

Supplementary Information

Synthesis, Characterization and Solid State Conductivity of Nano- Size Ionic Schiff
Base Polymers of Cu^{2+} , Zn^{2+} and VO^{2+} Containing Viologen Moiety

Zohreh. S. Ghavami^a, Gholamhossein Grivani^a

^aSchool of Chemistry, Damghan University, Damghan 36715-364, Iran

Corresponding author: e-.mail: grivani@du.ac.ir

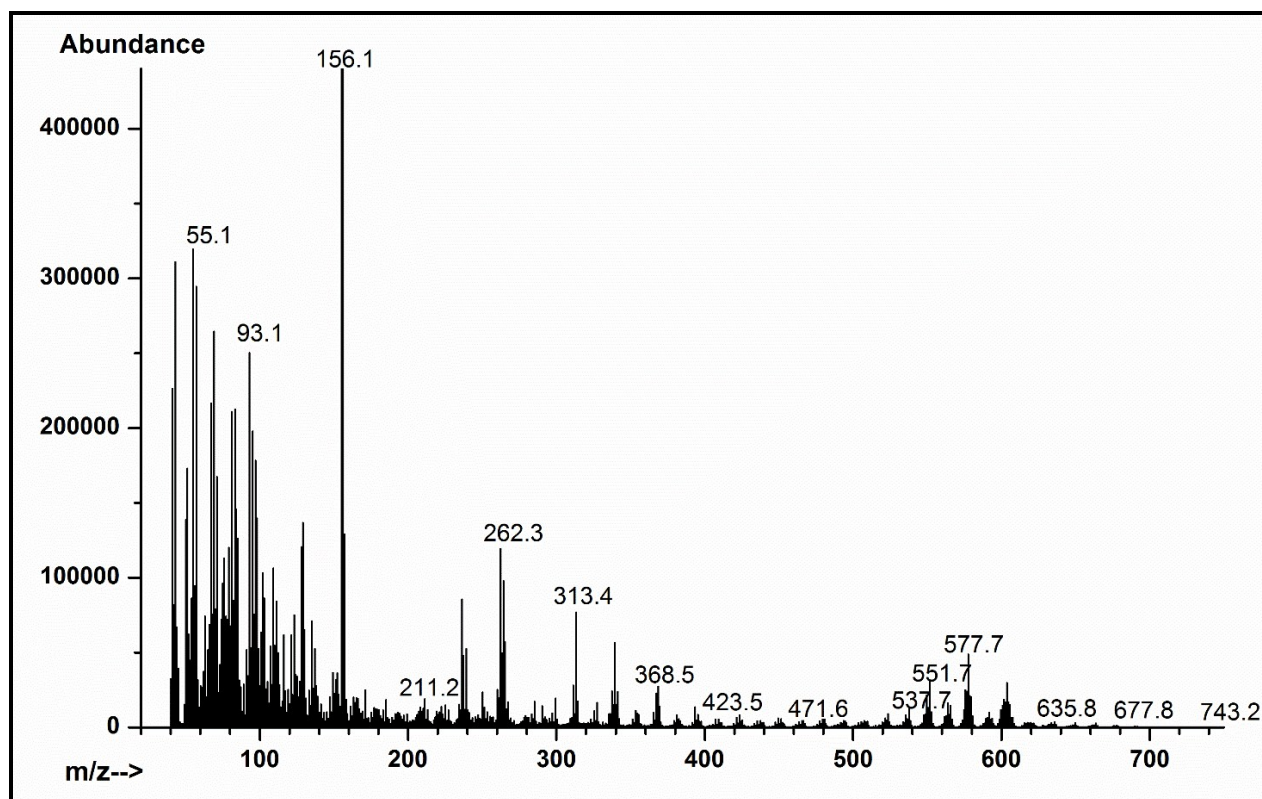


Figure S1. Mass spectrum of IVSP

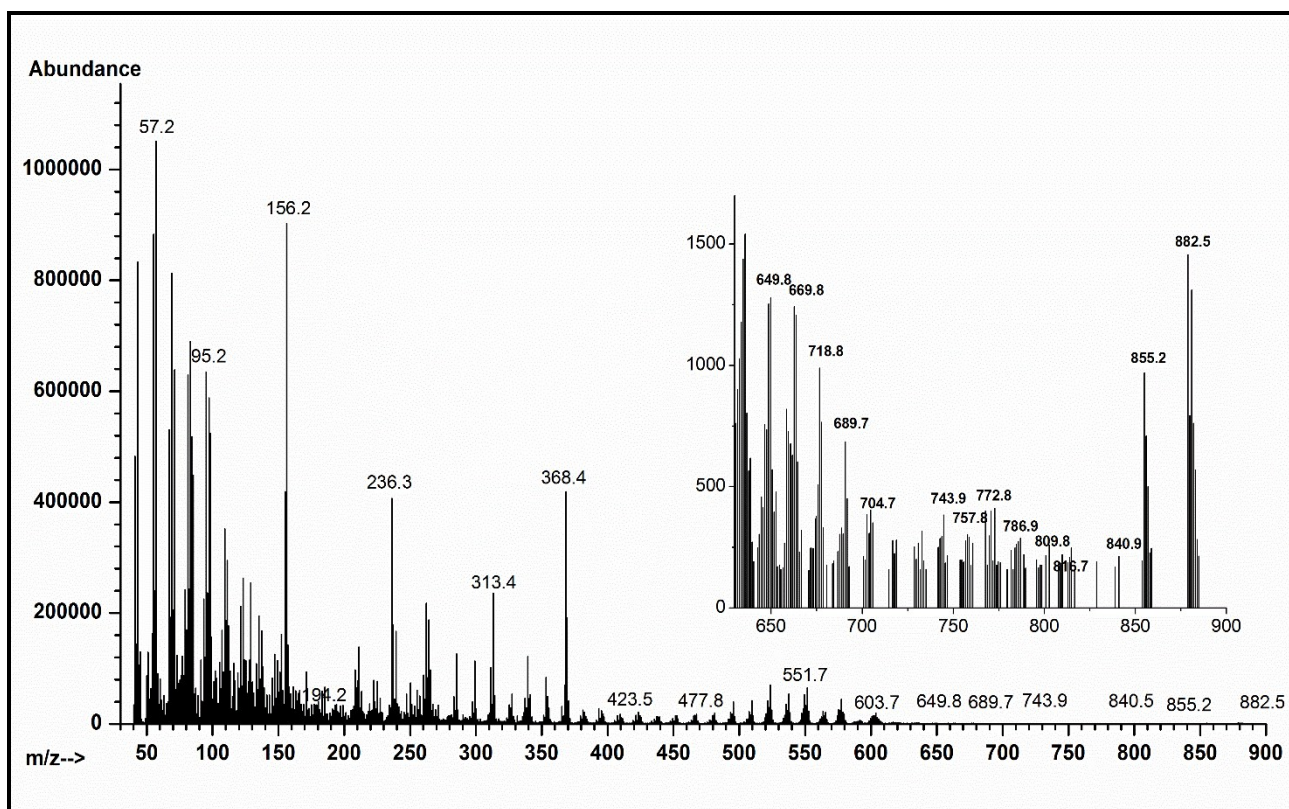


Figure S2. Mass spectrum of IZSP

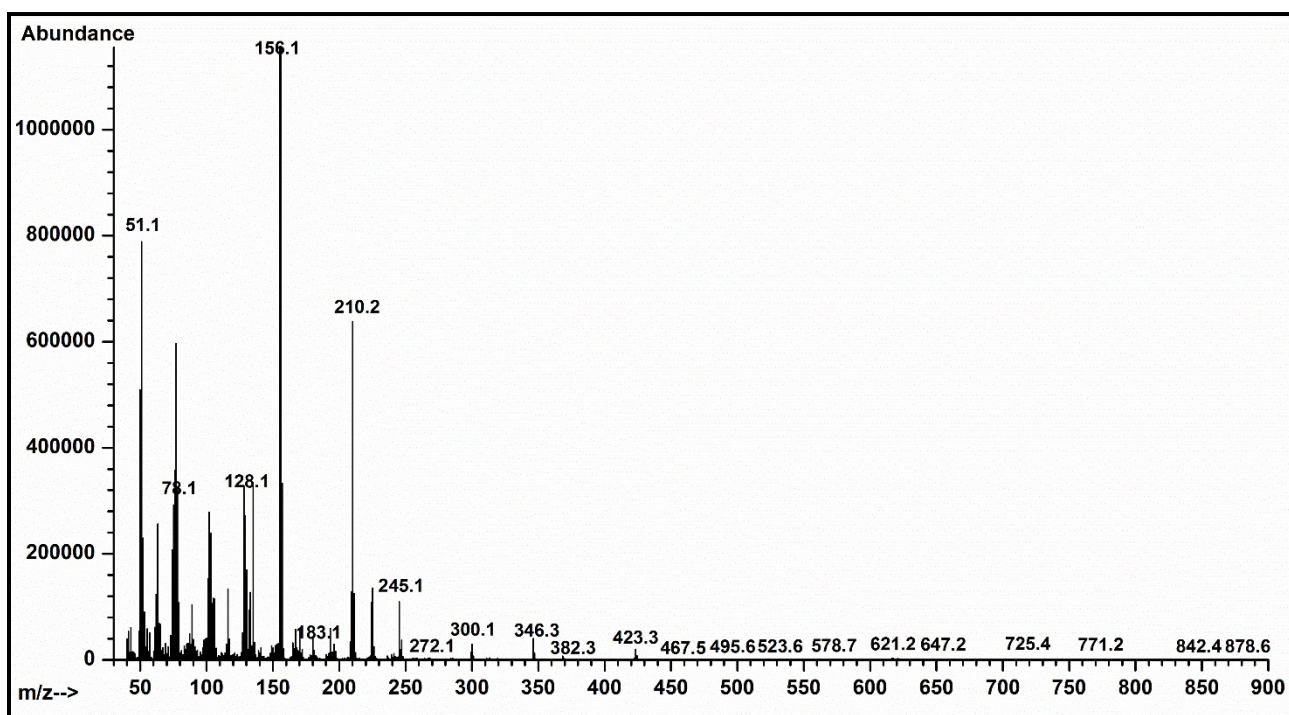


Figure S3. Mass spectrum of ICSP

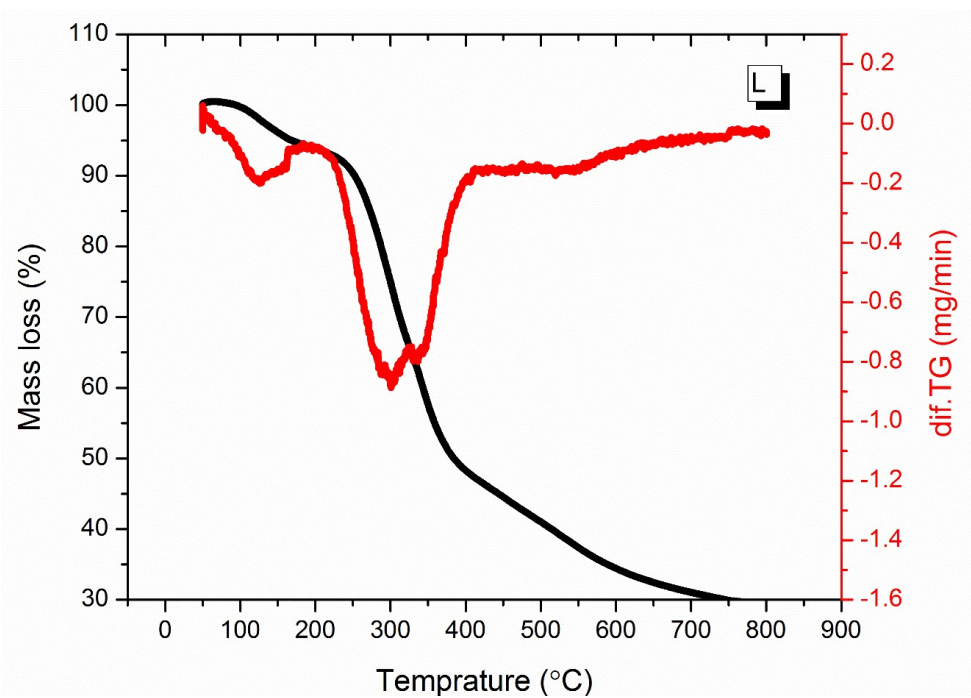


Figure S4. TGA and DTA curves of L.

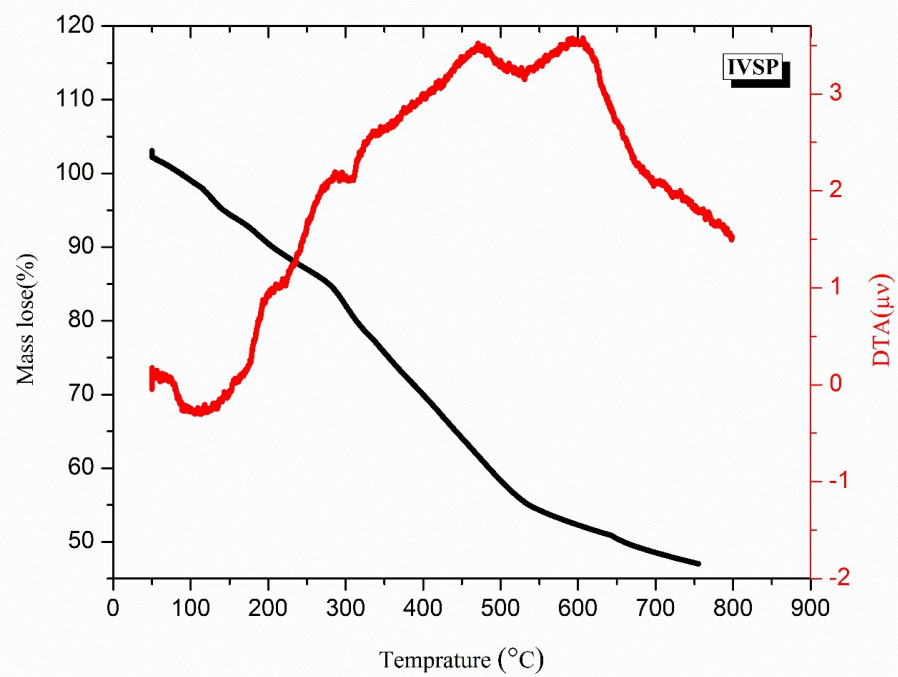


Figure S5. TGA and DTA curves of IVSP.

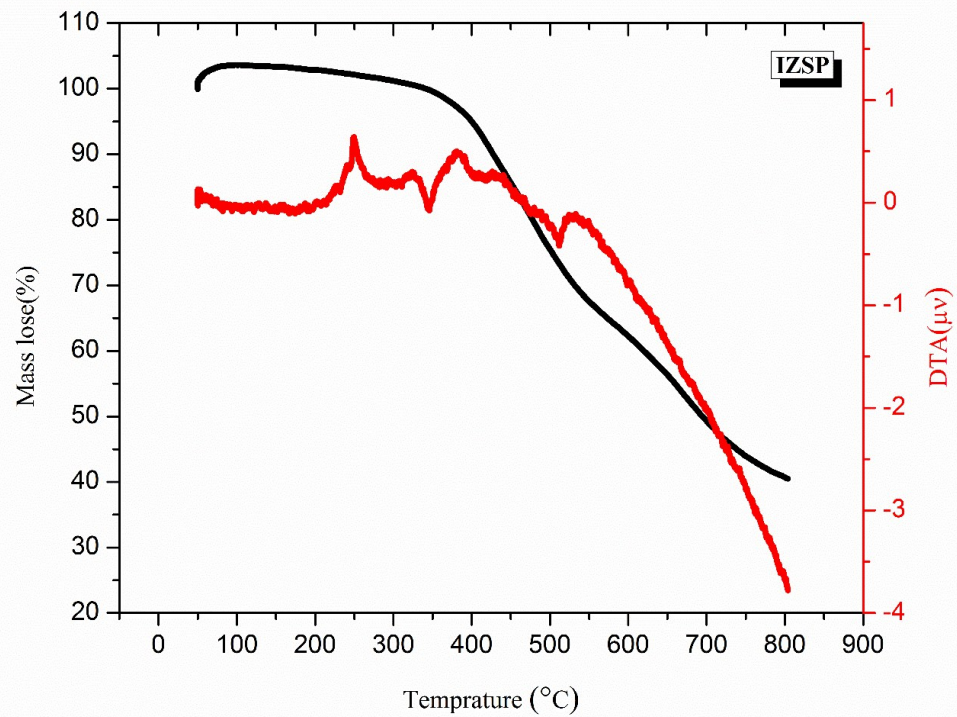


Figure S6. TGA and DTA curves of IZSP.

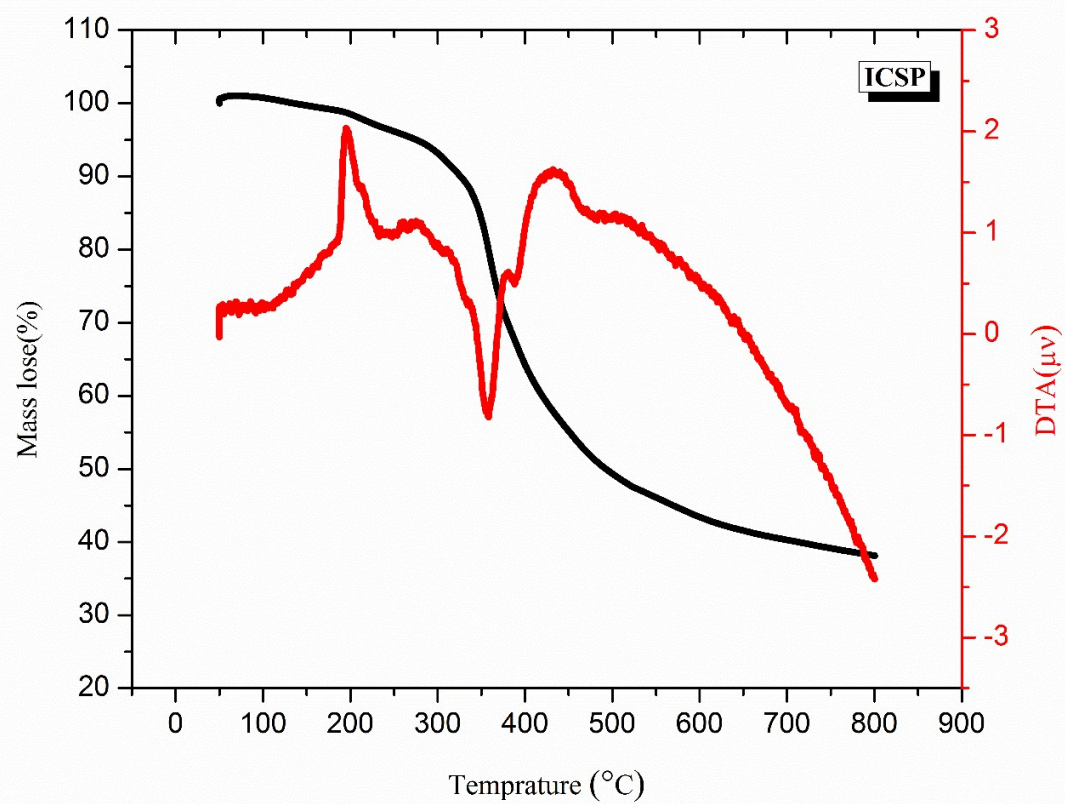


Figure S7. TGA and DTA curves of PSBL and ICSP.

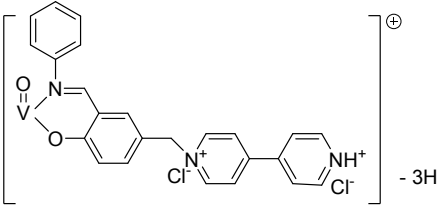
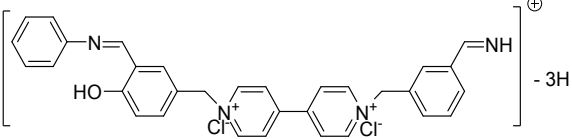
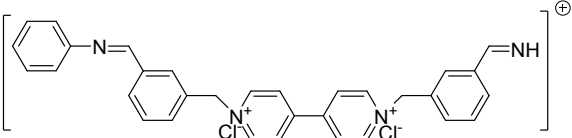
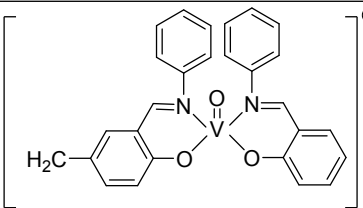
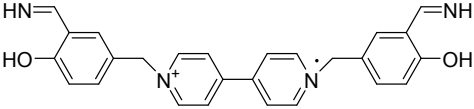
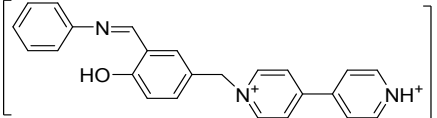
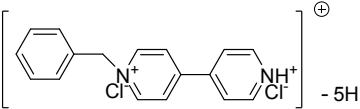
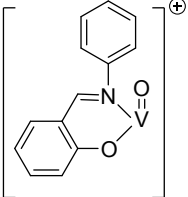
Table S1. Some main signals of the mass spectrum of the **L**

Number	M	Z	M/Z	Structure	Chemical formula
1	719	+1	719		$C_{38}H_{32}N_4O_2Cl_2 \cdot 4H_2O$
2	647.7	+1	647.7		$C_{38}H_{32}N_4O_2Cl_2$
3	630.5	+1	630.5		$C_{38}H_{32}N_4OCl_2$
4	617.7	+1	617.7		$C_{38}H_{32}N_4Cl_2$
5	577.7	+1	577.7		$C_{38}H_{32}N_4O_2$
6	551.7	+1	551.7		$C_{32}H_{28}ON_4Cl_2$
7	423.5	+1	423.5		$C_{26}H_{24}O_2N_4$
8	368.4	+1	368.4		$C_{24}H_{20}ON_3$
9	313.4	+1	313.4		$C_{17}H_{15}N_2Cl_2$
10	262.3	+1	262.3		$C_{17}H_{15}N_2Cl_2$

11	193.2	+1	193.2		C ₁₄ H ₁₁ N
12	156.2	+1	156.2		C ₁₀ H ₈ N ₂
13	93.1	+1	93.1		C ₆ H ₇ N
14	51.1	+1	51.1	[OCi] ⁺	OCI

Table S2. Some main signals of the mass spectrum of the **IVSP**

Number	M	Z	M/Z	Structure	Chemical formula ^a
1	743	+1	743		C ₃₉ H ₃₄ N ₄ O ₃ Cl ₂ V
2	712	+1	712		C ₃₈ H ₃₀ N ₄ O ₃ Cl ₂ V
3	677.8	+1	677.8		C ₃₈ H ₃₀ N ₄ O ₃ ClV
4	635.8	+1	635.8		C ₃₈ H ₃₀ N ₄ O ₃ V

5	577.7	+1	577.7		$C_{30}H_{21}N_3O_2Cl_2V$
6	551.7	+1	551.7		$C_{32}H_{27}ON_4Cl_2$
7	537.7	+1	537.7		$C_{32}H_{26}N_4Cl_2$
8	471.6	+1	471.6		$C_{27}H_{20}O_3N_2V$
9	423.5	+1	423.5		$C_{26}H_{24}O_2N_4$
10	368.5	+1	368.5		$C_{24}H_{20}ON_3$
11	313.4	+1	313.4		$C_{17}H_{15}N_2Cl_2$
12	262.3	+1	262.3		$C_{13}H_9NO_2V$

13	211.2	+1	211.2		C ₁₄ H ₁₃ NO
14	156.1	+1	156.1		C ₁₀ H ₈ N
15	93.1	+1	93.1		C ₆ H ₇ N
16	55.1	+1	55.1	[OCI] ⁺	OCI

Table S3. Some main signals of the mass spectrum of the IZSP

Number	M	Z	M/Z	Structure	Chemical formula
1	882.5	+1	882.5		C ₃₈ H ₃₀ Cl ₄ N ₄ O ₂ Zn ₂ ·2H ₂ O
2	864.2	+1	864.2		C ₃₈ H ₃₀ Cl ₄ N ₄ O ₂ Zn ₂ ·H ₂ O -9H
3	846.5	+1	846.5		C ₃₈ H ₃₀ Cl ₄ N ₄ O ₂ Zn ₂
4	809.8	+1	809.8		C ₃₈ H ₃₀ Cl ₃ N ₄ O ₂ Zn ₂

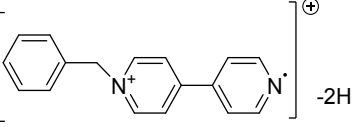
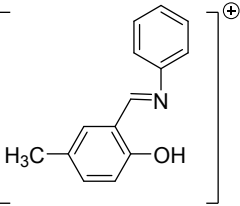
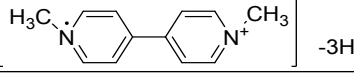
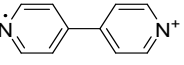
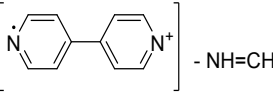
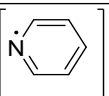
5	772	+1	772		$C_{38}H_{30}Cl_2N_4O_2Zn_2$
6	743.9	+1	743.9		$C_{31}H_{25}Cl_4N_3O_2Zn_2$
7	689.7	+1	689.7		$C_{26}H_{20}Cl_4N_4O_2Zn_2$
8	649.8	+1	649.8		$C_{25}H_{21}Cl_4N_3OZn_2$
9	551.7	+1	551.7		$C_{32}H_{27}ON_4Cl_2$
10	477.8	+1	477.8		$C_{26}H_{24}ON_4Cl_2$
11	423.5	+1	423.5		$C_{26}H_{24}O_2N_4$
12	368.4	+1	368.4		$C_{24}H_{20}ON_3$
13	313.4	+1	313.4		$C_{17}H_{15}N_2Cl_2$
14	236.3	+1	236.3		$C_{11}H_{10}N_2Cl_2$

15	194.2	+1	194.2		C ₁₄ H ₁₁ N
16	156	+1	156		C ₁₀ H ₈ N
17	95	+1	95		C ₆ H ₉ N

Table S4. Some main signals of the mass spectrum of the ICSP

Number	M	Z	M/Z	Structure	Chemical formula
1	878.6	+1	878.6		C ₃₈ H ₃₀ Cl ₄ N ₄ O ₂ Cu ₂ . 2H ₂ O
2	842.4	+1	842.4		C ₃₈ H ₃₀ Cl ₄ N ₄ O ₂ Cu ₂
3	771.2	+1	771.2		C ₃₈ H ₃₀ Cl ₂ N ₄ O ₂ Cu ₂
4	725.4	+1	725.4		C ₃₁ H ₂₆ Cl ₄ N ₃ OCu ₂
5	647.6	+1	647.6		C ₂₅ H ₂₁ Cl ₄ N ₃ OCu ₂
6	621.4	+1	621.4		C ₂₆ H ₂₂ Cl ₄ N ₄ O ₂ Cu

7	577.8	+1	577.8		$C_{25}H_{22}Cl_2N_3OCu_2$
8	551.6	+1	551.6		$C_{32}H_{28}ON_4Cl_2$
9	523.6	+1	523.6		$C_{31}H_{27}ON_3Cl_2$
10	495.6	+1	495.6		$C_{26}H_{24}O_2N_4Cl_2$
11	467.5	+1	467.5		$C_{25}H_{22}O_2N_3Cl_2$
12	440.4	+1	440.4		$C_{24}H_{22}O_2N_2Cl_2$
13	423.3	+1	423.3		$C_{26}H_{24}O_2N_4$
14	397.4	+1	397.4		$C_{25}H_{23}O_2N_3$
15	368.4	+1	368.4		$C_{24}H_{20}N_2O_2$
16	346.3	+1	346.3		$C_{24}H_{24}N_2$
17	300.1	+1	300.1		$C_7H_6NCu_2Cl_2$

18	245.1	+1	245.1	 $\left[\text{C}_6\text{H}_5\text{CH}_2\text{N}^+\text{C}_5\text{H}_4\text{N}^\bullet \right]^\oplus$	$\text{C}_{17}\text{H}_{15}\text{N}_2$
19	210.2	+1	210.2	 $\left[\text{H}_3\text{C}-\text{C}_6\text{H}_3(\text{OH})=\text{N}-\text{C}_6\text{H}_5 \right]^\oplus$	$\text{C}_{14}\text{H}_{12}\text{NO}$
20	183.1	+1	183.1	 $\left[\text{H}_3\text{C}-\text{N}^\bullet\text{C}_5\text{H}_4-\text{C}_5\text{H}_4-\text{N}^+\text{CH}_3 \right]^\oplus$	$\text{C}_{12}\text{H}_{14}\text{N}_2$
21	156.1	+1	156.1	 $\left[\text{N}^\bullet\text{C}_5\text{H}_4-\text{C}_5\text{H}_4-\text{N}^+ \right]^\oplus$	$\text{C}_{10}\text{H}_8\text{N}_2$
22	128.1	+1	128.1	 $\left[\text{N}^\bullet\text{C}_5\text{H}_4-\text{C}_5\text{H}_4-\text{N}^+ \right]^\oplus$	$\text{C}_9\text{H}_7\text{N}$
23	78.1	+1	78.1	 $\left[\text{N}^\bullet\text{C}_5\text{H}_5 \right]^\oplus$	$\text{C}_5\text{H}_5\text{N}$
24	51.1	+1	51.1	$[\text{OCl}]^+$	OCl