

Electronic Supplementary Information (ESI)

**Stable polymeric chain configuration producing high performance
PEBAX-1657 membrane for CO₂ separation**

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This supplementary data file includes:

1. Figures S1-S9
2. Tables S1-S3
3. References

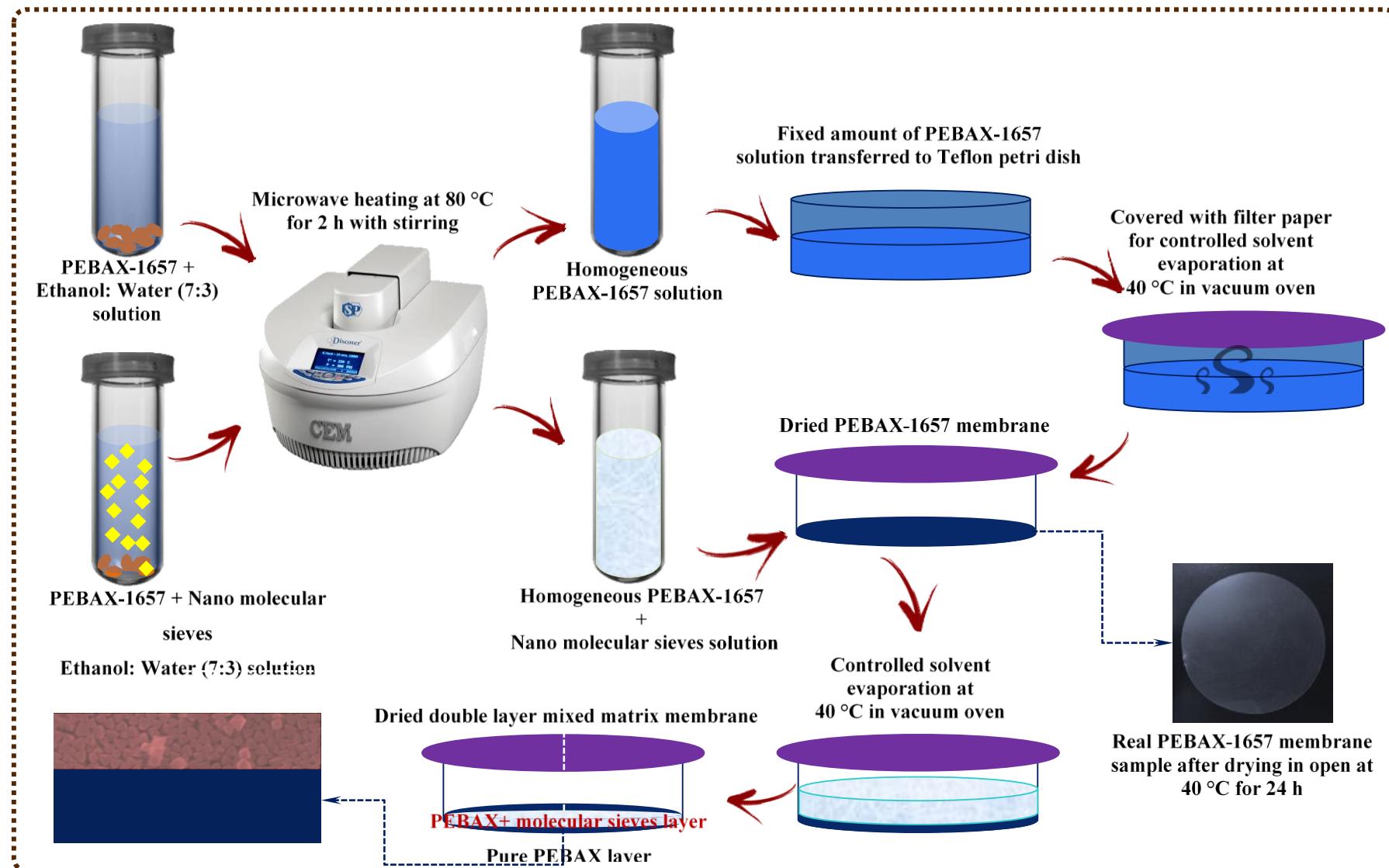


Fig. S1. Schematic description of pristine PEBAK and double layer PEBAK/nano molecular sieves MMMs preparation.

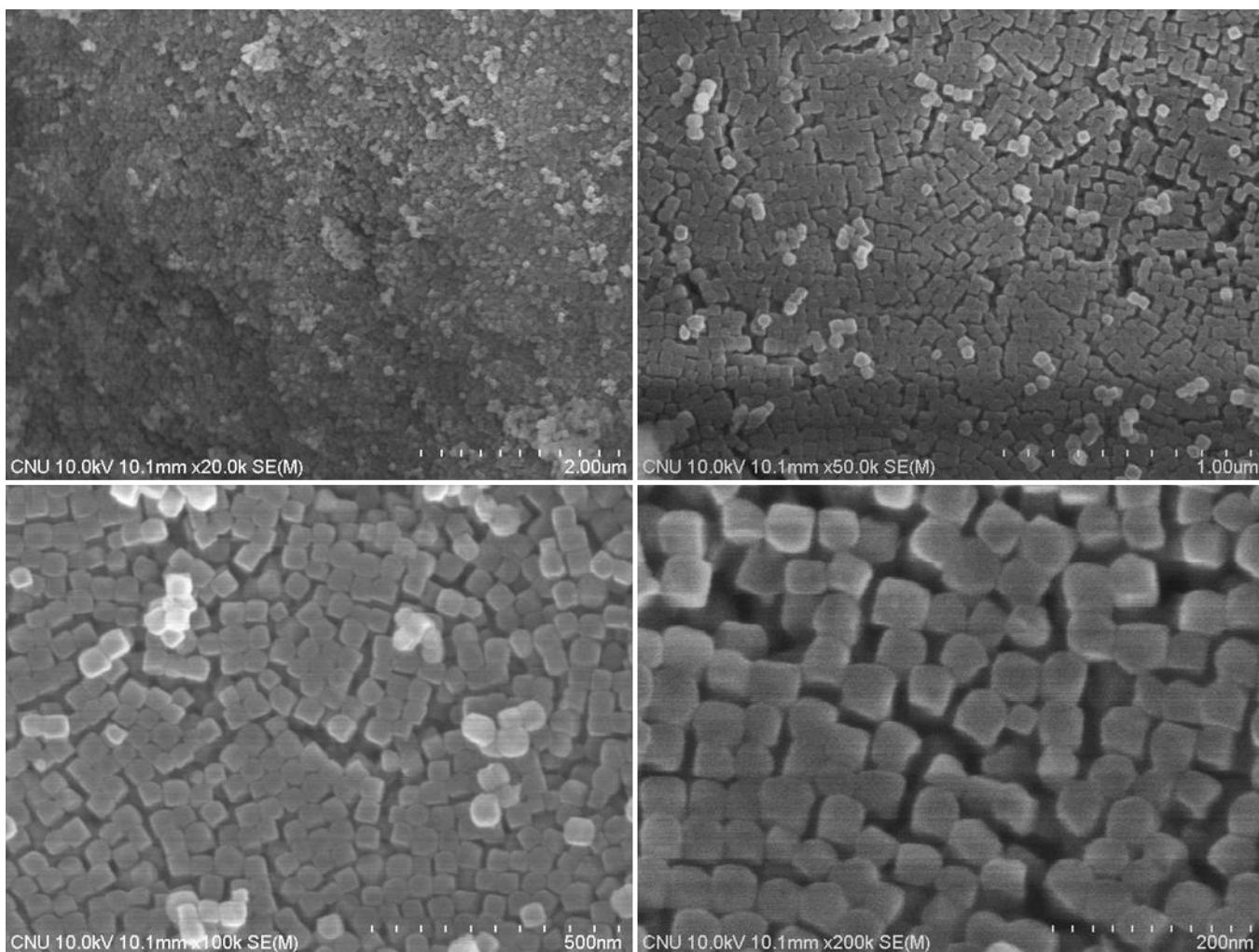


Fig. S2. Different magnification scanning electron micrographs of nano size zeolite A molecular sieve crystals.

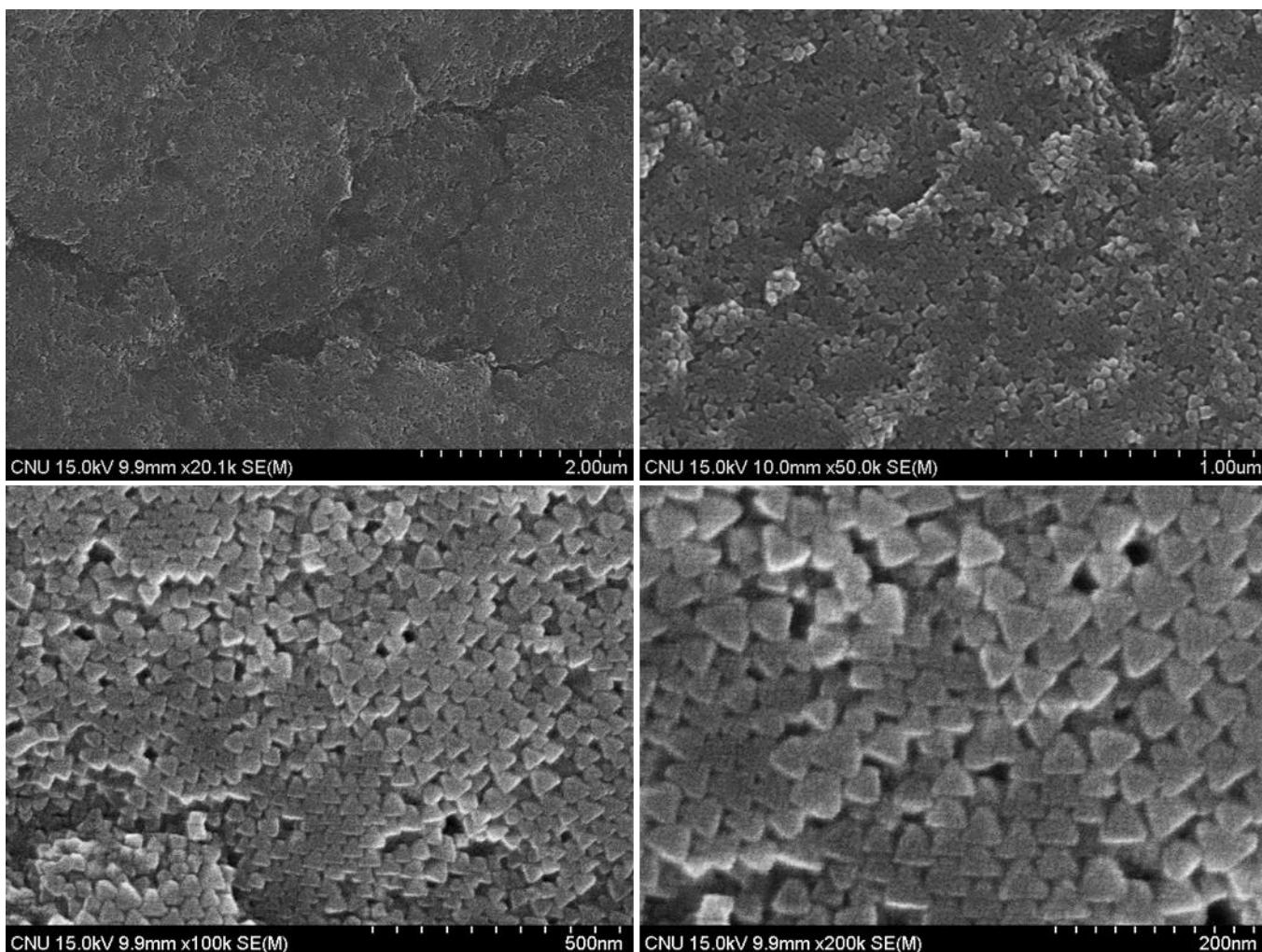


Fig. S3. Different magnification scanning electron micrographs of nano size zeolite Y molecular sieve crystals.

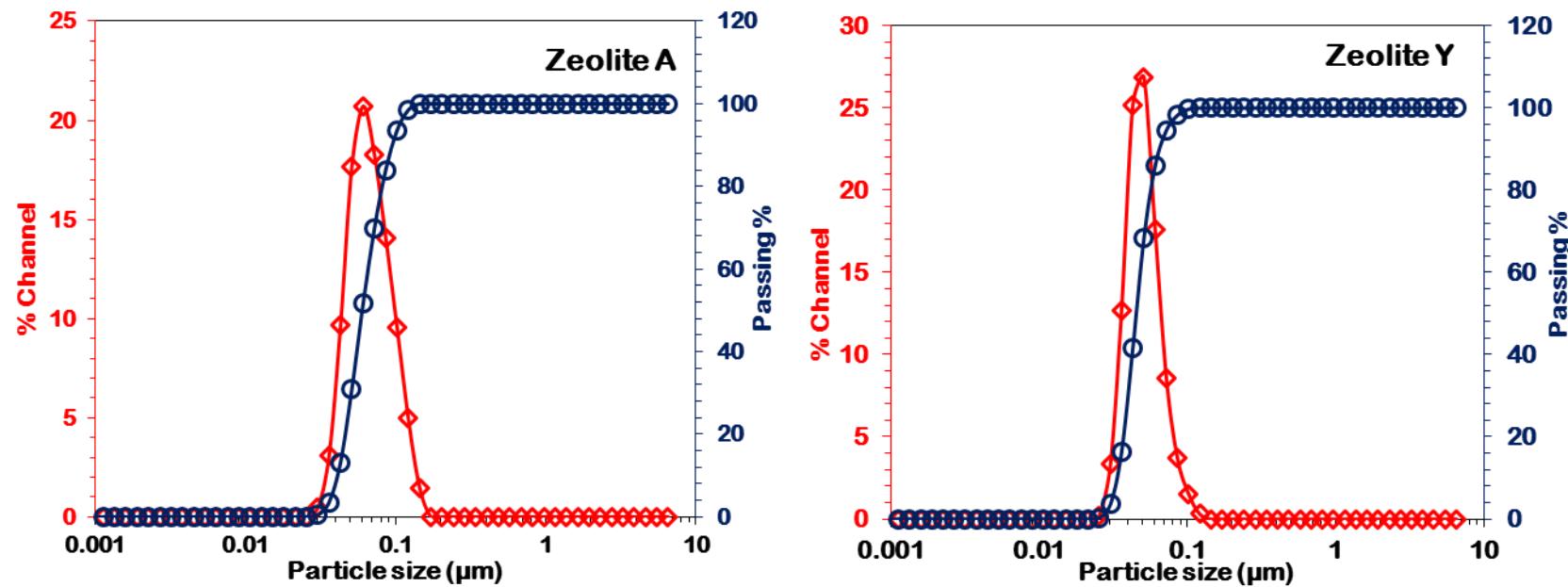


Fig. S4. PSD curves of as-synthesized zeolite molecular sieves A and Y.

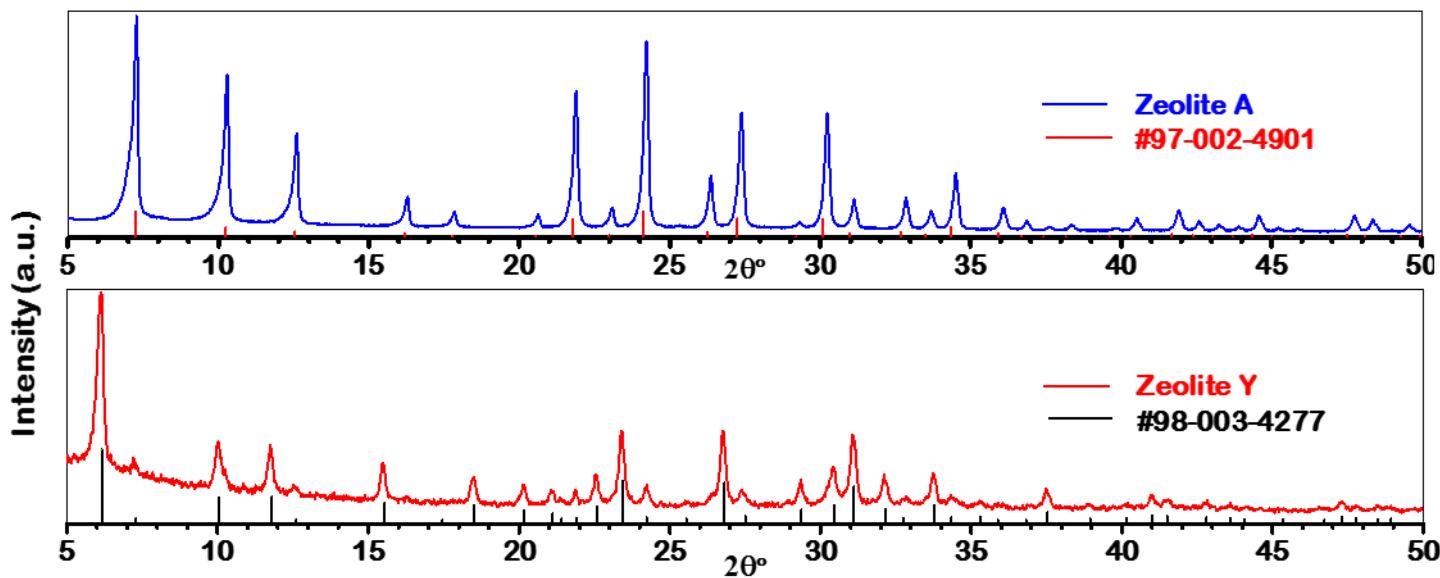


Fig. S5. X-ray diffraction patterns of as-synthesized zeolite molecular sieves A and Y along with their respective standard diffraction patterns.

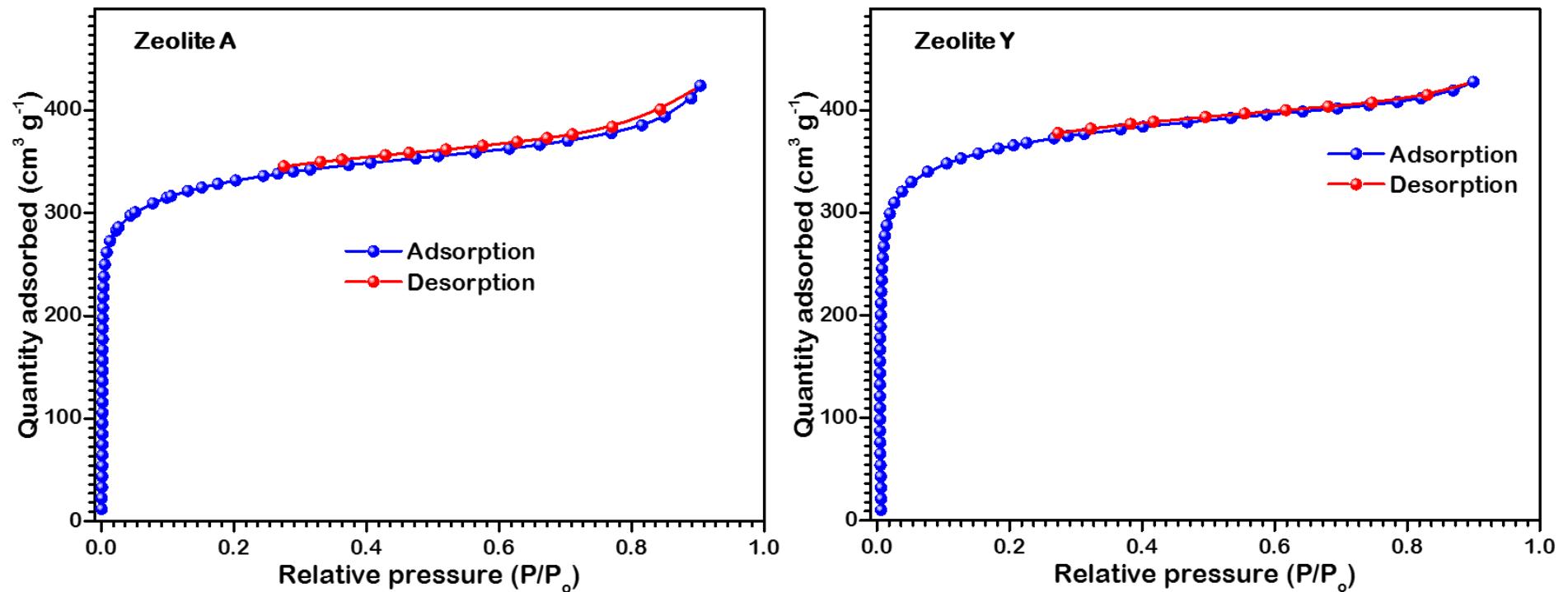


Fig. S6. Water vapor adsorption-desorption isotherms of nano size zeolites A and Y molecular sieve crystals at 298 K.

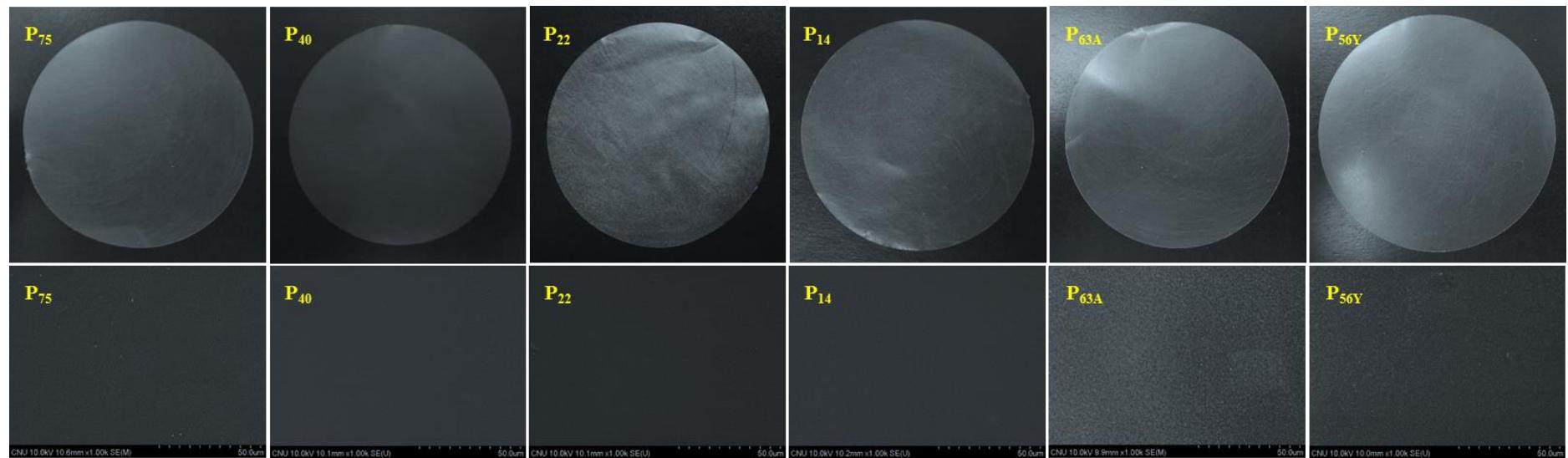


Fig. S7. Photographic camera images (top row), and surface SEM micrographs (bottom row) of various thickness PEBAKX-1657 membranes.

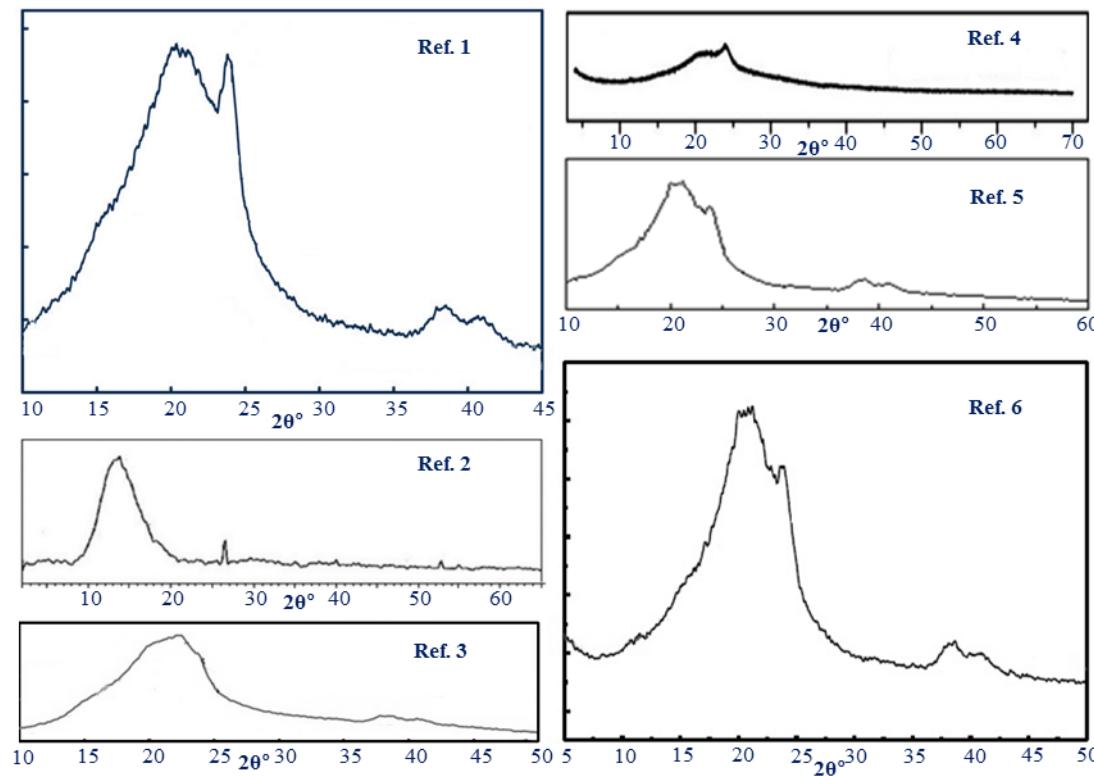


Fig. S8. Collective representation of X-ray diffraction patterns of pristine PEBAK-1657 membranes reported in literature to make a healthy comparison with diffraction patterns of the as-synthesized high crystallinity thin PEBAK membranes (**Fig. 4, main text**) (Adopted from ref.¹, and reproduced from refs.^{2–6} with permission from the Elsevier).

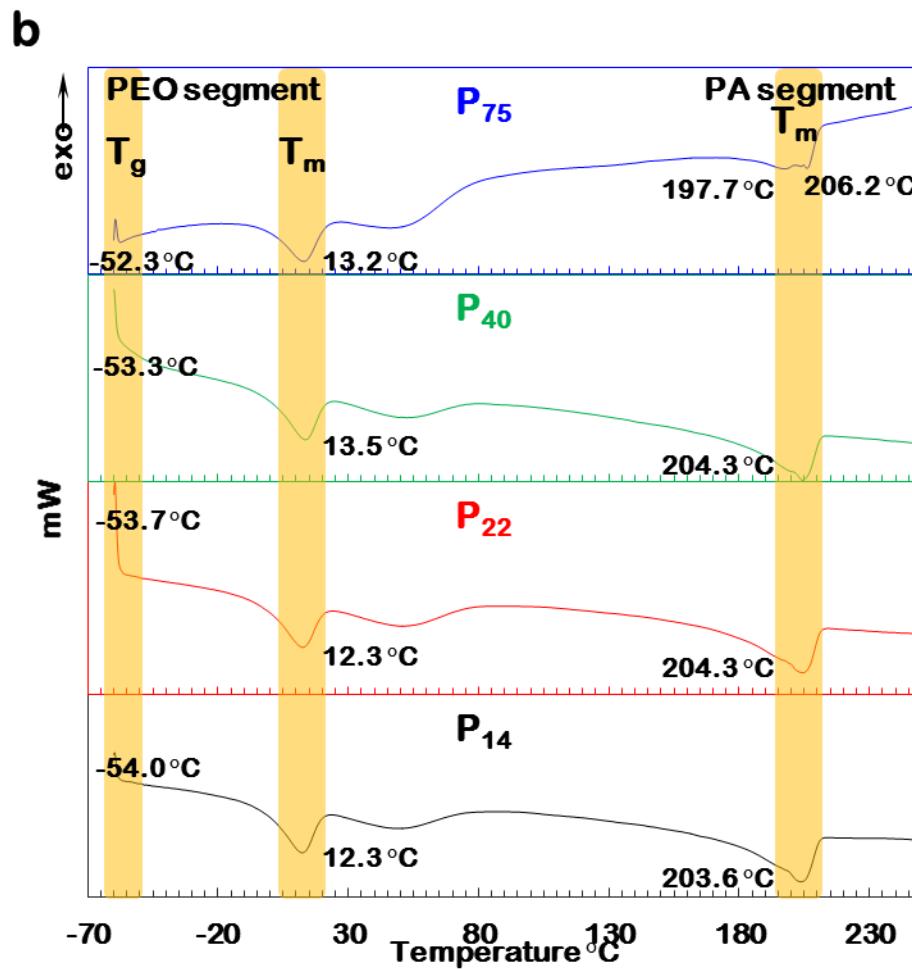
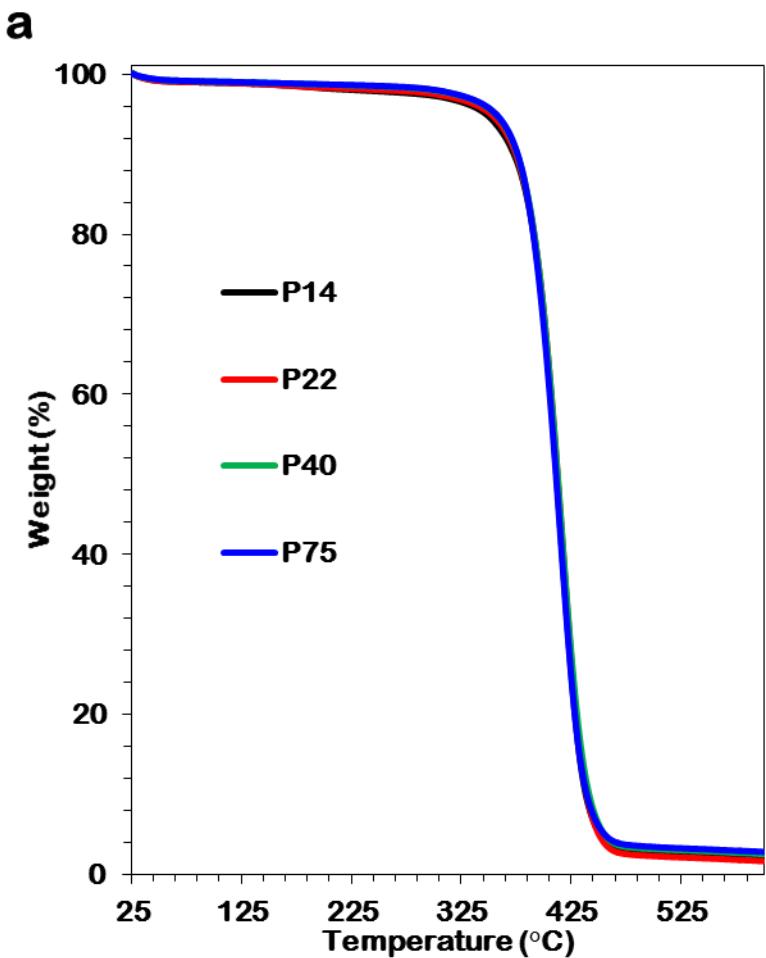


Fig. S9. (a) TGA curves, and (b) comparison of DSC profiles of various thickness PEBAx-1657 membranes.

Table S1. Textural parameters of zeolite molecular sieves determined by water vapor adsorption-desorption isotherms

Zeolite	Surface area, $\text{m}^2 \text{ g}^{-1}$	Pore volume ($P/P_0 = 0.90$), $\text{cm}^3 \text{ g}^{-1}$
NaA	982.15	0.34
NaY	1093.10	0.35

Table S2. Gas diffusivity and solubility coefficients of the different thickness pure PEBAx membranes

Membrane	Diffusivity ($\text{cm}^2 \text{S}^{-1}$)				Selectivity			Solubility $\times 10^{-4}$ ($\text{cm}^3(\text{STP})/\text{cm}^2.\text{cmHg}$)				Selectivity		
	CO_2	N_2	O_2	CH_4	CO_2/N_2	CO_2/O_2	CO_2/CH_4	CO_2	N_2	O_2	CH_4	CO_2/N_2	CO_2/O_2	CO_2/CH_4
P ₁₄	2.60×10^{-6}	9.29×10^{-7}	2.25×10^{-7}	2.88×10^{-7}	2.80	11.56	9.03	65.48	3.02	35.73	33.16	21.63	1.83	1.97
P ₂₂	1.47×10^{-6}	1.81×10^{-6}	1.89×10^{-5}	5.03×10^{-7}	8.12	0.78	29.22	9.73	1.46	0.41	18.85	6.65	23.85	0.52
P ₄₀	1.35×10^{-6}	1.66×10^{-6}	1.01×10^{-6}	4.65×10^{-7}	0.81	1.34	2.90	100.76	1.68	7.92	21.57	59.95	12.72	4.67
P ₇₅	1.21×10^{-6}	7.28×10^{-7}	7.63×10^{-7}	4.12×10^{-7}	1.66	1.59	2.94	70.37	3.31	8.62	20.19	21.26	8.16	3.48
P _{63A}	8.07×10^{-7}	2.67×10^{-7}	-	2.93×10^{-7}	3.02	-	2.75	67.31	5.77	-	17.99	11.67	-	3.74
P _{56Y}	1.24×10^{-6}	4.23×10^{-7}	1.26×10^{-6}	5.77×10^{-7}	2.93	0.98	2.15	93.64	8.32	9.13	19.36	11.25	10.25	4.84

Table S3. Comparison of literature reported pure PEBAx as well as MMM with as synthesize different thickness pure PEBAx membranes

Membrane	P_{CO_2} , Barrer	Selectivity			T, °C	Pressure, bar	Polymer Wt.%	Membrane thickness, μm	Ref.
		CO ₂ /N ₂	CO ₂ /O ₂	CO ₂ /CH ₄					
PEBAx	80.0	49.0	-	-	30.0	1	3.0	-	7
PEBAx-40% PEG-GDMS-POSS-THF	160.0	40.4	-	-	30.0	1	3.0	-	7
PEBAx	80.0	70.0	-	-	21.0	2.3	7.5	50.0-90.0	8
PEBAx-30% PS colloid	44.0	66.0	-	-	21.0	2.3	7.5	50.0-90.0	8
PEBAx-5% SWNT	102.0	73.0	-	-	21.0	2.3	7.5	50.0-90.0	8
PEBAx	55.8	40.2	11.8	18.0	25.0	24.5	4.0	-	2
PEBAx-10% 4A zeolite	97.0	54.0	12.4	26.5	25.0	24.5	4.0	-	2
PEBAx	88.4	49.4	-	20.4	25.0	2	4.5	80.0-100.0	3
PEBAx-2% MWCNT	119.3	51.5	-	17.6	25.0	2	4.5	80.0-100.0	3
PEBAx	56.0	40.0	-	-	35.0	4	10.0	25.0	4
PEBAx-2% ATP	77.0	52.0	-	-	35.0	4	10.0	30.0	4
PEBAx	70.3	63.9	-	18.5	25.0	2	6.0	-	9
PEBAx-30% TEOS	41.8	64.3	-	16.1	25.0	2	6.0	-	9
PEBAx	115.0	48.0	-	18.0	30.0	2	3.0	65.0-85.0	5
PEBAx-Pro(Silica)	161.5	82.8	-	65.5	25.0	1	3.0	65.0-85.0	5
P ₁₄	184.7	59.7	22.6	18.0	25.0	2.7	2.5	14.0	This work

References

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