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Support Information for:

**One-pot synthesis of a recyclable ratiometric fluorescent probe based on MOFs
for turn-on sensing of Mg^{2+} ions and bioimaging in live cells**

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Figure S19 Viabilities of HL-7702 cells and A373 cells in the presence of $Fe_3O_4/RhB@Al-MOFs$ composite as assayed by MTT.

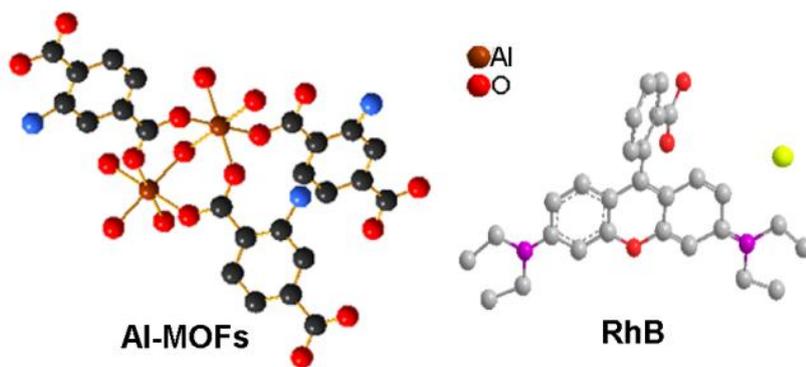


Figure S1 The structure of Al-MOFs and RhB

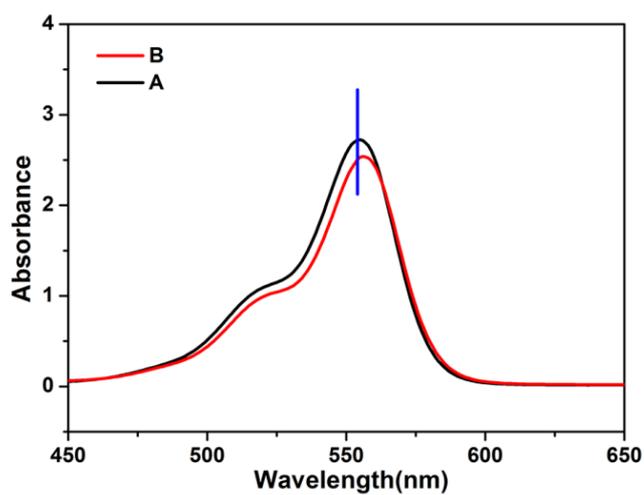


Figure S2 UV-vis absorption spectra of RhB (A) and RhB after interacting with Al^{3+} ions (B).

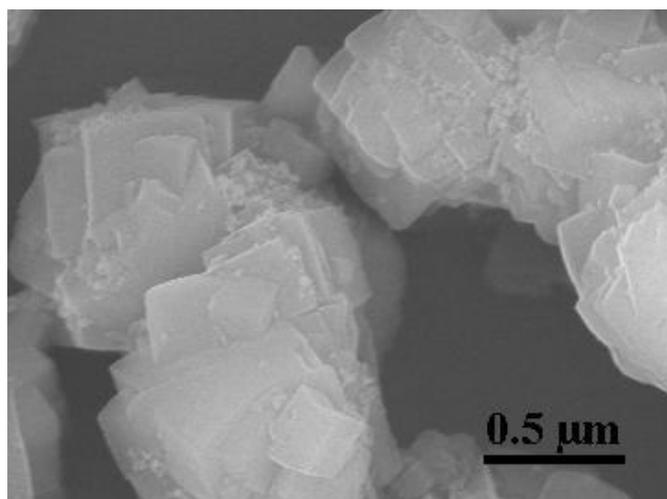


Figure S3 The enlarged SEM images of $\text{Fe}_3\text{O}_4/\text{RhB}@$ Al-MOFs

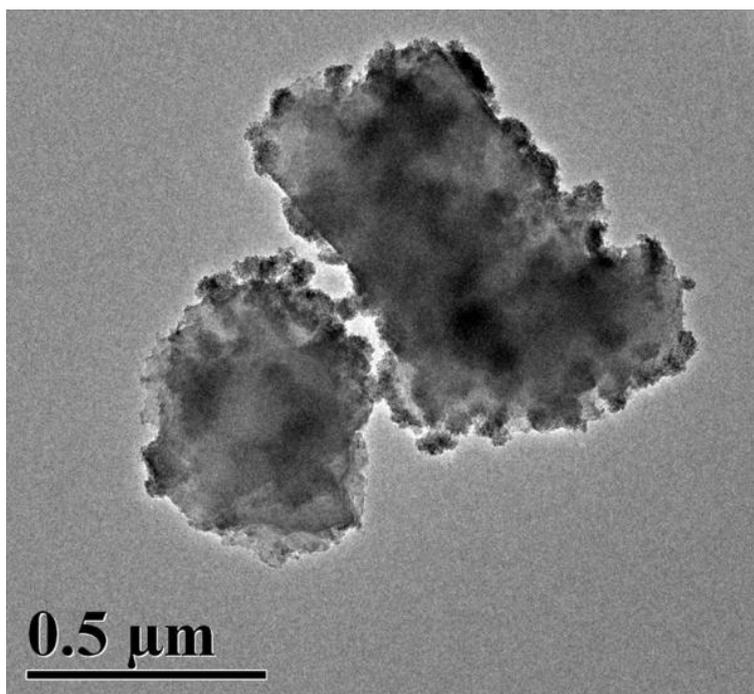


Figure S4 The TEM images of Fe₃O₄/RhB@Al-MOFs

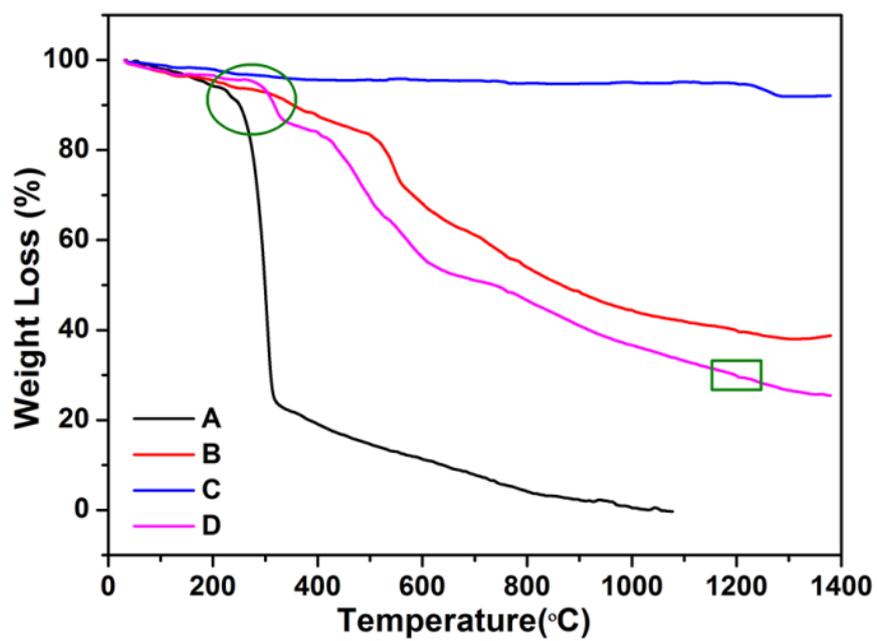


Figure S5 The TGA curves of RhB (A) and Al-MOFs (B), Fe₃O₄ (C) and Fe₃O₄/RhB@Al-MOFs (D)

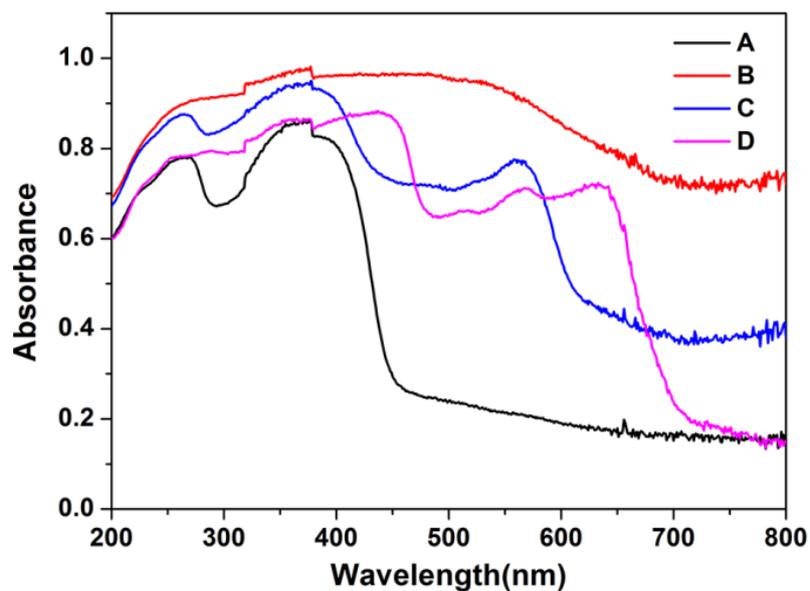


Figure S6 The solid-state UV-vis absorption spectra of Al-MOFs (A), Fe_3O_4 (B)
 $\text{Fe}_3\text{O}_4/\text{RhB}@$ Al-MOFs (C) and RhB (D)

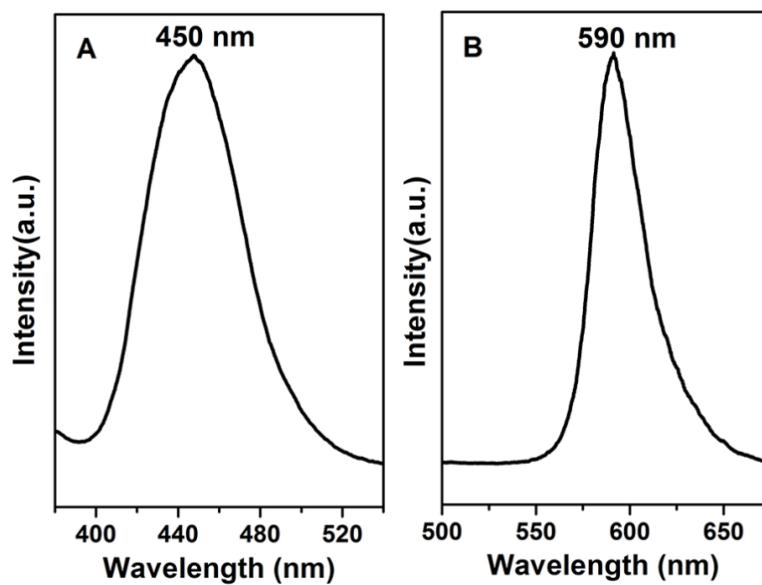


Figure S7 The emission spectra of Al-MOFs (A) and RhB (B) excited at 320 nm

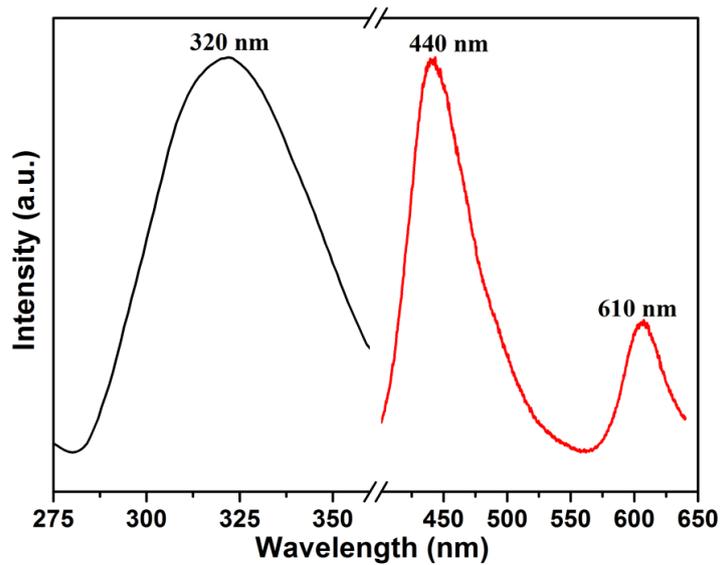


Figure S8 The excitation and emission spectra of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$

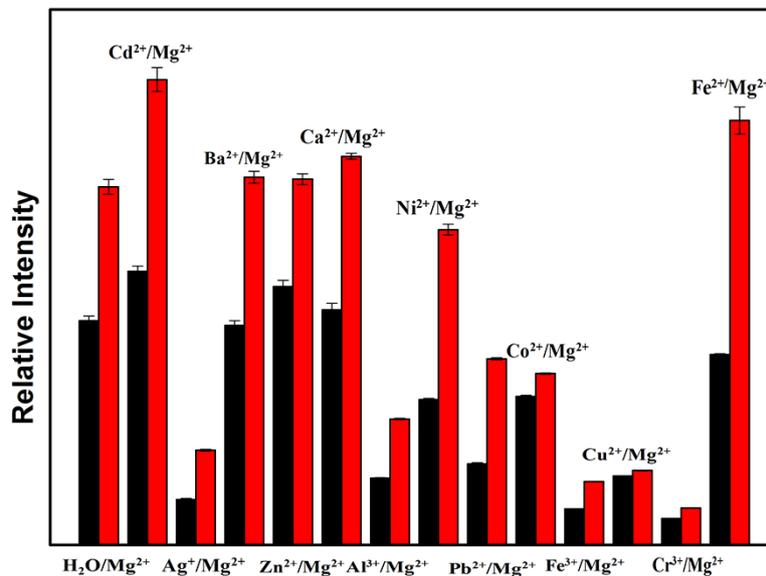


Figure S9 Fluorescence responses of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$ (I_{440}/I_{610}) to Mg^{2+} ions among various ions. The blue bars represent the emission intensities of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$ (I_{440}/I_{610}) in the presence of $1 \times 10^{-3} \text{ M}$ other metal ions. The red bars represent the change of the emission intensities of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$ (I_{440}/I_{610}) upon the subsequent addition of $1 \times 10^{-3} \text{ M}$ Mg^{2+} ions to the above solution.

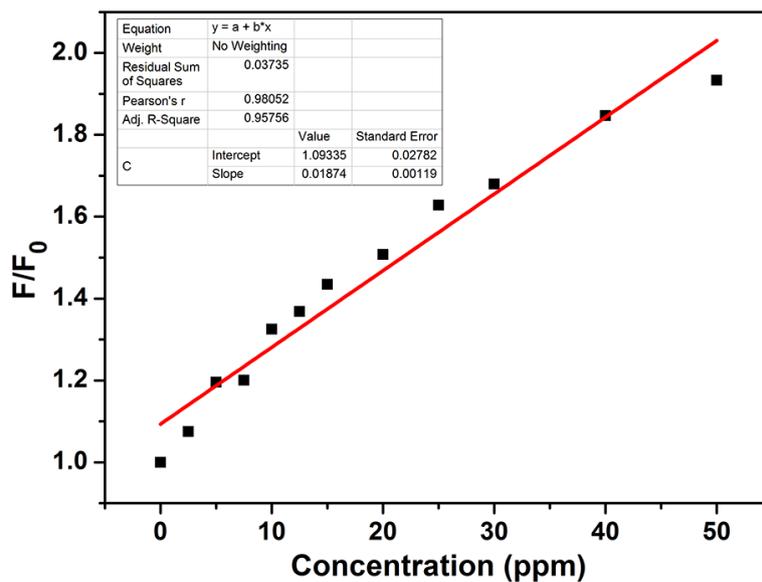


Figure S10 Linear relationship between the fluorescence intensity of Al-MOFs and Mg^{2+} ions concentration

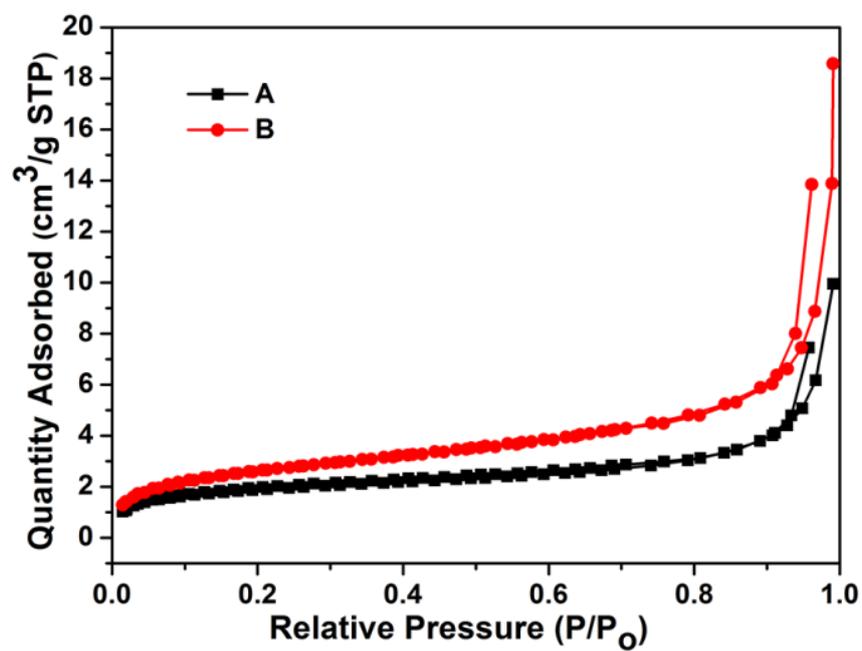


Figure S11 N_2 adsorption-desorption isotherms of Al-MOFs (A) and $Fe_3O_4/RhB@Al-MOFs$ (B)

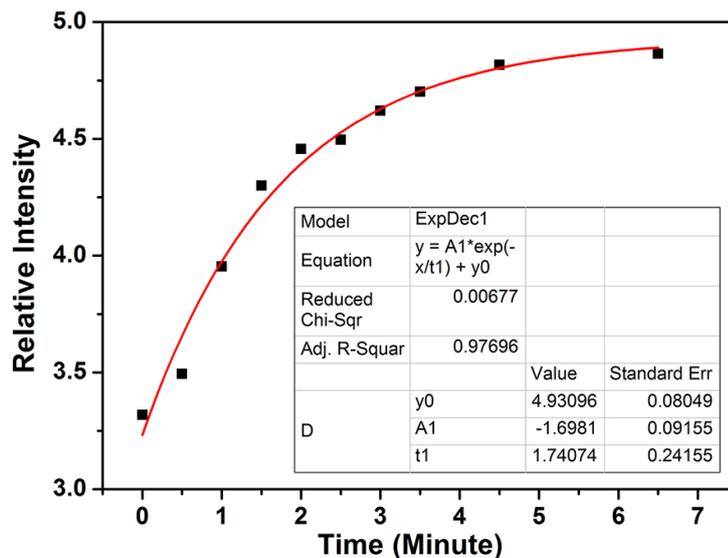


Figure S12 The kinetic study of the response of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$ to Mg^{2+} ions ($1 \times 10^{-3} \text{M}$) under pseudo-first-order conditions. Slope = -0.5745 min^{-1}

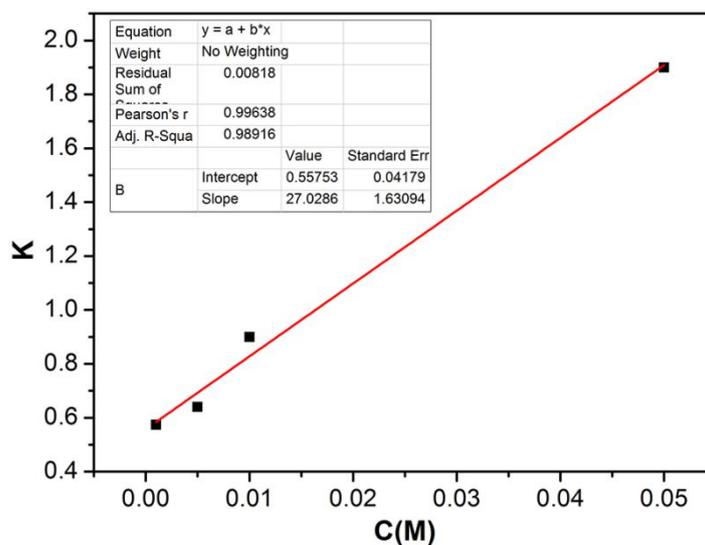


Figure S13 Plot of the observed k versus the concentration of Mg^{2+} ions for the pseudo first-order reaction of $\text{Fe}_3\text{O}_4/\text{RhB}@ \text{Al-MOFs}$ with varying concentration of

$$\text{Mg}^{2+} \text{ Slope} = 27.0286 \text{ M}^{-1} \text{ min}^{-1}$$

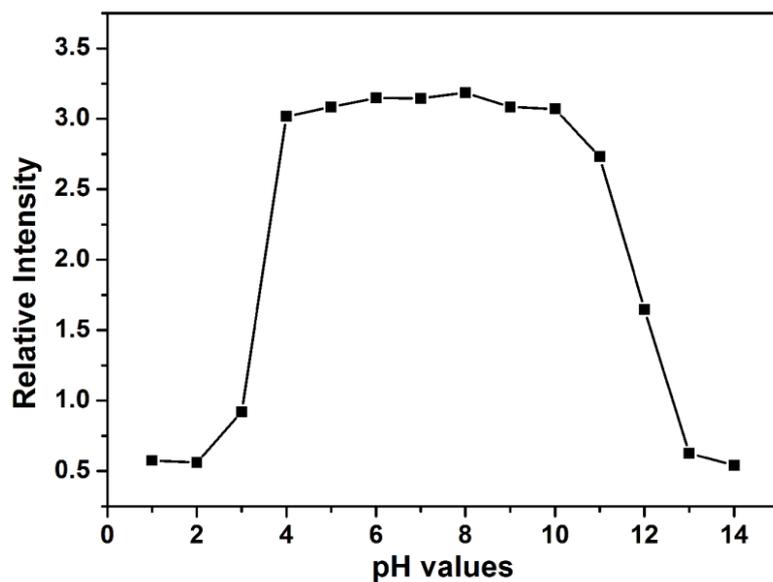


Figure S14 Fluorescence measurements of Fe₃O₄/RhB@Al-MOFs after treatment with different pH

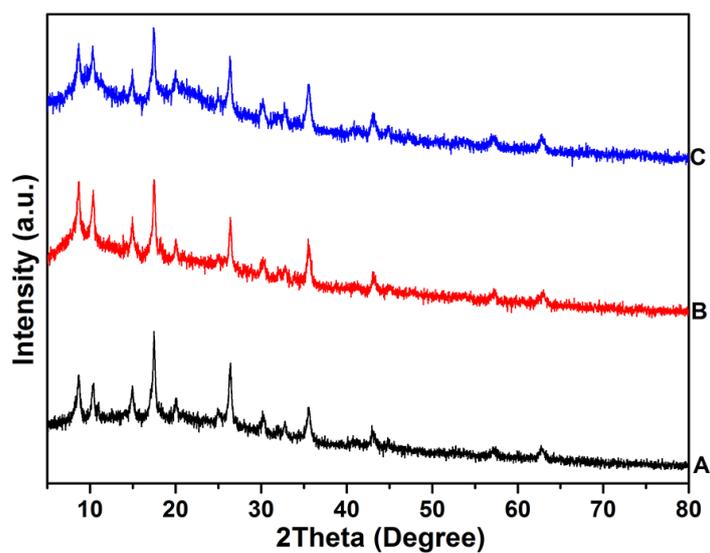


Figure S15 XRD patterns of as-synthesized Fe₃O₄/RhB@Al-MOFs (A), Fe₃O₄/RhB@Al-MOFs after detection of Mg²⁺ ions for five cycles (B) and Fe₃O₄/RhB@Al-MOFs after storing in water for 7 days (C)

Table 1 Comparison table for various probes for the detection of Mg²⁺ ions.

Probe	Metal	Detect limit	Application	Ref
4-hydroxy-5-isopropyl-2-thylisophthalaldehyde	Mg ²⁺	2.70 × 10 ⁻⁶ M	Fluorescent “turn-on” sensor	[14]
4-hydroxy-3-((2-hydroxy-5-methylphenyl)diazenyl)-2Hchromen-2-one	Mg ²⁺	2.4 × 10 ⁻⁸ M	Fluorescent “turn-off” sensor	[15]
8-hydroxyquinoline-5-carbaldehyde-(benzotriazol-1-acetyl)	Mg ²⁺	–	Fluorescent “turn-on” sensor	[16]
(Z)-2-hydroxy-N-(2-hydroxybenzylidene)benzohydrazide	Mg ²⁺	1.7 × 10 ⁻⁷ M	Fluorescent “turn-on” sensor	[17]
(E)-2-((2-(pyridin-2-yl)hydrazono)methyl)quinolin-8-ol	Mg ²⁺	1.9 × 10 ⁻⁷ M	Fluorescent “turn-on” sensor	[18]
{[Ln(L) ₃ Fe _{1.5} (H ₂ O) ₃] _{1.5} H ₂ O} _n	Mg ²⁺	–	Fluorescent “turn-on” sensor	[19]
[LnAg(PDA) ₂ (H ₂ O) ₃] ₃ H ₂ O _n	Mg ²⁺	–	Fluorescent “turn-on” sensor	[20]
8-HQC-PTH	Mg ²⁺	4.7 × 10 ⁻⁸ M	Ratiometric Fluorescent sensor	[21]
2-hydroxy-5-methyl-1,3-phenylenebis(methanylylidene)bis(isoquinoline-1-carbohydrazide)	Mg ²⁺	2.97 × 10 ⁻⁸ M	Fluorescent “turn-on” sensor Intracellular detection; plant tissues detection	[22]
Fe ₃ O ₄ /RhB@Al-MOFs	Mg ²⁺	8 × 10 ⁻⁷ M	Ratiometric Fluorescent “turn-on” sensor; Intracellular detection; Magnetic recycling	This work

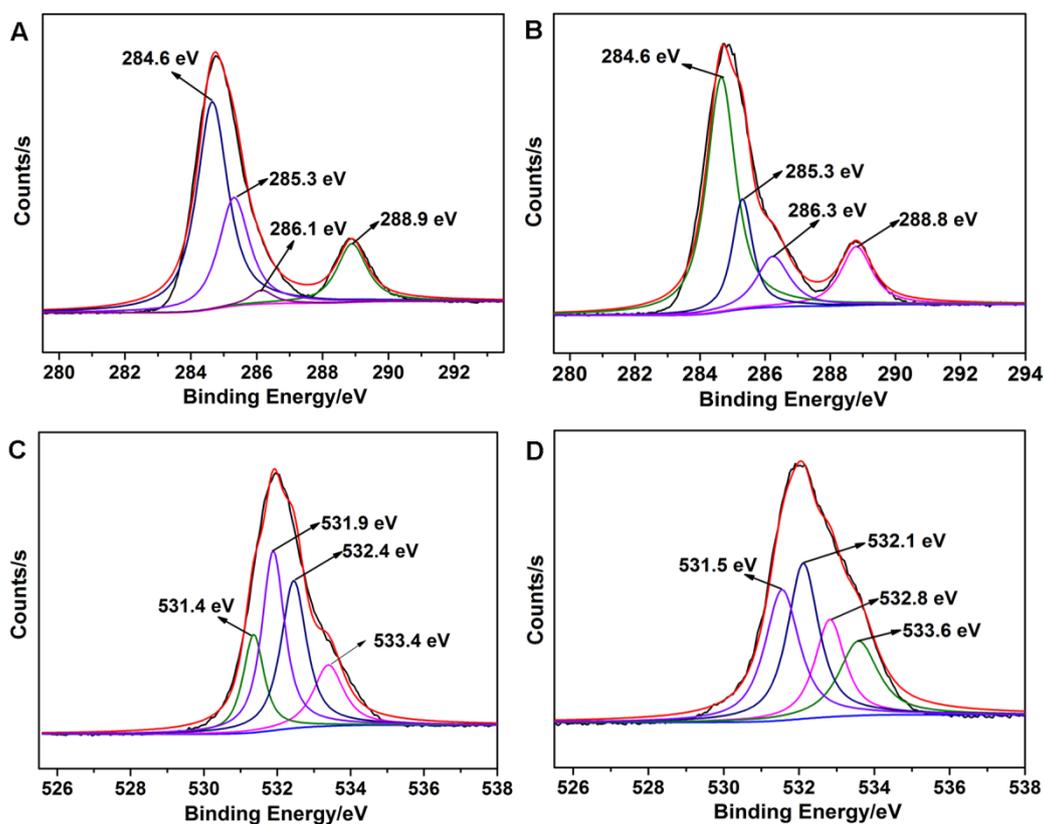


Figure S16 (A, B) C1s XPS for Fe₃O₄/RhB@Al-MOFs and Mg²⁺ treatment of Fe₃O₄/RhB@Al-MOFs; (C, D) O1s XPS for Fe₃O₄/RhB@Al-MOFs and Mg²⁺ treatment of Fe₃O₄/RhB@Al-MOFs

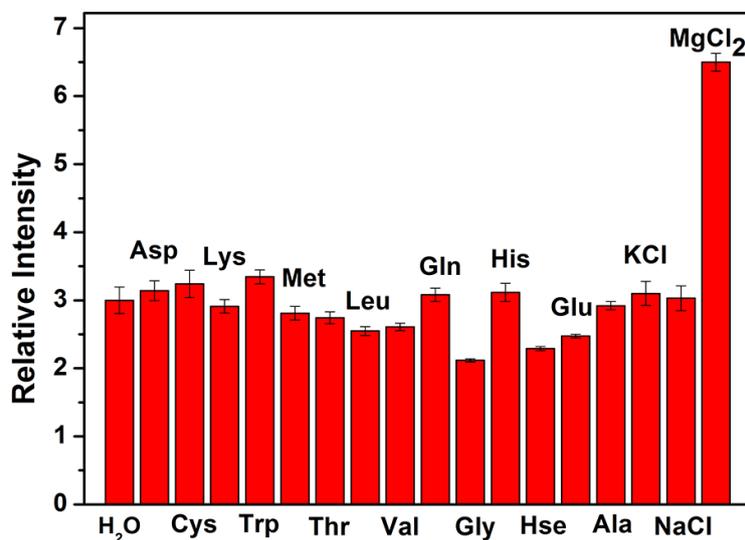


Figure S17 Comparison of the luminescence intensity of Fe₃O₄/RhB@Al-MOFs (I₄₄₀/I₆₁₀) in 10⁻³ M solutions of different biologically relevant substances

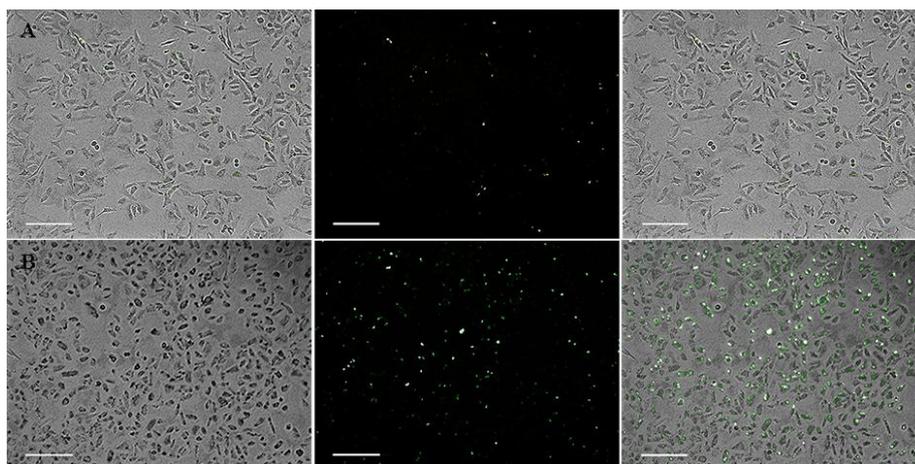


Figure S18 Fluorescence imaging of live A375 cells after being incubated with $\text{Fe}_3\text{O}_4/\text{RhB}@\text{Al-MOFs}$ composite material (A); Fluorescence imaging of live A375 cells after being incubated with $\text{Fe}_3\text{O}_4/\text{RhB}@\text{Al-MOFs}$ and Mg^{2+} ions (B). The left panels show dark-field fluorescence images, the middle panels show the corresponding bright-field images and the right panels are overlays of the left and middle panels. Scale bar: 200 μm .

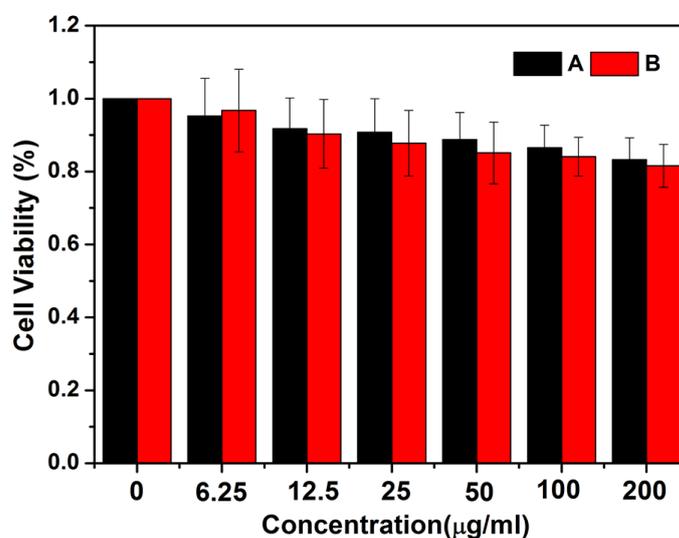


Figure S19 Viabilities of HL-7702 cells and A373 cells in the presence of $\text{Fe}_3\text{O}_4/\text{RhB}@\text{Al-MOFs}$ composite assessed by MTT.