Electronic Supplementary Material (ESI) for New Journal of Chemistry.
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Supporting information

Hydrothermal fabrication of hierarchical CuO nanoflower for dual-function amperometric sensing of hydrogen peroxide and glucose

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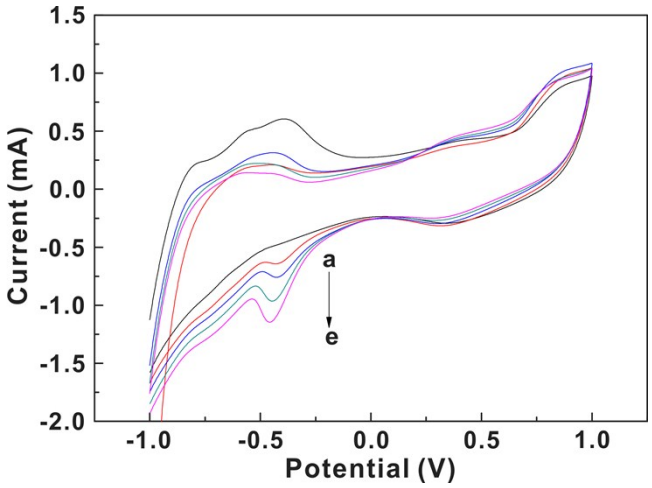

Fig. S1 CVs of the CuO NFs/GC electrode with different concentrations of H_2O_2 (from the top to bottom: 0 to 2.0 mM) in N_2-saturated 0.1 M PBS solution (pH=7) at a scan rate of 100 mV s^{-1}.

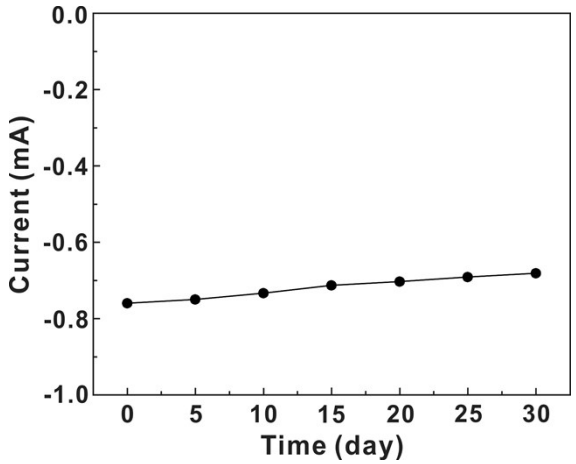

Fig. S2 Long-term stability of CuO NFs/GC electrode for H_2O_2 detection. Electrocatalytic current of H_2O_2 was recorded at -0.4V.

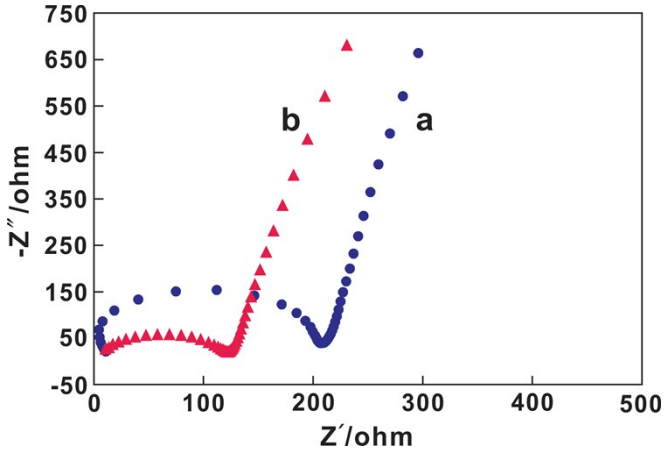

Fig. S3 Nyquist plots of (a) bare GCE and (b) CuO NFs/GC electrode in 0.1 M KCl containing 5 mM $[Fe(CN)_6]^{3-/4-}$.

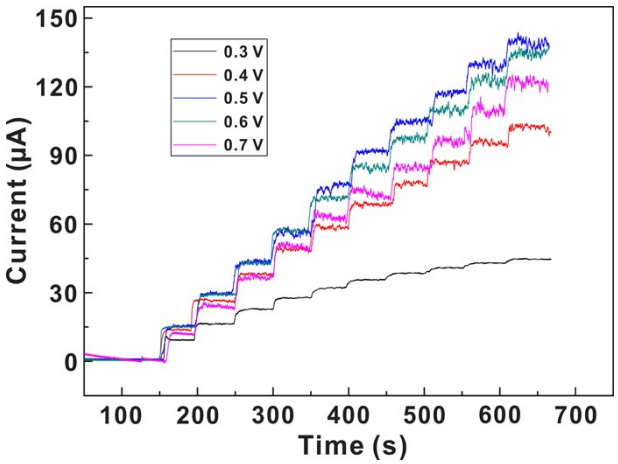

Fig. S4 Amperometric response of CuO NFs/GC electrode at different applied potentials with successive addition of 0.2 mM glucose.

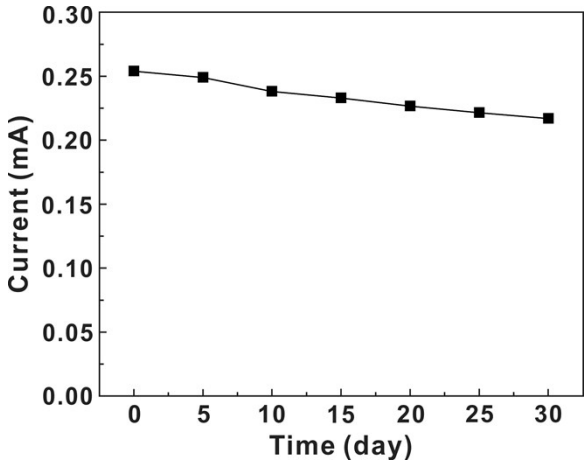

Fig. S5 Long-term stability of CuO NFs/GC electrode for glucose detection. Electrocatalytic current of glucose was recorded at 0.5 V.