

Electronic Supplementary Information (ESI) for

## Luminescent Sensing, Electrochemical, and Magnetic Properties of 2D Coordination Polymers Based on the Mixed Ligands of *p*-Terphenyl-2,2'',5'',5'''-tetracarboxylate Acid and 1,10-phenanthroline

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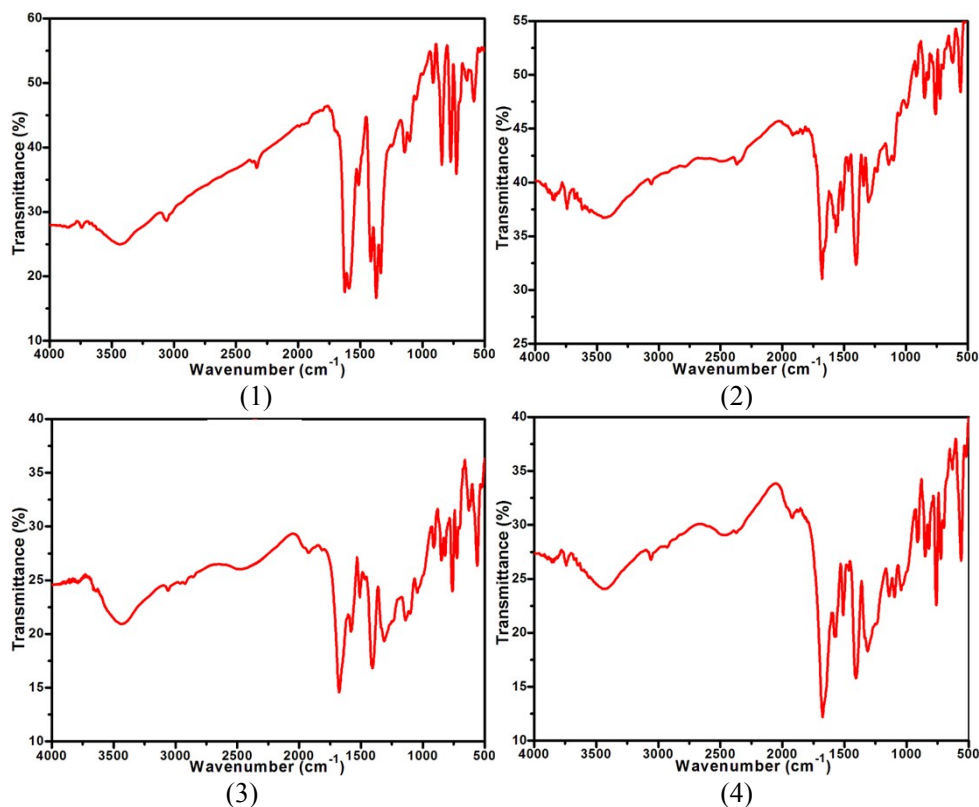
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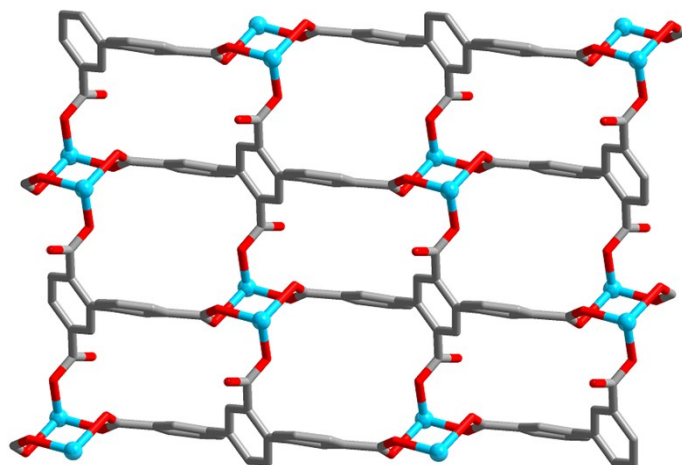
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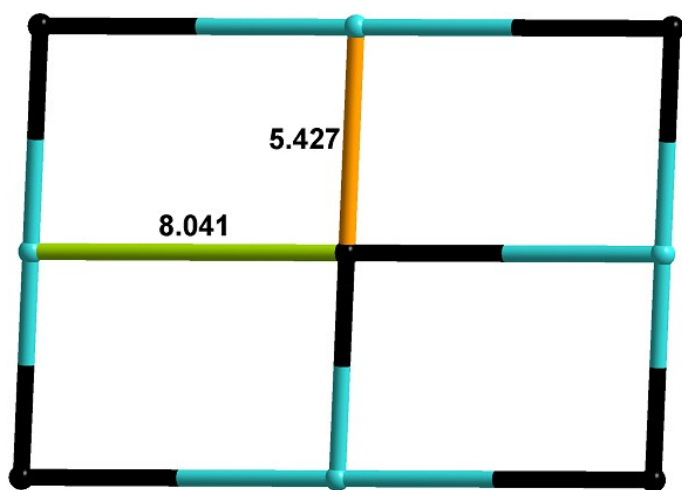


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**Fig. S1** The IR spectra of CPs **1-4**.

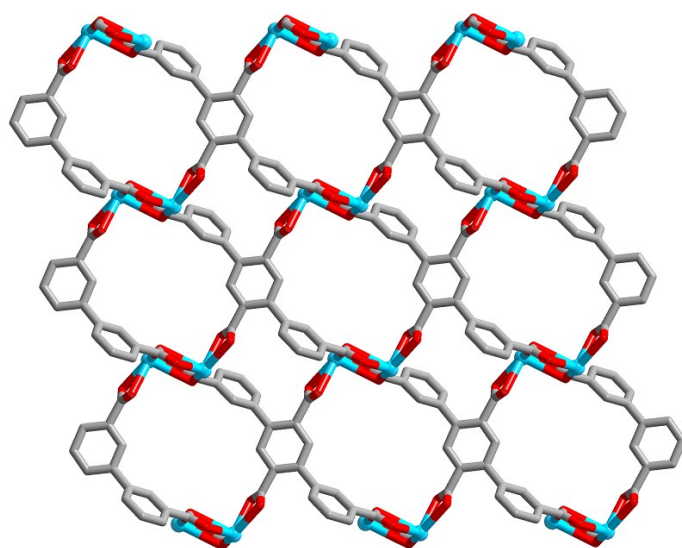


**Fig. S2** The 2D  $[\text{Cu}(\text{tptc})_{0.5}]_n$  sheet in CP 1.

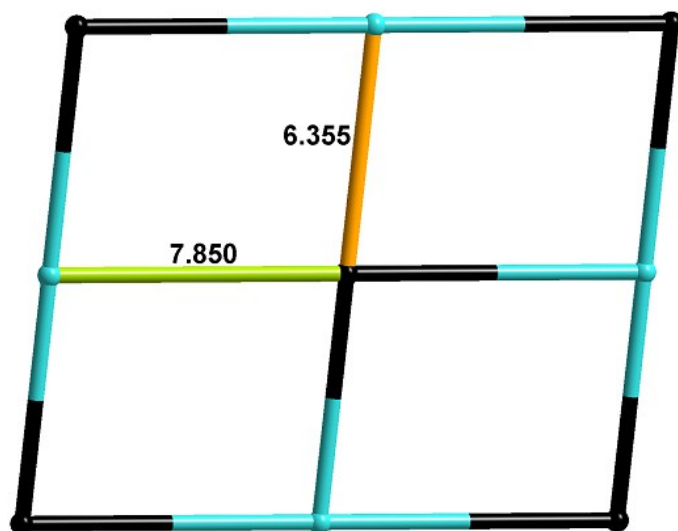


**Fig. S3** The (4,4)-connected  $\{4^4 \cdot 6^2\}$ -sql sheet of CP 1.

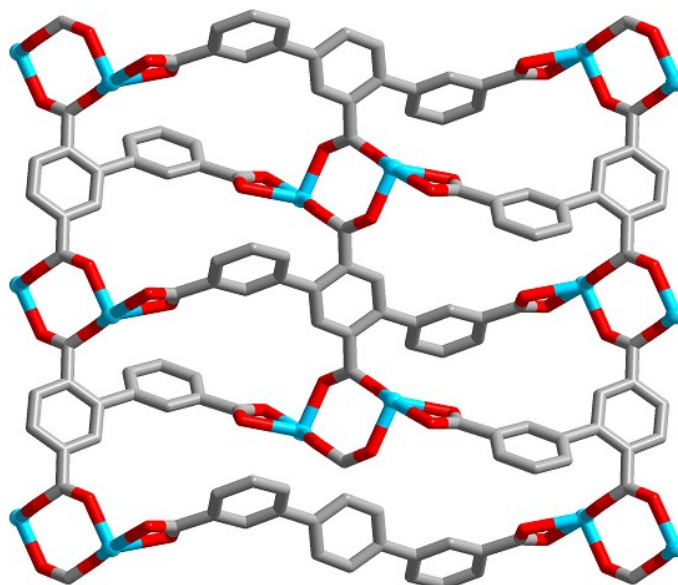
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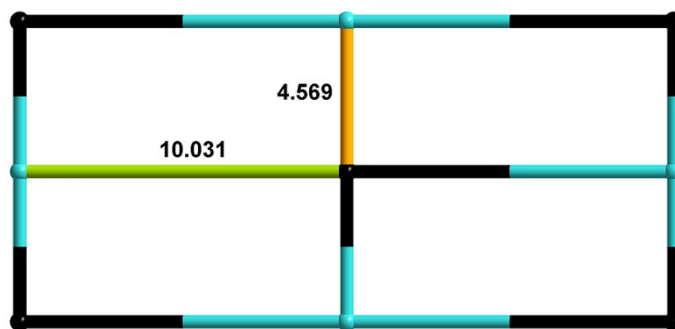
**Fig. S4** The 2D  $[\text{Cd}(\text{tptc})_{0.5}]_n$  sheet in CP 2.



**Fig. S5** The (4,4)-connected  $\{4^4 \cdot 6^2\}$ -sql sheet of CP 2.



**Fig. S6** The 2D  $[\text{Mn}(\text{tpc})_{0.5}]_n$  sheet in CP 3.



**Fig. S7** The (4,4)-connected  $\{4^4 \cdot 6^2\}$ -sql sheet of CP 3.

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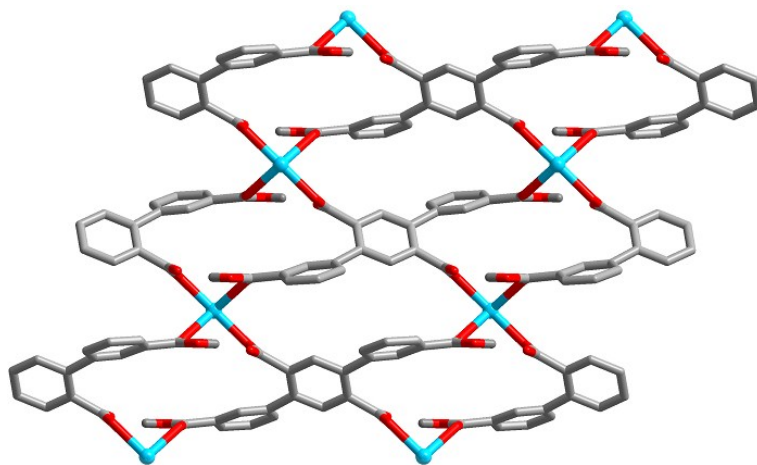


Fig. S8 The 2D  $[\text{Mn}(\text{H}_2\text{tptc})]_n$  sheet in CP 4.

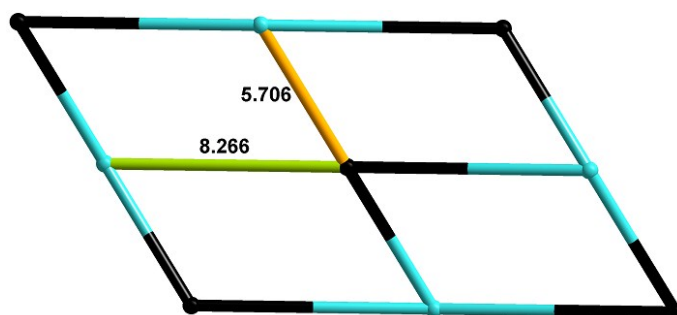


Fig. S9 The (4,4)-connected  $\{4^4-6^2\}$ -sql sheet of CP 4.

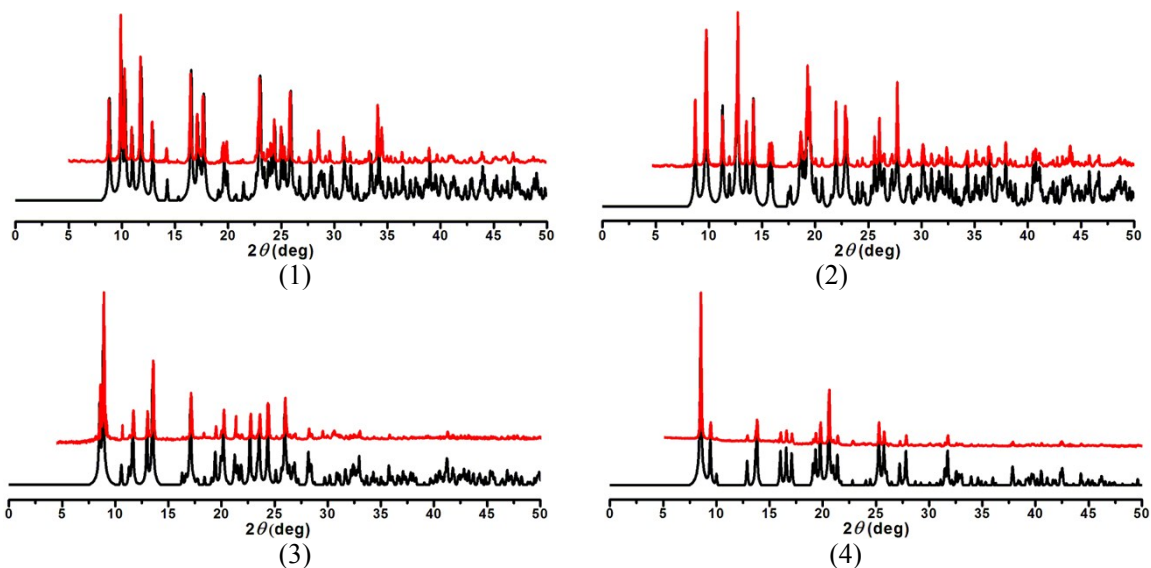


Fig S10. PXR D patterns of 1-4. Dark: calculated from the X-ray single-crystal data; Red: observed for the as-synthesized solids.

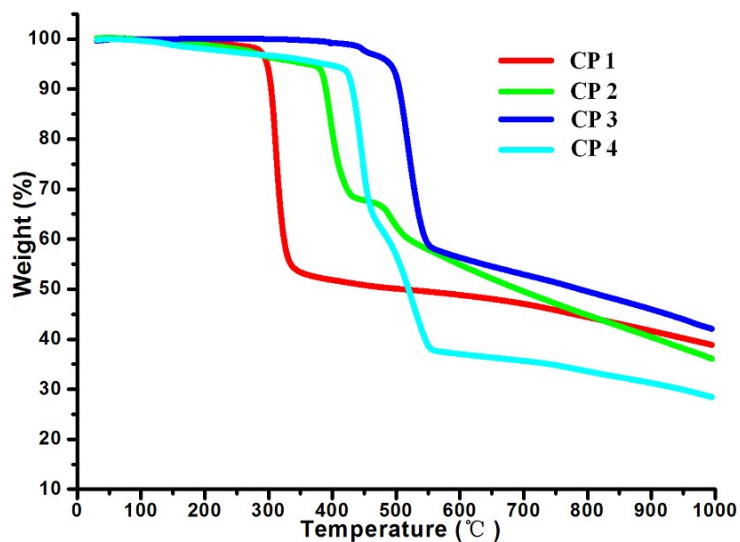


Fig S11. TGA curves for CPs 1-4.

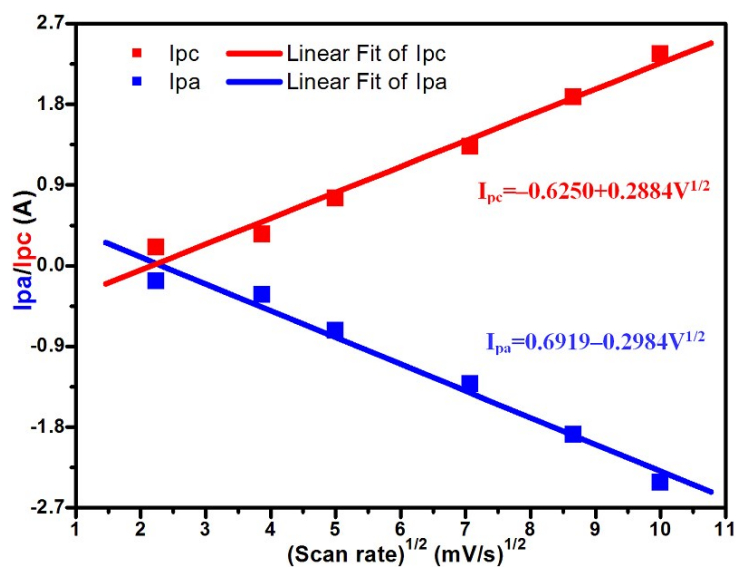


Fig S12. The voltammetric current as a function of square root of scan rate of CP 1 as electrode.

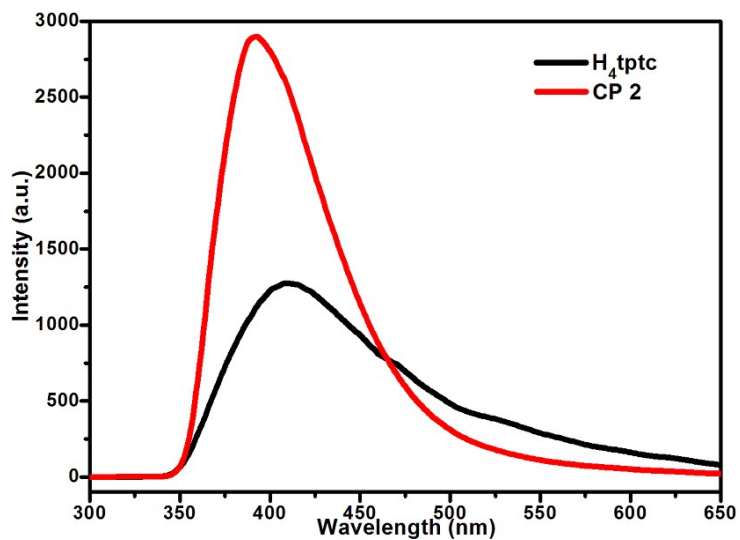


Fig S13. The fluorescence spectrum of free  $H_4tptc$  and CP 2 in solid state at room temperature.

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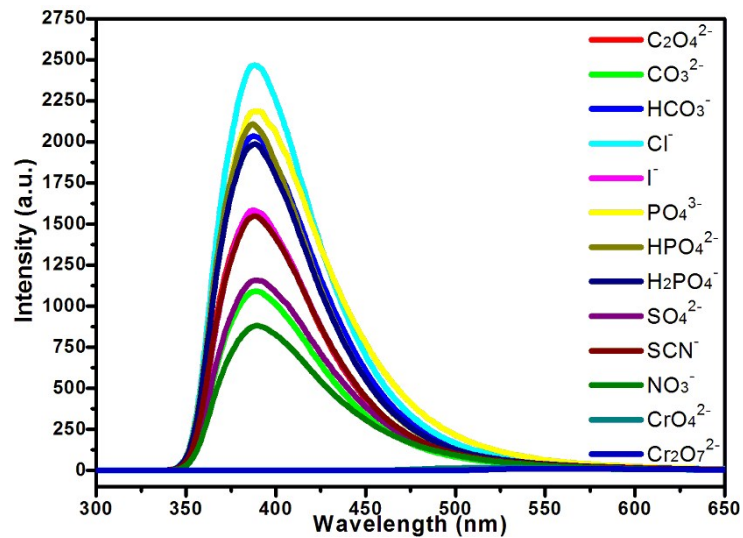


Fig S14. The luminescence intensities of CP 2 which were dispersed in the aqueous solution of different anions.

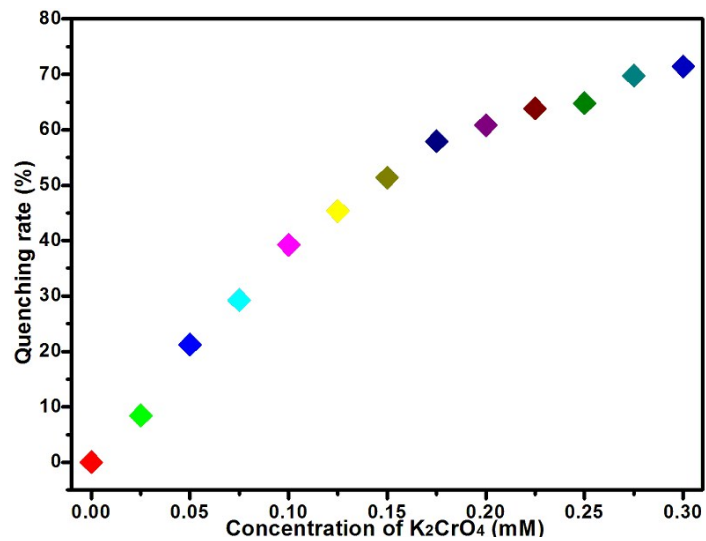


Fig S15. The quenching rate of CP 2 with different concentrations of  $K_2CrO_4$  in aqueous solution.

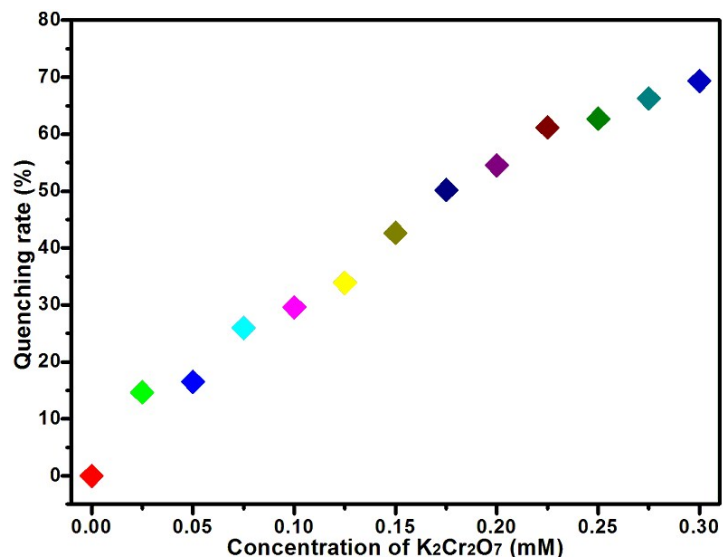


Fig S16. The quenching rate of CP 2 with different concentrations of  $K_2Cr_2O_7$  in aqueous solution.

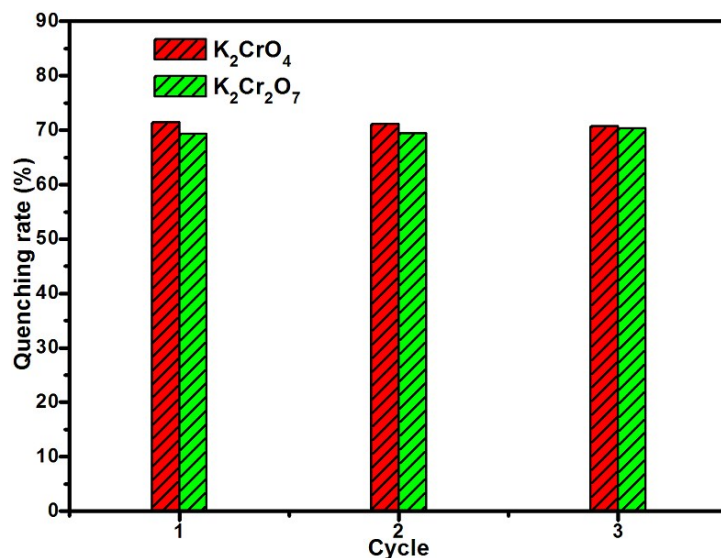
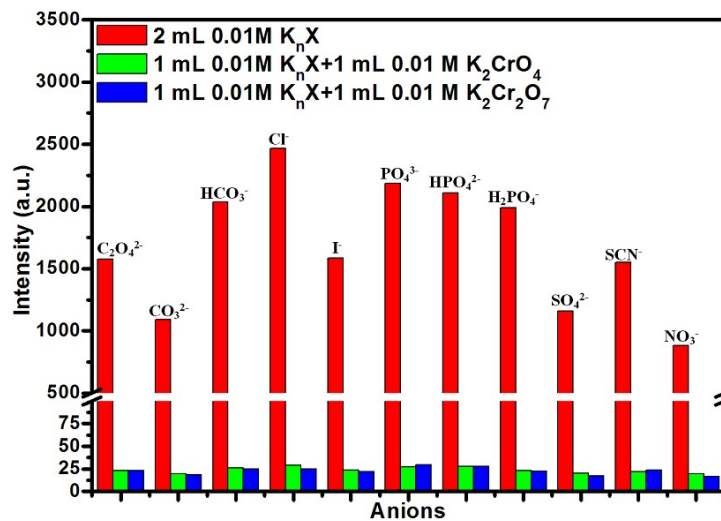


Fig. S17 The repeatability of Cr(VI) anions for CP 2 for three cycles.



5 Fig. S18 The anti-interferences of CP 2 in sensing of  $CrO_4^{2-}/Cr_2O_7^{2-}$  anions from normal anions in aqueous solution.

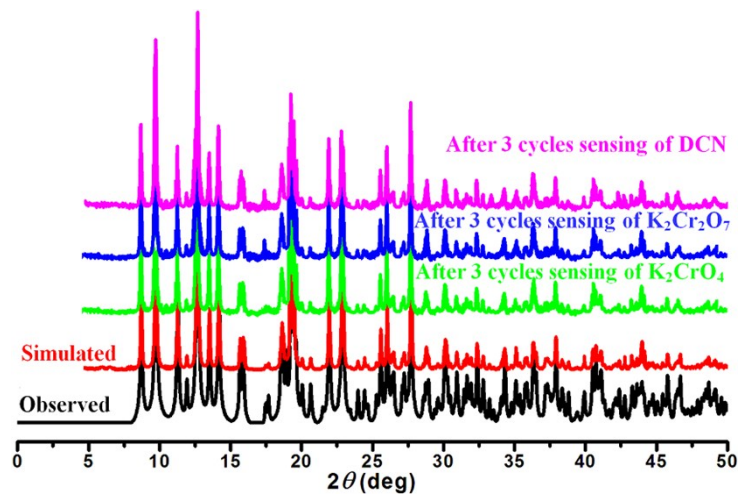
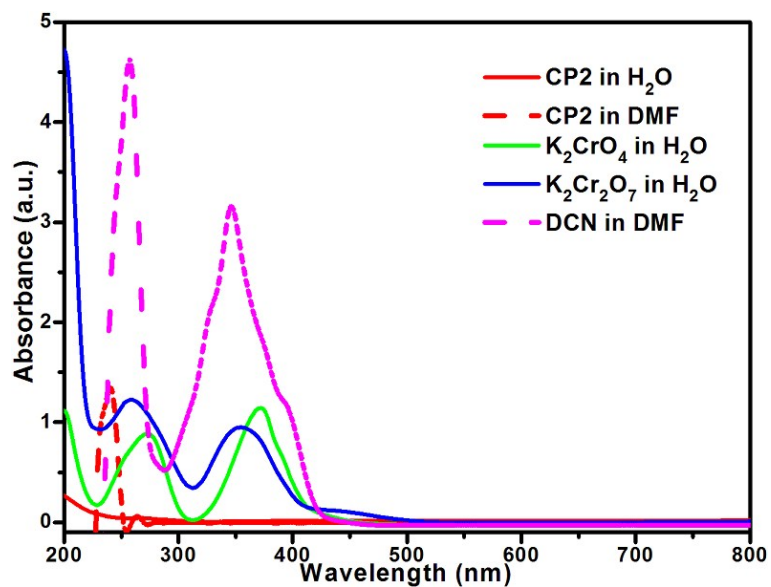
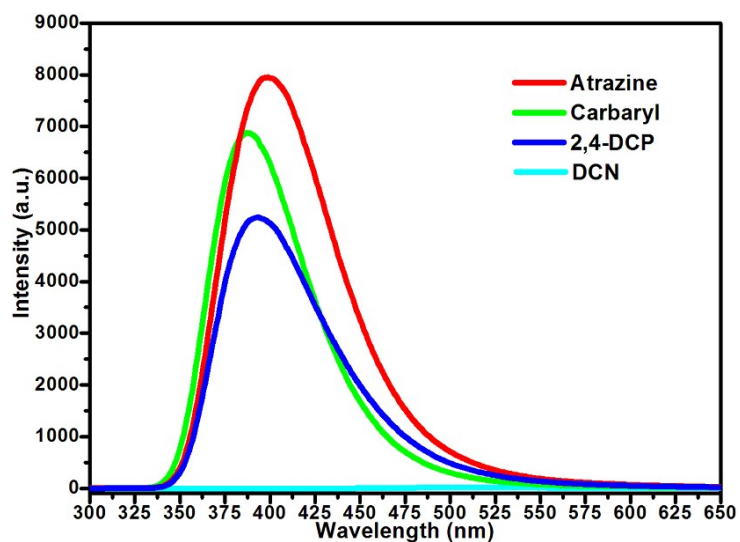


Fig S19. PXRD patterns of CP 2 after three cycles sensing of Cr(VI) anions in aqueous solution and DCN in DMF solution.





**Fig S20.** The UV-Vis absorbance spectrum of Cr(VI) anions in aqueous solution and DCN in DMF solution and the UV-Vis absorbance spectrum of CP 2.



5 **Fig S21.** The luminescence intensities of CP 2 which were dispersed in the DMF solution of different pesticides.

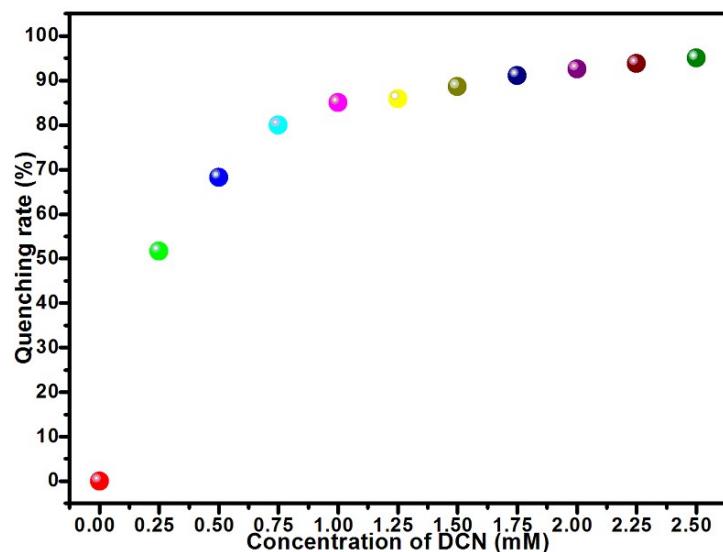


Fig S22. The quenching rates of CP 2 with different concentrations of DCN in DMF solution.

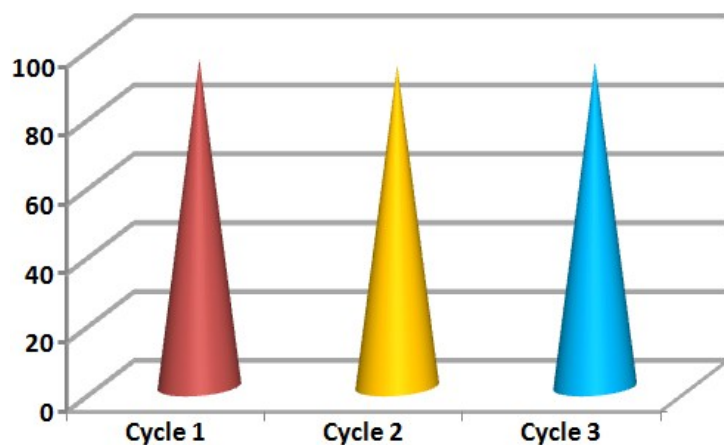


Fig. S23 The quenching rates of DCN for CP 2 for three cycles.

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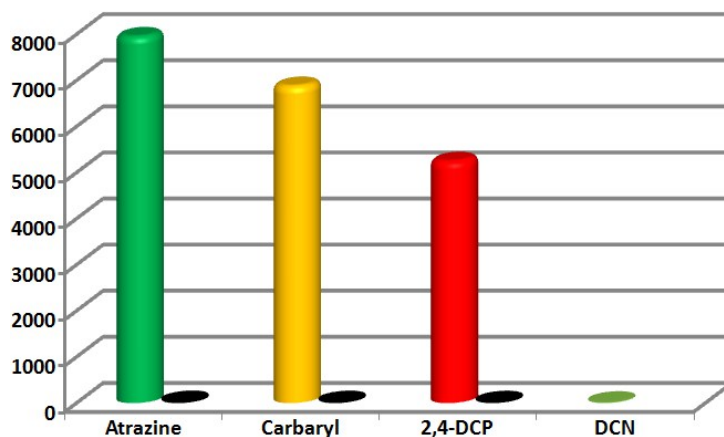


Fig. S24 The anti-interferences of CP 2 in sensing of DCN from normal pesticides in DMF solution.

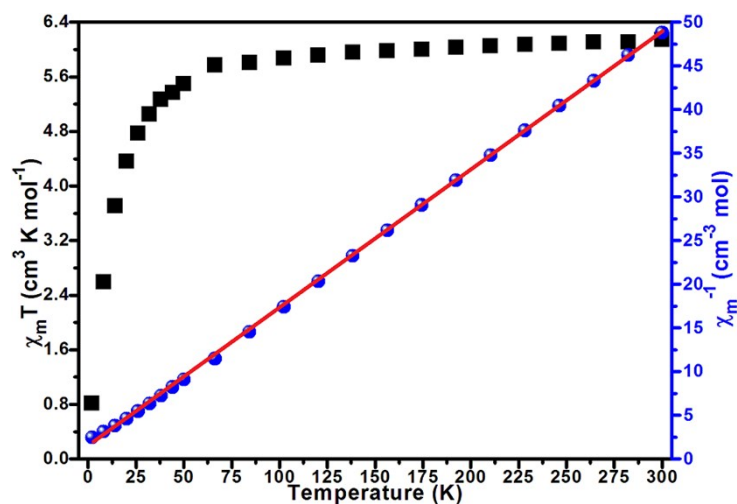


Fig S25. Temperature dependence of  $\chi_m T$  and  $\chi_m^{-1}$  for **3** under 1000 Oe.

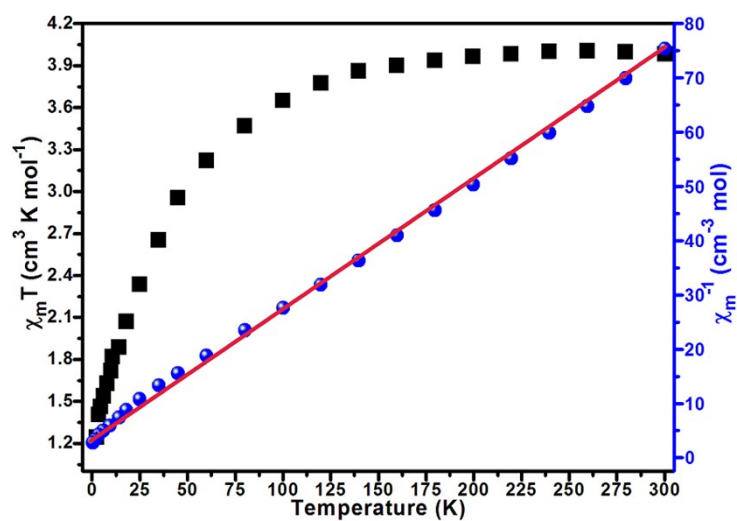


Fig S26. Temperature dependence of  $\chi_m T$  and  $\chi_m^{-1}$  for **4** under 1000 Oe.

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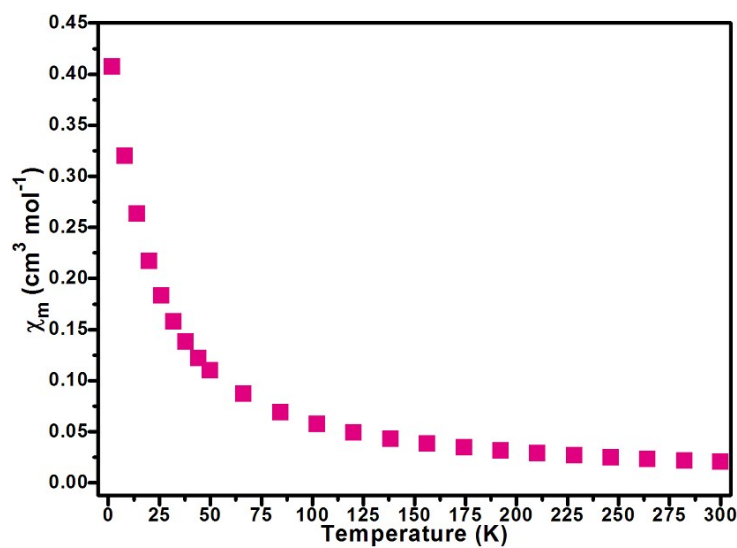


Fig S27. The  $\chi_m$  versus T of CP **3**.

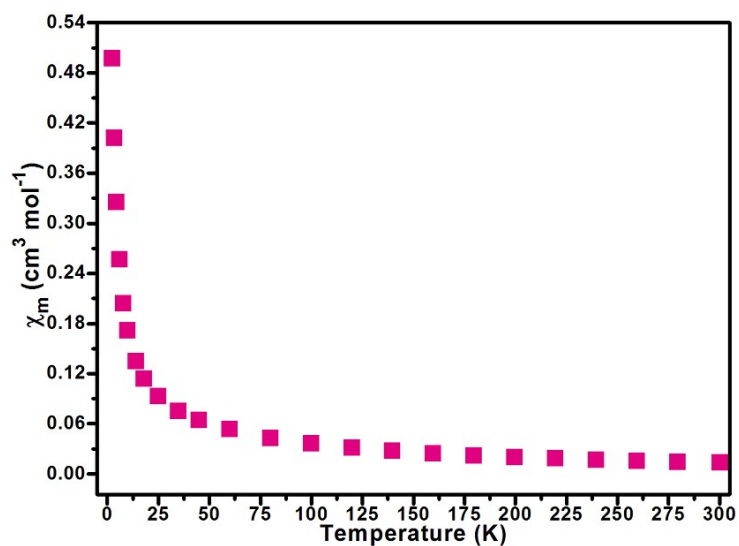


Fig S28. The  $\chi_m$  versus T of CP 4.

5 Table S1 Crystal data for 1–4.

CP	1	2	3	4
Formula	$C_{23}H_{13}CuN_2O_4$	$C_{46}H_{26}Cd_2N_4O_8$	$C_{23}H_{13}MnN_2O_4$	$C_{34}H_{20}MnN_2O_8$
Formula weight	444.89	987.56	436.29	639.46
Crystal system	Triclinic	Triclinic	Monoclinic	Monoclinic
Space group	$P-1$	$P-1$	$C2/c$	$I2/a$
$a$ (Å)	9.281(5)	9.7425(4)	20.831(16)	12.166(4)
$b$ (Å)	10.104(5)	9.7964(4)	9.137(6)	17.6501(10)
$c$ (Å)	10.837(6)	10.3959(5)	20.062(15)	14.104(7)
$\alpha$ (°)	67.98(2)	77.9630(10)	90	90
$\beta$ (°)	76.875(19)	81.9760(10)	96.34(3)	114.646(11)
$\gamma$ (°)	67.945(15)	69.6560(10)	90	90
$V$ (Å <sup>3</sup> )	868.7(8)	907.42(7)	3795(5)	2752.7(17)
$Z$	2	1	8	4
$D_{\text{calcd}}$ (Mg/m <sup>3</sup> )	1.701	1.807	1.527	1.543
$\mu$ (mm <sup>-1</sup> )	1.294	1.239	0.730	0.541
Temperature (K)	293(2)	296(2)	293(2)	295(2)
$F(000)$	452	489	1776	1308
$R_{\text{int}}$	0.0384	0.0253	0.0828	0.0581
$R_1$ [ $I > 2\sigma(I)$ ] <sup>a</sup>	0.0344	0.0236	0.0518	0.0658
$wR_2$ [ $I > 2\sigma(I)$ ] <sup>b</sup>	0.0829	0.0606	0.1044	0.1619
Gof	1.065	1.047	1.035	1.044

<sup>a</sup> $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$ , <sup>b</sup> $wR_2 = [\sum w(F_o^2 - F_c^2)^2] / \sum w(F_o^2)^2]^{1/2}$

**Table S2** Selected bond lengths (Å) and angles (°) for **1 – 4**.

<b>CP 1</b>							
Cu(1)-O(3)	1.9254(17)	Cu(1)-O(2) <sup>#2</sup>	2.264(2)	Cu(1)-N(1)	2.045(2)	Cu(1)-N(2)	2.025(2)
Cu(1)-O(1) <sup>#1</sup>	1.9269(17)	O(3)-Cu(1)-O(1) <sup>#1</sup>	98.41(8)	O(3)-Cu(1)-N(2)	88.35(8)	O(1) <sup>#1</sup> -Cu(1)-N(2)	166.61(8)
O(3)-Cu(1)-N(1)	166.50(7)	N(2)-Cu(1)-N(1)	80.78(8)	O(1) <sup>#1</sup> -Cu(1)-O(2) <sup>#2</sup>	99.08(8)	N(1)-Cu(1)-O(2) <sup>#2</sup>	83.15(8)
O(1) <sup>#1</sup> -Cu(1)-N(1)	90.73(8)	O(3)-Cu(1)-O(2) <sup>#2</sup>	105.01(8)	N(2)-Cu(1)-O(2) <sup>#2</sup>	90.22(8)		
Symmetry codes: #1 -x+1, -y+1, -z+1; #2 -x+1, -y, -z+1.							
<b>CP 2</b>							
Cd(1)-O(1)	2.2820(19)	Cd(1)-O(2) <sup>#2</sup>	2.3457(18)	Cd(1)-O(3) <sup>#1</sup>	2.4515(17)	Cd(1)-N(1)	2.376(2)
Cd(1)-O(4) <sup>#1</sup>	2.3006(17)	Cd(1)-N(2)	2.3571(19)	O(1)-Cd(1)-O(4) <sup>#1</sup>	116.18(8)	O(1)-Cd(1)-O(2) <sup>#2</sup>	122.48(7)
O(1)-Cd(1)-N(2)	82.54(7)	O(4) <sup>#1</sup> -Cd(1)-O(2) <sup>#2</sup>	85.97(7)	N(2)-Cd(1)-N(1)	70.39(7)	O(2) <sup>#2</sup> -Cd(1)-O(3) <sup>#1</sup>	139.33(6)
O(4) <sup>#1</sup> -Cd(1)-N(2)	160.71(8)	O(4) <sup>#1</sup> -Cd(1)-N(1)	97.98(8)	O(1)-Cd(1)-O(3) <sup>#1</sup>	87.48(7)	N(2)-Cd(1)-O(3) <sup>#1</sup>	135.90(6)
O(2) <sup>#2</sup> -Cd(1)-N(2)	79.42(7)	O(2) <sup>#2</sup> -Cd(1)-N(1)	92.54(7)	O(4) <sup>#1</sup> -Cd(1)-O(3) <sup>#1</sup>	54.48(6)	N(1)-Cd(1)-O(3) <sup>#1</sup>	84.90(7)
O(1)-Cd(1)-N(1)	130.83(7)						
Symmetry codes: #1 -x+1, -y, -z+1; #2 -x+1, -y+1, -z+1.							
<b>CP 3</b>							
Mn(1)-O(2)	2.096(2)	Mn(1)-O(4) <sup>#2</sup>	2.184(3)	Mn(1)-O(3) <sup>#2</sup>	2.332(2)	Mn(1)-N(2)	2.286(3)
Mn(1)-O(1) <sup>#1</sup>	2.113(2)	Mn(1)-N(1)	2.275(3)	O(2)-Mn(1)-O(1) <sup>#1</sup>	93.86(8)	O(2)-Mn(1)-O(4) <sup>#2</sup>	101.05(9)
O(2)-Mn(1)-N(1)	98.13(10)	O(1) <sup>#1</sup> -Mn(1)-O(4) <sup>#2</sup>	99.61(9)	N(1)-Mn(1)-N(2)	72.38(10)	O(4) <sup>#2</sup> -Mn(1)-O(3) <sup>#2</sup>	57.99(9)
O(1) <sup>#1</sup> -Mn(1)-N(1)	97.07(10)	O(1) <sup>#1</sup> -Mn(1)-N(2)	164.88(10)	O(2)-Mn(1)-O(3) <sup>#2</sup>	158.30(8)	N(1)-Mn(1)-O(3) <sup>#2</sup>	103.51(10)
O(4) <sup>#2</sup> -Mn(1)-N(1)	153.55(9)	O(4) <sup>#2</sup> -Mn(1)-N(2)	86.91(9)	O(1) <sup>#1</sup> -Mn(1)-O(3) <sup>#2</sup>	85.12(9)	N(2)-Mn(1)-O(3) <sup>#2</sup>	86.87(9)
O(2)-Mn(1)-N(2)	98.29(9)						
Symmetry codes: #1 -x+1/2, -y+3/2, -z+1; #2 -x+1/2, y+1/2, -z+1/2.							
<b>CP 4</b>							
Mn(1)-O(3)	2.114(3)	Mn(1)-N(1)	2.280(11)	Mn(1)-O(2) <sup>#2</sup>	2.268(3)	Mn(1)-N(2)	2.258(12)
O(3) <sup>#1</sup> -Mn(1)-N(1)	90.7(6)	O(3)-Mn(1)-O(3) <sup>#1</sup>	167.62(14)	O(3)-Mn(1)-O(2) <sup>#2</sup>	89.63(11)	N(2)-Mn(1)-O(2) <sup>#3</sup>	83.9(5)
N(2)-Mn(1)-N(1)	73.2(2)	O(3)-Mn(1)-N(2)	88.7(6)	N(2)-Mn(1)-O(2) <sup>#2</sup>	158.2(5)	O(2) <sup>#2</sup> -Mn(1)-O(2) <sup>#3</sup>	117.49(15)
O(2) <sup>#2</sup> -Mn(1)-N(1)	85.6(5)	O(3) <sup>#1</sup> -Mn(1)-N(2)	101.2(6)	O(3)-Mn(1)-O(2) <sup>#3</sup>	83.95(11)	O(3)-Mn(1)-N(1)	99.4(6)
O(2) <sup>#3</sup> -Mn(1)-N(1)	156.8(5)						
Symmetry codes: #1 -x+1/2, y, -z+1; #2 x-1, y, z; #3 -x+3/2, y, -z+1.							