## Catalytic Synthesis of Benzimidazoles and Organic Carbamates by Polymer Supported Zinc Catalyst through CO<sub>2</sub> Fixation

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## Materials

All chemicals were purchased from commercially available sources and used as received without further purification. Solvents were distilled and dried through standard methods before use.

## **Characterization Techniques**

Exeter Analytical Inc. model: CE 440 instrument was used to know elemental composition of ligand and catalyst. Fourier-transform infrared spectroscopy was carried out on a Perkin-Elmer FTIR 783 spectrophotometer using KBr pellets. Bruker D8 Advance X-ray diffractometer using Cu-K $\alpha$  radiation ( $\lambda$  = 1.5418 Å) operating at 40 kV and 40 mA was utilized to record powder X-ray diffraction (PXRD) data of samples. In order to understand the morphologies of both ligand and catalyst we have done FE-SEM through (ZEISS EVO40, England) equipped with EDAX facility. Thermal stability of samples was monitored by Mettler Toledo TGA/DTA 851 instrument. Bruker AMX- 400 instrument was operates for <sup>1</sup>H NMR spectra. To measure the metal loading in catalyst (both fresh and used catalyst) AAS analysis was performed on Spectra- AA 240 (Agilent Technologies).



Figure S1: <sup>1</sup>H NMR and <sup>13</sup>C spectra of [SALTETA] ligand.



Figure S2: Mass spectra of [SALTETA] ligand.



**Figure S3**: Energy dispersive X-Ray pattern of [PS-SALTETA] ligand (a) and [PS-Zn(II)SALTETA] catalyst (b).

## NMR Spectra of products





**Figure S4**: <sup>1</sup>H NMR spectra of 1H-benzo[d]imidazole (in CDCl<sub>3</sub>).





Figure S5: <sup>1</sup>H NMR spectra of 7-methyl-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).





**Figure S6**: <sup>1</sup>H NMR spectra of 5-methyl-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).





Figure S7: <sup>1</sup>H NMR spectra of 5,6-dimethyl-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).



**Figure S8**: <sup>1</sup>H NMR spectra of 5-fluoro-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).





Figure S9: <sup>1</sup>H NMR spectra of 5-chloro-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).





**Figure S10**: <sup>1</sup>H NMR spectra of benzo[d]oxazole (in CDCl<sub>3</sub>).



Figure S11: <sup>1</sup>H NMR spectra of 7-nitro-1H-benzo[d]imidazole (in CDCl<sub>3</sub>).

NO2





Figure S12: <sup>1</sup>H NMR spectra of butyl phenylcarbamate (in CDCl<sub>3</sub>).





**Figure S13**: <sup>1</sup>H NMR spectra of butyl 4-methoxyphenylcarbamate (in CDCl<sub>3</sub>).





Figure S14: <sup>1</sup>H NMR spectra of butyl 4-bromophenylcarbamate (in CDCl<sub>3</sub>).





Figure S15: <sup>1</sup>H NMR spectra of butyl 4-chlorophenylcarbamate (in CDCl<sub>3</sub>).





**Figure S16**: <sup>1</sup>H NMR spectra of butyl 2-iodophenylcarbamate (in CDCl<sub>3</sub>).





Figure S17: <sup>1</sup>H NMR spectra of butyl 5-fluoro-2-iodophenylcarbamate (in CDCl<sub>3</sub>).





Figure S18: <sup>1</sup>H NMR spectra of butyl 4-chloro-2-iodophenylcarbamate (in CDCl<sub>3</sub>).





Figure S19: <sup>1</sup>H NMR spectra of butyl methyl(phenyl)carbamate (in CDCl<sub>3</sub>).



Figure S20: IR spectra of fresh and reused catalyst.



Figure S21: Powder XRD spectra of fresh and reused catalyst.