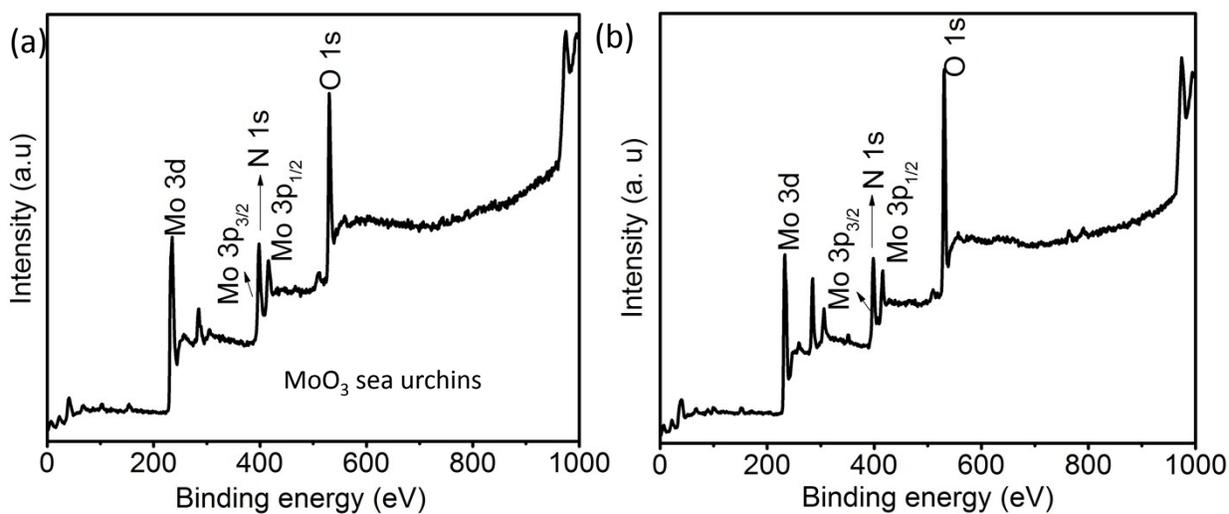


Figure S7. XPS Survey spectra of MoO₃ nanostructures (a) MoO₃ sea urchins (b) h-MoO₃ nanorods.

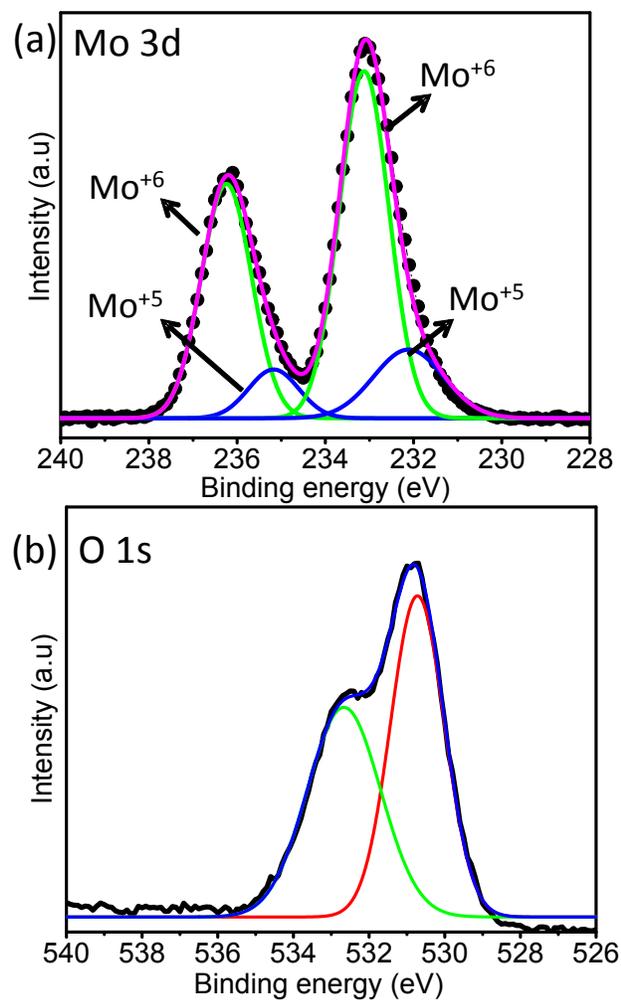


Figure S8. XPS spectra of h-MoO₃ nanorods (a) Mo 3d and (b) O 1s.

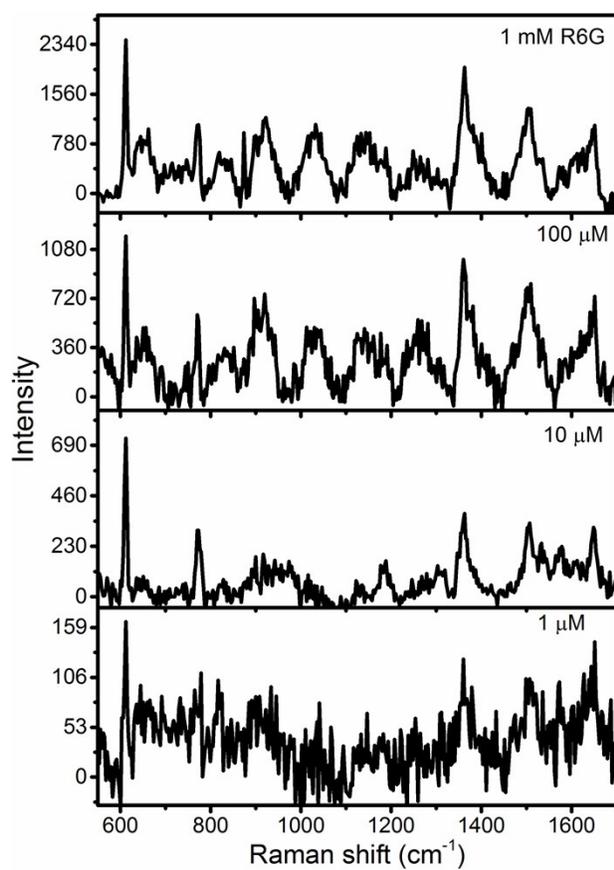


Figure S9. SERS spectra of R6G molecules with various concentrations on h-MoO₃ NRs.

Table S1. SERS of R6G enhancement factors for MoO₃ nanostructures as substrates.

<i>Peak positions</i>	<i>Enhancement factor (EF)</i>	
	<i>MoO₃ sea-urchins</i>	<i>MoO₃ vertically aligned nanorods</i>
612 cm ⁻¹	$\sim 1 \times 10^5$	$\sim 7.8 \times 10^3$
1358 cm ⁻¹	$\sim 7 \times 10^4$	$\sim 5.4 \times 10^3$

Table S2. Comparing the EF values for various semiconductors reported in the literature.

Material	Analyte	EF	Excitation Laser (nm)	Reference
MoO ₃ sea urchins h-MoO ₃ NRs	R6G (Rhodamine 6G)	$\sim 1 \times 10^5$ $\sim 7.8 \times 10^3$	532	This work
MoO ₃ NRs coated polymethacrylic acid shell	MB (Methylene blue)	1.6×10^4	532	Wang et al. RSC Adv., 2017, 7, 36201.
Plasmonic MoO ₃ . x@MoO ₃	MB	1.42×10^5	785	Tan et al. Chem. Commun., 2016, 52, 2893.
1D MoO ₃ nanoribbons	4-MBA (4- Mercaptobenzoic acid)	10^3	632.8	Dong et al. Chem. -Asian J., 2010, 5, 1824.
α -MoO ₃ Micro particles	R6G	9.3×10^5	532	Wu et al. Analyst, 2017, 142, 326.
α -MoO _{3-x} nanobelts		1.8×10^7		
Cu ₂ O superstructure	R6G	8×10^5	647	Lin et al. Adv. Mater., 2017, 29, 1604797.
Cu ₂ O nanostructure	4-ATP (4- Aminothiophenol)	3.4×10^4	532	Qiu et al. J. Phys. Chem. Lett., 2012, 3, 651.
CuO nanocrystals	4-MPY (4- Mercaptopyridine)	10^2	532	Wang et al. Anal. Sci., 2007, 23, 787.
Urchin-like W ₁₈ O ₄₉	R6G	3.4×10^5	522	Cong et al. Nat. Commun., 2015, 17, 7800.
ZnO nanocages	4-MBA	6.62×10^5	633	Wang et al. Angew. Chem. Int. Ed., 2017, 56, 9851.
ZnO nanocrystals	4-MPY	10^3	514.5	Wang et al. J. Raman Spectrosc., 2009, 40, 1072.
TiO ₂ microarrays	MB	2×10^4	532	Qi et al. J. Am. Chem. Soc., 2014, 136, 9886.
TiO ₂ NPs	Dopamine	10^3	442	Musumeci et al. J. Am. Chem. Soc., 2009, 131, 6040.
Fe ₂ O ₃	Pyridine	7.8×10^3	488 & 514.5	Zhang et al. Chem. Phys. Lett., 1988, 153, 215.
ZnS nanocrystals	4-MPY	10^3	514.5	Wang et al. J. Raman Spectrosc., 2007, 38, 34.

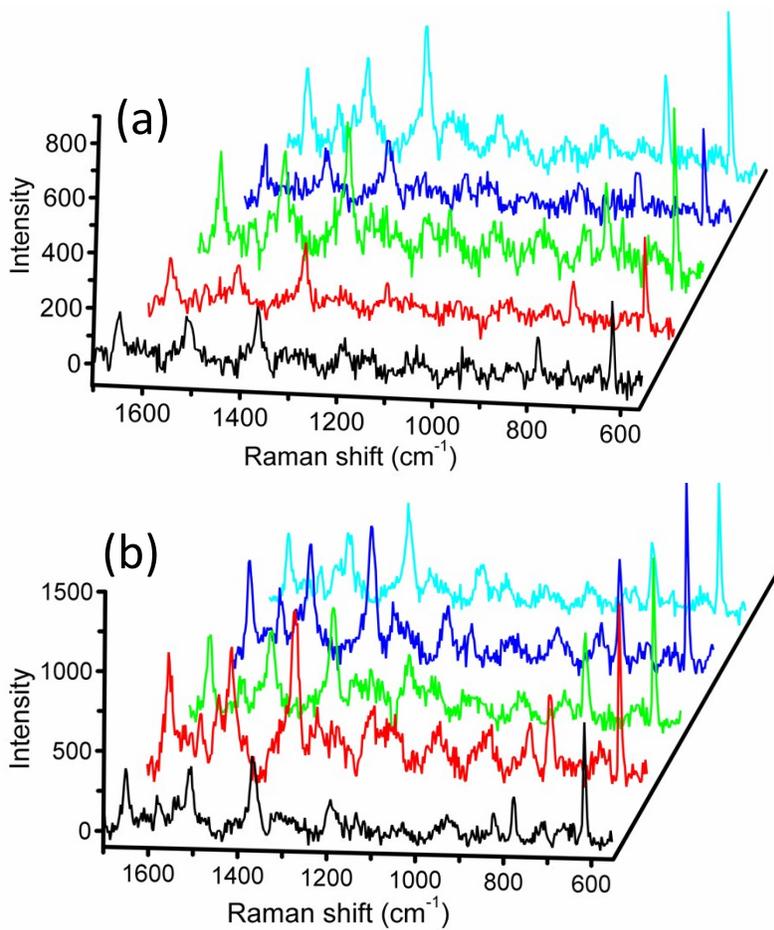


Figure S10. SERS spectra of R6G on MoO₃ sea urchin nanostructure (a) From the core or middle region (b) At the tips of the spiky features of the MoO₃ sea urchin.

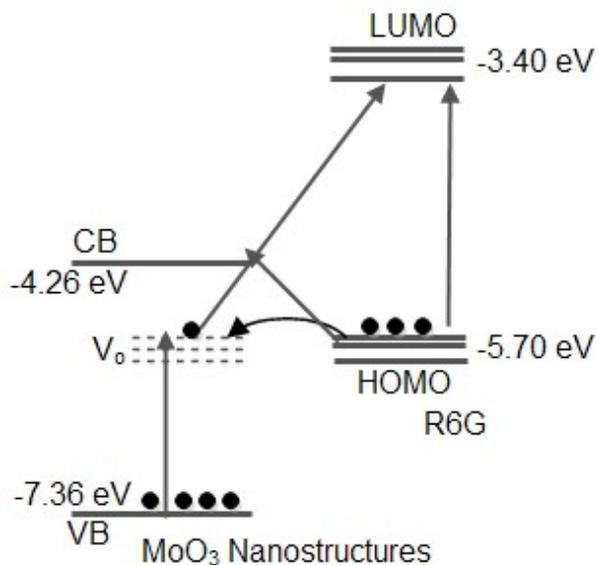


Figure S11. Schematic representation of charge transfer between the MoO₃ nanostructures and R6G molecules during the excitation with 532 nm laser light.

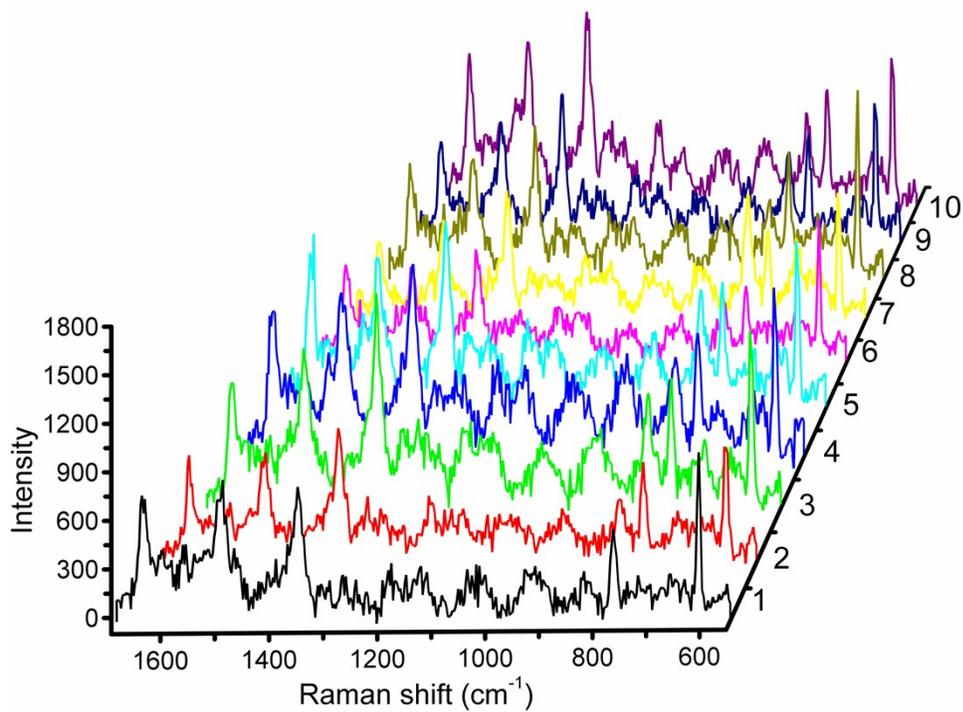


Figure S12. SERS spectra of R6G on MoO₃ sea urchin nanostructure surface at 10 random positions.

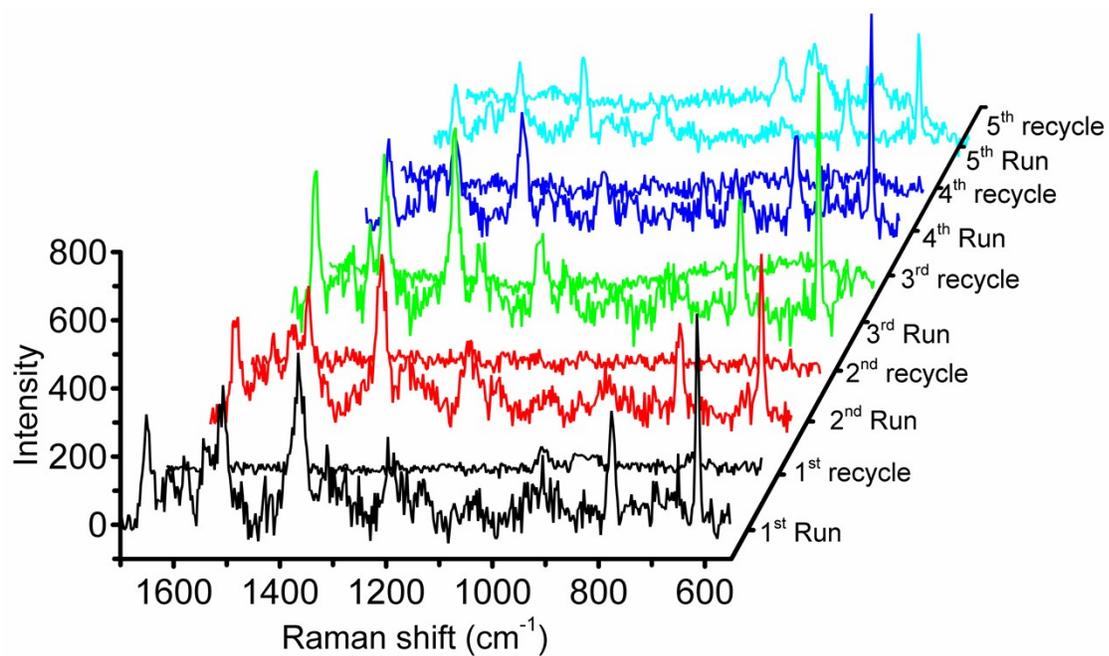


Figure S13. Reusable vertically aligned h-MoO₃ nanorods as SERS substrates; SERS of 100 μM R6G before and after UV treatment.