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Electronic Supplementary Material

A fast response TLC-SERS substrate for on-site detection of hydrophilic and hydrophobic adulterants in botanical dietary supplements

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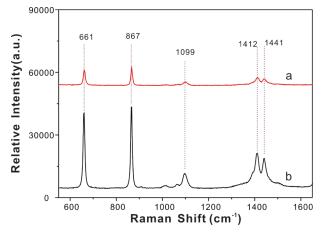


Fig. S1 (a) Blank SERS spectrum of DMF 2:1 silver colloid, and (b) Raman spectrum of DMF solvent.

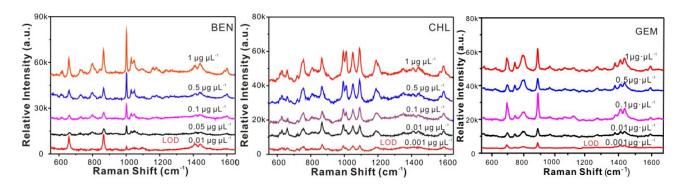


Fig. S2 (a) SERS spectra of different analytes with different concentrations using DMF 2:1 silver colloid.

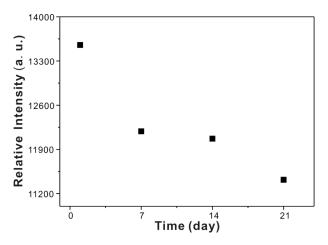


Fig. S3 SERS intensity of CHL at 1091 cm⁻¹ detected by Lee-Meisel colloid changing with time (day).

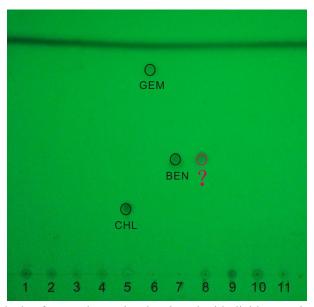


Fig. S4 Results from TLC analysis of ten real samples developed with dichloromethane-methanol-water 9:1:0.1 (*v: v: v)*, 1-4 (samples 1-4), 5-7 (references CHL, GEM, BEN), 8-11 (samples 5-8).

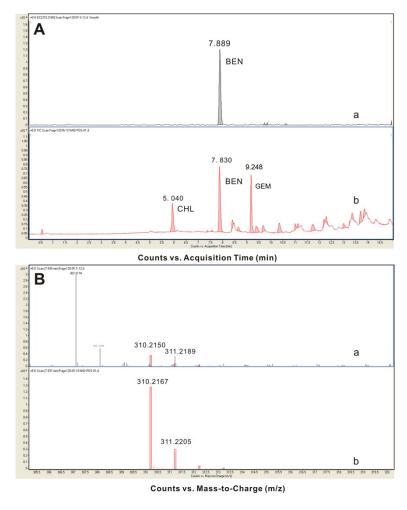


Fig. S5 (A) HPLC spectra of sample 5 (a) and standards of CHL, DIP, BEN (b); (B) MS spectra of sample 5 (a) and standard of BEN (b)

Table S1 Composition of reactants used.

Colloids	Mass of AgNO ₃ (mg)	Mass of PVP (mg)	Mass ratio of AgNO ₃ to PVP
DMF 1:1	8.5	8.5	1:1
DMF 2:1	17.0	8.5	2:1
DMF 3:1	25.5	8.5	3:1
DMF 4:1	34.0	8.5	4:1

Table S2 Validation of BEN in sample 5 by UPLC-QTOF/MS.

Name	Identification formula [M+H] ⁺	Mass	Error (ppm)
BEN standard	$C_{21}H_{28}NO$	310.2167	-0.58
Sample 5	$C_{21}H_{28}NO$	310.2153	4.01