

Supporting Information

Core@Shell Hollow Heterostructure of Co_3O_4 and Co_3S_4 : An Efficient Oxygen Evolution Catalyst

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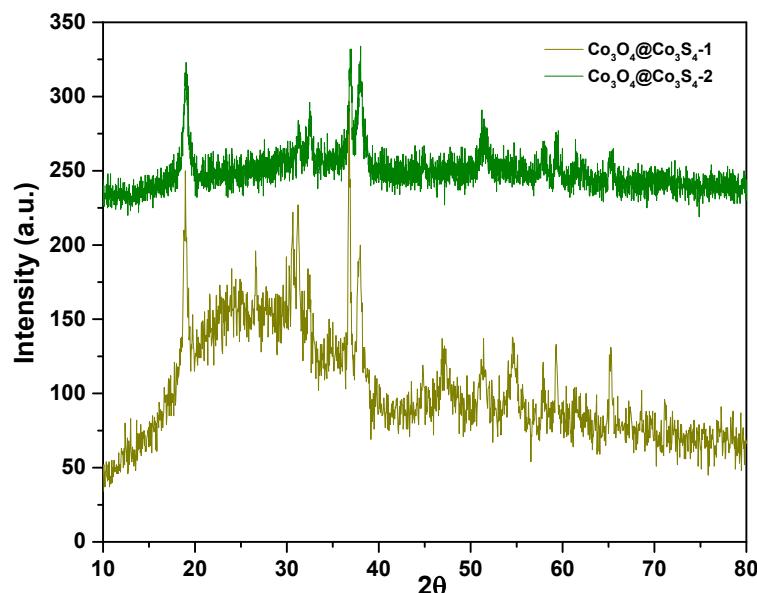


Fig. S1: XRD pattern of $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$ -1 and $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$ -2 sheets.

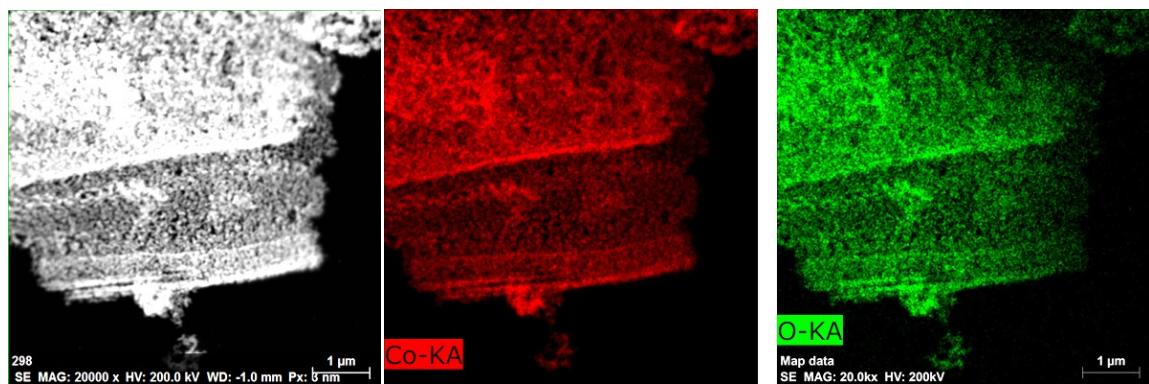


Fig. S2: EDS area mapping of Co_3O_4 sheets showing the uniform distribution of Co and O.

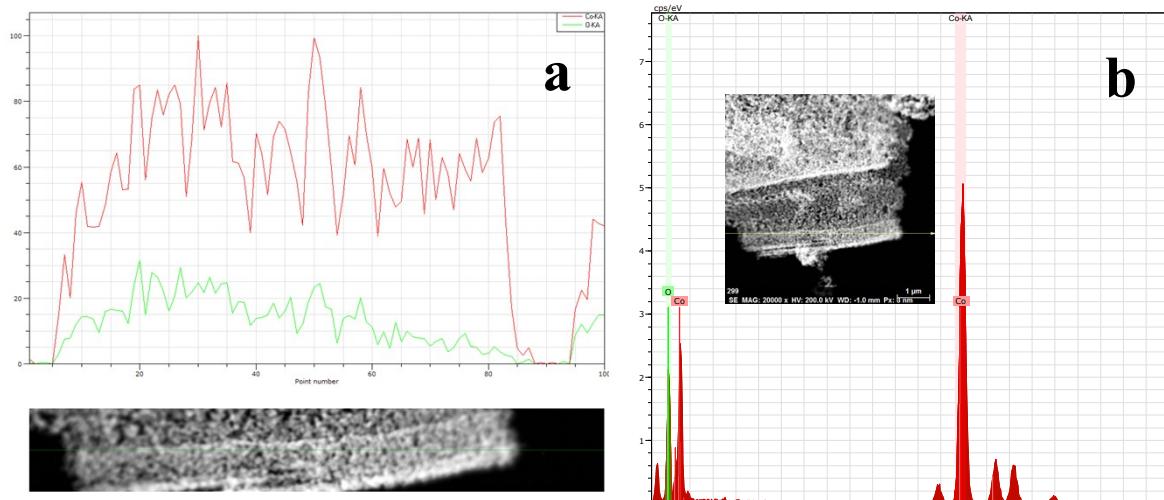


Fig. S3: (a) EDS line mapping shows comparative distribution of Co and O and (b) EDS spectrum indicating the presence of Co and O in Co_3O_4 sheets.

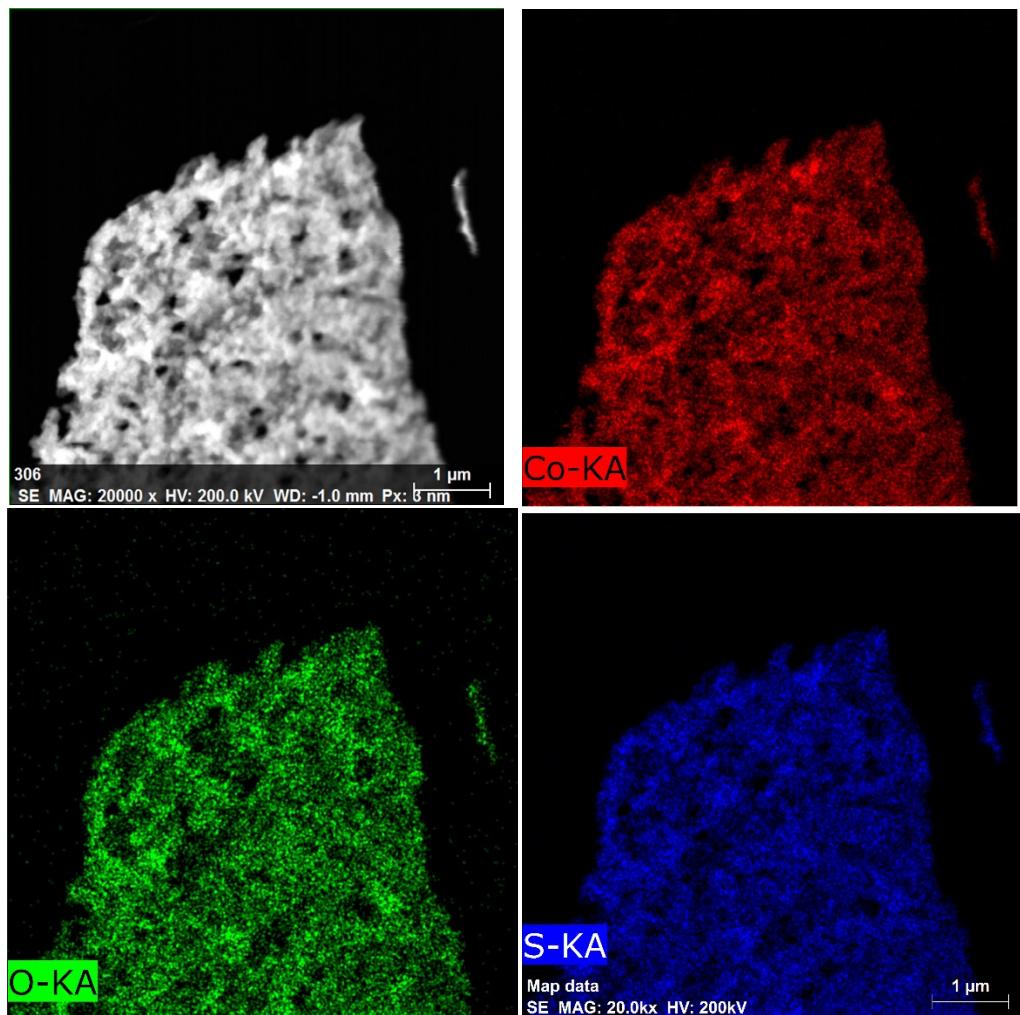


Fig. S4: EDS area mapping of $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$ -2 sheets showing the uniform distribution of Co, O and S.

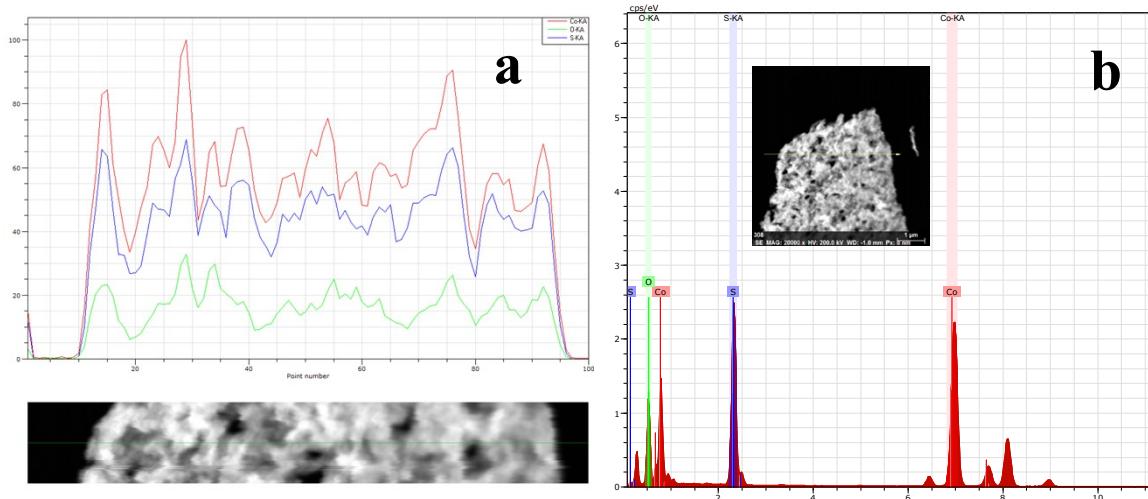


Fig. S5: (a) EDS line mapping shows comparative distribution of Co, S and O and (b) EDS spectrum indicating the signal of Co, S and O in $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$ -2 sheets.

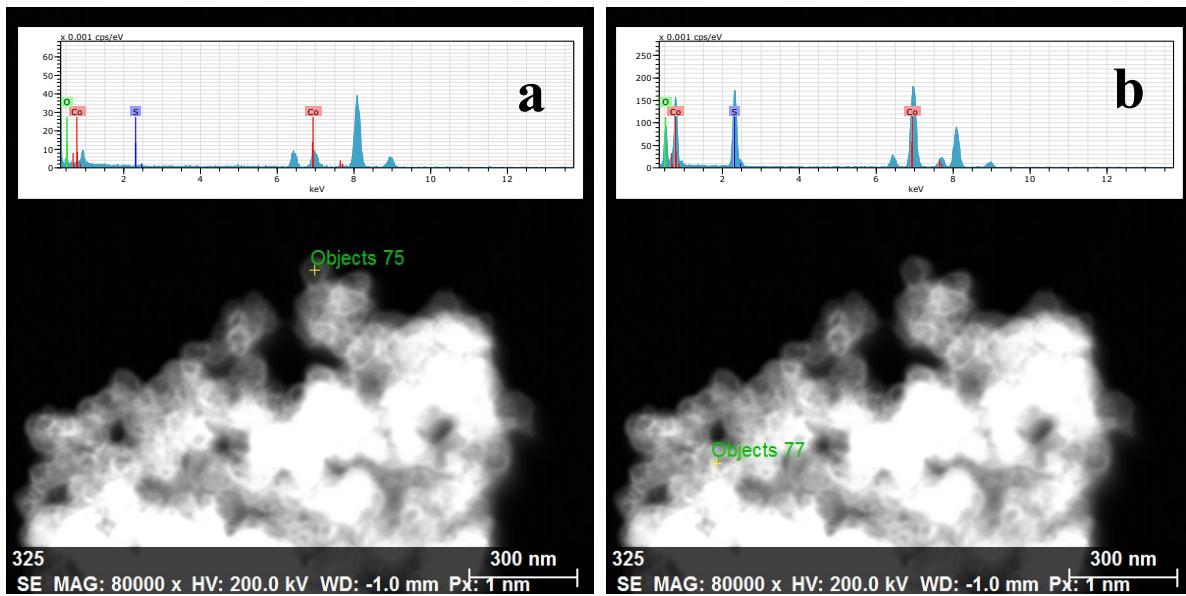


Fig. S6: ESD point mapping image taken in different points (a) at the centre of a particle, (b) at the periphery, showing hollow core@shell structure of $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$.

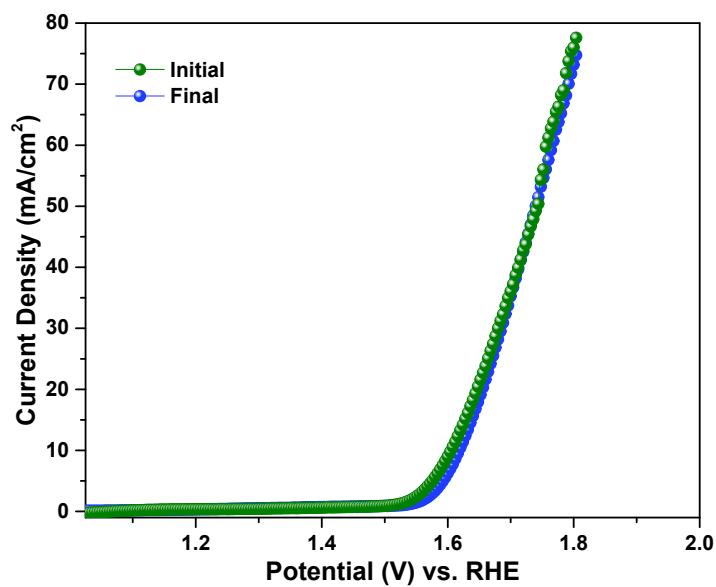


Fig. S7: Stability of $\text{Co}_3\text{O}_4@\text{Co}_3\text{S}_4$ -2 via consecutive runs up to 1000 cycles.

Table S1: Data Comparison for electrocatalytic activity of different reported catalysts, closely relating to the present work.

| S.N. | Electrocatalyst | Electrode reaction | Electrolyte | Potential (RHE) at 10 mA/cm ² Current | Tafel slope | Substrate | Reference |
|------|--|--------------------|--------------------------------------|--|-------------|-----------|-----------|
| 1 | Co _{1-x} S–Graphene Hybrid | ORR | 0.1 M KOH | 3.8 at 0.7 V | - | GC | 17 |
| 2 | Co ₃ O ₄ @Co ₃ S ₄ nanoarrays | OER and HER | 1 M KOH | 1.53 V | - | Ni-Foam | 22 |
| 3 | NiCo ₂ S ₄ @N/S-rGO | OER | 0.1 M KOH | 0.47 V | - | GC | R1 |
| 4 | CoO _x @CN | OER | 1 M KOH | 0.26 V | 115 | GC | R2 |
| 5 | Co ₃ O ₄ C-NA | OER | 0.1 M KOH | 1.52 V | 70 | Cu-foil | R3 |
| 6 | Mn ₃ O ₄ /CoSe ₂ | OER | 0.1 M KOH | 0.45 V | 49 | GC | R4 |
| 7 | Co ₃ O ₄ /N-doped-graphene | OER | 1 M KOH | 1.54 V | 67 | Ni-foam | 13 |
| 8 | N-doped graphene-CoO | OER | 1 M KOH | 1.57 V | 71 | GC | R5 |
| 9 | CeO ₂ /CoSe ₂ | OER | 0.1 M KOH | 0.288 V | 44 | GC | R6 |
| 10 | WS ₂ /WO ₂ | HER | 0.5 M H ₂ SO ₄ | -90 mV | 63 | GC | 19 |
| 11 | MoS ₂ /MoO ₂ | HER | 0.5 M H ₂ SO ₄ | -120 mV | 51 | GC | 19 |
| 12 | Co ₃ O ₄ @Ni ₃ S ₂ /NF composite | OER | 1 M KOH | 260@20 mA /cm ² | 171 | Ni-foam | 45 |
| 13 | Co ₃ O ₄ @Co ₃ S ₄ | OER | 0.5 M KOH | 1.606 V | 80 | GC | This work |

References:

- R1. Q. Liu, J. Jin and J. Zhang, *ACS Appl. Mater. Interfaces*, 2013, **5**, 5002–5008.
- R2. H. Jin, J. Wang, D. Su, Z. Wei, Z. Pang and Y. Wang, *J. Am. Chem. Soc.*, 2015, **137**, 2688–2694.
- R3. T. Y. Ma, S. Dai, M. Jaroniec and S. Z. Qiao, *J. Am. Chem. Soc.*, 2014, **136**, 13925–13931.

- R4. M. R. Gao, Y. F. Xu, J. Jiang, Y. R. Zheng and S. H. Yu, *J. Am. Chem. Soc.*, 2012, **134**, 2930-2933.
- R5. S. Mao, Z. Wen, T. Huang, Y. Hou and J. Chen, *Energy Environ. Sci.*, 2014, **7**, 609-616.
- R6. Y. R. Zheng, M. R. Gao, Q. Gao, H. H. Li, J. Xu, Z. Y. Wu and S. H. Yu, *Small*, 2015, **11**, 182-188.