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Supporting Information

Wrapping CuCo₂S₄ Arrays on Nickel Foam with Ni₂(CO₃)(OH)₂ Nanosheets as

High-performance Faradaic Electrode

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Figure S1 EDS spectrum and atom ratio of the NF/CuCo₂S₄-200@Ni₂(CO₃)(OH)₂ electrode with the CuCo₂S₄ synthesized at 200 °C.



Figure S2 (a) SEM images; (b) area full element mapping of NF/CuCo₂S₄- $200@Ni_2(CO_3)(OH)_2$ and Elemental mapping of C, Co, Cu, Ni, O and S.



Figure S3 XRD patterns of the NF/CuCo₂S₄@Ni₂(CO₃)(OH)₂ electrodes with CuCo₂S₄ synthesized at hydrothermal temperature of 120, 160, 200 °C, respectively.



Figure S4 EIS plots of NF/CuCo₂S₄-200 electrode (inset, equivalent circuit diagram of the NF/CuCo₂S₄-200@Ni₂(CO₃)(OH)₂ electrode fitting the electrochemical impedance spectra).



Figure S5 SEM images after charge/discharge 10000 cycles for the $Ni_2(CO_3)(OH)_2$ -wrapped electrode of NF/CuCo₂S₄-200@Ni₂(CO₃)(OH)₂ (a) and the bare electrode of NF/CuCo₂S₄-200 (b).



Figure S6 The brightness of light-emitting diode (LED) bulb at different time

Table S1. Comparison the electrochemical performance of NF/CuCo $_2S_4$ -200@Ni $_2(CO_3)(OH)_2$ with the electrode material in literature.

electrode material	electrolyte	specific capacity (C g ⁻¹)	cycling stability	ref.
flower-like CuCo ₂ S ₄	2 М КОН	363.6 (908.9 F g ⁻¹) at 5 mA cm ⁻²	91.1% (2000 cycles)	36
CuCo ₂ S ₄ nanospheres	3 М КОН	53.5 mF cm ⁻² at 0.17 mA cm ⁻²	86% (5500 cycles)	41
CuCo ₂ S ₄ /rGO composite	3 М КОН	262.5 (525 F g ⁻¹) at 1 A g ⁻¹	83% (1000 cycles)	46
NiO nanowall arrays	1 M KOH	148.5 (270 F g ⁻¹) at 0.67 A g ⁻¹	93% (4000 cycles)	47
CuCo ₂ S ₄ hollow spheres	6 M KOH	546 (1137.5 F g ⁻¹) at 2 A g ⁻¹	94.9% (6000 cycles)	48
NF/CuCo ₂ S ₄ -200@Ni ₂ (CO ₃)(OH) ₂ arrays	2 M KOH	343.9 at 0.3 A g ⁻¹	96.7% (10000 cycles)	This work