

# Ceria/Cobalt Borate Hybrids as Efficient Electrocatalysts for Water Oxidation in Neutral Conditions

Xuemei Zhou<sup>a</sup>, Sijia Guo<sup>a</sup>, Qiran, Cai<sup>b</sup>, Shaoming Huang<sup>\*,a</sup>

<sup>a</sup>School of Material and Energy, Guangdong University of Technology, Guangzhou, 510006, China.

E-mail: [smhuang@gdut.edu.cn](mailto:smhuang@gdut.edu.cn)

<sup>b</sup>Institute for Frontier Materials, Deakin University, Geelong Waurn Ponds Campus, Victoria, 3216, Australia.

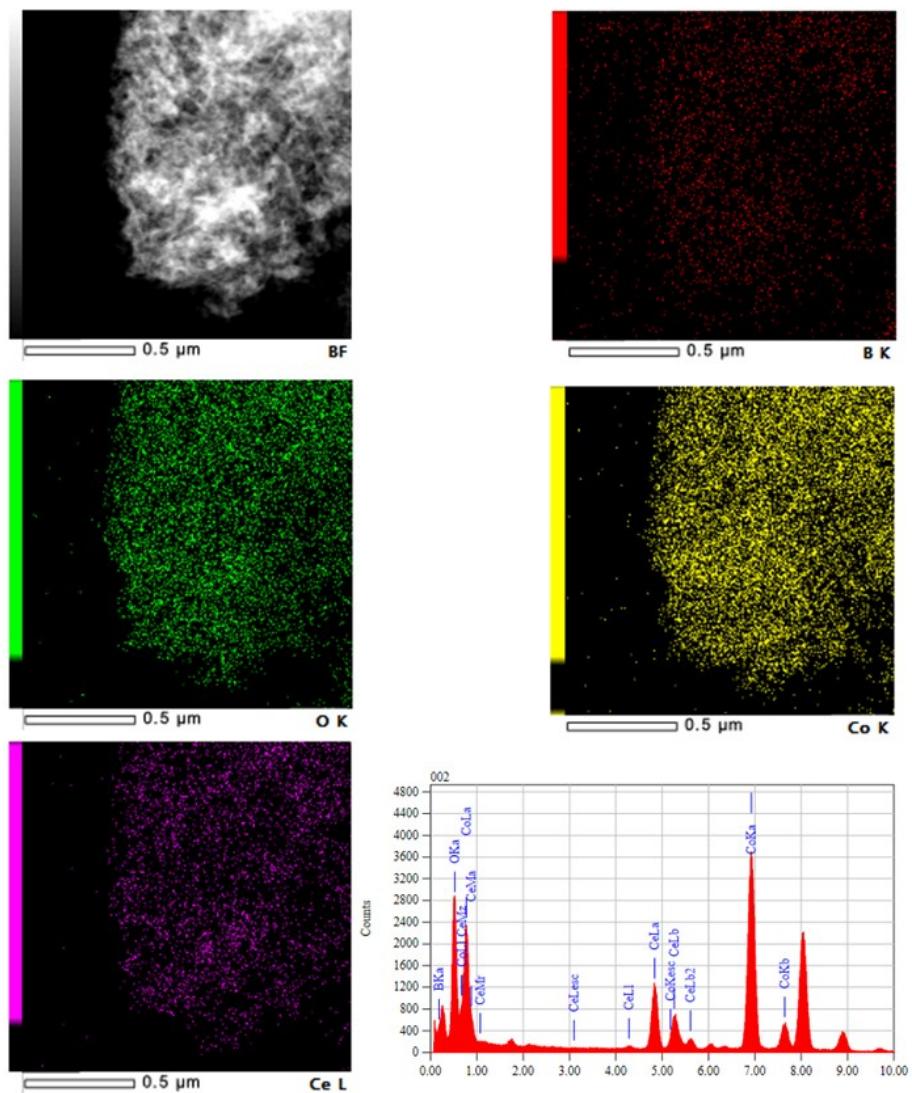


Fig. S1 EDS mapping images for 20CeO<sub>2</sub>/Co-Bi.

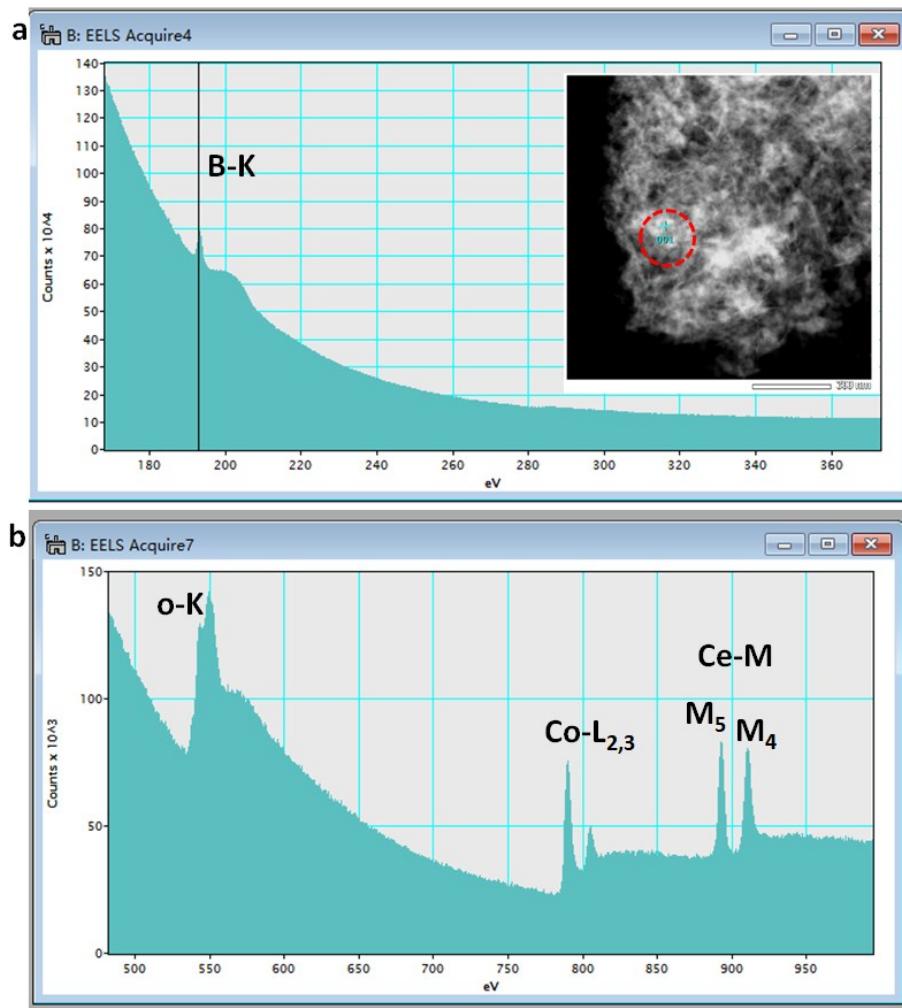


Fig. S2 EELS spectrum of 20CeO<sub>2</sub>/Co-Bi measured from the region of inset TEM images.

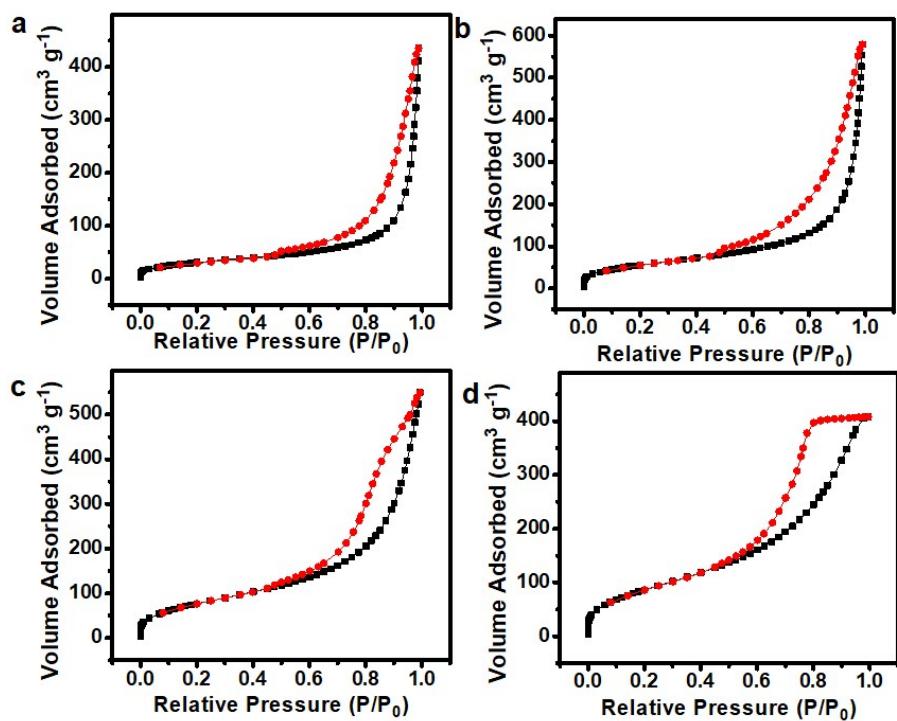


Fig. S3 Nitrogen adsorption/desorption isotherms of the Co-Bi (a), 10CeO<sub>2</sub>/Co-Bi (b), 20CeO<sub>2</sub>/Co-Bi (c) and 30CeO<sub>2</sub>/Co-Bi (d).

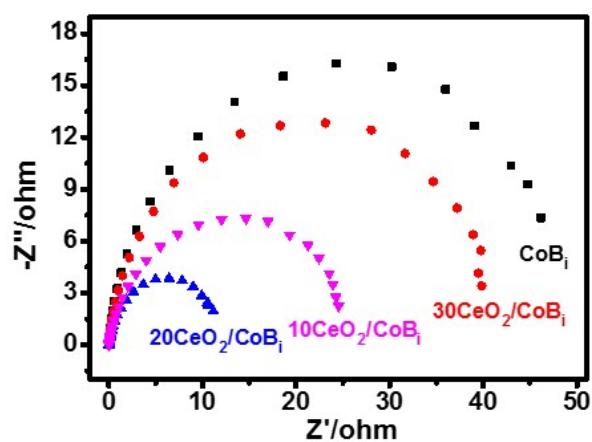


Fig. S4 Impedance Nyquist plots of Co-Bi,  $10\text{CeO}_2/\text{Co-Bi}$ ,  $20\text{CeO}_2/\text{Co-Bi}$  and  $30\text{CeO}_2/\text{Co-Bi}$  composites electrodes.

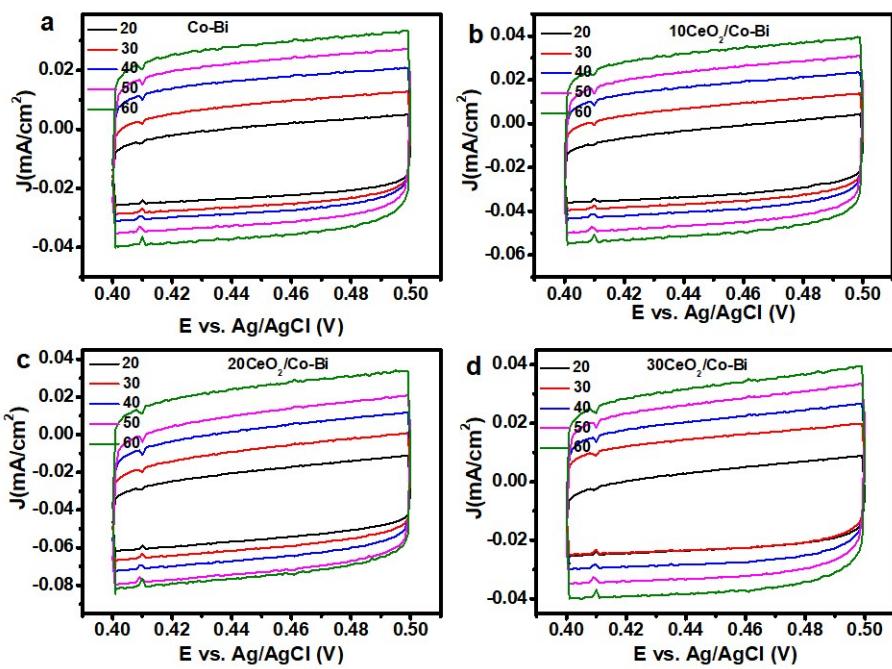


Fig. S5 Cyclic voltammograms for all the samples (Co-Bi, 10CeO<sub>2</sub>/Co-Bi, 20CeO<sub>2</sub>/Co-Bi and 30CeO<sub>2</sub>/Co-Bi composite) in the non-faradaic capacitance current range at scan rates of 10, 20, 30, 40, 50 and 60  $\text{mV s}^{-1}$ .

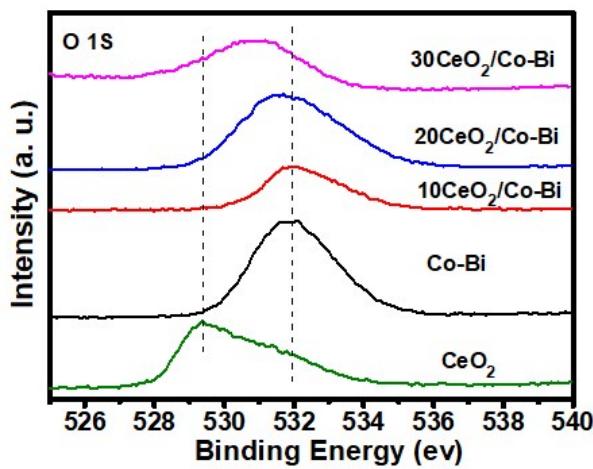


Fig. S6 High-resolution XPS spectra for O of pure CeO<sub>2</sub>, Co-Bi, 10CeO<sub>2</sub>/Co-Bi, 20CeO<sub>2</sub>/Co-Bi and 20CeO<sub>2</sub>/Co-Bi hybrid.

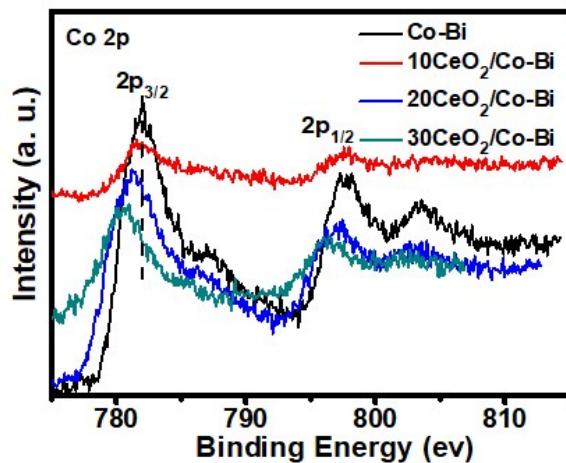


Fig. S7 High-resolution XPS spectra for Co of pure Co-Bi, 10CeO<sub>2</sub>/Co-Bi, 20CeO<sub>2</sub>/Co-Bi and 20CeO<sub>2</sub>/Co-Bi hybrid.

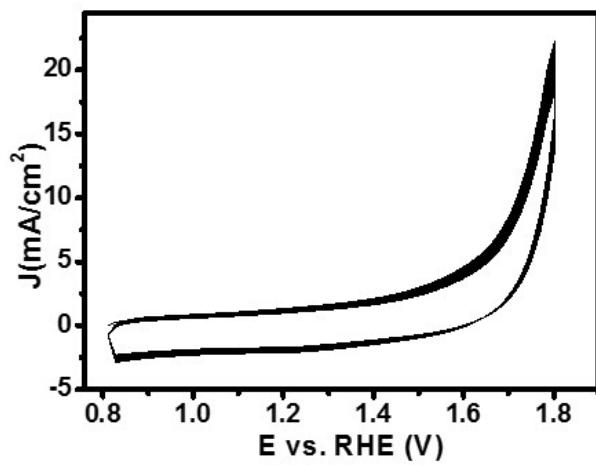


Fig. S8 Cyclic voltammograms for  $20\text{CeO}_2/\text{Co-Bi}$  composite electrode.

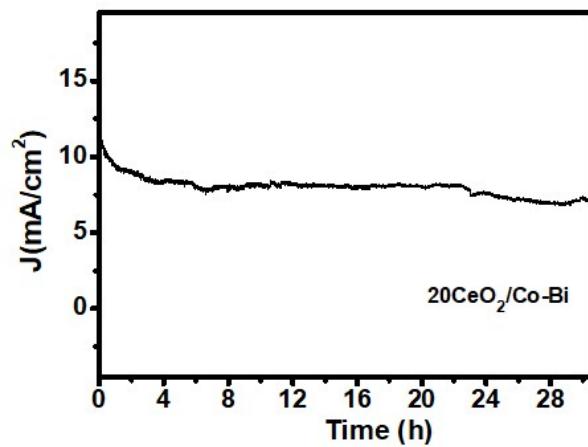


Fig. S9 Chronoamperometric response of of as-prepared  $20\text{CeO}_2/\text{Co-Bi}$  composite recorded at a constant potential of 1.70V vs. RHE.

Table S1 Comprision of cobalt-based catalysts of OER activities in neutral medium.

sample	Specific density (mA cm <sup>-2</sup> )	current (mV)	Overpotential	Electrolyte concentration	Ref
Co-Pi/ITO	1	410	0.1 M KP <sub>i</sub>		<sup>1</sup>
Co-Ni LDHs	1	490	0.1MP <sub>i</sub> +1.73MKNO <sub>3</sub>		<sup>2</sup>
Mn <sub>5</sub> O <sub>8</sub> Nanoparticles	5	580	0.3 M PBS		<sup>3</sup>
Co(PO <sub>3</sub> ) <sub>2</sub>	8	440	0.1 M phosphate		<sup>4</sup>
Co-W	1	420	0.05 M PBS		<sup>5</sup>
Sub-MnO <sub>x</sub>	5	530	0.3 M PBS		<sup>6</sup>
Fe-based film	1	480	0.1 M PBS		<sup>7</sup>
Ultrathin Co <sub>3</sub> S <sub>4</sub> nanosheets	3	620	0.1 M PBS		<sup>8</sup>
Mn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	0.32	680	0.5 M PBS		<sup>9</sup>
cobalt hexacyano-ferrate	1	580	0.05MKP <sub>i</sub> +1M KNO <sub>3</sub>		<sup>10</sup>
Au-Co(OH) <sub>2</sub>	1	410	0.1 M PBS		<sup>11</sup>
LiMnP <sub>2</sub> O <sub>7</sub>	0.5	680	0.5 M PBS		<sup>12</sup>
CoP <sub>i</sub> /GO	0.23	780	0.1 M KP <sub>i</sub>		<sup>13</sup>
CoP <sub>i</sub> Nanoarrays/Ti	10	450	0.1 M PBS		<sup>14</sup>
$\alpha$ -Co(OH) <sub>2</sub> NA/CC	10	536	1.0 M PBS		<sup>15</sup>
Fe-Co <sub>3</sub> O <sub>4</sub> @F-Co-Bi/CC	10	420	0.1 M K-Bi (pH=9.2)		<sup>16</sup>
Co-Bi NA/Ti	10	420	0.1 M K-Bi (pH=9.2)		<sup>17</sup>
NiB <sub>i</sub> /CC	10	470	0.1 M KB <sub>i</sub> (pH=9.2)		<sup>18</sup>
Co@Co-Bi/Ti	10	470	0.4 M NaBi		<sup>19</sup>
Co-Bi NS/G	10	490	0.1 M PBS		<sup>20</sup>
20CeO <sub>2</sub> /CoB <sub>i</sub> composite	10	453	0.1 M PBS (pH=7.4)		This work

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