

Supporting Information

## Di-*tert*-butylcatecholate derivatives of titanocene

Pavel A. Petrov,<sup>\*ab</sup> Taisiya S. Sukhikh,<sup>ab</sup> Vladimir A. Nadolinny,<sup>a</sup>  
Artem S. Bogomyakov,<sup>c</sup> Yuliya A. Laricheva<sup>a</sup> and Alexandre V. Piskunov<sup>d</sup>

<sup>a</sup> Nikolaev Institute of Inorganic Chemistry, Siberian Branch of Russian Academy of Sciences, Novosibirsk, Russia

<sup>b</sup> Novosibirsk State University, Novosibirsk, Russia

<sup>c</sup> International Tomography Centre, Siberian Branch of Russian Academy of Sciences, Novosibirsk, Russia

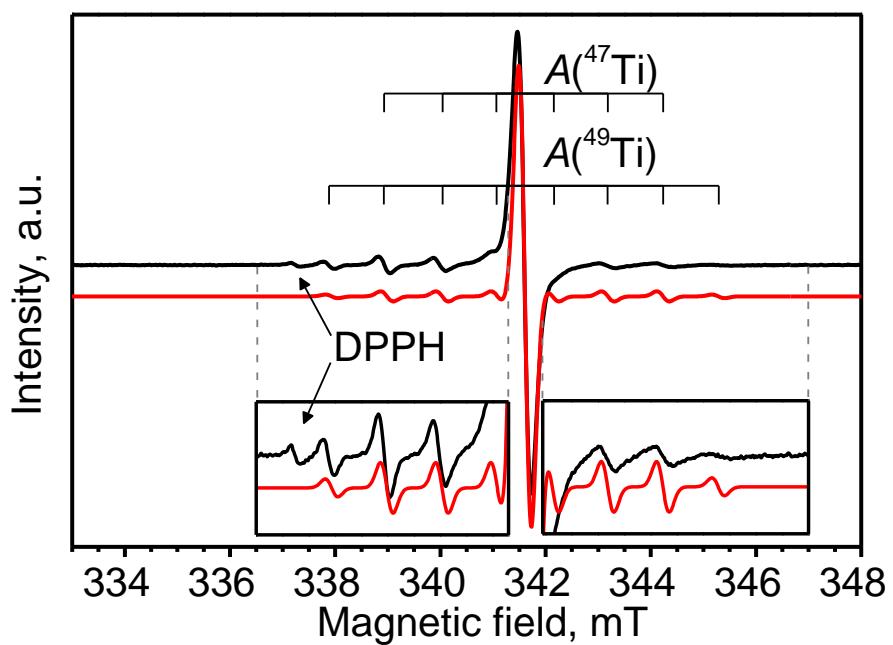
<sup>d</sup> G.A. Razuvayev Institute of Organometallic Chemistry, Russian Academy of Sciences, Nizhny Novgorod, Russia

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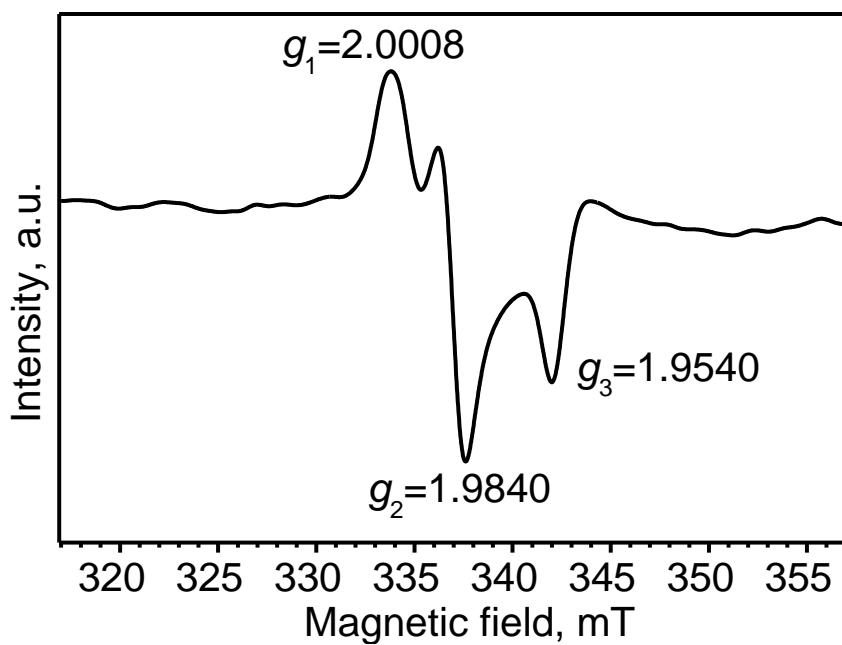
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**Table S1. Crystal Data and Data Collection and Refinement Details**

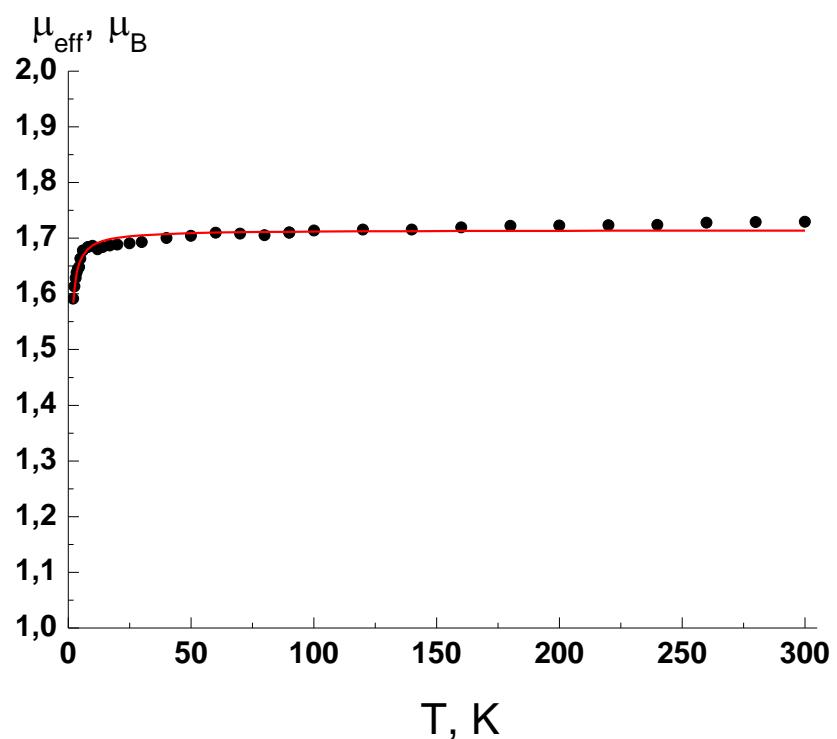
	<b>1</b>	<b>2</b>
Chemical formula	C <sub>24</sub> H <sub>30</sub> O <sub>2</sub> Ti	C <sub>43</sub> H <sub>55</sub> O <sub>4</sub> Ti <sub>2</sub>
M <sub>r</sub>	398.38	731.67
Crystal system, space group	Orthorhombic, <i>Pnma</i>	Monoclinic, <i>P2</i> <sub>1</sub>
Temperature (K)	150	173
<i>a</i> (Å)	14.9844(6)	10.1699(18)
<i>b</i> (Å)	17.5943(7)	32.789(6)
<i>c</i> (Å)	7.8829(3)	11.604(2)
β (°)	90	96.672(4)
<i>V</i> (Å <sup>3</sup> )	2078.25(14)	3843.3(12)
<i>Z</i>	4	4
μ (mm <sup>-1</sup> )	0.43	0.46
Crystal size (mm)	0.42 × 0.40 × 0.40	0.25 × 0.12 × 0.10
<i>F</i> (000)	848	1556
Θ range (°)	2.719–26.353	1.767–26.372
<i>h, k, l</i> limits	−18 ≤ <i>h</i> ≤ 17, −21 ≤ <i>k</i> ≤ 14, −9 ≤ <i>l</i> ≤ 9	−12 ≤ <i>h</i> ≤ 12, −40 ≤ <i>k</i> ≤ 40, −14 ≤ <i>l</i> ≤ 14
Reflections measured, independent and observed [ <i>I</i> > 2σ( <i>I</i> )]	11391, 2183, 2115	30499, 14812, 12535
<i>R</i> <sub>int</sub>	0.0198	0.0359
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> [ <i>F</i> <sup>2</sup> > 2σ( <i>F</i> <sup>2</sup> )]	0.0341, 0.0885	0.0408, 0.0870
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all data)	0.0353, 0.0896	0.0531, 0.0910
<i>GOOF</i>	1.063 (1.214)	1.031
No. of parameters	180	908
No. of restraints	48	1
Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )	0.46, −0.37	0.49, −0.31
CCDC	1895786	1895787



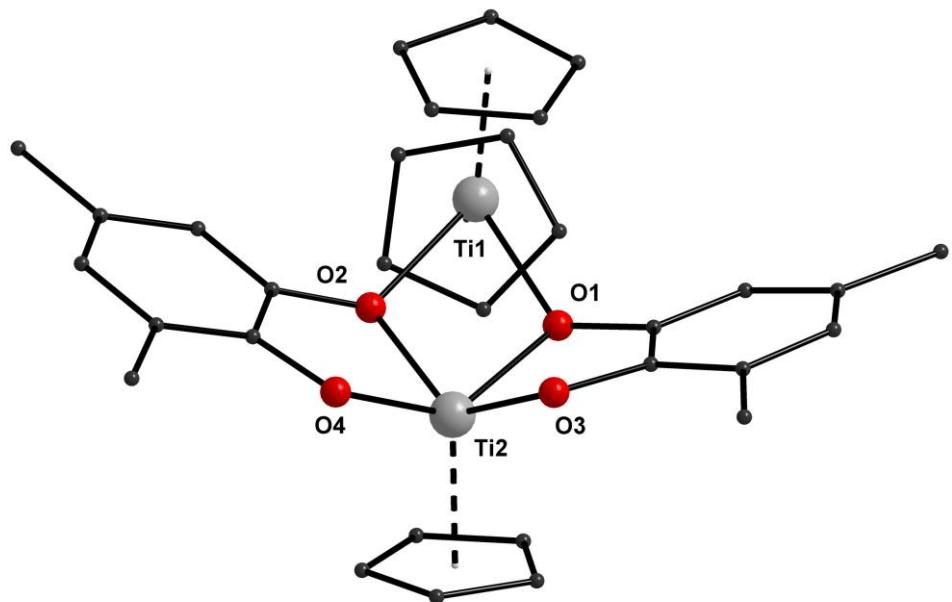
**Fig. S1.** ESR spectrum of THF solution of complex **2** at 300K in X-frequency band (black – experimental spectrum, red – simulation). In insets, the hyperfine structure of the high-gain ESR spectrum is shown.



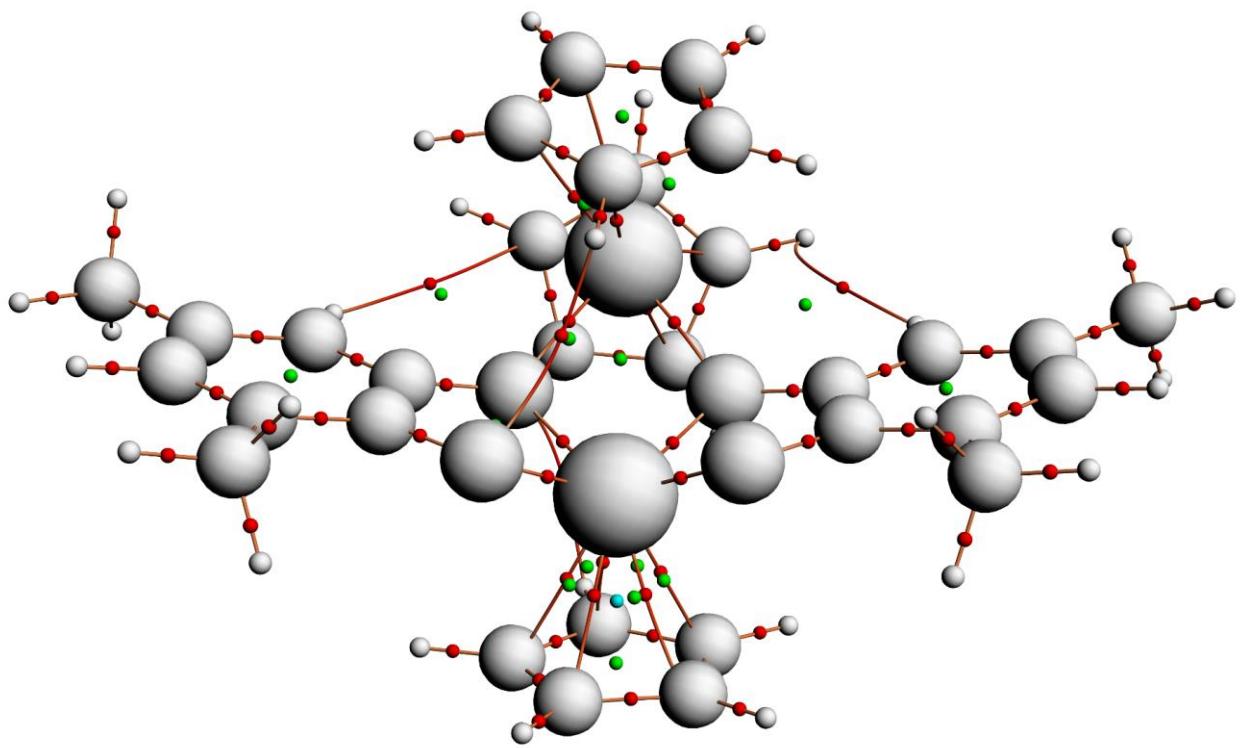
**Fig. S2.** EPR spectrum of frozen THF solution of complex **2** at 77K



**Fig. S3.**  $\mu_{\text{eff}}$  vs. T plot for the polycrystalline sample of complex **2** (solid line – theoretical curve).



**Fig. S4.** Structure of model complex **2m** used for DFT calculations.



**Fig. S5.** Critical points of electron density in complex **2m**.

**Table S2.** The energy values (eV) and occupations (electrons) of frontier orbitals of **2m**.

	BP86	B3LYP	M06L
LUMO+1	-3.480 (0.00)	-2.535 (0.00)	-2.903 (0.00)
LUMO	-3.465 (0.00)	-2.533 (0.00)	-2.894 (0.00)
SOMO	-3.836 (1.00)	-5.333 (1.00)	-3.616 (1.00)
SOMO-1 $\alpha\beta$	-5.000 (2.00)	-5.505 (2.00)	-4.650 (2.00)

**Table S3.** Hirshfeld and Bader charge analysis of **2m**.

Hirshfeld charges	BP86	B3LYP	M06L
Ti1	0.3110	0.3899	0.3630
Ti2	0.4701	0.5657	0.5365
O1	-0.1824	-0.2185	-0.2087
O2	-0.1815	-0.2177	-0.2080
O3	-0.2060	-0.2366	-0.2371
O4	-0.2051	-0.2368	-0.2363

Bader charges	BP86	B3LYP	M06L
Ti1	1.5555	1.6522	1.6191
Ti2	1.9185	2.0564	1.9953
O1	-1.0631	-1.1273	-1.0545
O2	-1.0637	1.1217	-1.0495
O3	-0.9923	-1.0341	-0.9446
O4	-0.9889	-1.0441	-0.9422

**Table S4.** BCP properties for Ti bonds, values in a.u.

Bond	Distance, Å	BP length, Å	$\rho(r)$	$\Delta\rho(r)$	BP86		$-\frac{1}{4}\nabla^2\rho(r)$	$H(r)/\rho(r)$	$ V(r) /G(r)$	
Ti2-C21	2,3965	2,4228	0,0469	0,1389	0,0406	-0,0466	-0,0059	-0,0347	-0,1264	1,1458
Ti2-C22	2,3812	2,4140	0,0482	0,1475	0,0429	-0,0489	-0,0060	-0,0369	-0,1249	1,1403
Ti2-C23	2,3634	2,3683	0,0500	0,1410	0,0430	-0,0508	-0,0078	-0,0352	-0,1552	1,1805
Ti2-C25	2,4129	2,4205	0,0455	0,1394	0,0399	-0,0449	-0,0050	-0,0348	-0,1102	1,1257
Ti2-O5	1,9062	1,9082	0,1205	0,4661	0,1620	-0,2076	-0,0455	-0,1165	-0,3778	1,2808
Ti2-O6	1,9050	1,9074	0,1202	0,4696	0,1624	-0,2073	-0,0450	-0,1174	-0,3739	1,2769
Ti2-O3	2,0621	2,0631	0,0812	0,3109	0,0956	-0,1134	-0,0178	-0,0777	-0,2196	1,1866
Ti2-O4	2,0613	2,0624	0,0812	0,3126	0,0958	-0,1134	-0,0176	-0,0782	-0,2173	1,1841
Ti1-O3	2,1176	2,1214	0,0638	0,2963	0,0787	-0,0832	-0,0046	-0,0741	-0,0718	1,0582
Ti1-O4	2,1220	2,1247	0,0637	0,2920	0,0778	-0,0826	-0,0048	-0,0730	-0,0756	1,0618
Ti1-C9	2,3505	2,3687	0,0496	0,1593	0,0458	-0,0518	-0,0060	-0,0398	-0,1204	1,1304
Ti1-C10	2,3481	2,3521	0,0523	0,1430	0,0448	-0,0539	-0,0091	-0,0358	-0,1736	1,2025
Ti1-C12	2,3603	2,3669	0,0485	0,1576	0,0448	-0,0502	-0,0054	-0,0394	-0,1110	1,1202
Ti1-C13	2,3675	2,3713	0,0494	0,1390	0,0423	-0,0498	-0,0075	-0,0348	-0,1520	1,1777
Ti1-C16	2,3688	2,3854	0,0499	0,1408	0,0429	-0,0505	-0,0077	-0,0352	-0,1536	1,1787

**Table S4 (continued)**

Bond	Distance, Å	BP length, Å	$\rho(r)$	$\Delta\rho(r)$	G(r)	V(r)	H(r)	$-\frac{1}{4}\nabla^2\rho(r)$	H(r)/ρ(r)	$ V(r) /G(r)$
Ti2-C25	2,4452	2,4993	0,0472	0,1277	0,0390	-0,0461	-0,0071	-0,0319	-0,1497	1,1812
Ti2-C23	2,3931	2,3997	0,0425	0,1281	0,0362	-0,0404	-0,0042	-0,0320	-0,0990	1,1162
Ti2-C21	2,4467	2,4824	0,0422	0,1259	0,0357	-0,0399	-0,0042	-0,0315	-0,0994	1,1176
Ti2-O5	1,8892	1,8914	0,1260	0,4838	0,1716	-0,2223	-0,0507	-0,1209	-0,4020	1,2952
Ti2-O6	1,8918	1,8940	0,1252	0,4805	0,1701	-0,2200	-0,0499	-0,1201	-0,3988	1,2936
Ti2-O3	2,0667	2,0675	0,0801	0,3088	0,0942	-0,1112	-0,0170	-0,0772	-0,2122	1,1804
Ti2-O4	2,0619	2,0627	0,0809	0,3133	0,0957	-0,1131	-0,0174	-0,0783	-0,2145	1,1814
Ti1-O3	2,1607	2,1681	0,0560	0,2705	0,0686	-0,0697	-0,0010	-0,0676	-0,0182	1,0149
Ti1-O4	2,1537	2,1605	0,0571	0,2755	0,0703	-0,0716	-0,0014	-0,0689	-0,0241	1,0196
Ti1-C8	2,4583	2,4704	0,0400	0,1258	0,0344	-0,0373	-0,0029	-0,0315	-0,0734	1,0854
Ti1-C10	2,4123	2,4179	0,0459	0,1248	0,0377	-0,0442	-0,0065	-0,0312	-0,1417	1,1725
Ti1-C12	2,4164	2,4225	0,0428	0,1434	0,0390	-0,0421	-0,0031	-0,0359	-0,0723	1,0795
Ti1-C13	2,4114	2,4158	0,0451	0,1243	0,0371	-0,0432	-0,0061	-0,0311	-0,1343	1,1632
Ti1-C16	2,4259	2,4630	0,0444	0,1249	0,0368	-0,0423	-0,0056	-0,0312	-0,1253	1,1511

**Table S4 (continued)**

Bond	Distance, Å	BP length, Å	$\rho(r)$	$\Delta\rho(r)$	G(r)	V(r)	H(r)	$-\frac{1}{4}\nabla^2\rho(r)$	H(r)/ρ(r)	$ V(r) /G(r)$
Ti2-C21	2,3967	2,4070	0,0457	0,1389	0,0400	-0,0452	-0,0052	-0,0347	-0,1141	1,1307
Ti2-C22	2,3820	2,3940	0,0472	0,1455	0,0419	-0,0475	-0,0056	-0,0364	-0,1178	1,1325
Ti2-C23	2,3605	2,3643	0,0492	0,1417	0,0426	-0,0497	-0,0071	-0,0354	-0,1451	1,1676
Ti2-C24	2,3795	2,3999	0,0474	0,1421	0,0415	-0,0475	-0,0060	-0,0355	-0,1266	1,1446
Ti2-C25	2,4022	2,4085	0,0455	0,1394	0,0399	-0,0449	-0,0050	-0,0349	-0,1104	1,1260
Ti2-O5	1,8927	1,8945	0,1162	0,5309	0,1679	-0,2031	-0,0352	-0,1327	-0,3030	1,2097
Ti2-O6	1,8927	1,8946	0,1160	0,5319	0,1679	-0,2028	-0,0349	-0,1330	-0,3010	1,2080
Ti2-O3	2,0510	2,0519	0,0784	0,3553	0,1004	-0,1120	-0,0116	-0,0888	-0,1481	1,1156
Ti2-O4	2,0536	2,0544	0,0778	0,3528	0,0995	-0,1109	-0,0113	-0,0882	-0,1457	1,1139
Ti1-O3	2,1426	2,1472	0,0564	0,3042	0,0745	-0,0730	0,0015	-0,0760	0,0271	0,9795
Ti1-O4	2,1451	2,1490	0,0563	0,3016	0,0740	-0,0726	0,0014	-0,0754	0,0249	0,9811
Ti1-C8	2,4053	2,4125	0,0442	0,1396	0,0392	-0,0434	-0,0043	-0,0349	-0,0961	1,1086
Ti1-C9	2,3503	2,3650	0,0489	0,1614	0,0457	-0,0510	-0,0053	-0,0403	-0,1093	1,1170
Ti1-C10	2,3431	2,3457	0,0523	0,1404	0,0444	-0,0537	-0,0093	-0,0351	-0,1776	1,2091
Ti1-C12	2,3548	2,3592	0,0482	0,1604	0,0451	-0,0500	-0,0050	-0,0401	-0,1030	1,1102
Ti1-C13	2,3500	2,3522	0,0509	0,1377	0,0430	-0,0516	-0,0086	-0,0344	-0,1686	1,1994
Ti1-C16	2,3719	2,3873	0,0487	0,1398	0,0420	-0,0490	-0,0070	-0,0349	-0,1441	1,1673

G(r) — kinetic energy densities at the bcp were estimated using the Abramov's approximation [1]

$$G(r) = 3/10(3\pi^2)^{2/3}\rho^{5/3}(r) + 1/6\nabla^2\rho(r)$$

$V(r)$  — potential energy densities at the BCP from local virial theorem

$$V(r) + 2G(r) = -\frac{1}{4}\nabla^2\rho(r)$$

$H(r)$  — total energy densities, the sum of the  $G(r)$ , a positive quantity, and the  $V(r)$ , a negative quantity [2,3]

If  $r$  is bcp,

$$H(r) = G(r) + V(r)$$

**Table S5.** Cartesian coordinates (in Å) of **2m** (BP86).

Ti	4.27705931	6.33456484	3.21541726
O	5.77218113	6.60440127	4.69048227
O	5.71330499	7.73126307	2.51592877
C	3.11092547	8.45788150	3.38393769
C	3.25077260	7.94012226	4.68997408
C	2.54455928	6.70432688	4.76032857
C	1.94953818	6.48225457	3.48778132
C	2.32772053	7.55005689	2.62318708
C	4.71490558	5.12623764	1.23566409
C	5.71712563	4.76333037	2.18457489
C	5.08445852	4.07706659	3.25805243
C	3.69746870	4.03156850	2.98940245
C	3.47042442	4.67109233	1.73445246
Ti	7.39592044	7.29735133	3.62479358
C	6.16358952	5.69470096	5.62680826
C	5.99317561	7.91799386	1.19537923
H	3.60410198	9.34820243	2.99837479
H	3.86607128	8.36454324	5.48083283
H	2.44662010	6.06331376	5.63457144
H	1.33045235	5.63042112	3.21649023
H	2.04952416	7.65642519	1.57565151
H	4.88058091	5.67251858	0.30799204
H	6.78047478	4.97780408	2.10351156
H	5.57392794	3.69795712	4.15430386
H	2.93612788	3.60174737	3.63909460

H	2.50376834	4.81192576	1.25404371
O	8.10495430	5.67486471	4.33084881
O	8.06314376	7.07830710	1.85391011
C	7.46077534	5.18394471	5.41560028
C	7.29875788	7.53634501	0.83555409
C	6.96826473	9.46506577	4.55285719
C	7.94534455	9.61258378	3.53564748
C	9.10496703	8.89893977	3.94077526
C	8.84679111	8.32747890	5.21876928
C	7.52930403	8.68063291	5.59734099
C	5.36998944	5.24127840	6.67567620
C	5.08786583	8.38987401	0.24911443
C	7.97775898	4.18762005	6.25389982
C	7.71856162	7.60886283	-0.49942626
H	5.94423375	9.83196165	4.50869770
H	7.79988267	10.10150180	2.57244840
H	10.00947781	8.76446091	3.34939182
H	9.51545316	7.67370588	5.77593509
H	7.00748989	8.32801635	6.48615205
C	5.87378998	4.25047130	7.53386898
H	4.36696311	5.65227307	6.81755642
C	5.49009994	8.47734760	-1.09295198
H	4.07688796	8.67244284	0.55475827
C	7.16119005	3.74021130	7.30432177
C	9.34631199	3.62529605	5.99061528
C	6.79188803	8.08259543	-1.44123391
C	9.10881123	7.17701175	-0.87208379

C	5.05412904	3.77249627	8.70622330
C	4.52148552	8.94799257	-2.14965653
H	7.54878870	2.96075891	7.96975747
H	5.36513242	2.76302361	9.02237904
H	3.98009273	3.73775918	8.45558020
H	5.17485651	4.44944961	9.57234806
H	7.10238098	8.14712199	-2.49010518
H	9.41080107	3.22508834	4.96369877
H	9.58246185	2.81764477	6.70192990
H	10.11857478	4.40961404	6.07902399
H	9.86375082	7.81033242	-0.37355452
H	9.29368624	6.13808689	-0.54791644
H	9.26056179	7.24056976	-1.96159599
H	5.05336361	9.41867650	-2.99331249
H	3.81208377	9.68613320	-1.73887655
H	3.93349400	8.10286524	-2.55379155

**Table S6.** Cartesian coordinates (in Å) of **2m** (B3LYP).

Ti	4.16458193	6.30644610	3.23535751
O	5.79526808	6.47086124	4.64327902
O	5.71139672	7.58180852	2.44835528
C	3.11678094	8.53438152	3.45644137
C	3.27570895	8.02128586	4.75595753
C	2.47667249	6.85799464	4.87249567
C	1.81417788	6.66935208	3.63928981
C	2.22786458	7.69260465	2.75263694
C	4.45597977	4.97899000	1.23732600
C	5.41827785	4.53819491	2.17887047
C	4.73363884	3.93642456	3.25960431
C	3.35323375	4.02244952	3.00060056
C	3.18178975	4.66309369	1.74596163
Ti	7.38533264	7.30492662	3.61992201
C	6.22724061	5.57979427	5.57907700
C	6.04887933	7.81916164	1.15023573
H	3.62460001	9.39570663	3.04812134
H	3.92383775	8.42021846	5.52202333
H	2.35570791	6.25052585	5.75810664
H	1.10349984	5.88766875	3.41599263
H	1.89008005	7.82773521	1.73451610
H	4.66673989	5.47350862	0.29993210
H	6.49009898	4.63491928	2.08327024
H	5.18842886	3.50636150	4.14029618
H	2.56449731	3.64975118	3.63927385

H	2.23882761	4.86522494	1.25786301
O	8.19578012	5.77039490	4.36651123
O	8.09939163	7.02632868	1.89036079
C	7.56158551	5.18793798	5.40742246
C	7.37892159	7.49523454	0.84891329
C	6.92741794	9.60076563	4.33124940
C	8.00737456	9.64275354	3.42669470
C	9.08807414	8.94103166	4.00830551
C	8.67909371	8.48681360	5.28431836
C	7.34345071	8.88980547	5.48154153
C	5.46243270	5.05806673	6.60976832
C	5.19999642	8.31601710	0.17491962
C	8.14419469	4.24215985	6.24821960
C	7.87281970	7.63705815	-0.44582838
H	5.94602117	10.01950456	4.16336016
H	7.99515190	10.08750191	2.44118216
H	10.05031492	8.76374376	3.54881076
H	9.27130588	7.89672924	5.96849763
H	6.73536049	8.66014508	6.34501636
C	6.02611100	4.11506942	7.47582468
H	4.43466233	5.38006418	6.73855451
C	5.67405840	8.47851301	-1.13152516
H	4.17613455	8.57320107	0.42641883
C	7.34957340	3.72113345	7.27450378
C	9.56679802	3.81008495	6.02798536
C	6.99458964	8.13017099	-1.41680594
C	9.29349204	7.25919112	-0.75835402

C	5.21989596	3.56421186	8.62521415
C	4.76774957	9.03629719	-2.20026785
H	7.78599284	2.98313311	7.94263290
H	5.66234680	2.64367846	9.01466931
H	4.19305679	3.33721529	8.32156744
H	5.16913757	4.28309454	9.45155690
H	7.36204430	8.24935572	-2.43274120
H	9.70133982	3.40637929	5.01944751
H	9.85848599	3.04188348	6.74815577
H	10.25706399	4.65393682	6.12880098
H	10.00016541	7.87457659	-0.19196416
H	9.48944395	6.21666429	-0.48852289
H	9.50698053	7.38540468	-1.82249454
H	5.24093726	8.99031605	-3.18428371
H	4.51890730	10.08478670	-2.00041337
H	3.82673961	8.47882109	-2.25800163

**Table S7.** Cartesian coordinates (in Å) of **2m** (M06L).

Ti	4.18446030	6.29171033	3.21628776
O	5.79296055	6.47505126	4.61978906
O	5.71730091	7.58344476	2.45253700
C	3.17616263	8.47302843	3.47532285
C	3.29605276	7.91422571	4.75373001
C	2.50888028	6.74138403	4.80181795
C	1.88495709	6.60002335	3.54399359
C	2.32146200	7.65364987	2.70972727
C	4.51715283	5.04215656	1.24830374
C	5.48955883	4.62862177	2.19000395
C	4.82249958	3.99767787	3.26140454
C	3.44496948	4.03750157	2.99759583
C	3.25735685	4.67482137	1.74933479
Ti	7.38946847	7.26154164	3.60033785
C	6.20045446	5.61552803	5.58064978
C	6.03529916	7.83596883	1.16324989
H	3.70825073	9.33547509	3.11004005
H	3.93451239	8.27291441	5.54427505
H	2.37043838	6.09756607	5.65389593
H	1.20440294	5.81494811	3.26461549
H	2.02663569	7.81769008	1.68656316
H	4.70905174	5.55884639	0.32260347
H	6.55422038	4.78746505	2.11352297
H	5.28541706	3.59655808	4.14779047
H	2.66666733	3.66411770	3.64251942

H	2.31012173	4.87043862	1.27545255
O	8.16044159	5.70671956	4.35559825
O	8.07748314	7.00160640	1.85636704
C	7.52892557	5.19589139	5.42554809
C	7.35903267	7.50746498	0.84054091
C	6.88810939	9.43827429	4.46900614
C	7.84239946	9.59501669	3.44715950
C	9.02439855	8.94495324	3.85581687
C	8.80215385	8.40140810	5.13885094
C	7.48263011	8.70653148	5.51710260
C	5.41704523	5.13860010	6.61529552
C	5.17277300	8.34542773	0.20963407
C	8.09604566	4.28895973	6.31446998
C	7.84108183	7.68629666	-0.45174771
H	5.86638451	9.77744821	4.43474776
H	7.67617958	10.06101596	2.48931483
H	9.92332458	8.84389415	3.27127901
H	9.49970777	7.80661255	5.70416916
H	6.99212966	8.37835152	6.41901249
C	5.96646384	4.22820804	7.52007445
H	4.39164535	5.47115090	6.71219095
C	5.63613766	8.53511658	-1.09361711
H	4.15263463	8.58724875	0.47999341
C	7.28781591	3.82264408	7.35142141
C	9.50908597	3.84380382	6.13070646
C	6.95393828	8.20252433	-1.39648584
C	9.25173135	7.32695941	-0.78276791

C	5.14757478	3.71924623	8.66678716
C	4.71886589	9.07663890	-2.14747257
H	7.70993456	3.11344816	8.05334173
H	5.61677601	2.86075883	9.14028389
H	4.14989956	3.42176361	8.34774195
H	5.01809251	4.48360198	9.43362865
H	7.31070422	8.35281372	-2.40840963
H	9.65344885	3.38409400	5.15399125
H	9.79970788	3.12729242	6.89384079
H	10.19777286	4.68738070	6.17337167
H	9.95693475	7.88997721	-0.17194386
H	9.44776343	6.27480626	-0.58077010
H	9.47364375	7.52214513	-1.82819187
H	5.26001650	9.32101147	-3.05775255
H	4.20791802	9.97764069	-1.81119140
H	3.94422752	8.35601893	-2.41140020

## References

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1. Y.A. Abramov, *Acta Cryst.* 1997, **A53**, 264.
2. R.F.W. Bader, *Atoms in Molecules. A Quantum Theory* (Clarendon, Oxford, 1990).
3. P. Popelier, *Atoms in Molecules. An Introduction* (Prentice-Hall, Harlow, UK, 2000).