Electronic Supplementary Information

Investigation on the antrypanosomal effects of 2-formyl-8-hydroxyquinoline-derived hydrazones and their antimony(III) and bismuth(III) complexes

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| | 10 |





g. S1 ¹H NMR spectrum of H2Q4NO₂Im (H₂La, 1) in DMF- d_7 (400 MHz) at room

temperature



Fig. S2 ¹³C{¹H} NMR spectrum of H2Q4NO₂Im (H₂La, 1) in DMF- d_7 (100 MHz) at room

temperature



Fig. S3 ¹H NMR spectrum of H2Q4NO₂Ph (H₂Lb, 2) in DMF- d_7 (400 MHz) at room

temperature



Fig. S4 ¹³C{¹H} NMR spectrum of H2Q4NO₂Ph (H₂Lb, 2) in DMF- d_7 (100 MHz) at room

temperature



Fig. S5 ¹H NMR spectrum of $[Sb(HQ4NO_2Im)Cl_2]$ (3) in DMF- d_7 (400 MHz) at room



Fig. S6 ¹³C{¹H} NMR spectrum of [Sb(HQ4NO₂Im)Cl₂] (3) in DMF- d_7 (100 MHz) at

room temperature



Fig. S7 ¹H NMR spectrum of [Sb(HQ4NO₂Ph)Cl₂] (4) in DMF-d₇ (400 MHz) at room

temperature



Fig. S8 ${}^{13}C{}^{1}H$ NMR spectrum of [Sb(HQ4NO₂Ph)Cl₂] (4) in DMF- d_7 (100 MHz) at

room temperature



Fig. S9 ¹H NMR spectrum of [Bi(HQ4NO₂Im)Cl₂] (**5**) in DMF-*d*₇ (400 MHz) at room temperature



Fig. S10 ¹³C{¹H} NMR spectrum of [Bi(HQ4NO₂Im)Cl₂] (5) in DMF- d_7 (100 MHz) at

room temperature



Fig. S11 ¹H NMR spectrum of [Bi(HQ4NO₂Ph)Cl₂] (6) in DMF-d₇ (400 MHz) at room

temperature

Infrared spectra of compounds (1-6)



Fig. S12 FT-IR spectrum of H2Q4NO₂Im (H₂La, 1) (ATR technique)



Fig. S13 FT-IR spectrum of H2Q4NO₂Ph (H₂Lb, 2) (ATR technique)



Fig. S14 FT-IR spectrum of [Sb(HQ4NO₂Im)Cl₂] (3) (ATR technique)



Fig. S15 FT-IR spectrum of [Sb(HQ4NO₂Ph)Cl₂] (4) (ATR technique)



Fig. S16 FT-IR spectrum of [Bi(HQ4NO₂Im)Cl₂] (5) (ATR technique)



Fig. S17 FT-IR spectrum of [Bi(HQ4NO₂Ph)Cl₂] (6) (ATR technique)

Thermogravimetry



Fig. S18 TG curve of the H2Q4NO₂Ph (H₂Lb, 2) at a heating rate of 10°C min⁻¹ under N₂ flow of 50 mL min⁻¹

Crystal packing of [Sb(HQ4NO₂Ph)(DMF)Cl₂]·3DMF (4a)

The crystal packing of complex (4a) shows an intermolecular hydrogen bond involving the hydrazone nitrogen and the oxygen of a DMF molecule $[d(N3...O41^i) =$ 2.737(3) Å, \angle (N3-H3N...O41ⁱ) = 157° (symmetry operator: i = 1-x,-y,1-z)]. Non classical C-H···Cl and C-H···O hydrogen intermolecular interactions forming a dimeric arrangement were also observed, Figure S19 and Table S1.



Figure S19. Dimers formed by NH···O and CH···Cl intermolecular interactions in the crystal packing of complex [Sb(HQ4NO₂Ph)(DMF)Cl₂]·3DMF (**4a**). Symmetry code: i (1 - x, -y, 1 - z).

 Table S1. Geometrical parameters of intermolecular interactions in the structure of

 [Sb(HQ4NO₂Ph)(DMF)Cl₂]·3DMF (4a).

| H (D-H···A) | d(D-H) / Å | $d(H\cdots A) / Å$ | $d(D \cdots A) / Å$ | D-H…A / ° |
|-------------------------|------------|--------------------|----------------------|-----------|
| | 0.97 | 1.02 | 2 727(2) | 157.0 |
| $C3-H(3)\cdots Cl2^{i}$ | 0.86 | 2.81 | 2.737(3) 3.692(2) | 157.0 |
| C400-H400…Cl2 | 0.93 | 2.62 | 3.535(3) | 167.0 |

Symmetry code: *i* (1 - *x*, -*y*, 1 - *z*).