

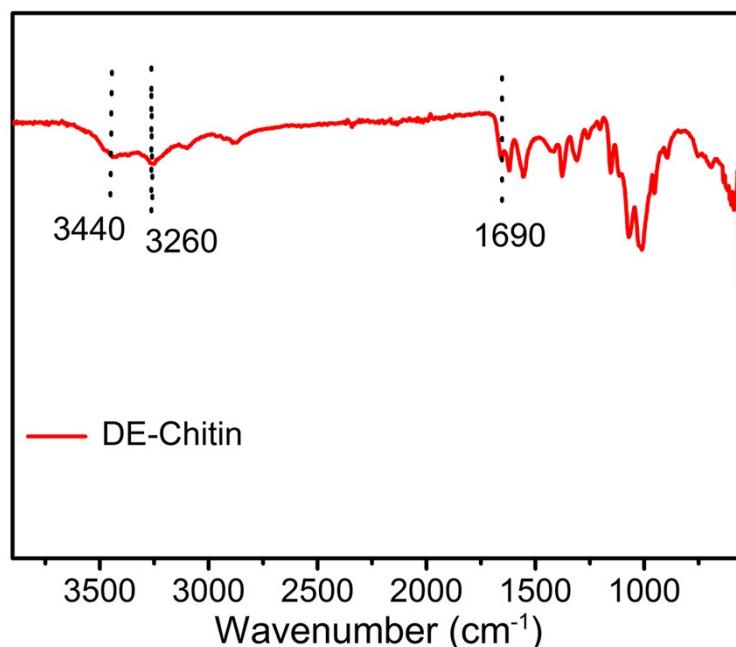
## Supporting information

### ***In situ synthesis of small Pt nanoparticles on chitin aerogel derived N doped ultra-thin carbon nanofibers for superior hydrogen evolution catalysis***

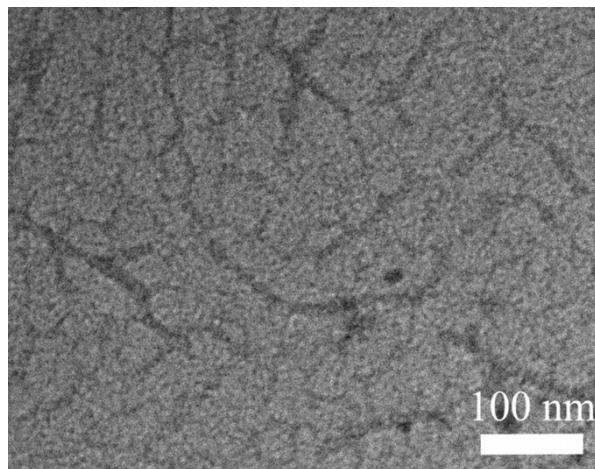
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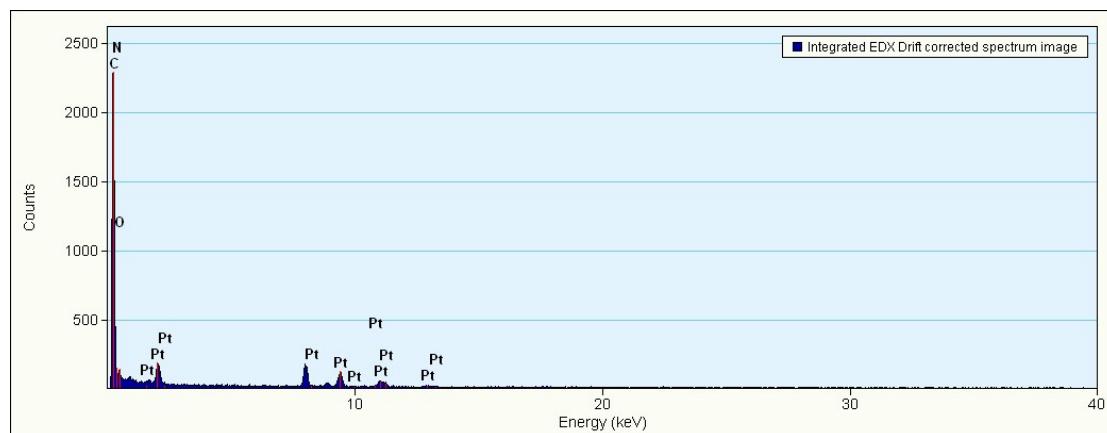
E-mail: zhysw@jiangnan.edu.cn; [du@jiangnan.edu.cn](mailto:du@jiangnan.edu.cn)



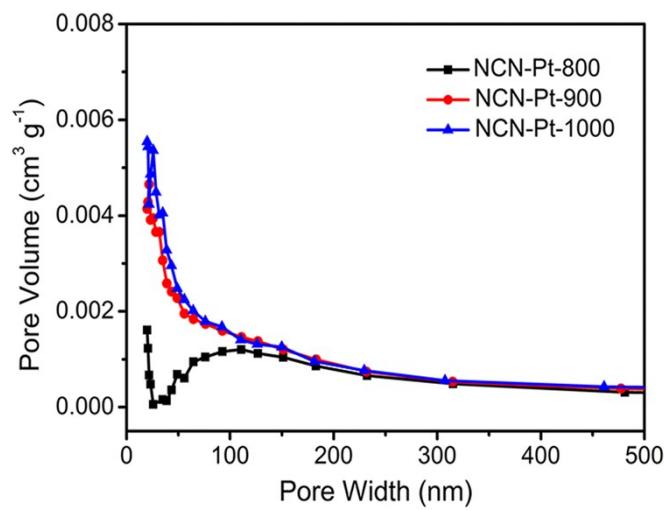
**Figure S1.** FT-IR spectra of the DE-chitin.



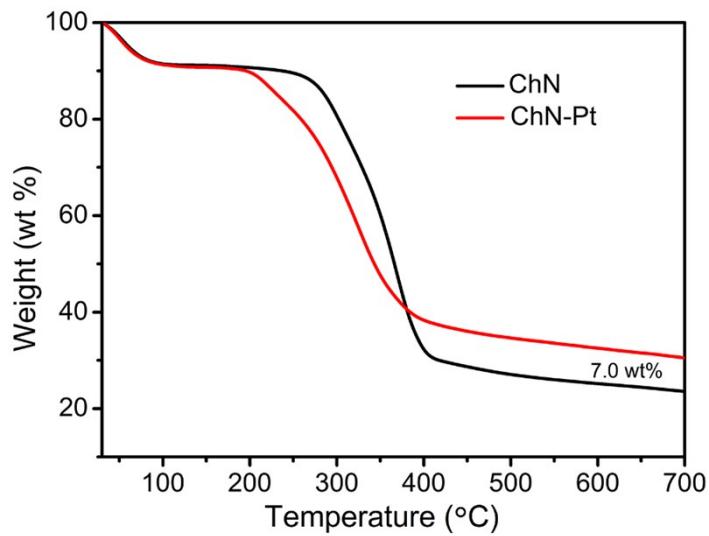
**Figure S2.** TEM image of chitin nanofibers.



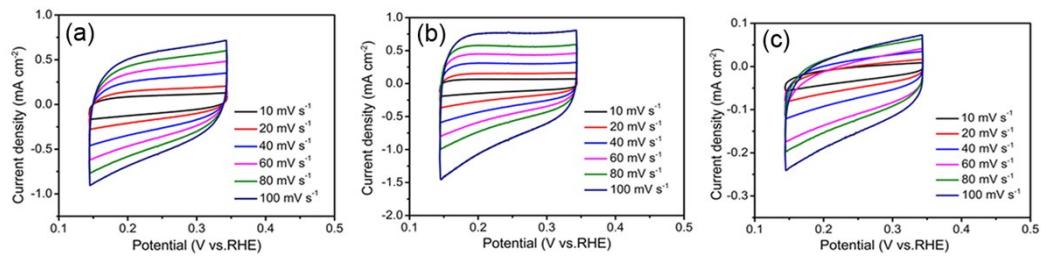
**Figure S3.** The EDX spectra of the NCN-Pt.



**Figure S4.** The corresponding BET adsorption pore size distribution of the NCN-Pt prepared at different temperatures.



**Figure S5.** TGA curves of ChN and ChN-Pt.



**Figure S6.** CV curves for NCN-Pt-800, NCN-Pt-900 and NCN-Pt-1000 with different rates from 10 to  $100 \text{ mV s}^{-1}$  in the potential range of 0.144 to 0.344 V

**Table S1. HER performance of Pt-based catalysts.**

Sample	Electrolyte s	Pt loading ( $\mu\text{g cm}^{-2}$ )	Overpotential (mV @ 10 mA $\text{cm}^{-2}$ )	Tafel slope (mV $\text{dec}^{-1}$ )	Reference
NCN-Pt-900	0.5 M $\text{H}_2\text{SO}_4$	8.3	34	39	This work
10Pt@HN-BC	0.5 M $\text{H}_2\text{SO}_4$	12	47	35	<sup>1</sup>
Pt @N-doped hollow porous carbon	0.1 M $\text{HClO}_4$	2	57	27	<sup>2</sup>
SWNT	0.5 M $\text{H}_2\text{SO}_4$	19.4	27	38	<sup>3</sup>
WC @ C @ Pt	0.5 M $\text{H}_2\text{SO}_4$ 1 M KOH	70.7	30	26	<sup>4</sup>
YS-Pt-CoP	1 M KOH	11	48	54	<sup>5</sup>
Pt <sub>1</sub> /NPC	0.5 M $\text{H}_2\text{SO}_4$	3.8	25	28	<sup>6</sup>
MoS <sub>2</sub> @Pt-3	0.5 M $\text{H}_2\text{SO}_4$	17	70	36	<sup>7</sup>
Pt/BCF	0.5 M $\text{H}_2\text{SO}_4$	7	135	70	<sup>8</sup>
Pt <sub>66</sub> Ni <sub>34</sub> NFs	0.5 M $\text{H}_2\text{SO}_4$	146	43	33	<sup>9</sup>
Pt <sub>2.6</sub> Co <sub>1</sub> NFs	0.5 M KOH	152	40	42	<sup>10</sup>

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