

# Supramolecular tripodal Au(I) assemblies in water. Interactions with pyrene fluorescent probe.

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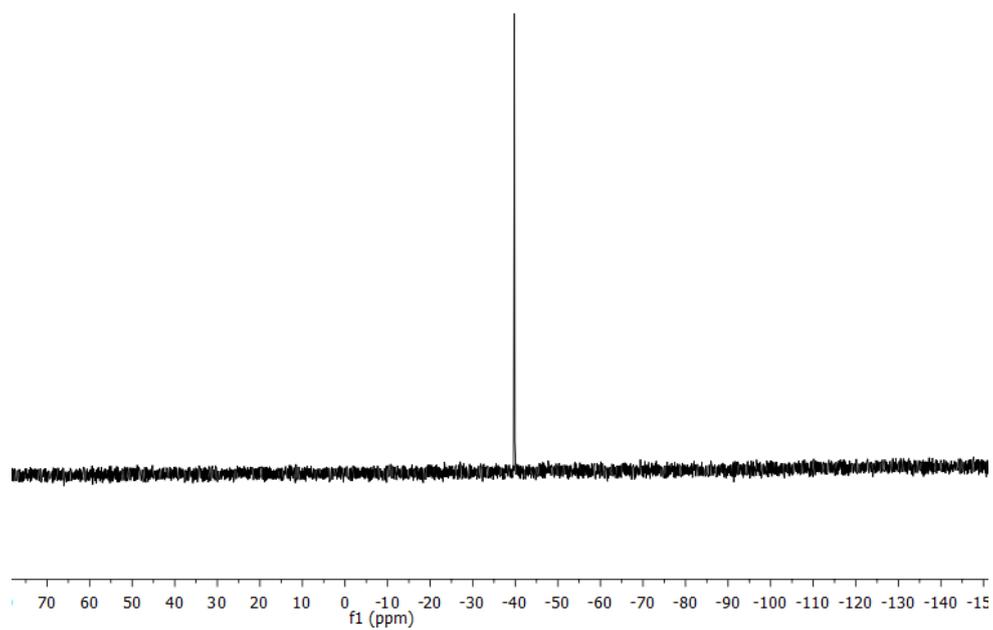
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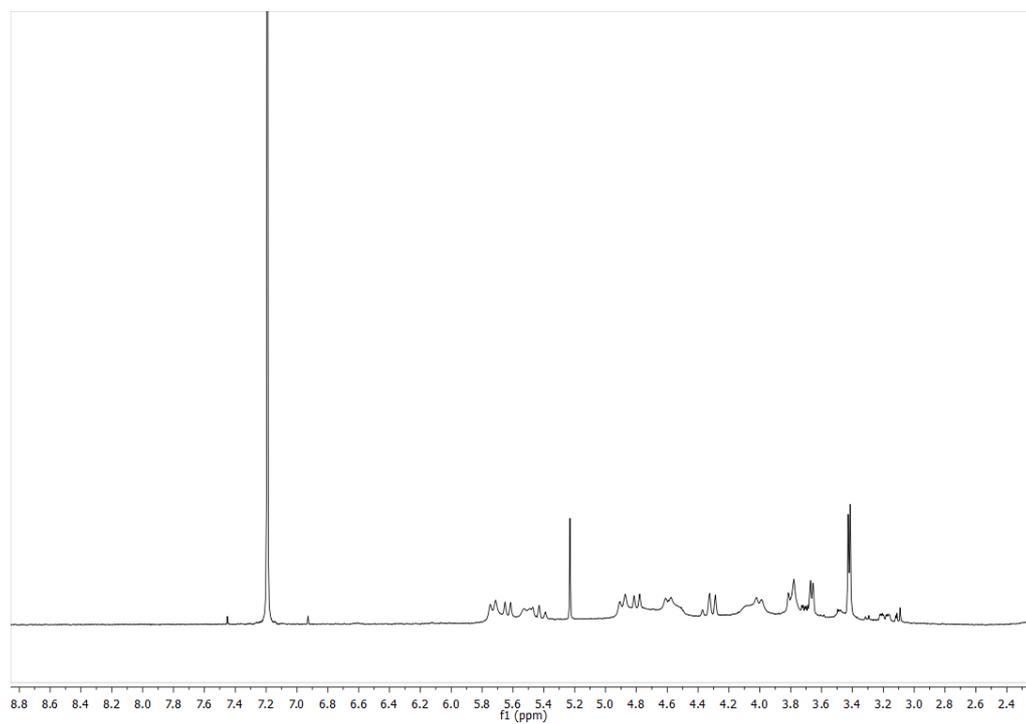
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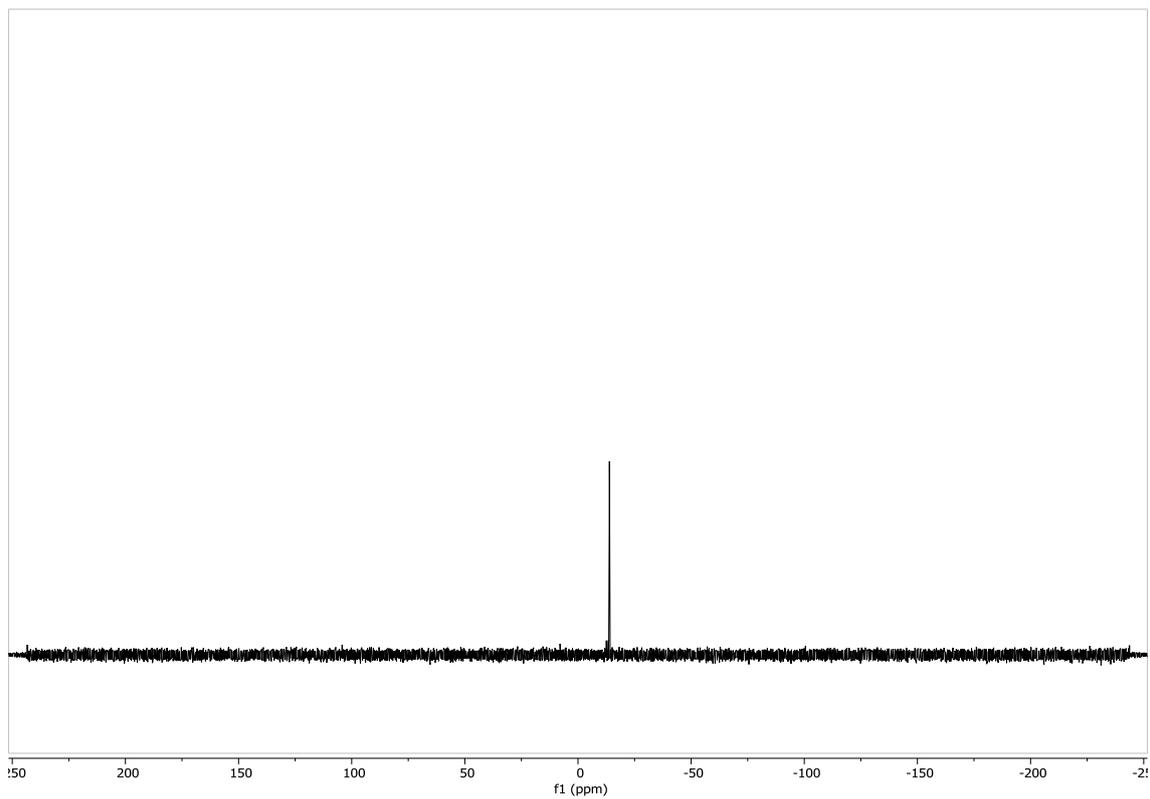
## Supporting Information



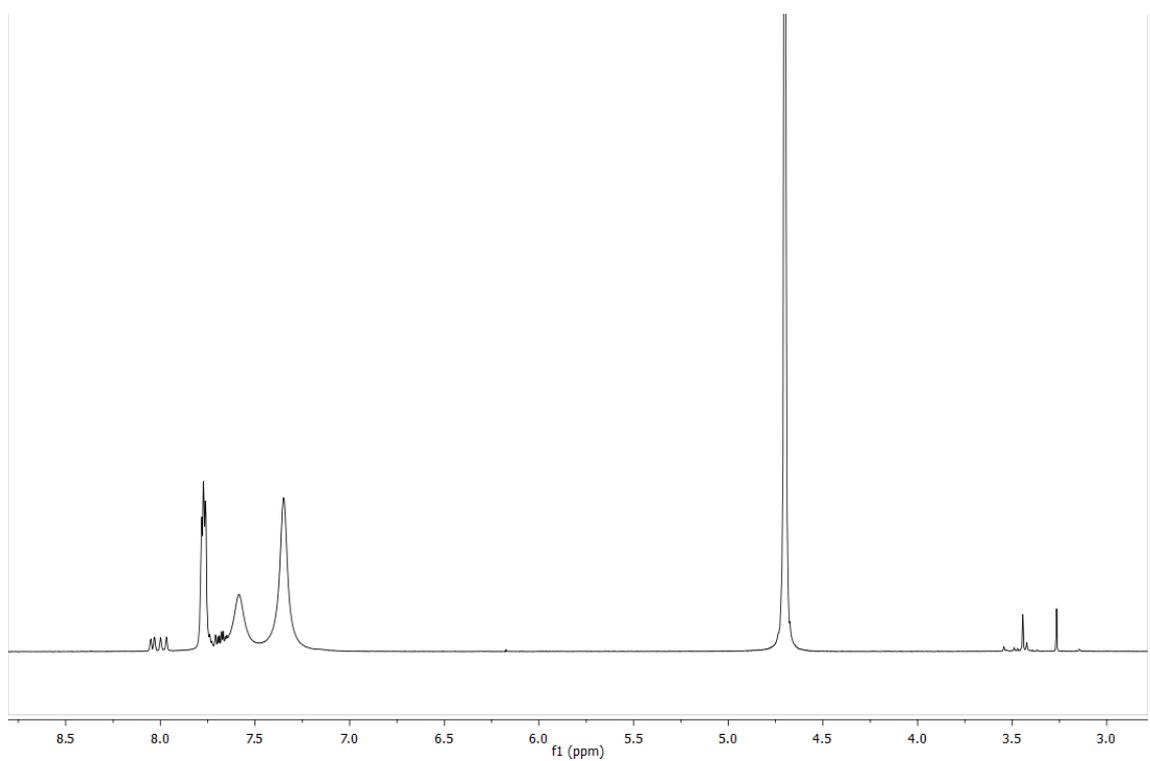
**Figure S1.**  $^{31}\text{P}$  NMR spectrum of **1** in  $\text{CDCl}_3$ .



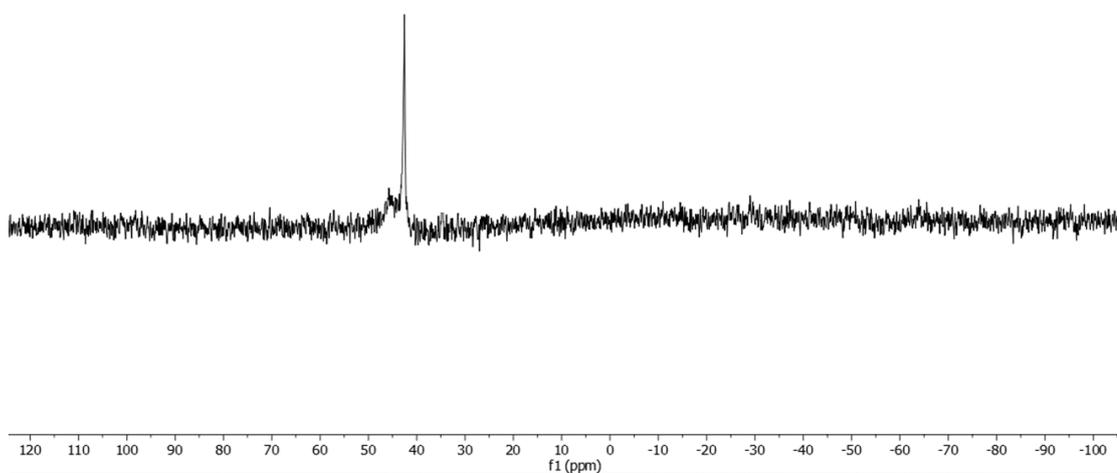
**Figure S2.**  $^1\text{H}$  NMR spectrum of **2** in  $\text{CDCl}_3$ .



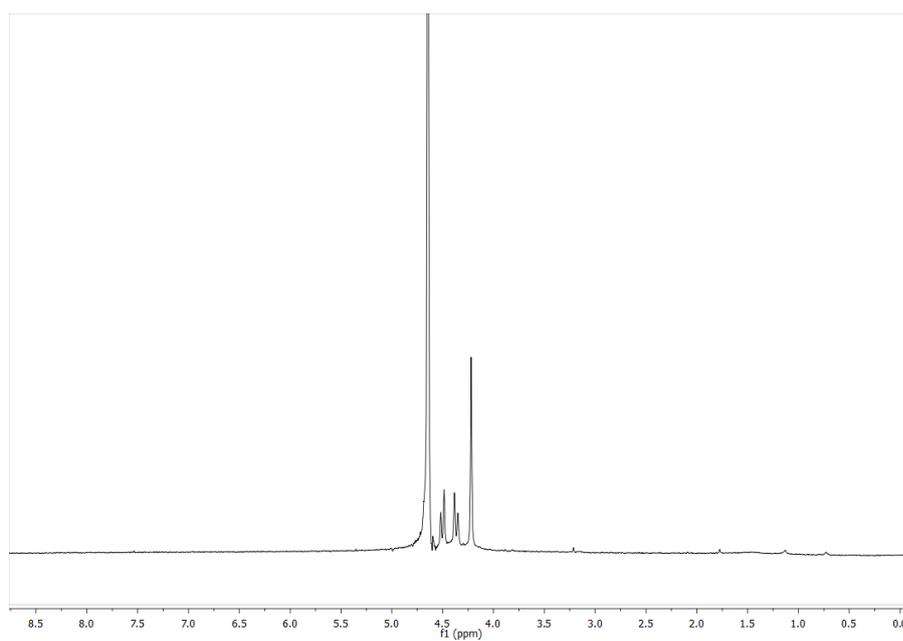
**Figure S3.**  $^{31}\text{P}$  NMR spectrum of **2** in  $\text{D}_2\text{O}$ .



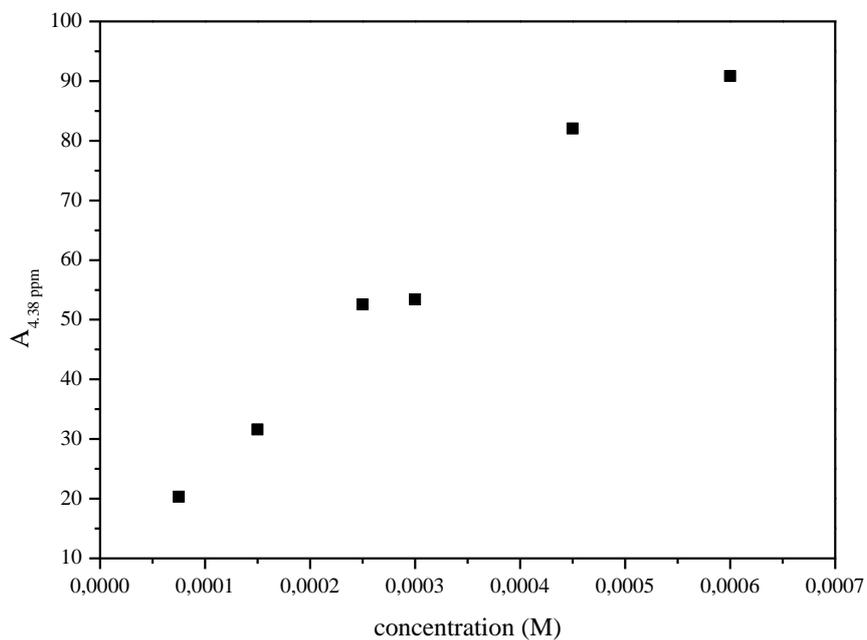
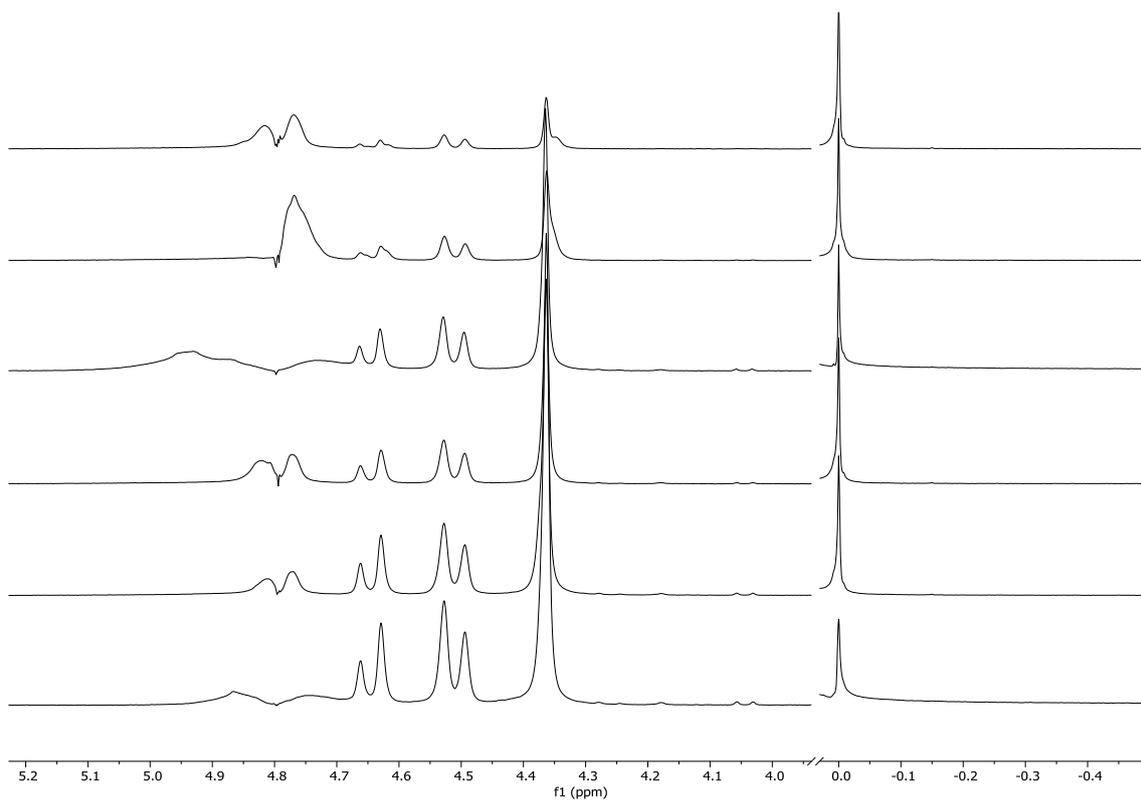
**Figure S4.**  $^1\text{H}$  NMR spectrum of **3** in  $\text{D}_2\text{O}$ .



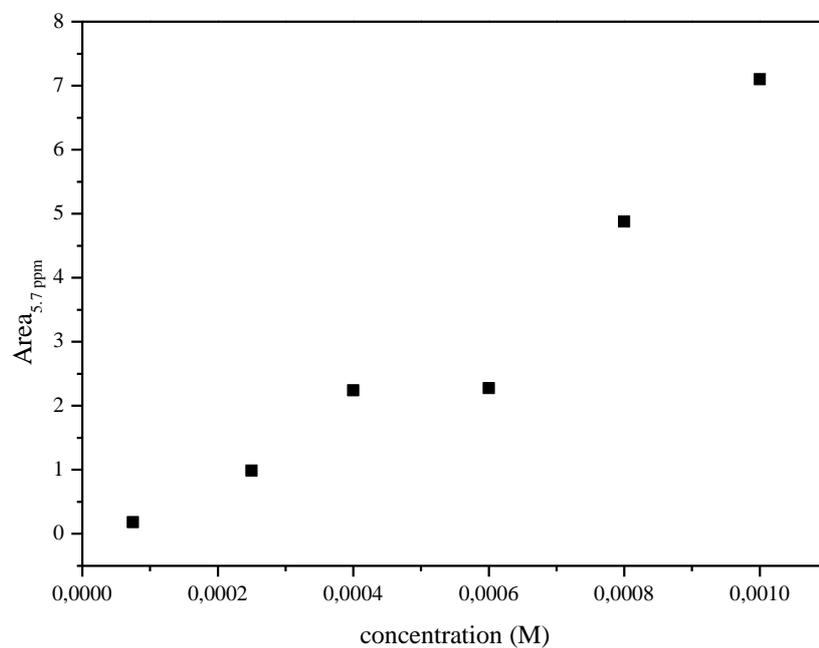
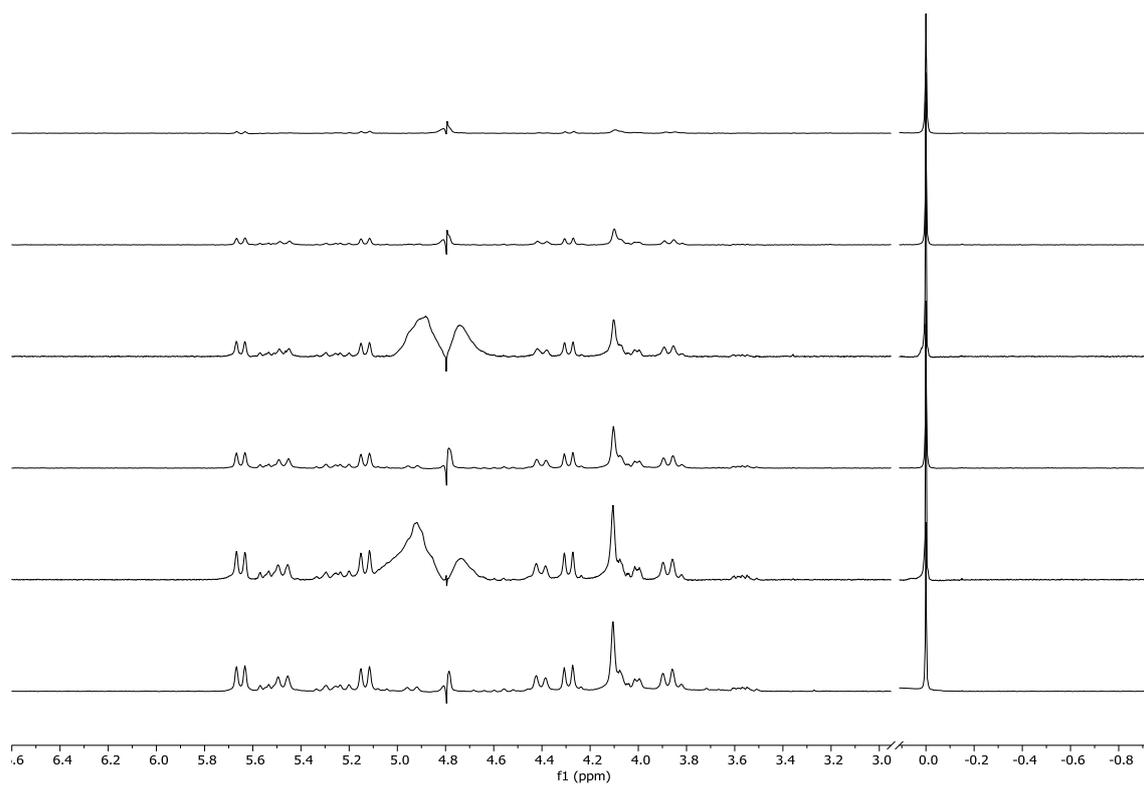
**Figure S5.**  $^{31}\text{P}$  NMR spectrum of **3** in  $\text{D}_2\text{O}$ .



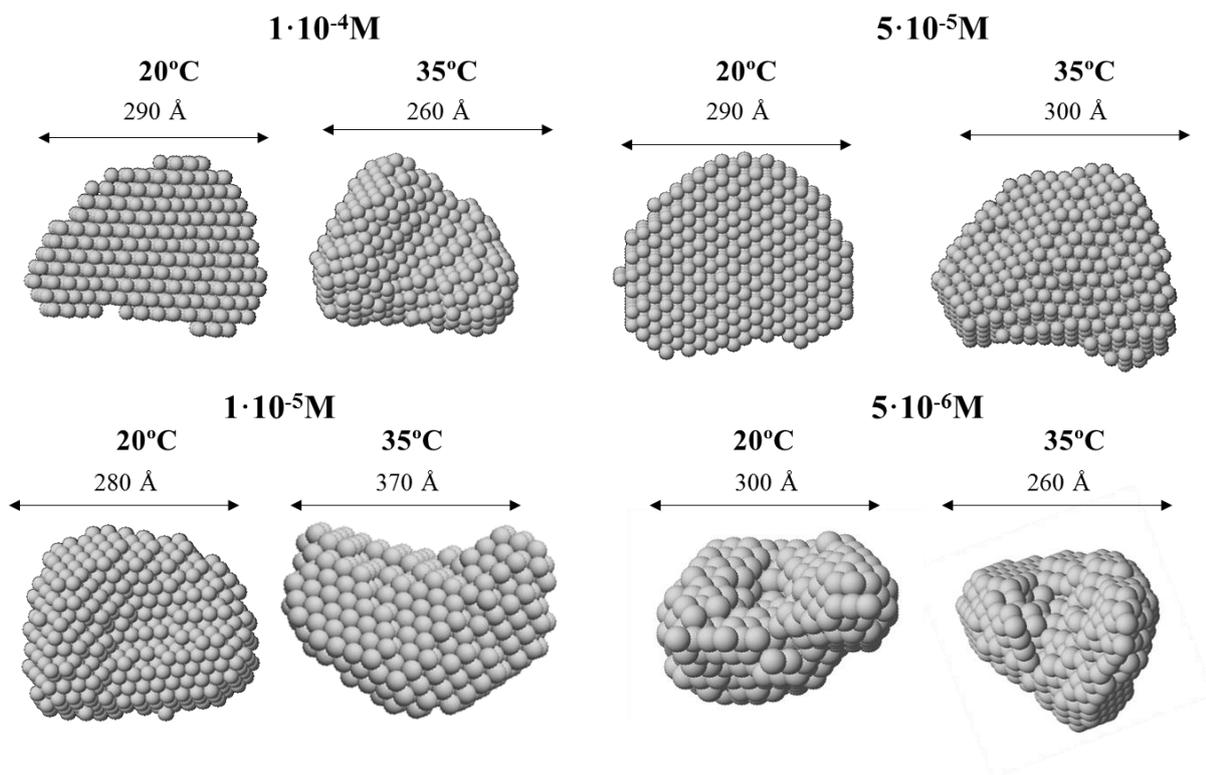
**Figure S6.**  $^1\text{H}$  NMR spectrum of **1** in  $\text{D}_2\text{O}$ .



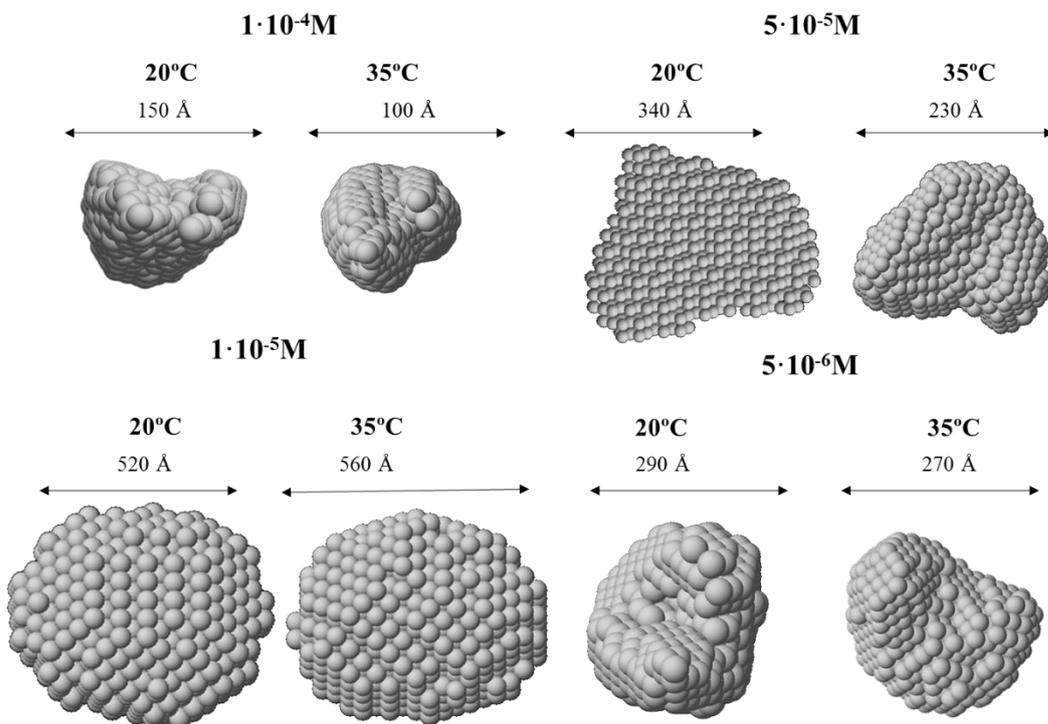
**Figure S7.** <sup>1</sup>H NMR spectra variations on the phosphine protons of **1** using TSP as standard reference (above) and on the area of the protons (below) with concentration. Deviations on the linearity are only observed at the highest recorded concentrations.



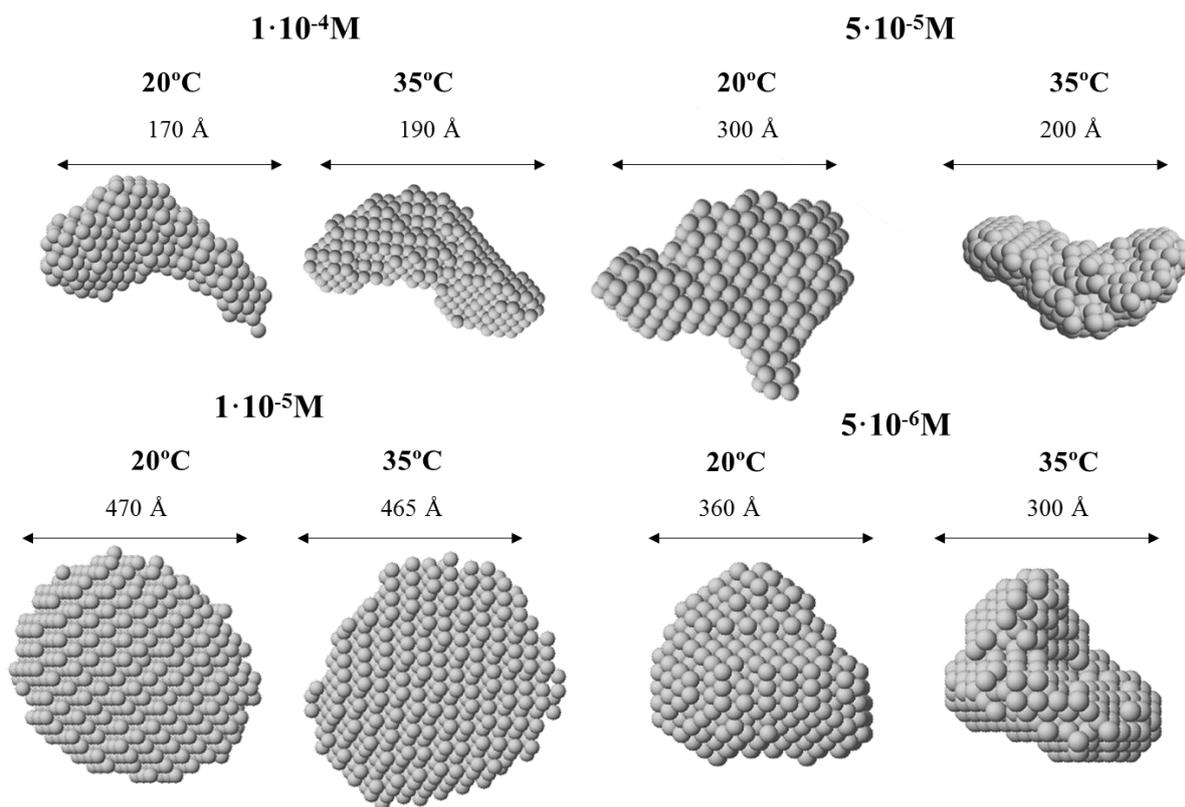
**Figure S8.**  $^1\text{H}$  NMR spectra variations on the phosphine protons of **2** using TSP as standard reference (above) and on the area of the protons (below) with concentration in  $\text{D}_2\text{O}$ .



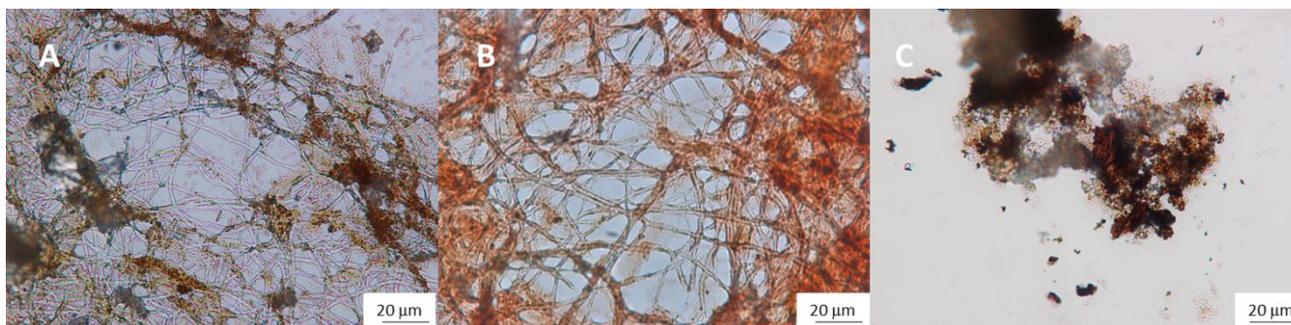
**Figure S9.** DAMMIN low-resolution structures reconstructed from SAXS patterns for of **1** at different concentrations in water.



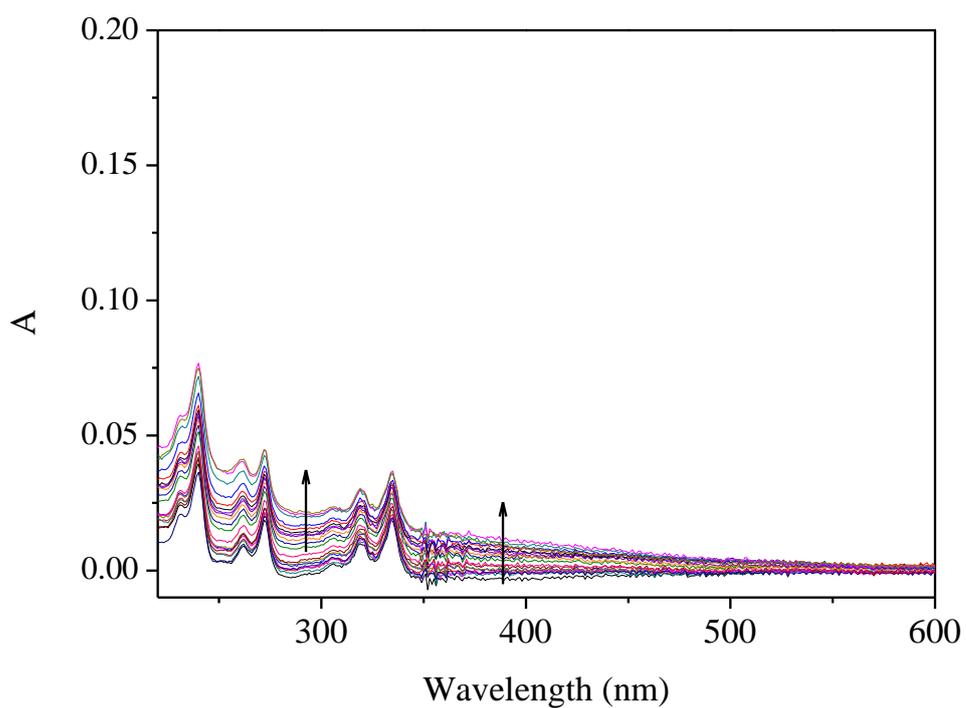
**Figure S10.** DAMMIN low-resolution structures reconstructed from SAXS patterns for of **2** at different concentrations in water.



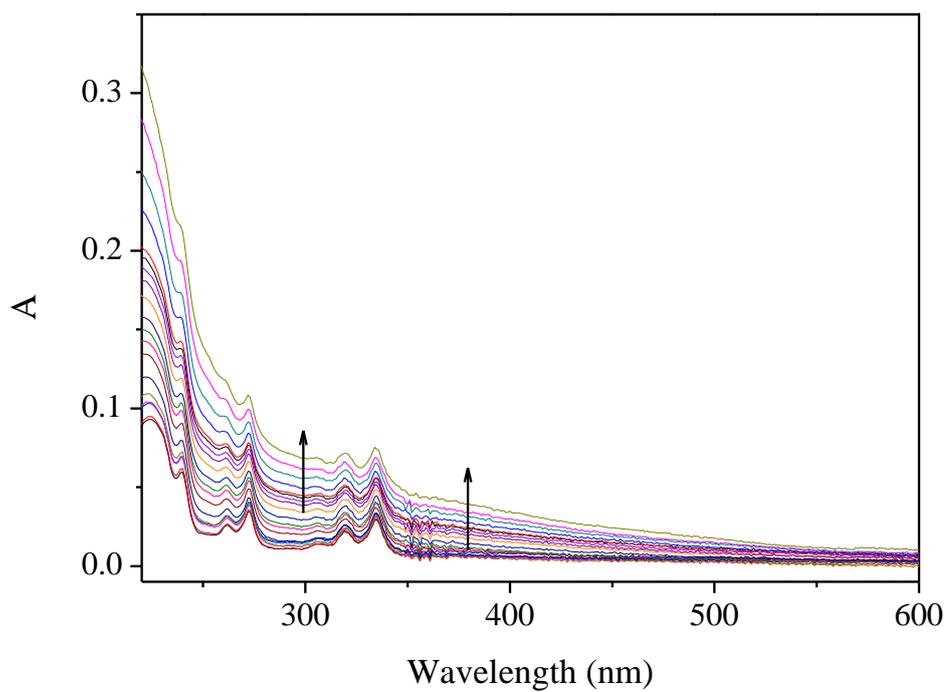
**Figure S11.** DAMMIN low-resolution structures reconstructed from SAXS patterns for of **3** at different concentrations in water.



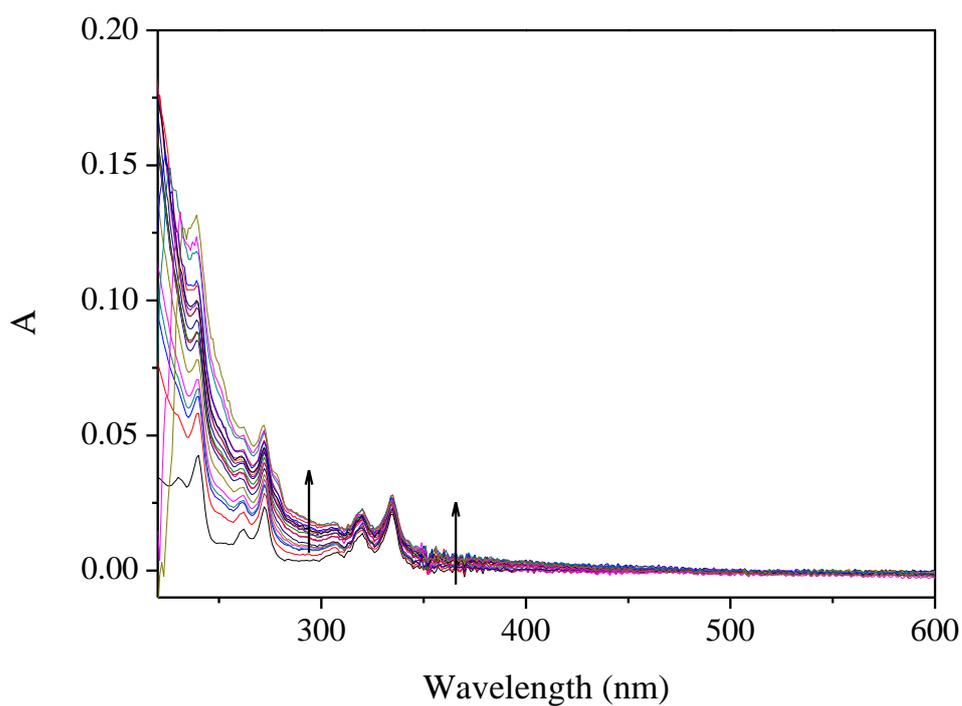
**Figure S12.** Optical microscopy images of fibers obtained from  $1 \cdot 10^{-4} \text{M}$  aqueous solutions of **1** (A), **2** (B) and **3** (C). 100x magnification.



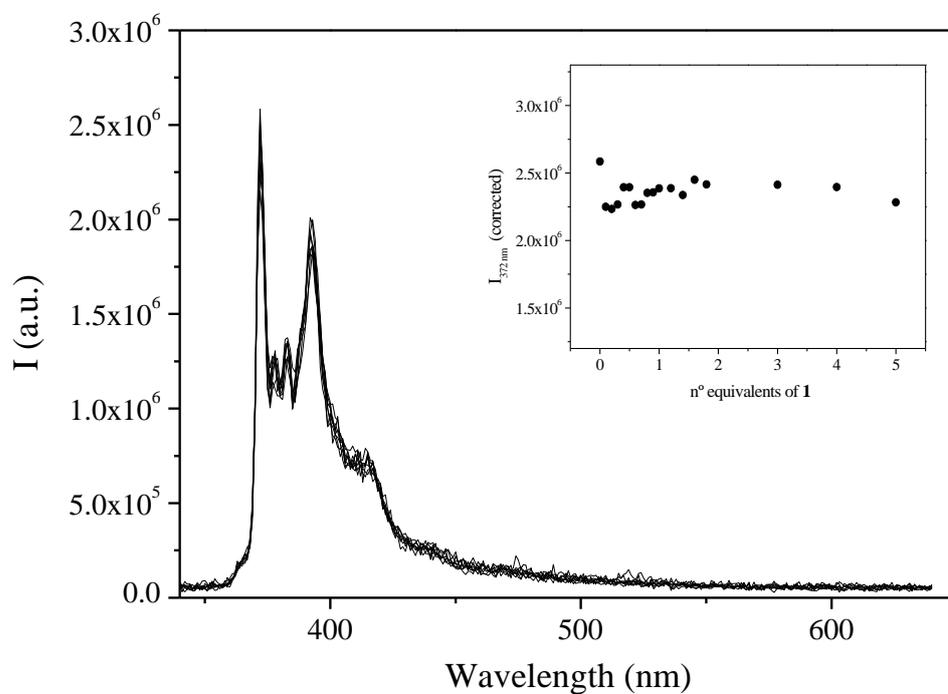
**Figure S13.** Absorption spectra of a  $1 \cdot 10^{-5} \text{ M}$  solution of pyrene in the presence of increasing amounts of **1**. [pyrene] =  $1 \cdot 10^{-5} \text{ M}$ . Solvent: water. pH  $\sim 7$ .



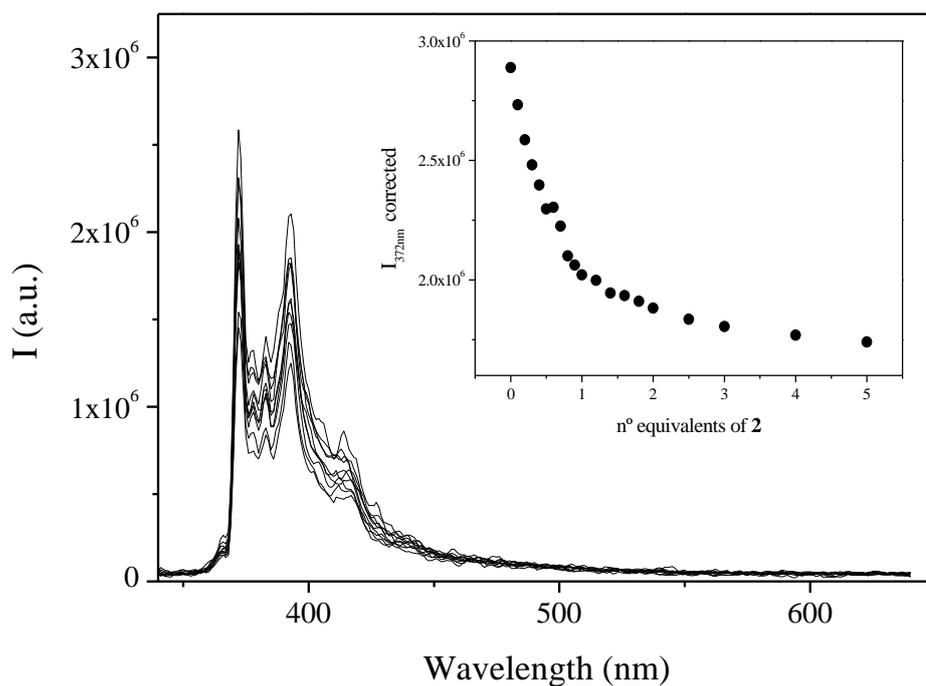
**Figure S14.** Absorption spectra of a  $1 \cdot 10^{-5} \text{ M}$  solution of pyrene in the presence of increasing amounts of **2**. [pyrene] =  $1 \cdot 10^{-5} \text{ M}$ . Solvent: water. pH  $\sim 7$ .



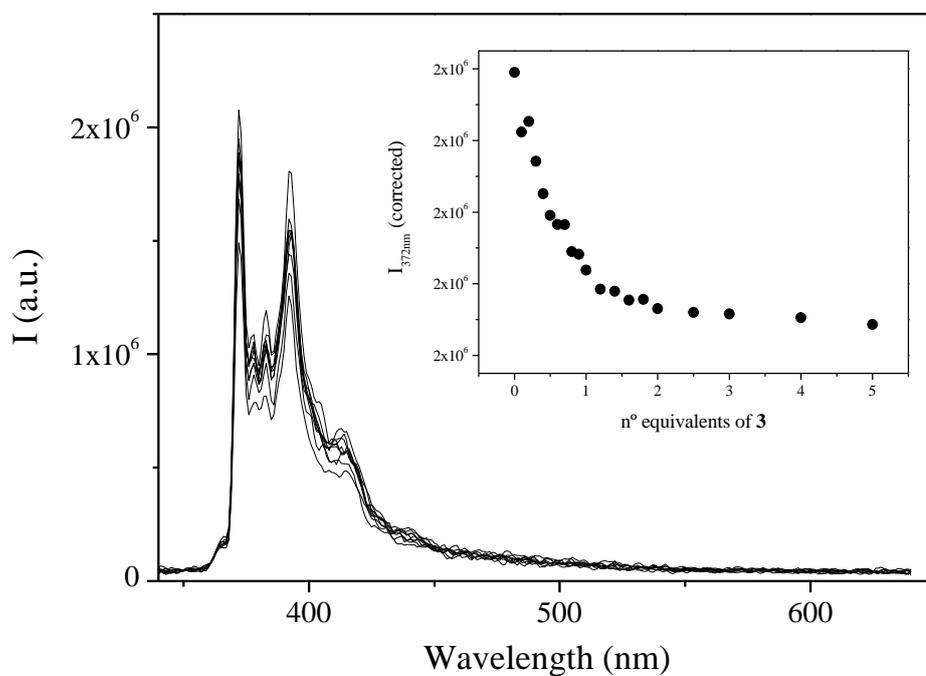
**Figure S15.** Absorption spectra of a  $1 \cdot 10^{-5}$ M solution of pyrene in the presence of increasing amounts of **3**. [pyrene] =  $1 \cdot 10^{-5}$ M. Solvent: water. pH  $\sim 7$ .



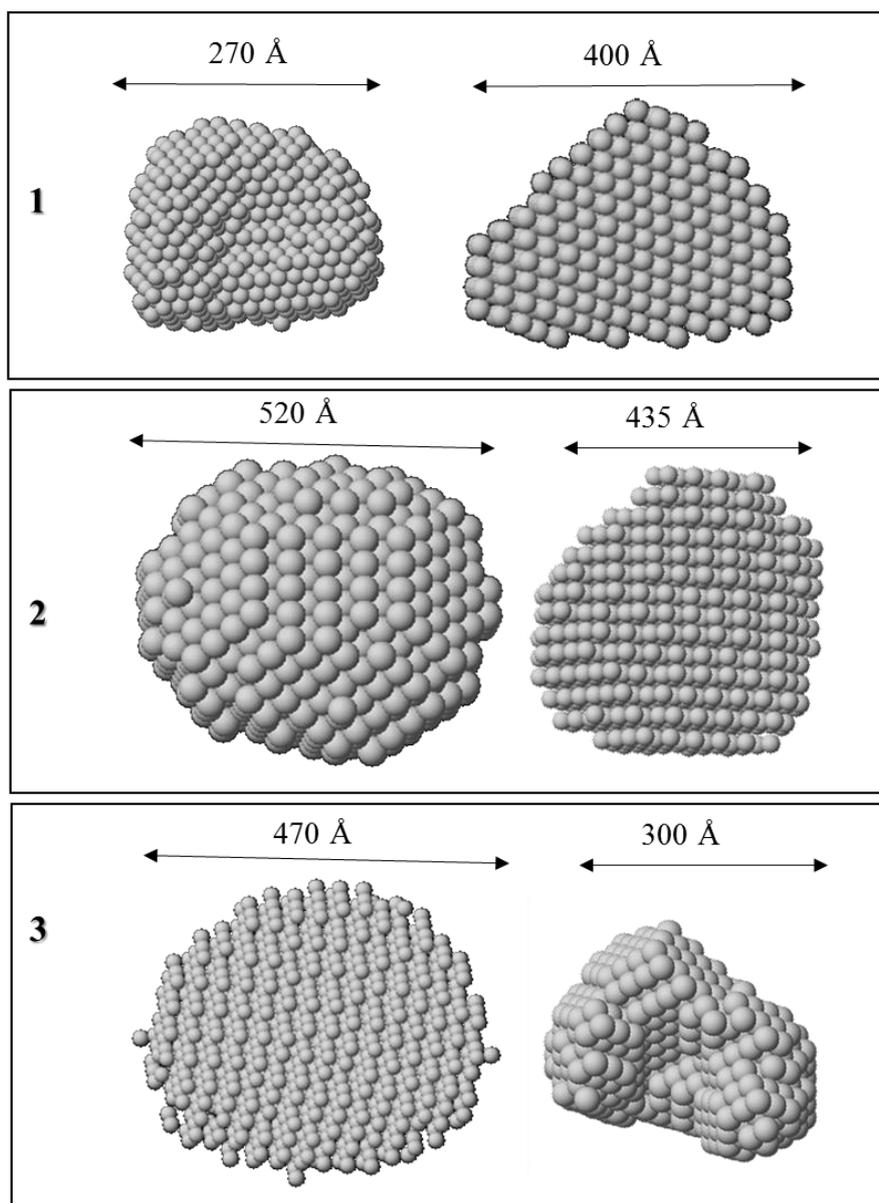
**Figure S16.** Emission spectra of a  $1 \cdot 10^{-5}$ M solution of pyrene in the presence of increasing amounts of **1**. Inset: plot of the variation of  $I_{372 \text{ nm}}$  vs number of equivalents of host. [pyrene] =  $1 \cdot 10^{-5}$ M. Solvent: water. pH  $\sim 7$ .



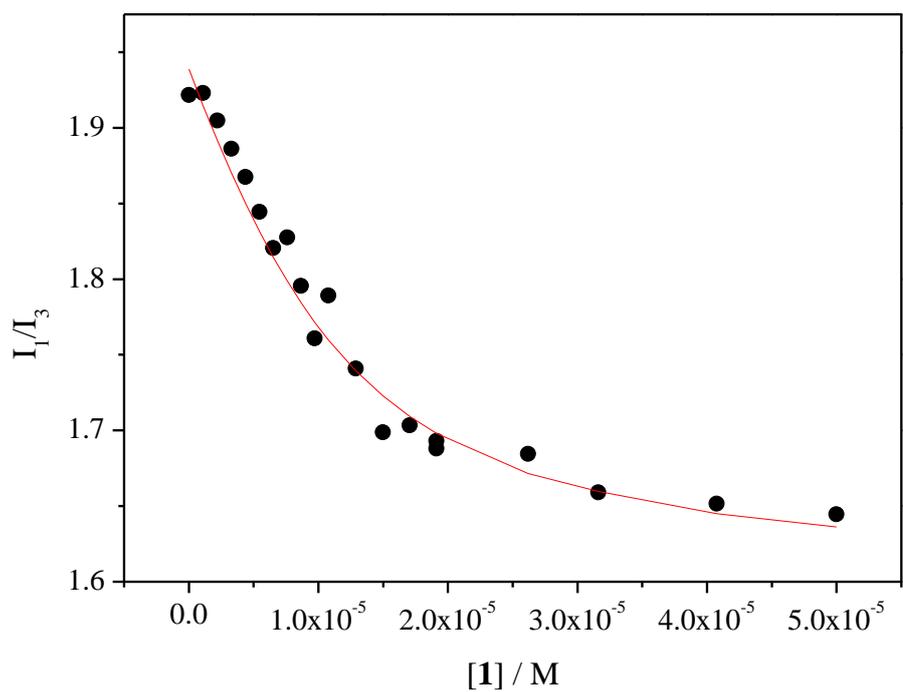
**Figure S17.** Emission spectra of a  $1 \cdot 10^{-5} \text{ M}$  solution of pyrene in the presence of increasing amounts of **2**. Inset: plot of the variation of  $I_{372 \text{ nm}}$  vs number of equivalents of host. [pyrene] =  $1 \cdot 10^{-5} \text{ M}$ . Solvent: water. pH  $\sim 7$ .



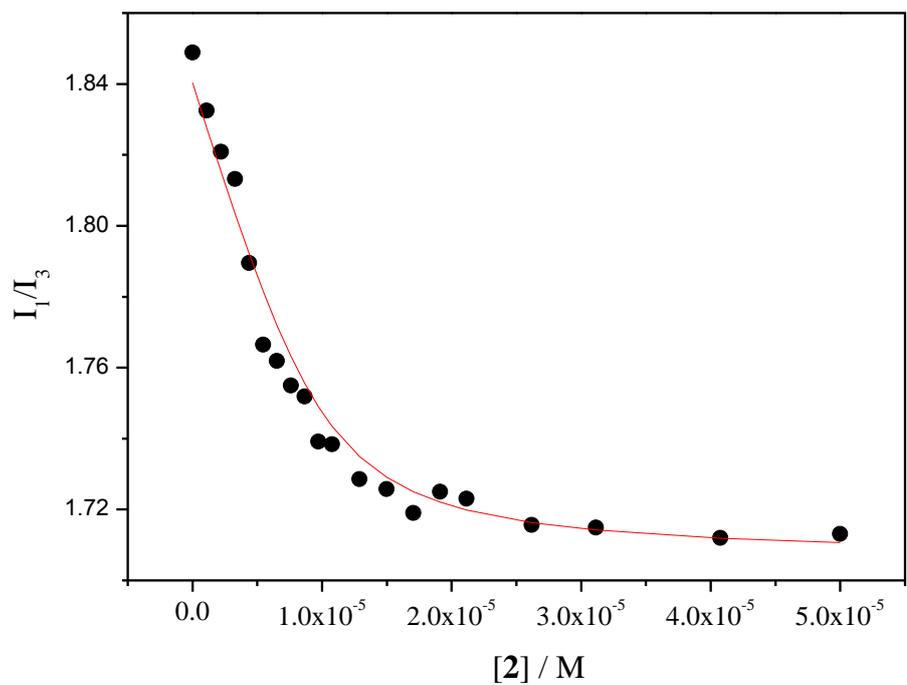
**Figure S18.** Emission spectra of a  $1 \cdot 10^{-5} \text{ M}$  solution of pyrene in the presence of increasing amounts of **3**. Inset: plot of the variation of  $I_{372 \text{ nm}}$  vs number of equivalents of host. [pyrene] =  $1 \cdot 10^{-5} \text{ M}$ . Solvent: water. pH  $\sim 7$ .



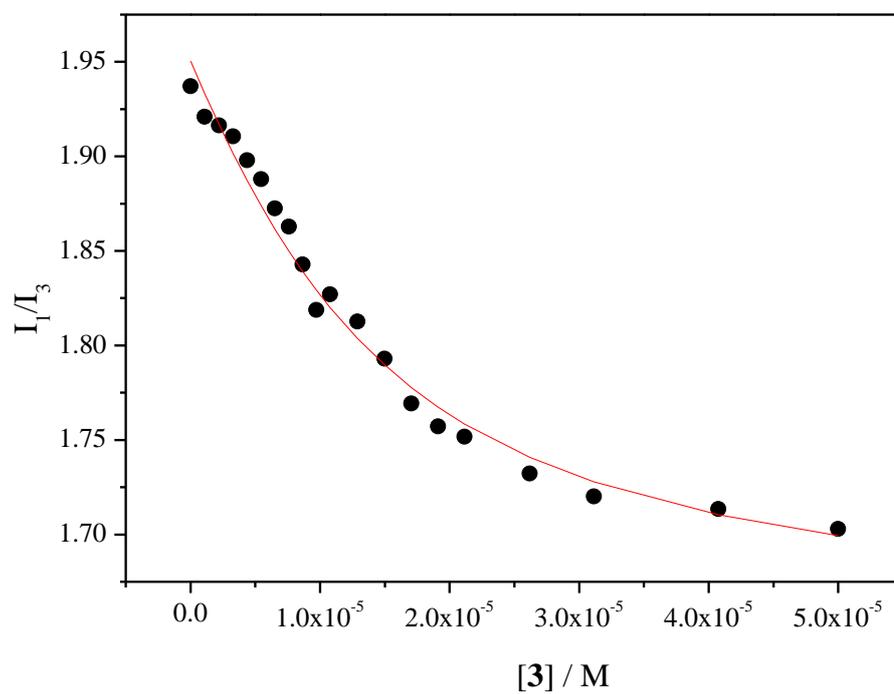
**Figure S19.** DAMMIN low-resolution structures reconstructed from SAXS patterns for  $1 \cdot 10^{-5}$  M solutions of **1** (top), **2** (middle) and **3** (bottom) in the presence of one equivalent of pyrene (right). The corresponding patterns of the solutions of the hosts have been also included in the left column for better comparison purposes.



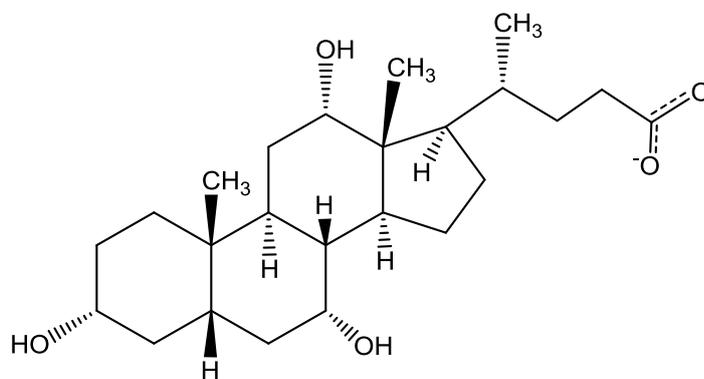
**Figure S20.** Plot of  $I_1/I_3$  in the emission spectra vs concentration of **1** (black dots) and fitting of the  $I_1/I_3$  emission data, assuming a 1:1 stoichiometry (red line).



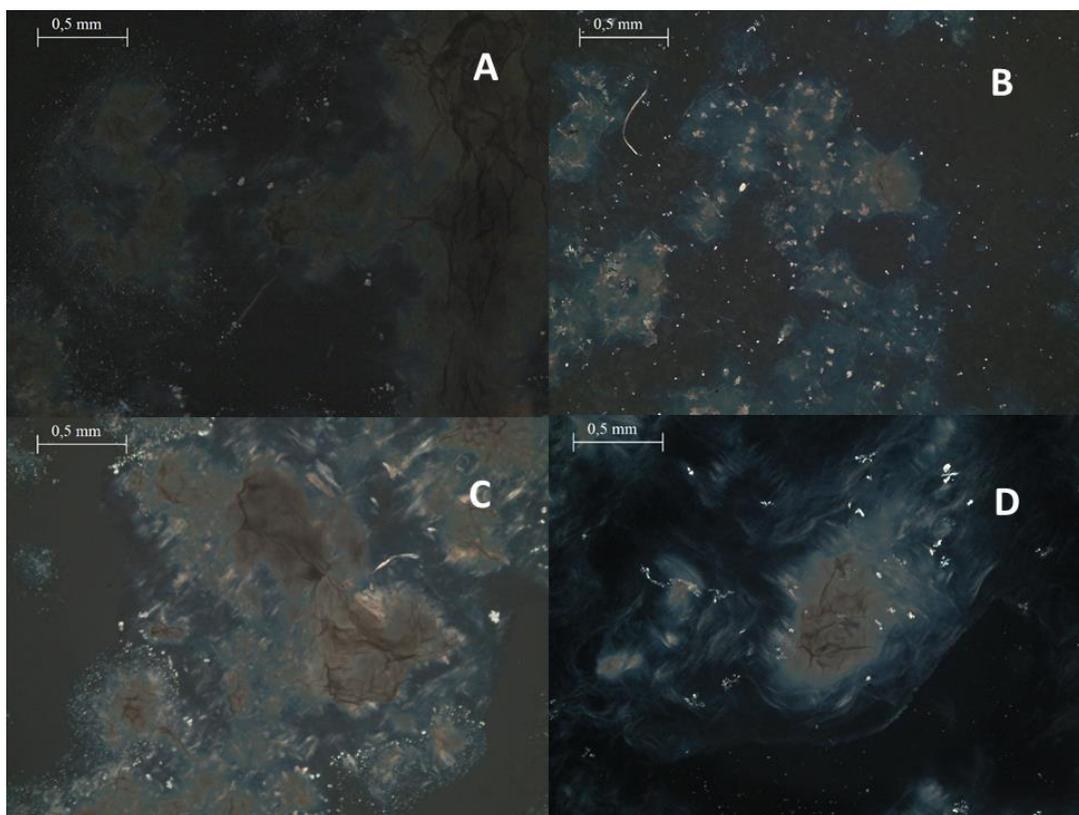
**Figure S21.** Plot of  $I_1/I_3$  in the emission spectra vs concentration of **2** (black dots) and fitting of the  $I_1/I_3$  emission data, assuming a 1:1 stoichiometry (red line).



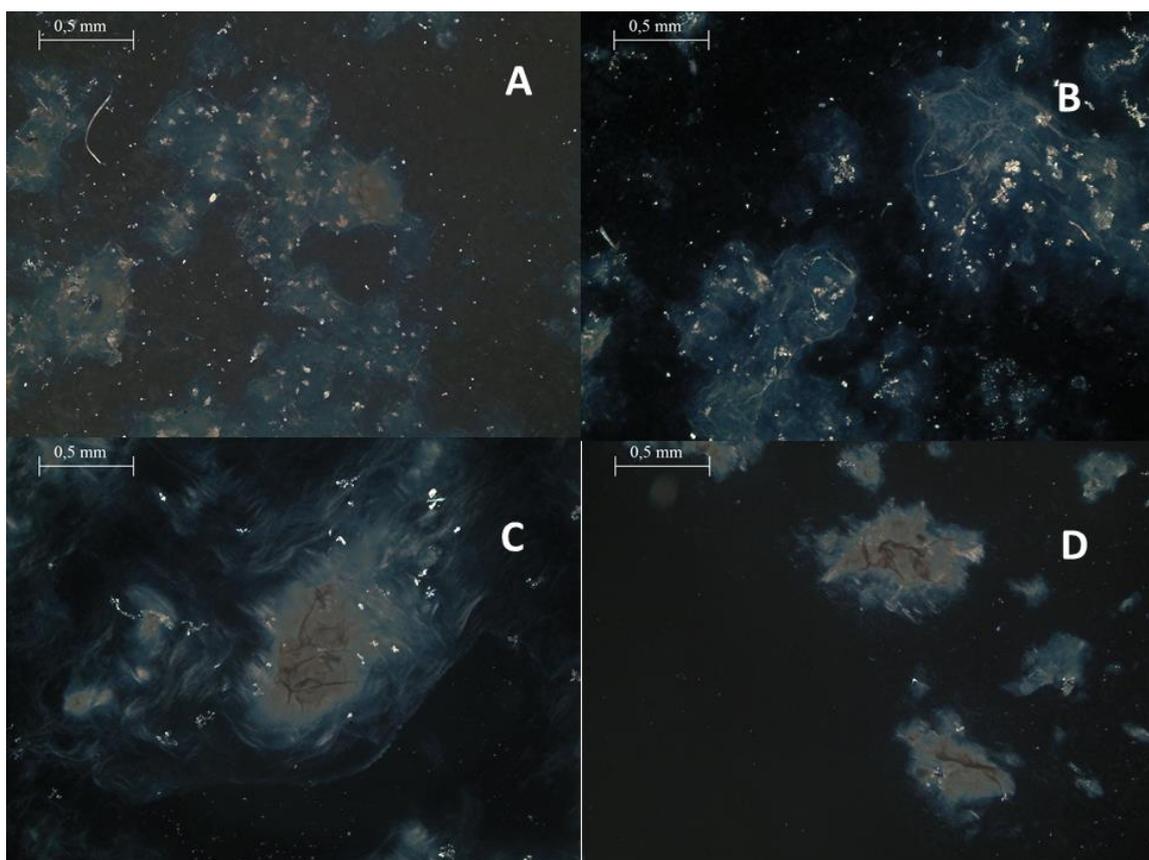
**Figure S22.** Plot of  $I_1/I_3$  in the emission spectra vs concentration of **3** (black dots) and fitting of the  $I_1/I_3$  emission data, assuming a 1:1 stoichiometry (red line).



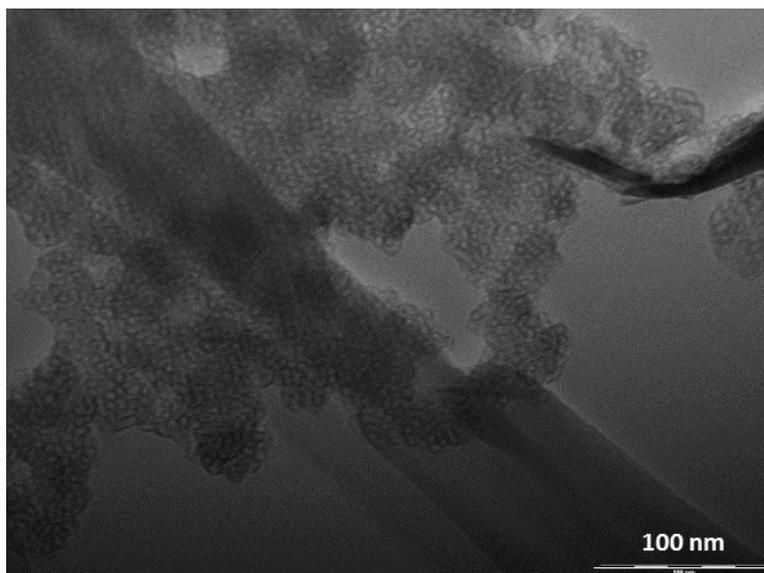
**Figure S23.** Chemical structure representation of cholate anion.



**Figure S24.** Optical microscopy image of dried samples of cholate hydrogel (A); **1** @ cholate (B); **2** @ cholate (C) and **3** @ cholate (D). Smaller and more brilliant aggregates are indicative of the presence of gold(I) complexes.



**Figure S25.** Optical microscopy image of dried samples of **1 @ cholate** (A); **1 : pyrene @ cholate** (B); **3 @ cholate** (C) and **3 : pyrene @ cholate** (D).



**Figure S26.** Transmission electron microscopy image of **1**: pyrene @ cholate after electron beam irradiation.