

Supporting Information

A surface carbonization strategy to MoS₂ microspheres for enhancing electrochemical hydrogen evolution activity

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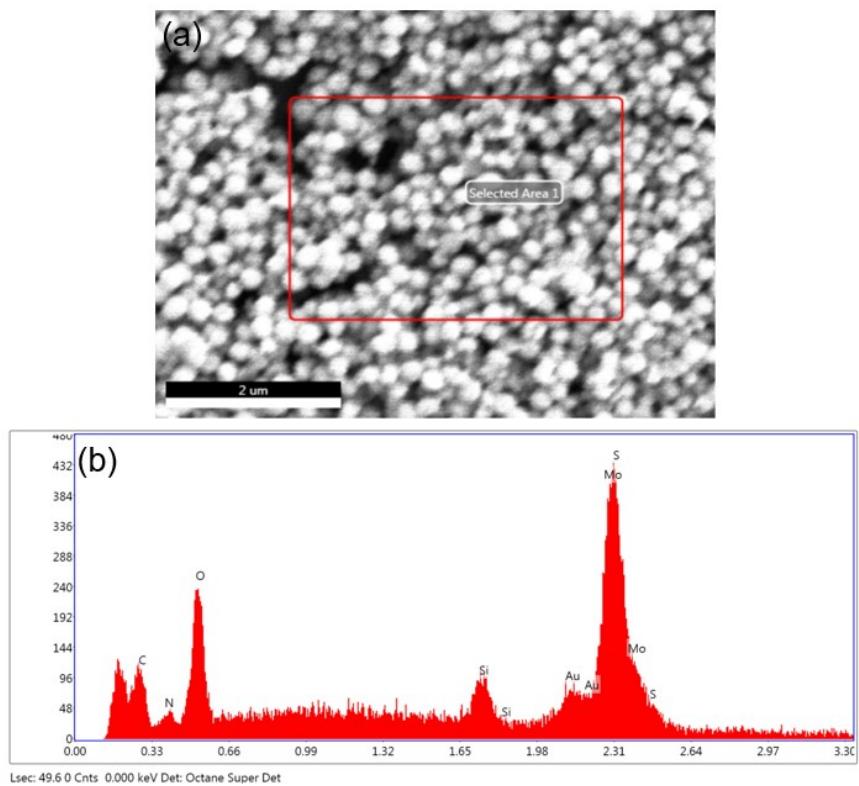


Fig. S1 SEM image with a low magnification (a) of MD-1:1 and EDX spectrum (b). Si and Au are from silicon substrate and conductive coating respectively.

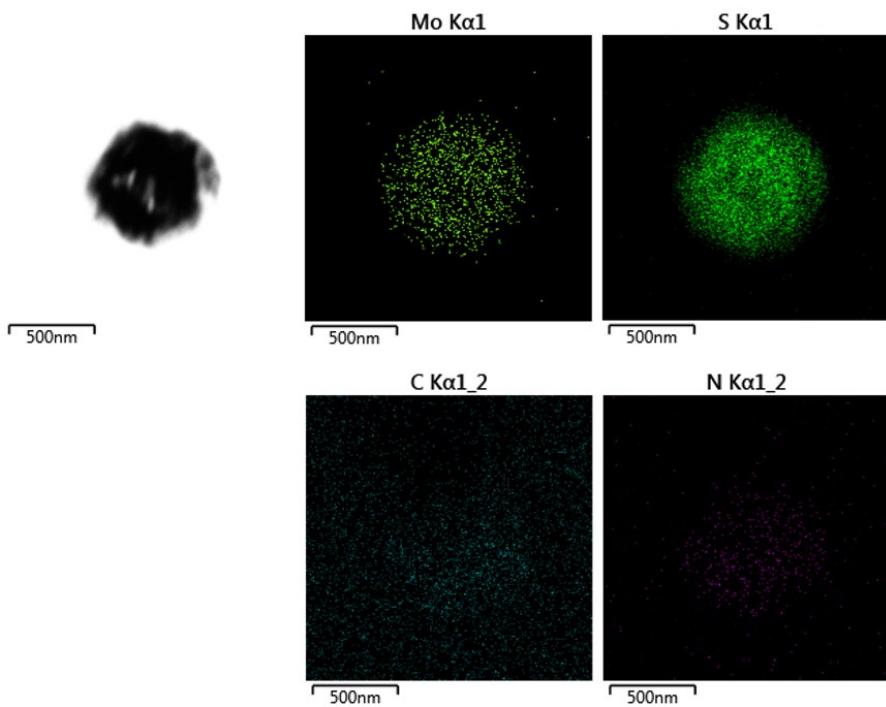


Fig. S2 Bright-field TEM image and elemental mapping of Mo, S, C and N in MD-1:1.

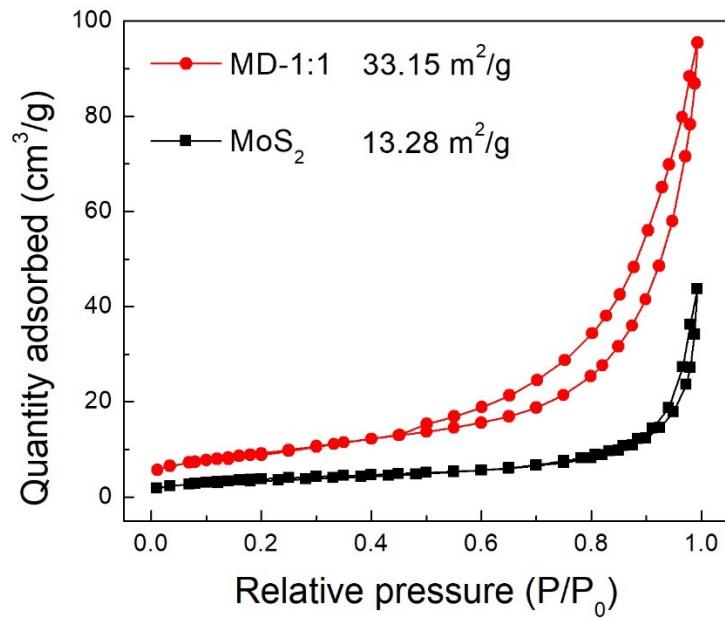


Fig. S3 N₂ adsorption-desorption isotherm of MoS₂ and MD-1:1.

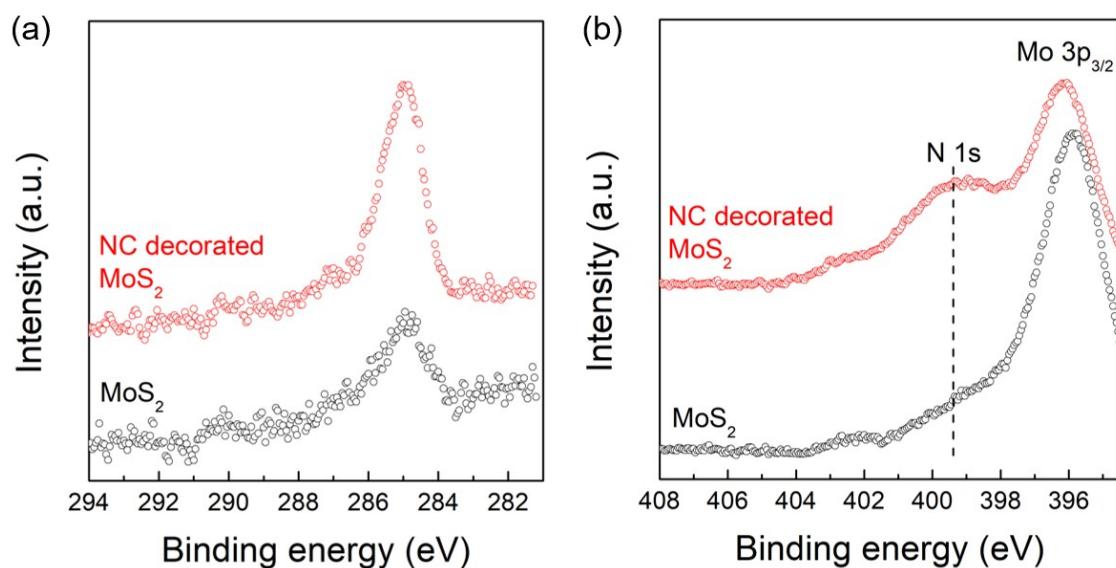


Fig. S4 The C 1s (a) and N 1s (b) spectra of MoS₂ and NC decorated MoS₂. The peak at 396 corresponds to Mo 3p_{3/2}.

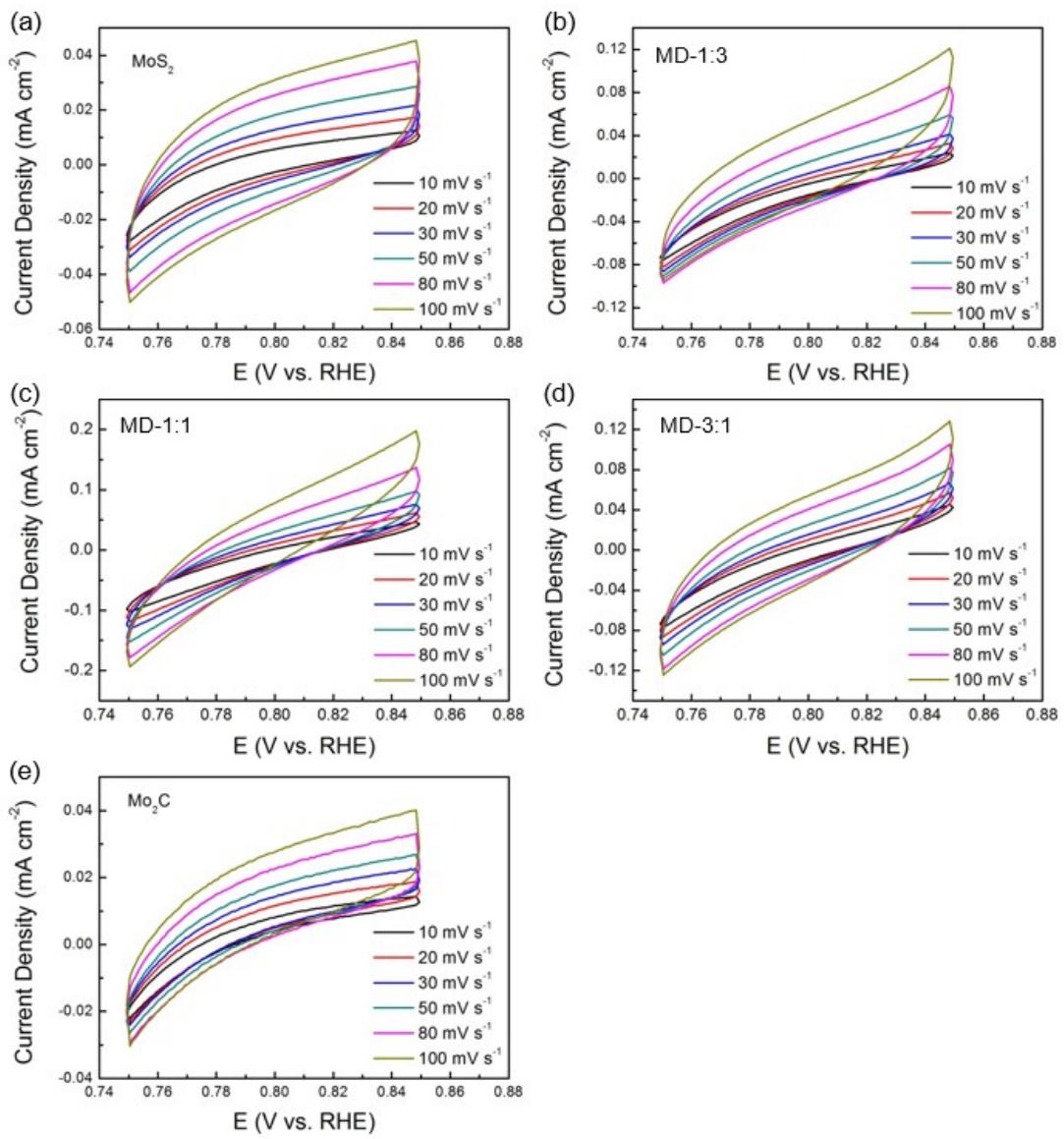


Fig. S5 CV curves (a-e) of the electrocatalysts at different scan rates in 1 M KOH.

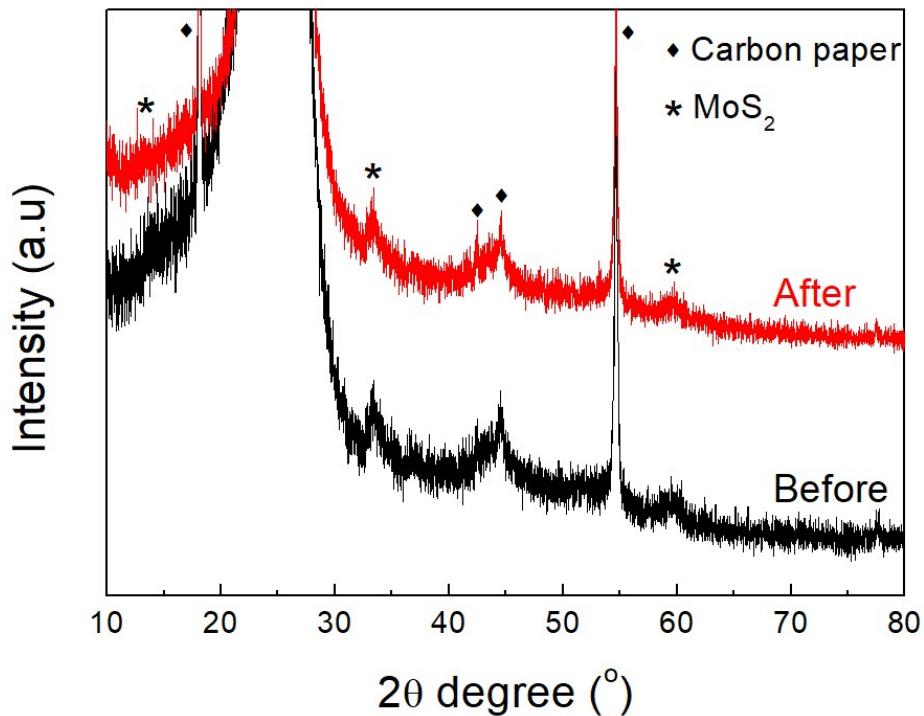


Fig. S6 XRD patterns of MD-1:1 catalyst before and after long-term HER test under alkaline condition (1 M KOH).

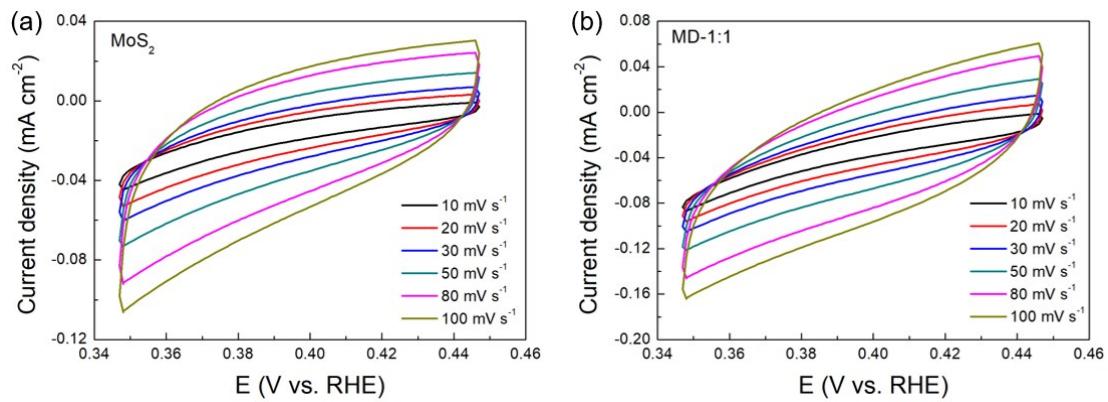


Fig. S7 CV curves of MoS₂ (a) and MD-1:1 (b) at different scan rates in 0.5 M H₂SO₄.

Table S1 Comparison of the HER performance of various MoS₂-based electrocatalysts.

Catalysts	Electrolyte	Current density @ 300 mV (mA cm ⁻²)	Ref.
MoS ₂ nanosheet-C ₃ N ₄ heterojunction ^a	0.5 M H ₂ SO ₄	-0.508	[1]
(PMo ₁₂ /MoS ₂) ₄	0.5 M H ₂ SO ₄	-0.499	[2]
Ni-doped MoS ₂	0.5 M H ₂ SO ₄	-2.368	[3]
Ta-doped MoS ₂	0.5 M H ₂ SO ₄	-0.205	[4]
NC decorated MoS ₂	0.5 M H ₂ SO ₄	-3.172	This work

^a Treated by oxygen plasma treatment for 15 min.

References

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