Carbon/TiO₂/Fe₂O₃ Hybrid Shells toward Efficient Visible Light

Photocatalysts

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Figure S1 TEM images of (a, b) TiO_2 hollow spheres and (c, d) TiO_2/Fe_2O_3 hollow hybrids calcined at

600 °C in air.



Figure S2 XRD pattern of TiO₂ hollow spheres.



Figure S3 Bar plot showing the adsorption of RhB with different photocatalysts.



Figure S4 The photocatalytic activity normalized with the specific surface area.



Figure S5 The photocatalystic activity under visible light irradiation of (a) RhB, (b) MB and (c) phenol using C/TiO₂/Fe₂O₃ hollow hybrids synthesized at different FeCl₃·6H₂O precursor concentration.



Figure S6 PL spectra of TiO₂ and TiO₂/Fe₂O₃, C/TiO₂/Fe₂O₃ hollow hybrids excited at 360 nm.



Figure S7 Fluorescence spectra of $C/TiO_2/Fe_2O_3$ hollow hybrids in a 5×10^{-4} mol L⁻¹ basic solution of

terephthalic acid under visible light irradiation at a fixed 30 min.

sample	<i>k</i> (min ⁻¹)	irradiation	organic dye	ref
TiO ₂ /Fe ₂ O ₃ /CNTs	0.0211	visible light	tetracycline	1
TiO ₂ &γ-Fe ₂ O ₃ @GO	0.00200	visible light	RhB	2
Fe ₂ O ₃ -Fe ₃ O ₄ @SiO ₂ @TiO ₂ -	0.0201	visible light	RhB	3
TNS-GR				
Fe ₂ O ₃ -TiO ₂ -graphene	0.0176	visible light	RhB	4
C/TiO ₂ /Fe ₂ O ₃ hollow	0.0289	visible light	RhB	this work
hybrids				

Table S1 Comparison of the photocatalytic activity of different C/TiO₂/Fe₂O₃ photocatalysts.

References

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