

## Electronic supplementary information

### Large magnetocaloric effect and remarkable single-molecule-magnet behavior in triangle-assembled Ln<sup>III</sup><sub>6</sub> clusters

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#### Experimental Section

##### Synthesis of Ln(acac)<sub>3</sub>·2H<sub>2</sub>O

The synthesis of Ln(acac)<sub>3</sub>·2H<sub>2</sub>O (Ln<sup>III</sup> = Gd and Dy) is according the previous literature.<sup>1</sup> A solution of Ln(NO<sub>3</sub>)<sub>3</sub>·6H<sub>2</sub>O (Ln<sup>III</sup> = Gd and Dy, 12 mmol) in distilled water (18 mL) was stirred for 30 min at room temperature, then 36 mmol acetylacetonone (Hacac) was added to it, and the mixture was adjusted the pH to 7.0 by using NaOH solution (1 mol/L). Finally, the mixture was stirred for about 6.0 h. The mixture stood for about 3.0 hours, a precipitate formed was collected through filtration and washed with distilled water, vacuum drying for 2 day, obtaining the Ln(acac)<sub>3</sub>·2H<sub>2</sub>O (Ln<sup>III</sup> = Gd and Dy) crude product.

##### Synthesis of H<sub>3</sub>L

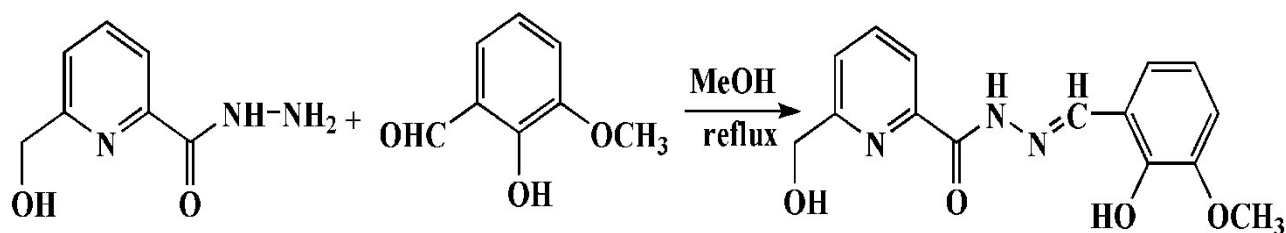
The synthesis of H<sub>3</sub>L is according the reported reference (Scheme. S1).<sup>2</sup> A 30 mL methanolic solution of 3-methoxysalicylaldehyde (10 mmol) and 6-(hydroxymethyl) picolinohydrazide (10 mmol) was stirred at room temperature for about 30 min. After that, the reaction mixture was heated under reflux overnight, then cooled to room temperature. A precipitate formed was collected through filtration and washed with slight MeOH. Yield: 1.9 g (63%). Anal. Calcd. for C<sub>15</sub>H<sub>16</sub>N<sub>3</sub>O<sub>4</sub>: C, 59.60; H, 5.30; N, 13.91. Found: C, 59.64; H, 5.26; N, 13.87.

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Scheme S1. Detailed outline of the synthesis of the ligand ( $H_3L$ ).

Table S1 Selected bond lengths (Å) and angles ( $^\circ$ ) for cluster **1**<sup>a</sup>

<b>Bond lengths</b>			
Gd(2)-O(4)	2.359(5)	Gd(2)-O(14)	2.767(6)
Gd(2)-O(11)	2.316(5)	Gd(2)-O(10)#1	2.461(5)
Gd(2)-O(10)	2.599(6)	Gd(2)-O(15)	2.382(6)
Gd(2)-N(3)	2.566(7)	Gd(2)-O(3)	2.409(5)
Gd(2)-N(5)	2.579(7)	Gd(1)-O(2)	2.421(5)
Gd(1)-O(5)	2.376(6)	Gd(1)-O(14)#1	2.324(5)
Gd(1)-O(11)#1	2.376(5)	Gd(1)-O(10)	2.468(5)
Gd(1)-O(6)	2.363(6)	Gd(1)-O(3)	2.369(6)
Gd(1)-N(1)	2.497(7)	Gd(3)-O(2)#1	2.399(6)
Gd(3)-O(4)	2.278(5)	Gd(3)-O(14)	2.397(6)
Gd(3)-O(11)	2.562(5)	Gd(3)-O(7)	2.378(6)
Gd(3)-O(1)#1	2.602(6)	Gd(3)-O(8)	2.376(6)
Gd(3)-O(13)	2.627(7)	Gd(3)-N(4)	2.579(8)
<b>Bond angles</b>			
O(4)-Gd(2)-O(14)	67.19(18)	O(4)-Gd(2)-O(10)	137.73(19)
O(4)-Gd(2)-O(10)#1	133.87(18)	O(4)-Gd(2)-O(15)	87.8(2)
O(4)-Gd(2)-O(3)	127.22(18)	O(11)-Gd(2)-O(4)	74.57(19)
O(11)-Gd(2)-O(14)	58.81(18)	O(11)-Gd(2)-O(10)#1	74.20(18)
O(11)-Gd(2)-O(10)	77.74(18)	O(11)-Gd(2)-O(15)	134.2(2)
O(11)-Gd(2)-O(3)	146.44(19)	O(10)-Gd(2)-O(14)	121.96(16)
O(10)#1-Gd(2)-O(14)	67.78(17)	O(10)#1-Gd(2)-O(10)	64.5(2)
O(15)-Gd(2)-O(14)	75.42(19)	O(15)-Gd(2)-O(10)#1	90.4(2)
O(15)-Gd(2)-O(10)	134.0(2)	O(15)-Gd(2)-O(3)	76.8(2)
O(3)-Gd(2)-O(14)	147.99(18)	O(3)-Gd(2)-O(10)#1	96.90(18)
O(3)-Gd(2)-O(10)	69.40(17)	O(2)-Gd(1)-O(10)	138.76(18)
O(5)-Gd(1)-O(2)	73.5(2)	O(5)-Gd(1)-O(11)#1	131.06(19)
O(5)-Gd(1)-O(10)	132.8(2)	O(14)#1-Gd(1)-O(2)	80.00(19)
O(14)#1-Gd(1)-O(5)	81.4(2)	O(14)#1-Gd(1)-O(11)#1	65.00(19)
O(14)#1-Gd(1)-O(10)	75.20(19)	O(14)#1-Gd(1)-O(6)	116.4(2)
O(14)#1-Gd(1)-O(3)	144.35(19)	O(11)#1-Gd(1)-O(2)	66.72(18)
O(11)#1-Gd(1)-O(10)	73.04(18)	O(6)-Gd(1)-O(2)	138.12(19)
O(6)-Gd(1)-O(10)	82.89(18)	O(6)-Gd(1)-O(3)	73.8(2)
O(3)-Gd(1)-O(2)	116.4(2)	O(3)-Gd(1)-O(5)	132.38(19)

O(3)-Gd(1)-O(11)#1	91.47(19)	O(3)-Gd(1)-O(10)	72.32(18)
O(2)#1-Gd(3)-O(11)	64.14(17)	O(2)#1-Gd(3)-O(1)#1	63.26(18)
O(2)#1-Gd(3)-O(13)	67.9(2)	O(4)-Gd(3)-O(2)#1	135.16(19)
O(4)-Gd(3)-O(14)	75.28(19)	O(4)-Gd(3)-O(11)	71.34(18)
O(4)-Gd(3)-O(7)	80.3(2)	O(4)-Gd(3)-O(1)#1	151.3(2)
O(4)-Gd(3)-O(8)	84.6(2)	O(4)-Gd(3)-O(13)	132.9(2)
O(14)-Gd(3)-O(2)#1	79.00(18)	O(14)-Gd(3)-O(11)	61.13(18)
O(14)-Gd(3)-O(1)#1	133.37(19)	O(14)-Gd(3)-O(13)	70.7(2)
O(11)-Gd(3)-O(1)#1	117.57(18)	O(11)-Gd(3)-O(13)	116.22(19)
O(7)-Gd(3)-O(2)#1	97.6(2)	O(7)-Gd(3)-O(14)	140.1(2)
O(7)-Gd(3)-O(11)	81.47(19)	O(7)-Gd(3)-O(1)#1	74.6(2)
O(7)-Gd(3)-O(13)	144.9(2)	O(1)#1-Gd(3)-O(13)	70.3(2)
O(8)-Gd(3)-O(2)#1	138.14(19)	O(8)-Gd(3)-O(14)	133.8(2)
O(8)-Gd(3)-O(11)	147.5(2)	O(8)-Gd(3)-O(7)	73.1(2)
O(8)-Gd(3)-O(1)#1	75.01(19)	O(8)-Gd(3)-O(13)	96.1(2)

<sup>a</sup> Symmetry transformations used to generate equivalent atoms: #1 -x+1, -y+1, -z+2.

**Table S2** Selected bond lengths (Å) and angles (°) for cluster **2**<sup>a</sup>

<b>Bond lengths</b>			
Dy(1)-O(3)#1	2.417(6)	Dy(1)-O(10)	2.612(6)
Dy(1)-O(10)#1	2.461(6)	Dy(1)-O(11)	2.322(6)
Dy(1)-O(4)#1	2.358(6)	Dy(1)-O(14)	2.769(6)
Dy(1)-N(4)	2.582(8)	Dy(1)-O(15)	2.363(7)
Dy(1)-N(3)#1	2.569(7)	Dy(2)-O(3)	2.359(6)
Dy(2)-O(10)#1	2.463(6)	Dy(2)-O(11)	2.378(6)
Dy(2)-O(5)	2.372(6)	Dy(2)-O(2)	2.419(6)
Dy(2)-O(6)	2.377(6)	Dy(2)-O(14)	2.318(6)
Dy(2)-N(1)	2.503(7)	Dy(3)-O(11)	2.559(6)
Dy(3)-O(4)#1	2.275(6)	Dy(3)-O(2)	2.407(6)
Dy(3)-O(14)	2.402(6)	Dy(3)-O(7)	2.379(7)
Dy(3)-O(1)	2.622(7)	Dy(3)-O(8)	2.362(7)
Dy(3)-O(13)	2.633(8)	Dy(3)-N(5)	2.558(10)
<b>Bond angles</b>			
O(3)#1-Dy(1)-O(10)	69.1(2)	O(3)#1-Dy(1)-O(14)	148.2(2)
O(10)#1-Dy(1)-O(10)	64.8(2)	O(10)-Dy(1)-O(14)	121.91(17)
O(10)#1-Dy(1)-O(14)	67.41(19)	O(11)-Dy(1)-O(3)#1	146.2(2)
O(11)-Dy(1)-O(10)#1	74.2(2)	O(11)-Dy(1)-O(10)	77.8(2)
O(11)-Dy(1)-O(4)#1	74.5(2)	O(11)-Dy(1)-O(14)	58.8(2)
O(11)-Dy(1)-O(15)	134.2(2)	O(4)#1-Dy(1)-O(3)#1	127.4(2)
O(4)#1-Dy(1)-O(10)#1	133.6(2)	O(4)#1-Dy(1)-O(10)	137.8(2)
O(4)#1-Dy(1)-O(14)	67.2(2)	O(4)#1-Dy(1)-O(15)	87.1(3)
O(15)-Dy(1)-O(3)#1	77.3(2)	O(15)-Dy(1)-O(10)	134.6(2)
O(15)-Dy(1)-O(10)#1	90.9(2)	O(15)-Dy(1)-O(14)	75.5(2)

O(3)-Dy(2)-O(10)#1	72.6(2)	O(3)-Dy(2)-O(11)	91.7(2)
O(3)-Dy(2)-O(5)	73.8(2)	O(3)-Dy(2)-O(2)	116.4(2)
O(3)-Dy(2)-O(6)	132.2(2)	O(11)-Dy(2)-O(10)#1	73.22(19)
O(11)-Dy(2)-O(2)	66.7(2)	O(5)-Dy(2)-O(10)#1	83.0(2)
O(5)-Dy(2)-O(11)	155.0(2)	O(5)-Dy(2)-O(2)	137.9(2)
O(5)-Dy(2)-O(6)	71.6(2)	O(2)-Dy(2)-O(10)#1	138.9(2)
O(6)-Dy(2)-O(10)#1	132.8(2)	O(6)-Dy(2)-O(11)	130.9(2)
O(6)-Dy(2)-O(2)	73.2(2)	O(14)-Dy(2)-O(3)	144.5(2)
O(14)-Dy(2)-O(10)#1	75.0(2)	O(14)-Dy(2)-O(11)	65.1(2)
O(14)-Dy(2)-O(5)	116.2(2)	O(14)-Dy(2)-O(2)	80.2(2)
O(14)-Dy(2)-O(6)	81.4(2)	O(11)-Dy(3)-O(1)	117.3(2)
O(11)-Dy(3)-O(13)	116.2(2)	O(4)#1-Dy(3)-O(11)	71.4(2)
O(4)#1-Dy(3)-O(2)	135.2(2)	O(4)#1-Dy(3)-O(14)	75.3(2)
O(4)#1-Dy(3)-O(7)	80.6(2)	O(4)#1-Dy(3)-O(1)	151.8(2)
O(4)#1-Dy(3)-O(8)	84.4(2)	O(4)#1-Dy(3)-O(13)	132.5(2)
O(2)-Dy(3)-O(11)	64.11(19)	O(2)-Dy(3)-O(1)	62.9(2)
O(2)-Dy(3)-O(13)	68.0(2)	O(14)-Dy(3)-O(11)	61.2(2)
O(14)-Dy(3)-O(2)	78.8(2)	O(14)-Dy(3)-O(1)	132.9(2)
O(14)-Dy(3)-O(13)	70.5(2)	O(7)-Dy(3)-O(11)	81.4(2)
O(7)-Dy(3)-O(2)	97.6(2)	O(7)-Dy(3)-O(14)	140.2(2)
O(7)-Dy(3)-O(1)	74.8(2)	O(7)-Dy(3)-O(13)	145.0(2)
O(1)-Dy(3)-O(13)	70.3(2)	O(8)-Dy(3)-O(11)	147.3(2)
O(8)-Dy(3)-O(2)	138.3(2)	O(8)-Dy(3)-O(14)	133.9(2)
O(8)-Dy(3)-O(7)	73.0(2)	O(8)-Dy(3)-O(1)	75.5(2)
O(8)-Dy(3)-O(13)	96.3(2)		

<sup>a</sup> Symmetry transformations used to generate equivalent atoms: #1 -x+2, -y+1, -z.

**Table S3** The Gd<sup>III</sup> geometry analysis by SHAPE 2.0 for cluster 1.

Cluster 3	$D_{4d}$ SAPR	$D_{2d}$ TDD	$C_{2v}$ JBTPR	$C_{2v}$ BTPR	$D_{2d}$ JSD
eight-coordinated Gd1 <sup>III</sup>	1.487	3.197	3.173	2.769	5.001
	$C_{4v}$ JCSAPR	$C_{4v}$ CSAPR	$D_{3h}$ JTCTPR	$D_{3h}$ TCTPR	$C_s$ MFF
nine-coordinated Gd2 <sup>III</sup>	2.336	2.179	2.342	2.515	2.340
nine-coordinated Gd3 <sup>III</sup>	2.039	1.460	2.308	1.335	1.384

**SAPR-8** = Square antiprism; **TDD-8** = Triangular dodecahedron; **JBTPR-8** = Biaugmented trigonal prism J50; **BTPR-8** = Biaugmented trigonal prism; **JSD-8** = Snub diphenoid J84.

**JCSAPR-9** = Capped square antiprism J10; **CSAPR-9** = Spherical capped square antiprism;

**JTCTPR-9** = Tricapped trigonal prism J51; **TCTPR-9** = Spherical tricapped trigonal prism;

**MFF-9** = Muffin.

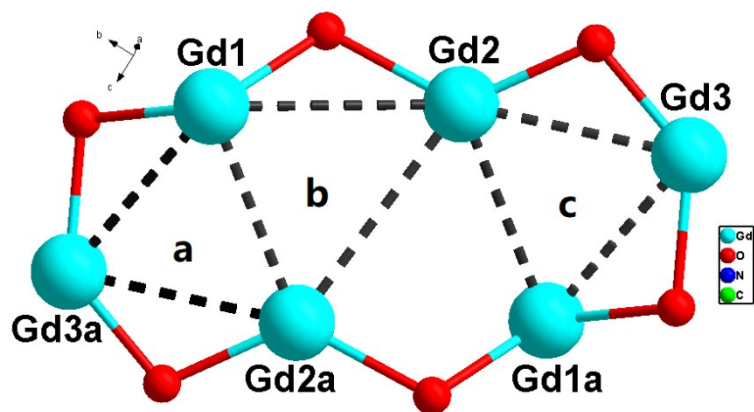


Fig. S1 The Gd<sub>6</sub> core of 1 highlighting the triangular units (dashed lines).

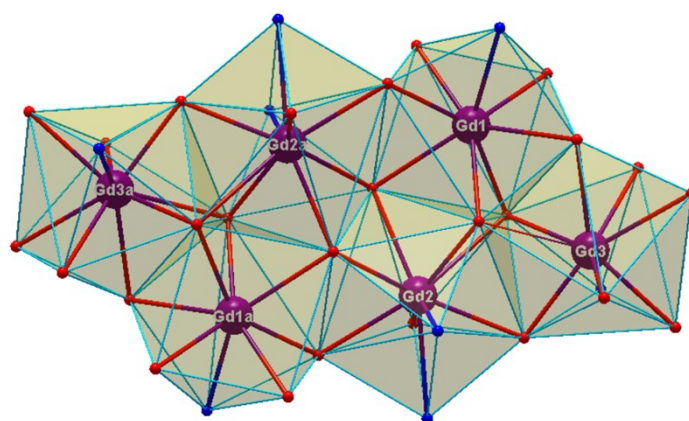


Fig. S2 The geometric polyhedra of Gd<sup>III</sup> ions observed in cluster 1.

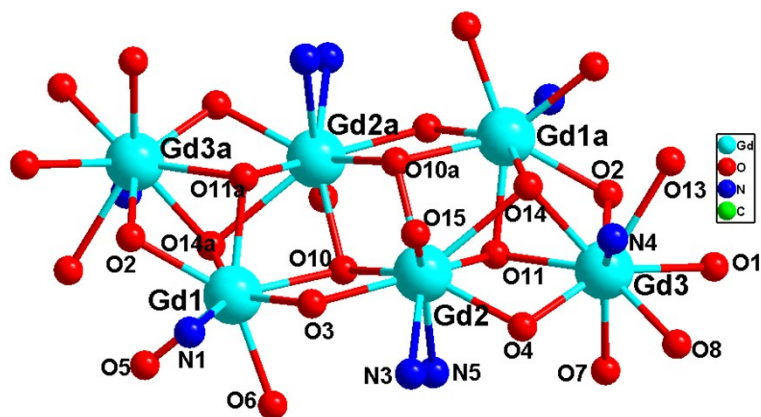


Fig. S3 Coordination atoms labels of the Gd<sub>6</sub> core in 1.

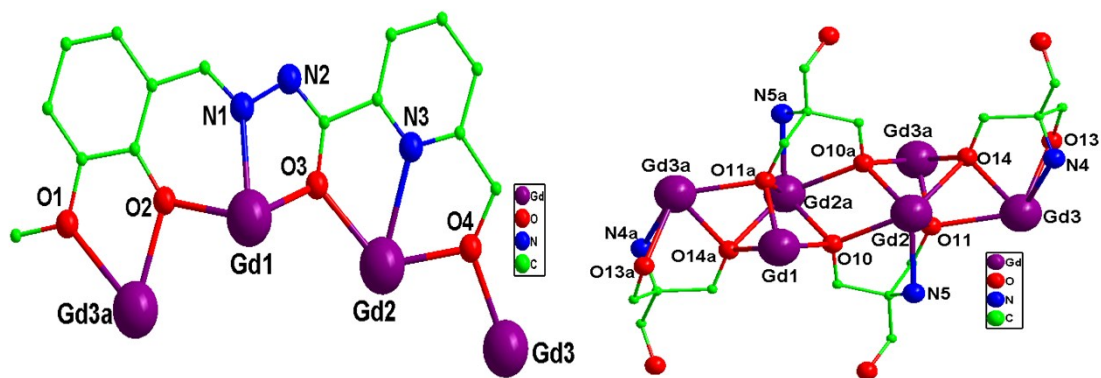


Fig. S4 Binding modes of the ligand H<sub>3</sub>L (left) and THAM (right) in cluster 1.

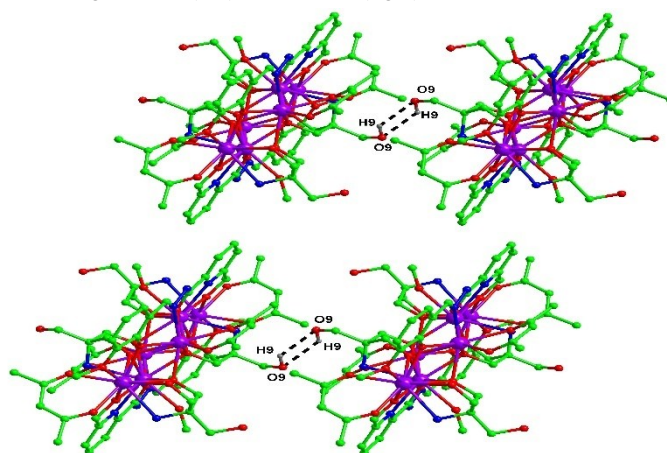


Fig. S5 The hydrogen bond connected mode and crystal packing in cluster 1.

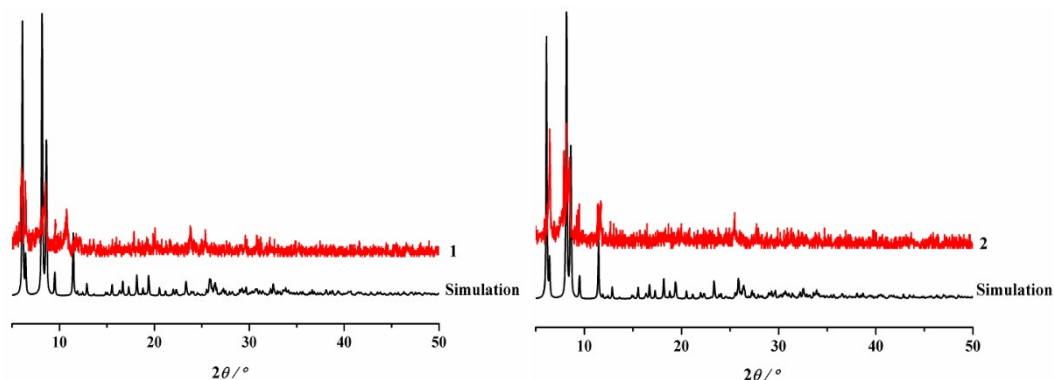


Fig. S6 PXRD patterns for clusters 1 and 2.

## References

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- 2 (a) V. Chandrasekhar, S. Hossain, S. Das, S. Biswas and J. P. Sutter, *Inorg. Chem.*, 2013, **52**, 6346; (b) W. M. Wang, X. Z. Li, L. Zhang, J. L. Chen, J. H. Wang, Z. L. Wu and J. Z. Cui, *New J. Chem.*, 2019, **43**, 7419.