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### Electronic Supplementary Information for:

#### Enzymatically Triggered Chromogenic Cross-Linking Agents Under Physiological Conditions

Hikaru Fujita, Jinghuai Dou, Nobuyuki Matsumoto, Zhiyuan Wu and Jonathan S. Lindsey\*

Department of Chemistry, North Carolina State University, Raleigh, NC 27695-8204, USA

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## (1) X-ray single crystal structures

Summary of the crystal data for	1-acetyl-4-bromo-5-hydroxy-1H-indol-3-yl 2,3,4,6-tetra-O-
acetyl-β-D-glucopyranoside (17)	

CCDC registry	1902489
Formula	C <sub>26</sub> H <sub>30</sub> BrNO <sub>12.50</sub>
Formula Weight (g/mol)	636.42
Crystal Dimensions (mm)	0.130×0.229×0.321
Crystal System	orthorhombic
Space Group	P 21 21 21
Temperature (K)	100(2)
<i>a</i> (Å)	13.3401(8)
b (Å)	14.3756(9)
c (Å)	28.9203(19)
$\alpha$ (°)	90
$\beta$ (°)	90
γ(°)	90
$V(Å^3)$	5546.1(6)
Number of reflections to determine final unit cell	9951
Min and Max 2θ for cell determination (°)	4.398, 51.23
Ζ	8
F(000)	2624
$\rho$ (g/cm <sup>3</sup> )	1.524
λ (Å, Μο Κα)	0.71073
$\mu (\mathrm{mm}^{-1})$	1.550
Max 2θ for data collection (°)	56.66
Measured fraction of data	0.999
Number of reflections measured	113542
Unique reflections measured	13811
R <sub>merge</sub>	6.44%
Number of parameters in least-squares	788
R <sub>1</sub>	0.0324
wR <sub>2</sub>	0.0684
R <sub>1</sub> (all data)	0.0423
$wR_2$ (all data)	0.0712

Summary of the crystal data for 1-acetyl-4,6-dibromo-5-hydroxy-1*H*-indol-3-yl 2,3,4,6-tetra-*O*-acetyl-β-D-glucopyranoside (18)

CCDC registry	1902488
Formula	C <sub>25.50</sub> H <sub>28</sub> Br <sub>2</sub> NO <sub>12.50</sub>
Formula Weight (g/mol)	708.31
Crystal Dimensions (mm)	0.072×0.290×0.302
Crystal System	orthorhombic
Space Group	P 21 21 21
Temperature (K)	100
<i>a</i> (Å)	13.4601(5)
b (Å)	14.6763(5)
<i>c</i> (Å)	12.7603 (4)
α (°)	90
$\beta$ (°)	90
γ (°)	90
$V(Å^3)$	5670.5(3)
Number of reflections to determine final unit cell	9915
Min and Max 2θ for cell determination (°)	4.991, 62.78
Ζ	8
<i>F</i> (000)	2864
$\rho$ (g/cm <sup>3</sup> )	1.659
λ (Å, Μο Κα)	0.71073
$\mu (\mathrm{mm}^{-1})$	2.926
Max 2θ for data collection (°)	66.38
Measured fraction of data	0.999
Number of reflections measured	116273
Unique reflections measured	21707
R <sub>merge</sub>	4.48%
Number of parameters in least-squares	757
R <sub>1</sub>	0.0297
wR <sub>2</sub>	0.0604
R <sub>1</sub> (all data)	0.0380
wR <sub>2</sub> (all data)	0.0627

#### (2) Time course of indigogenic reactions selected from Table 1

Figure S1 shows the time course of the indigogenic reaction of known indoxyl-glucoside 1 or 45 with  $\beta$ -glucosidase from almonds. The formation of indigoid products was monitored by absorption spectroscopy over the course of 16 h due to the slow reaction rate.



**Figure S1.** Time course of absorption spectra of indigogenic reactions with  $\beta$ -glucosidase from almonds (1 unit/mL) in 40 mM acetate buffer (pH 5.0, containing 5% DMF) at 37 °C. (A) The reaction of **1** (1 mM). After incubation, the reaction mixture was diluted four times with DMF and used for absorption spectroscopy. (B) The reaction of **45** (1 mM). After incubation, the reaction mixture was diluted ten times with DMF and used for absorption spectroscopy.

#### (3) Indigogenic reactions examined in the presence of oxidants or with tritosomes

To improve the yield and reaction rate of the indigogenic reaction, addition of oxidizing agents was attempted (Figure S2). However,  $K_2Fe(CN)_6/K_3Fe(CN)_6$ , peroxidase/H<sub>2</sub>O<sub>2</sub>, 2-(4-iodophenyl)-3-(4-nitrophenyl)-5-phenyl tetrazolium chloride (S2, Chart S1), or phenazine methosulfate (S3) did not increase the yield of the indigogenic reaction with 1 (Figure S2A). As shown in Figure S2B, the reaction rate of the indigogenic reaction with 45 was not increased by inclusion of  $K_2Fe(CN)_6/K_3Fe(CN)_6$  or peroxidase/H<sub>2</sub>O<sub>2</sub>.



**Figure S2.** Indigogenic reactions in the presence of oxidizing agents. (A) Absorption spectra from the reaction with 1 (1 mM) without additives or with K<sub>2</sub>Fe(CN)<sub>6</sub> (3 mM)/K<sub>3</sub>Fe(CN)<sub>6</sub> (3 mM), peroxidase (1 unit/mL)/H<sub>2</sub>O<sub>2</sub> (5 mM), **S2** (3 mM), or **S3** (3 mM). After incubation, the reaction mixture was diluted with DMF 4 times (for no additive; K<sub>2</sub>Fe(CN)<sub>6</sub>)/K<sub>3</sub>Fe(CN)<sub>6</sub>; or peroxidase), 12 times (**S2**) or 20 times (**S3**) and used for absorption spectroscopy. (B) Time course of the reaction with **45** (1 mM) without additives, with K<sub>2</sub>Fe(CN)<sub>6</sub> (3 mM)/K<sub>3</sub>Fe(CN)<sub>6</sub> (3 mM), and with peroxidase (1 unit/mL)/H<sub>2</sub>O<sub>2</sub> (5 mM). Yields were estimated from absorption spectroscopy with  $\varepsilon = 2.00 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$  reported for 5,5'-dibromo-4,4'-dichloroindigo.<sup>40</sup>



Chart S1. Structures of additives for indigogenic reactions.

A set of indigogenic reactions was examined where oxidants were included along with the  $\beta$ -glucosidase.



 Table S1. Indigogenic reactions of indoxyl derivatives.



<sup>*a*</sup>The yield was estimated by absorption spectroscopy with  $\varepsilon = 1.27 \times 10^4 \text{ M}^{-1} \text{cm}^{-1}$  (DMF/H<sub>2</sub>O = 2:1) measured for the parent indigo 4 unless otherwise noted. A yield of <1% implies no color was observed by visual inspection and no absorption was observed spectroscopically. A yield of <5% implies a faint blue color was observed by visual inspection but the absorption was too weak for reliable spectroscopic determination. <sup>*b*</sup>A mixture of the indoxyl (1 mM), **S2** (3 mM), and β-glucosidase from almonds (1 unit/mL) in 40 mM acetate buffer (pH 5.0, containing 5% DMF) was incubated at 37 °C for 16–19 h. The yield of indigoid compounds was roughly evaluated by

absorption spectroscopy as products from **S2** have a shoulder absorption in the region 600–650 nm. <sup>c</sup>A mixture of the indoxyl (1 mM) and tritosomes (0.13 mg protein/mL) in 20 mM acetate buffer (pH 5.0, containing 5% DMF) was incubated at 37 °C for 16–19 h. <sup>d</sup>A mixture of the indoxyl (1 mM) and tritosomes (0.13 mg protein/mL) in 20 mM acetate buffer (pH 5.0, containing 5% DMF) in the presence of 1x catabolic buffer was incubated at 37 °C for 16–19 h. <sup>e</sup>The yield was estimated from absorption spectroscopy with  $\varepsilon = 2.00 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$  reported for 5,5'-dibromo-4,4'-dichloroindigo.<sup>40</sup> <sup>f</sup>Not conducted. <sup>g</sup>The yield was estimated from absorption spectroscopy with  $\varepsilon = 2.6 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$  (DMF/H<sub>2</sub>O = 2:1) measured for 46.

Figure S3 shows the effect of the nonionic surfactant tyloxapol (Triton WR1339) over a 1000-fold concentration range on the indigogenic reaction of **45** with  $\beta$ -glucosidase from almonds. The effect was measurable but rather small over the very large range of concentration examined.



Figure S3. Effect of tyloxapol on the indigogenic reaction of 45 with  $\beta$ -glucosidase from almonds.

#### (4) pH Profile of β-glucosidase from *Agrobacterium*.

Following a reported procedure<sup>43</sup> with slight modification, a solution of 4-nitrophenyl  $\beta$ -D-glucopyranoside (2 mM) in 45 mM phosphate buffers (pH 4.0, 5.0, 6.0, 7.0, 8.0, or 9.0) was incubated at 37 °C for 10 min. A solution of  $\beta$ -glucosidase from *Agrobacterium* in NaPi-NaCl buffer (10 µL, 1 µM, pH 7.0, 10 mM NaPi and 50 mM NaCl) was added. The reaction mixture was incubated at 37 °C for 5 min and then quenched by the addition of 1.2 M aqueous NaOH (120, 100, 90, 60, 30, or 20 µL for the reaction at pH 4.0, 5.0, 6.0, 7.0, 8.0, or 9.0, respectively) at room temperature. The solution was diluted two times with H<sub>2</sub>O. The absorption at 400 nm was measured to calculate the activity of the enzyme (Figure S4).



Figure S4. pH profile of  $\beta$ -glucosidase from *Agrobacterium*.

#### (5) HPLC data from oligomerization studies



Figure S5. Calibration curve for SEC analysis.



Figure S6. Analytical SEC traces for the supernatant and precipitate samples from 50 µM of 49.

#### (6) Mass spectrometry of the oligomer supernatant

ESI-MS analysis was carried out of the supernatant in the negative-ion mode using MeCN/H<sub>2</sub>O (0.1% TFA). The spectrum is shown in Figure S7. Peaks consistent with the monomer and the dimer were detected. The putative monomer and dimer structures are shown in Figure S8.



Figure S7. ESI-MS spectrum (negative-ion mode) of the supernatant derived from 300  $\mu$ M of 49.



 $(C_{41}H_{49}Br_4N_7O_{14}S)_n$ , *m/z*: 1214.98 when n = 1

Figure S8. Cyclic monomer and dimer structures where n = 1 and 2, respectively.

MALDI-MS analysis (positive-ion mode) was carried out of the supernatant derived from 300  $\mu$ M of **49** using CHCA as a matrix and calibration with peptide standards. The spectrum is shown in Figure S9. The observed peak masses are compared with calculated masses in Table S2.



Figure S9. Peaks in the MALDI-MS spectrum of the supernatant derived from 300 µM of 49.

n	Calculated $m/z$ for $[M + Na]^+$	Observed	Difference
1	1238	1241	3
2	2453	2478	25
3	3669	3695	26
4	4884	4943, 4945	59, 61
5	6098	6161	63
6	7315	7406	91
7	8528	8642	114
8	9746	9866	120

**Table S2.** Comparison of observed and calculated peak masses.

#### (7) Final concentrations of each ingredient in buffered solutions

#### Table 1

#### **Reactions with β-glucosidase from almonds.**

Ingredient	<b>Final concentration</b>
Indoxyl compound	1 mM
β-Glucosidase from almonds	1 unit/mL
Sodium acetate buffer (pH 5.0)	42.5 mM
DMF	5%

#### **Reactions with β-glucosidase from** *Agrobacterium*.

Ingredient	Final concentration
Indoxyl compound	100 μM
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer (pH 7.0)	48.2 mM
DMF	2%
NaCl	1 mM
$(NH_4)_2SO_4$	12 mM

#### Reactions in rat liver homogenate.

Ingredient	Final concentration
Indoxyl compound	100 μM
Rat liver homogenate	95%
DMF	5%

#### Reaction of 33 with $\beta$ -glucosidase from *Agrobacterium* in rat liver homogenate.

Ingredient	<b>Final concentration</b>
33	100 μM
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer (pH 7.0)	200 μM
Rat liver homogenate	96%
DMF	2%
NaCl	1 mM
$(NH_4)_2SO_4$	12 mM

#### Figure 6 Effect of pH (Figure 6A).

Ingredient	Final concentration
33	100 μM
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer	48.2 mM
(pH 4.0, 5.0, 6.0, 7.0, 8.0, or 9.0)	
DMF	2%
NaCl	1 mM
(NH4)2SO4	12 mM

#### Effect of the enzyme concentration (Figure 6B).

Ingredient	Final concentration
33	100 μM
β-Glucosidase from <i>Agrobacterium</i>	10 nM
Sodium phosphate buffer	48.6 mM
(pH 4.0, 5.0, 6.0, 7.0, 8.0, or 9.0)	
DMF	2%
NaCl	0.5 mM
$(NH_4)_2SO_4$	0.59 mM

#### Effect of the concentration of 33 (Figure 6C).

Ingredient	Final concentration
33	1.00, 1.78, 3.16, 5.62, 10.0, 17.8, 31.6, or 56.2 μM)
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer (pH 7.0)	9.8 mM
DMF	2%
NaCl	50 mM
$(NH_4)_2SO_4$	12 mM

# Table 2Oligomerization of 49 in NaPi-NaCl buffer.

Ingredient	Final concentration
49	300, 100, 50 or 10 μM
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer (pH 7.0)	10 mM
NaCl	50 mM
$(NH_4)_2SO_4$	12 mM

## Low-salt oligomerization of 49.

Ingredient	Final concentration
49	300 µM
β-Glucosidase from <i>Agrobacterium</i>	200 nM
Sodium phosphate buffer (pH 7.0)	9.9 mM
DMF	0.6%
NaCl	1 mM
$(NH_4)_2SO_4$	12 mM

## Figure S4

pH Profile of β-glucosidase from Agrobacterium.

Ingredient	Final concentration
4-nitrophenyl β-D-glucopyranoside	2 mM
β-Glucosidase from Agrobacterium	5 nM
Sodium phosphate buffer	45 mM
(pH 4.0, 5.0, 6.0, 7.0, 8.0, or 9.0)	
NaCl	0.025 mM
$(NH_4)_2SO_4$	0.029 mM

## (8) Spectral data































QAc
















S37













ppm


































































S76









S80































