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## Supplementary Information for:

## Synthesis of Macrocyclic Polyphenol Resin by Methylene Crosslinked Calix[4]arene (MC-[4]H) for The Adsorption of Palladium and Platinum Ions

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Scheme S1 The synthetic route of 3 ([4]H)





Figure S1 FT-IR spectra of <sup>t</sup>Bu[4]H (2)

Wave number / cm <sup>-1</sup>	Assignment
781 & 815	aromatic C-H ( <mark>a</mark> )
1201	methylene bridge CH <sub>2</sub> ( <b>b</b> )
1480	aromatic C=C ( <mark>c</mark> )
2956	alkyl C-H sp <sup>3</sup> ( <b>d</b> )
3192	О-Н ( <mark>е</mark> )
	Wave number / cm <sup>-1</sup> 781 & 815 1201 1480 2956 3192

Table S1 FT-IR interpretation of <sup>t</sup>Bu[4]H (2)



Figure S2 <sup>1</sup>H-NMR spectra of <sup>t</sup>Bu[4]H (2)

<sup>1</sup> H-NMR	<i>δ</i> (ppm)	Assignment	Multiplicity	Integral ratio
300 MHz	1.26	а	S	8.91 (36H)
CDCl₃	3.64	b	S	0.69 (4H)
TMS	4.34	С	S	0.62(4H)
	7.05	d	S	1.95 (8H)
	10.3	е	S	1.00 (4H)

Table S2 <sup>1</sup>H-NMR interpretation of  ${}^{t}Bu[4]H$  (2)



Figure S3 The	FT-IR spectra	of [4]H (3)
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FT-IR	Wave number / cm <sup>-1</sup>	Assignment
KBr	751	aromatic C-H ( <mark>a</mark> )
	1268	Methylene bridge CH <sub>2</sub> (b)
	1595	aromatic C=C ( <mark>c</mark> )
	3208	О-Н ( <b>d</b> )

Table S3	FT-IR	inter	pretation	of	[4]H (	3)
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Figure S4 <sup>1</sup>H-NMR spectra of [4]H (3)

<sup>1</sup> H-NMR	<i>δ</i> (ppm)	Assignment	Multiplicity	<i>J</i> (Hz)	Integral ratio
400 MHz	3.53	а	S	0	1.00 (4H)
	4.23	b	S	0	0.87 (4H)
TIMS	6.72	С	t	8.00 & 8.00	0.94 (4H)
	7.04	d	d	8.00	1.81 (8H)
	10.8	е	S	0	0.95 (4H)

Table S4 <sup>1</sup>H-NMR interpretation of [4]H (3)



Figure S5	<sup>13</sup> C-NMR spectra	of [4]H	(3)	
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<sup>13</sup> C-NMR	<i>δ</i> (ppm)	Assignment
100 MHz	31.7	а
CDCl <sub>3</sub> TMS	122	b
11113	128	С
	129	d
	149	е

Table S5 <sup>13</sup>C-NMR interpretation of [4]H (3)



Figure S6 FT-IR spectra of [4]CH<sub>3</sub> (4)

Table S6 FT-IR	interpretation	of [4]CH <sub>3</sub>	(4)
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FT-IR	Wave number / cm <sup>-1</sup>	Assignment
KBr	772	aromatic C-H ( <mark>a</mark> )
	1022 & 1085	ether C-O ( <mark>b</mark> )
	1204 & 1247	methylene bridge CH <sub>2</sub> (c)
	1466	aromatic C=C ( <mark>d</mark> )
	2932	alkyl C-H sp <sup>3</sup> ( <b>e</b> )



Figure S7 <sup>1</sup> H-NMF	spectra of	[4]CH <sub>3</sub> (4)
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<sup>1</sup> H-NMR	<i>δ</i> (ppm)	Assignment	Multiplicity	Integral ratio
400 MHz	3.59	а	S	2.00 (4H)
CDCl <sub>3</sub>	4.05	b	S	3.35 (6H)
TMS	4.24	С	S	2.58 (6H)
	4.79	d	S	1.58 (4H)
	7.12	е	d	2.98 (6H)
	7.41	f	m	2.80 (6H)

Table S7  $^{1}$ H-NMR interpretation of [4]CH<sub>3</sub> (4)



<b>Figure S8</b>	<sup>13</sup> C-NMR s	pectra of	[4]CH <sub>3</sub> (4)
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<sup>13</sup> C-NMR	<i>δ</i> (ppm)	Assignment
100 MHz	30.6	а
	61.5	b
11015	122	С
	128	d
	135	е
	158	f

Table S8 <sup>13</sup>C-NMR interpretation of [4]CH<sub>3</sub> (4)



Figure S9 FT-IR Spectra of MC-[4]H (direct synthesis from 3 with s-trioxane)





Chemical Formula	% Massa	% Atomic	σ	Integral intensity	Ratio K
С	74.20	84.29	0.02	434372	0.2090725
0	14.91	12.71	0.04	64916	0.0919040
F	2.05	1.47	0.03	10067	0.0248272
S	1.80	0.77	0.02	16155	0.0319447
I	7.05	0.76	0.12	14990	0.0993228
Total	100	100			

Figure S10 The SEM-EDX of MC-[4]H





Chemical	% Massa	% Atomic	σ	Integral	Patio K
Formula	70 IVIA55A	70 Alonnic	0	intensity	Ratio R
С	72.33	82.26	0.02	395627	0.1904234
0	18.14	15.49	0.05	69063	0.0977741
Na	1.06	0.63	0.02	11518	0.0146870
S	1.52	0.65	0.02	12759	0.0252298
Zn	1.06	0.22	0.05	6535	0.0109805
Pd	5.88	0.76	0.09	19758	0.0750599
Total	100	100			

Figure S11 The SEM-EDX of complex MC-[4]H-Pd(II) (Low concentration of Pd(II))





Chemical Formula	% Massa	% Atomic	σ	Integral intensity	Ratio K
С	66.87	78.60	0.02	321772	0.1548754
0	20.24	17.86	0.05	72423	0.1025313
Mg	3.13	1.82	0.03	33257	0.0428476
S	1.40	0.62	0.02	10758	0.0212740
Pd	8.35	1.11	0.10	25764	0.0978793
Total	100	100			

Figure S12 The SEM-EDX of complex MC-[4]H-Pd(II) (High concentration of Pd(II))