

Electronic Supporting Information

Design of core-shell titania-heteropolyacid-metal nanocomposites for photocatalytic reduction of CO₂ to CO at ambient temperature

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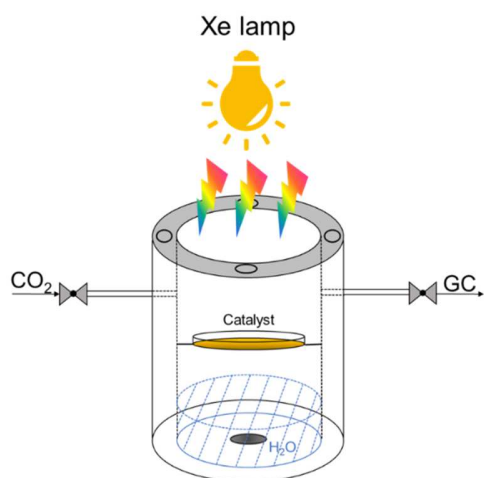


Figure S1. The schema of photocatalytic reactor

Table S1. Zn content in the Zn-HPW/TiO₂ composites from XRF.

Catalyst	Calculated Zn contents (wt.%)	Measured Zn contents (wt.%)
Zn-HPW/TiO ₂	0.5	0.5
	1.5	1.6
	3.0	2.9
	6.0	6.1
	10	10.3
	12	12.2

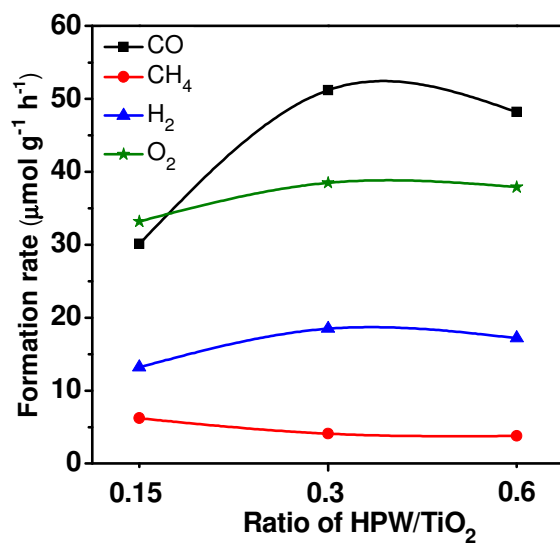


Figure S2. CO, O₂, CH₄ and H₂ formation rates over Zn-HPW/TiO₂ composites with different ratio of HPW/TiO₂. Reaction conditions: catalyst, 0.1 g; Gas phase pressure, CO₂ 0.2 MPa; H₂O, 15mL; irradiation time, 6h.

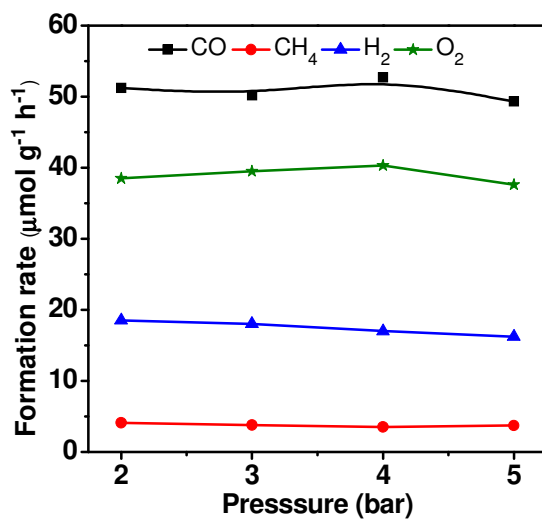


Figure S3. CO, O₂, CH₄ and H₂ formation rates over Zn-HPW/TiO₂ composites with different pressure. Reaction conditions: catalyst, 0.1 g; Gas phase pressure, CO₂ 0.2 ~ 0.5 MPa; H₂O, 15mL; irradiation time, 6h.

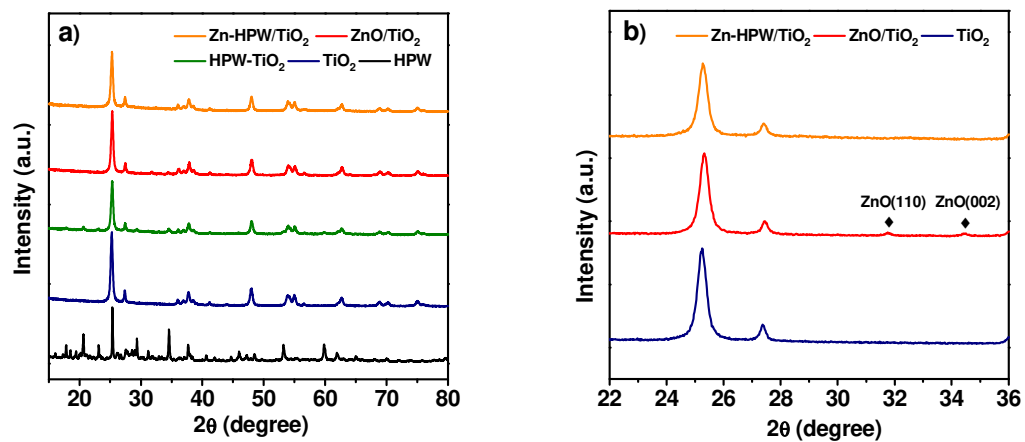


Figure S4. XRD patterns of different nanocomposites catalysts.

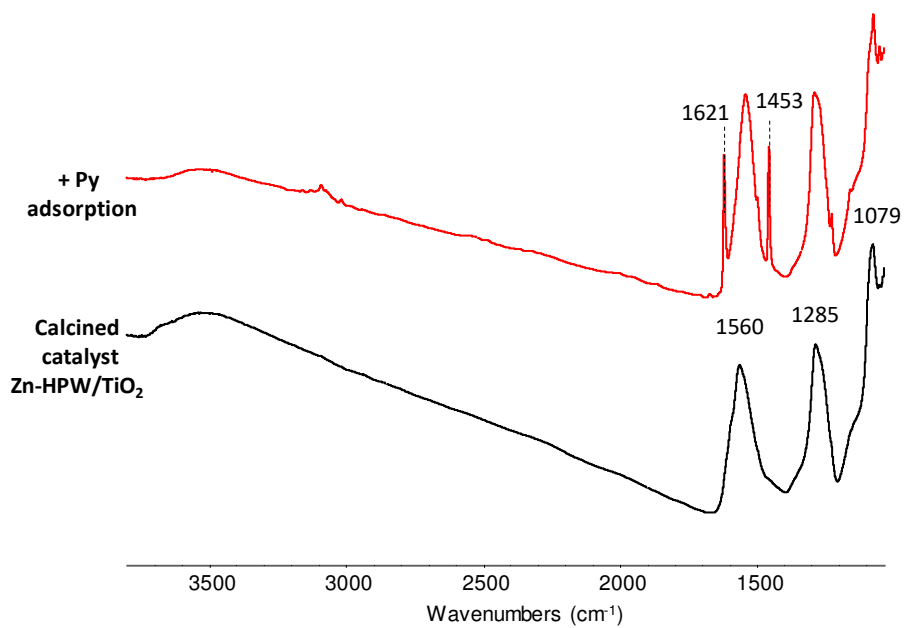


Figure S5. FTIR spectra of the calcined Zn-HPW/TiO₂ (6wt.%) catalyst before and after adsorption of pyridine on the sample dehydrated at 200°C.

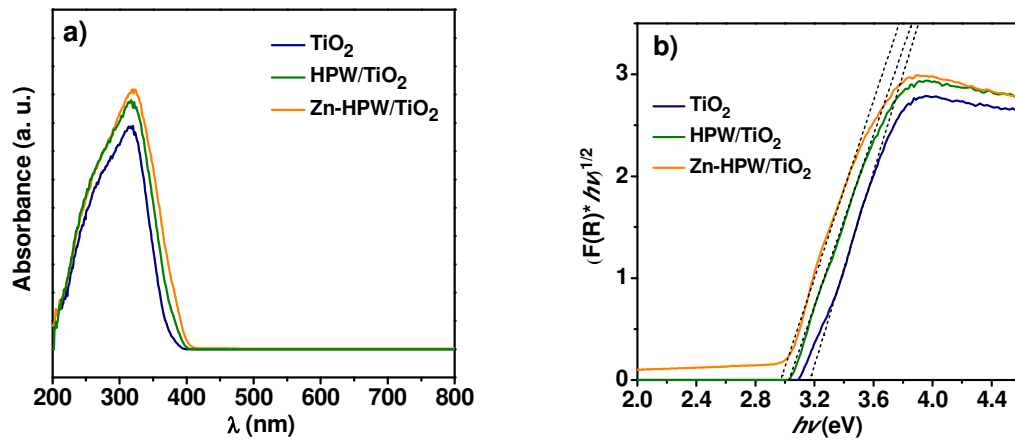


Figure S6 a. UV-Vis spectra of Zn-HPW/TiO₂ sample b. $[F(R_\infty)h\nu]^{1/2}$ versus $h\nu$ for Zn-HPW/TiO₂ sample

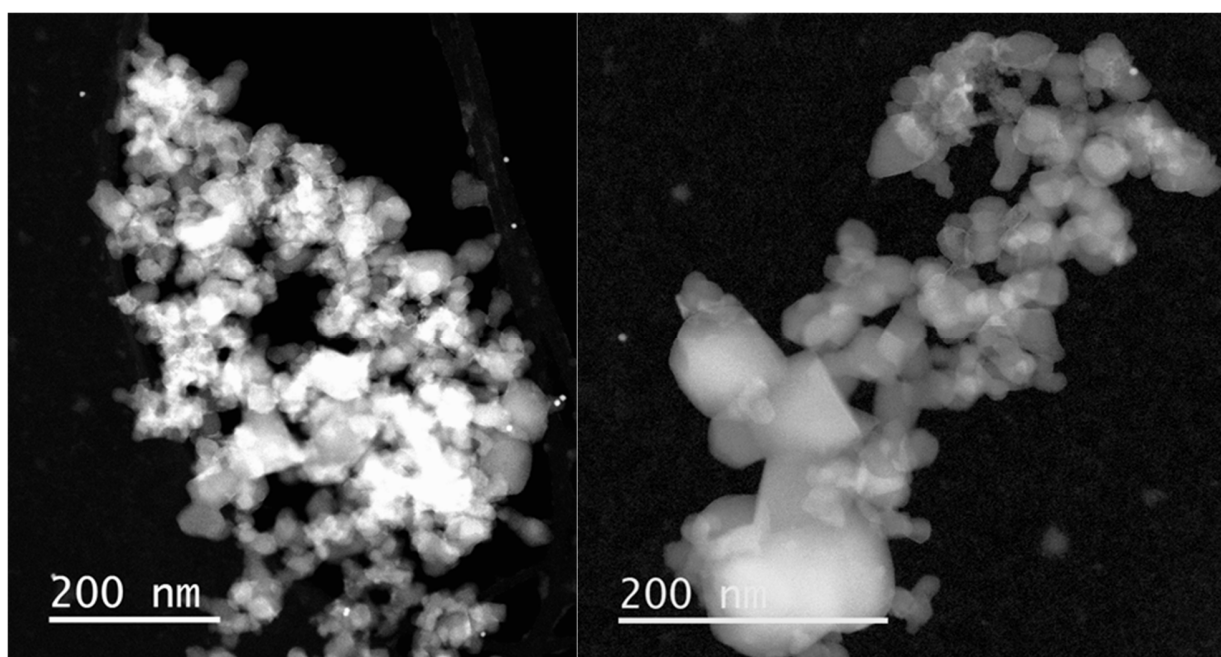


Figure S7. TEM images of the Zn-HPW/TiO₂ composite

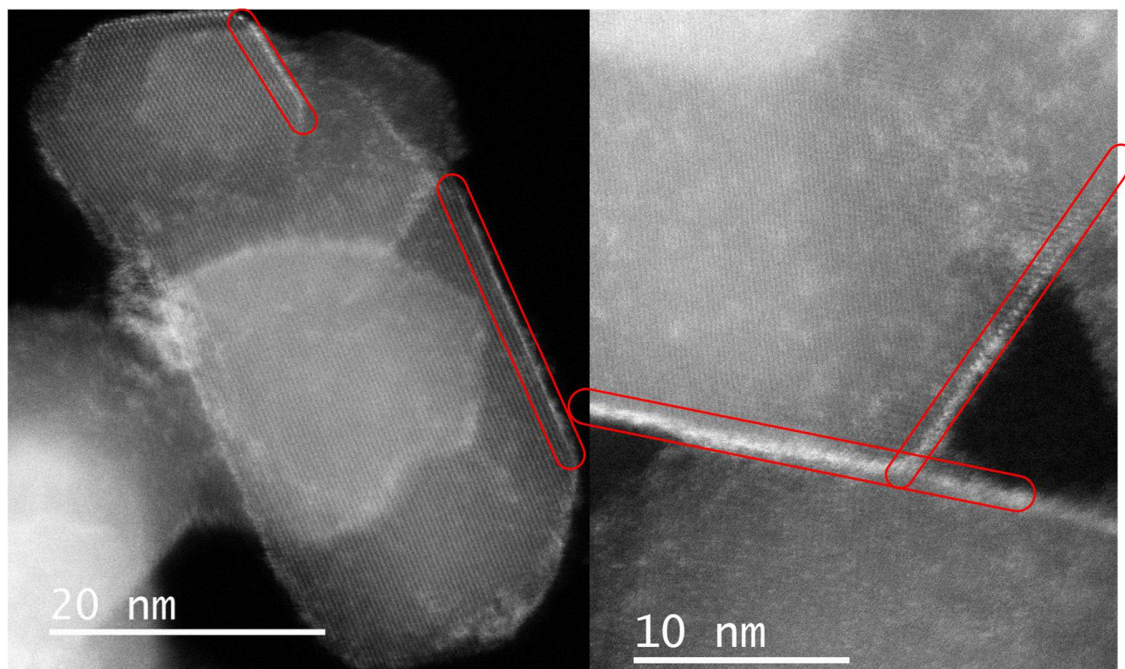


Figure S8. High resolution STEM-HAADF images showing the formation of the core shell structures in Zn-HPW/TiO₂

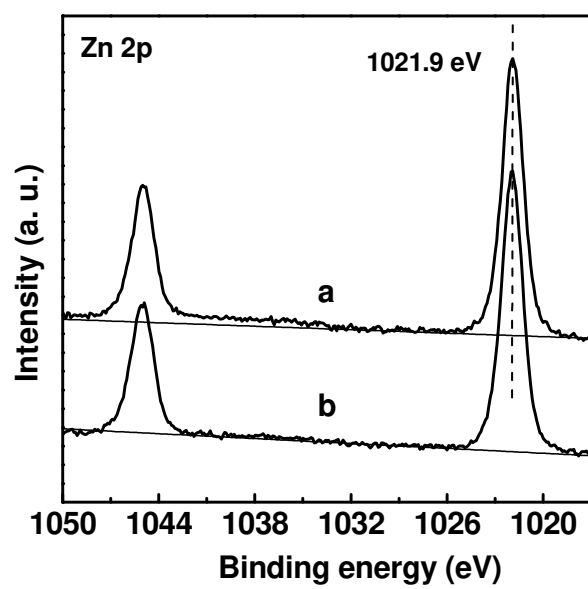


Figure S9. Zn 2p XPS spectra of Zn-HPW/TiO₂ catalysts: a. fresh catalyst, b. used catalyst.

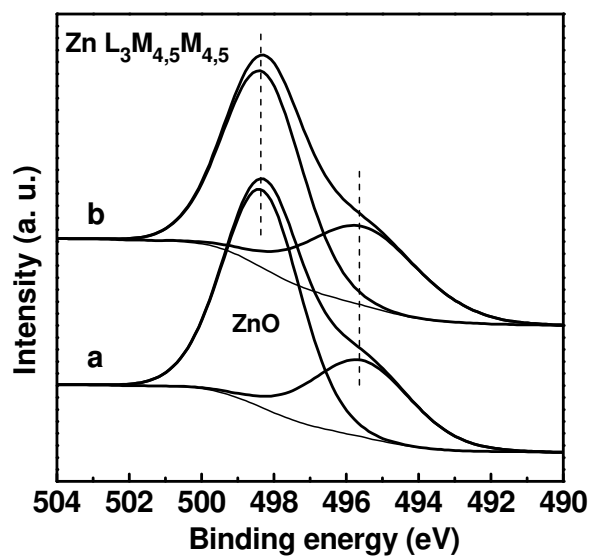


Figure S10. Zn $L_{3}M_{4.5}M_{4.5}$ Auger peaks for the Zn-HPW/TiO₂ catalysts; a. fresh catalyst, b. used catalyst.

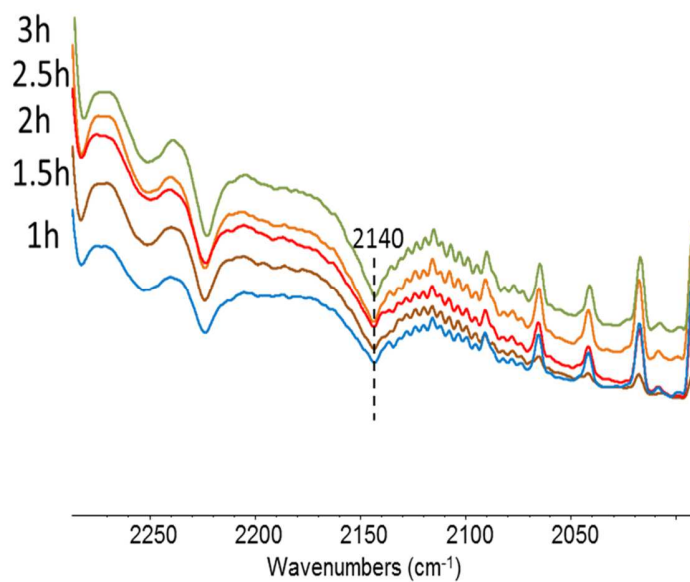


Figure S11. In situ FTIR spectra of the gas phase over the Zn-HPW/TiO₂ catalyst measured under light at different reaction times in CO₂ and H₂O vapor.