

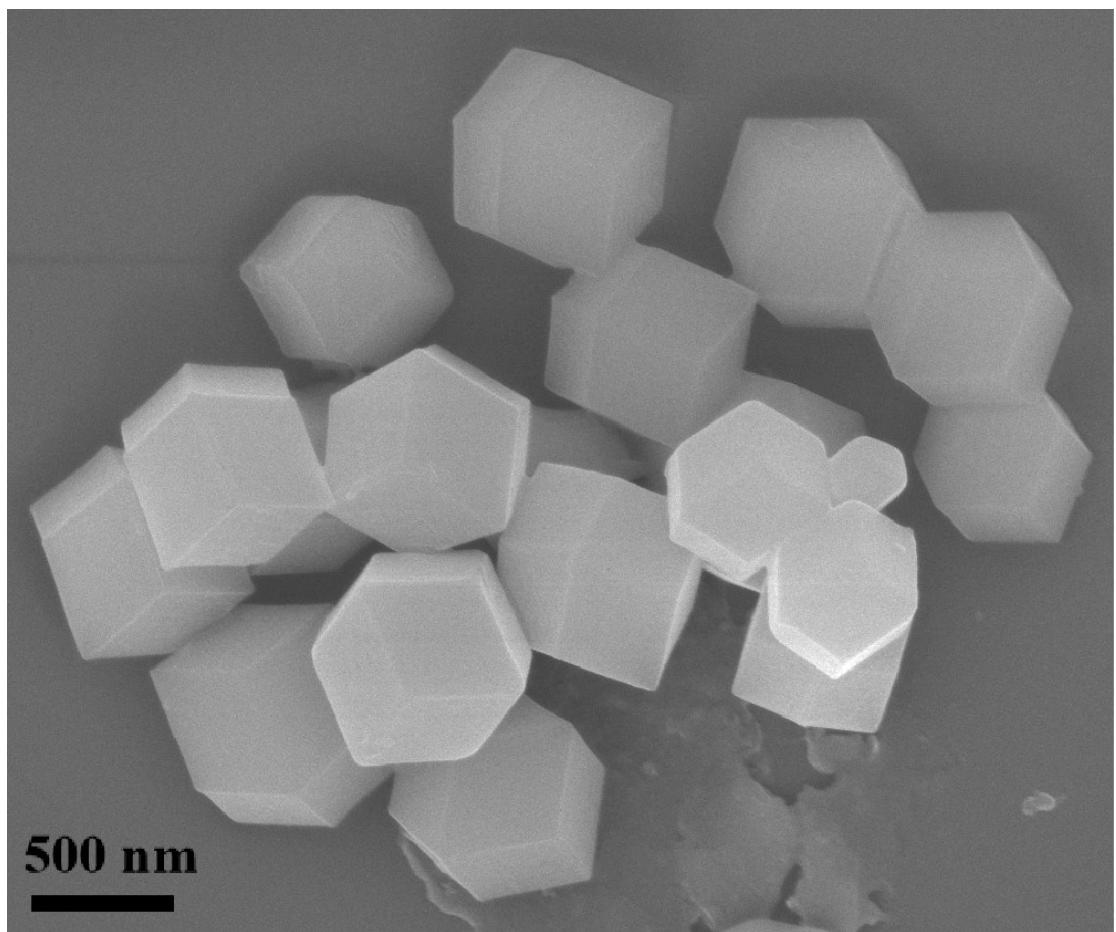
## Supplementary Information

### **MOF-derived NiO-NiCo<sub>2</sub>O<sub>4</sub>@PPy hollow polyhedron as a sulfur immobilizer for lithium-sulfur batteries**

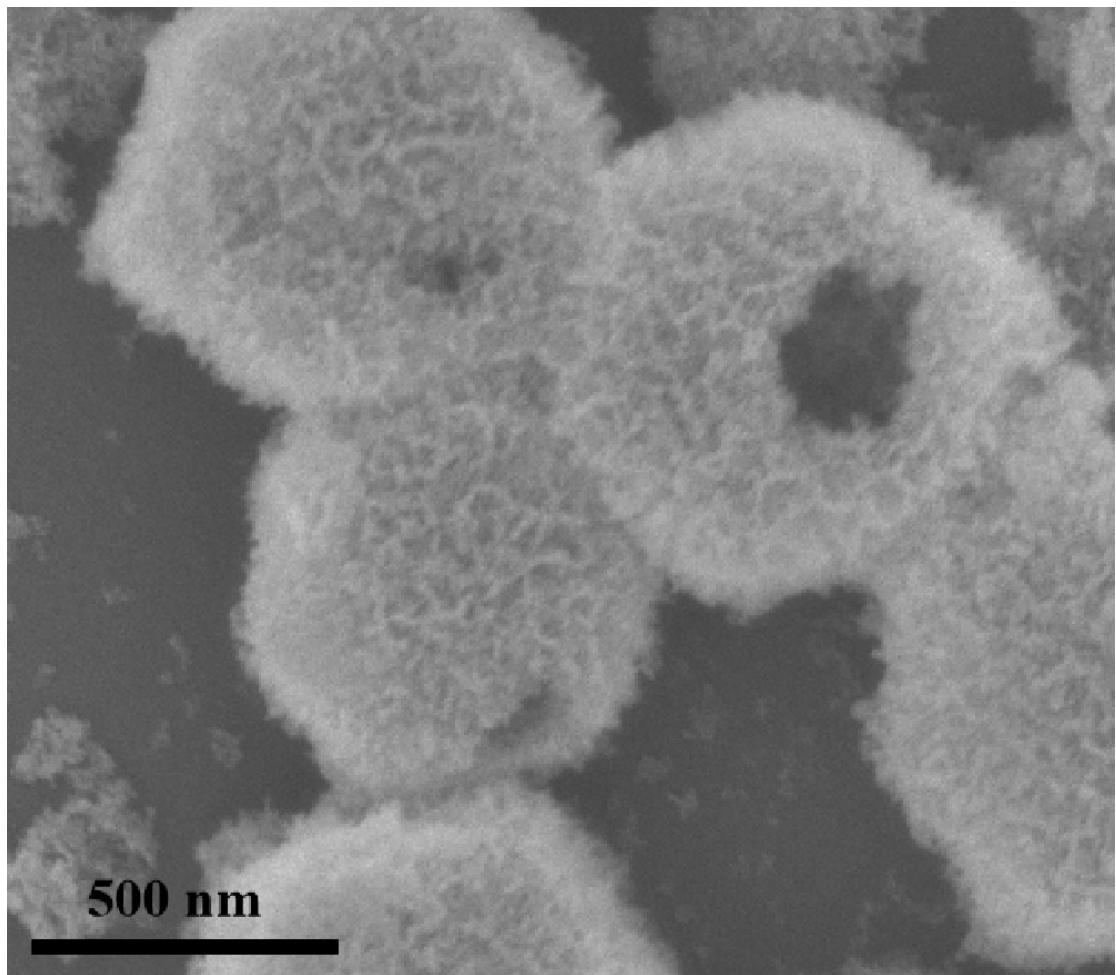
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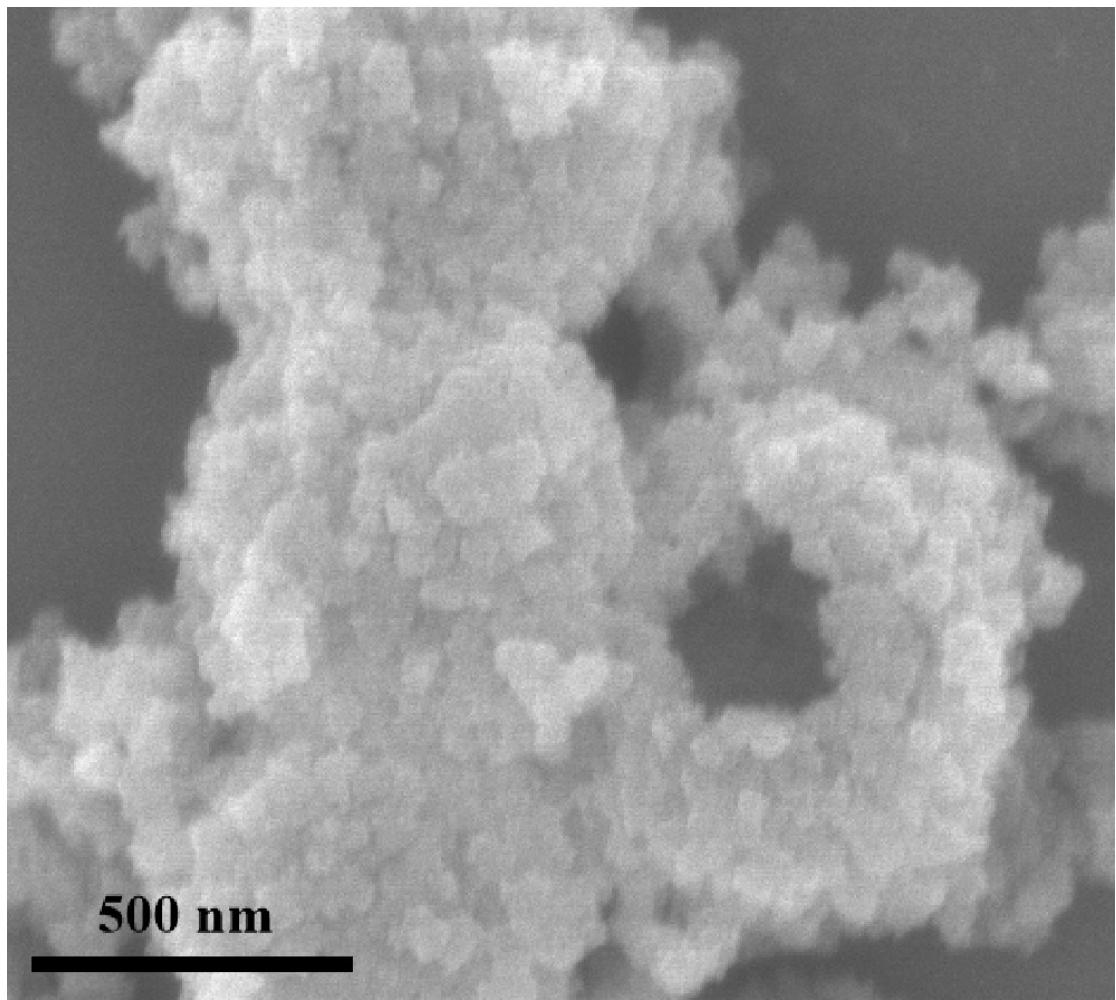
E-mail: [jinbo@jlu.edu.cn](mailto:jinbo@jlu.edu.cn) (B. Jin)



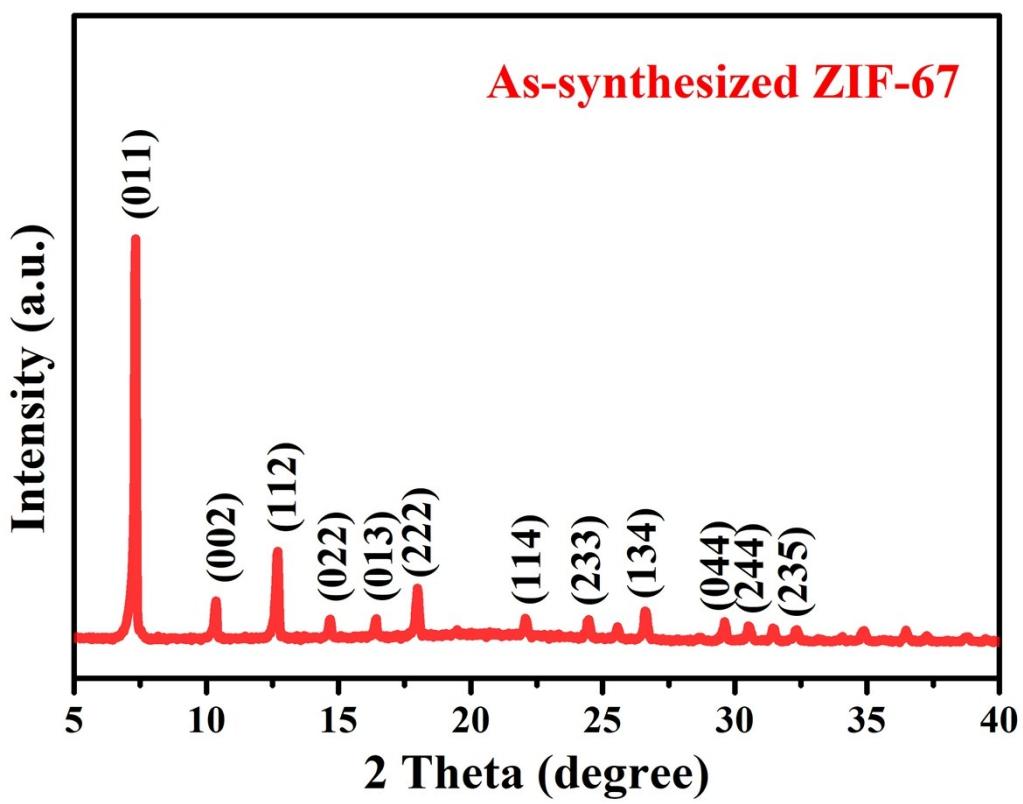
**Figure S1.** FESEM image of ZIF-67.



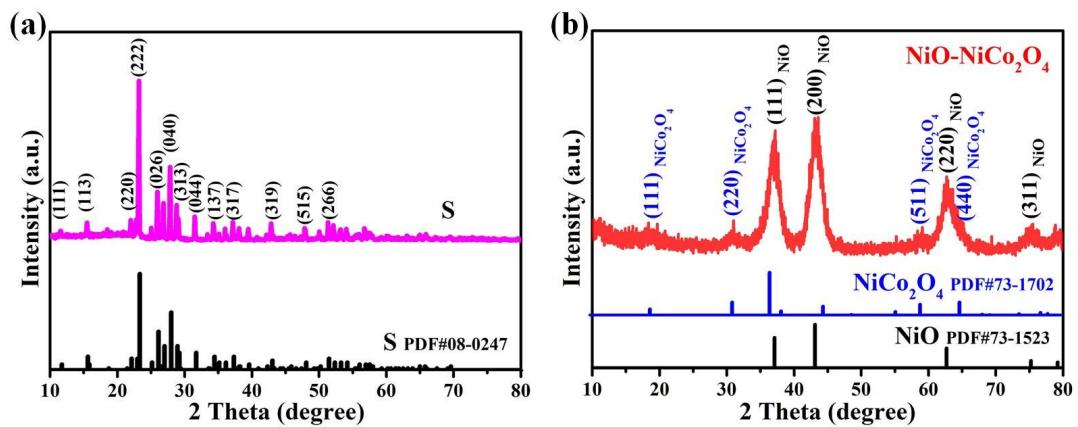
**Figure S2.** FESEM image of NiO-NiCo<sub>2</sub>O<sub>4</sub> polyhedron.



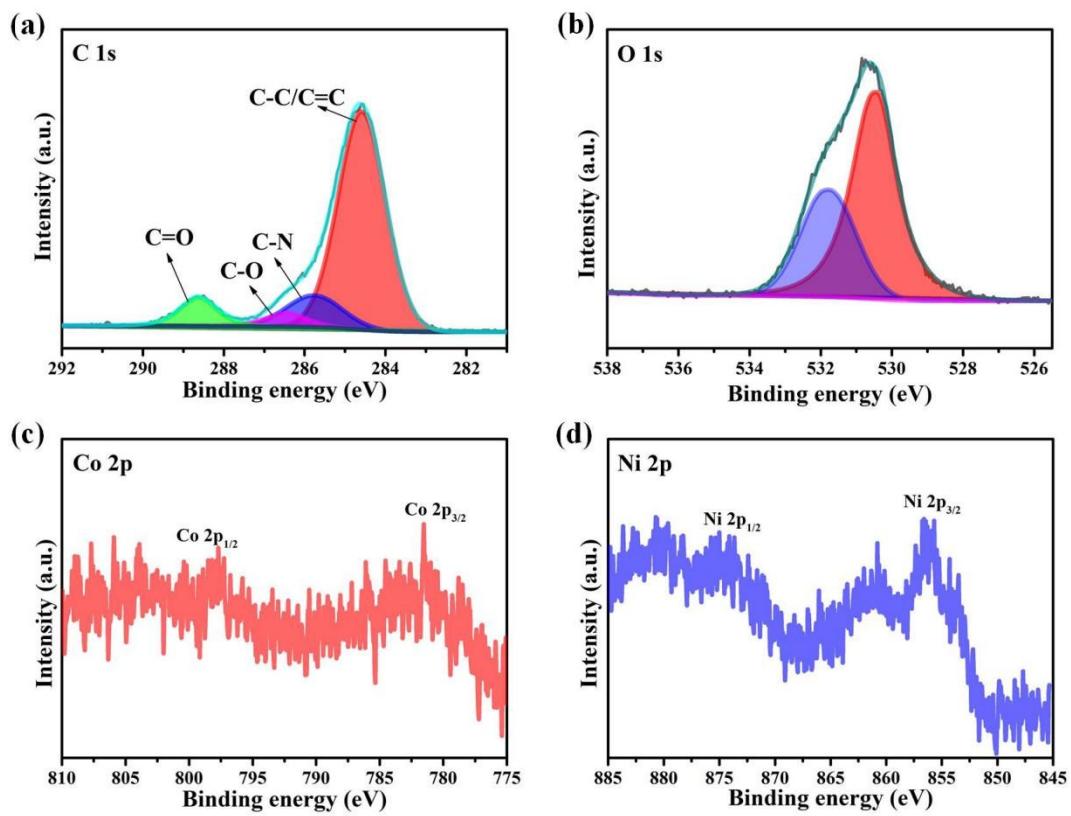
**Figure S3.** FESEM image of NiO-NiCo<sub>2</sub>O<sub>4</sub>@PPy.



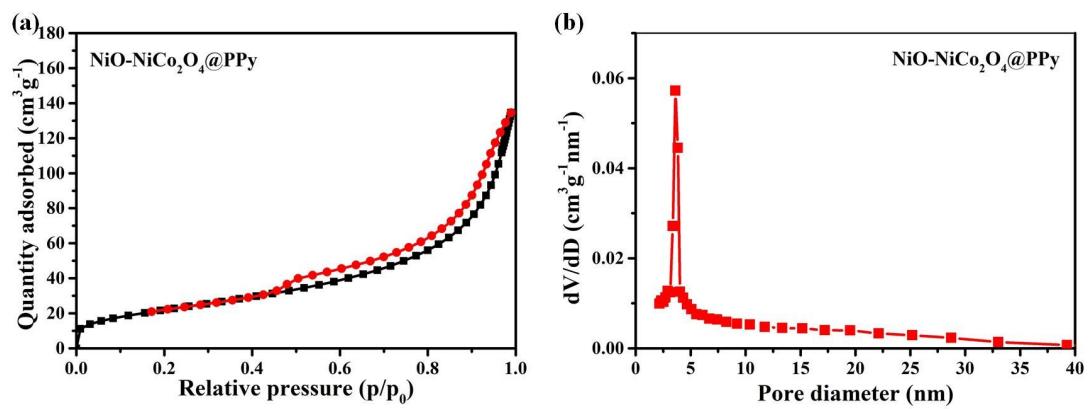
**Figure S4.** XRD pattern of the as-synthesized ZIF-67.



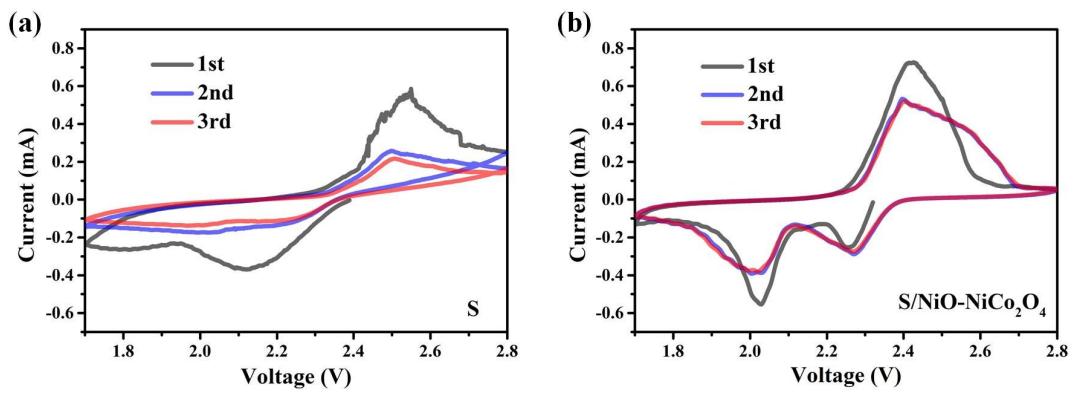
**Figure S5.** XRD patterns of (a) S and (b) NiO-NiCo<sub>2</sub>O<sub>4</sub>.



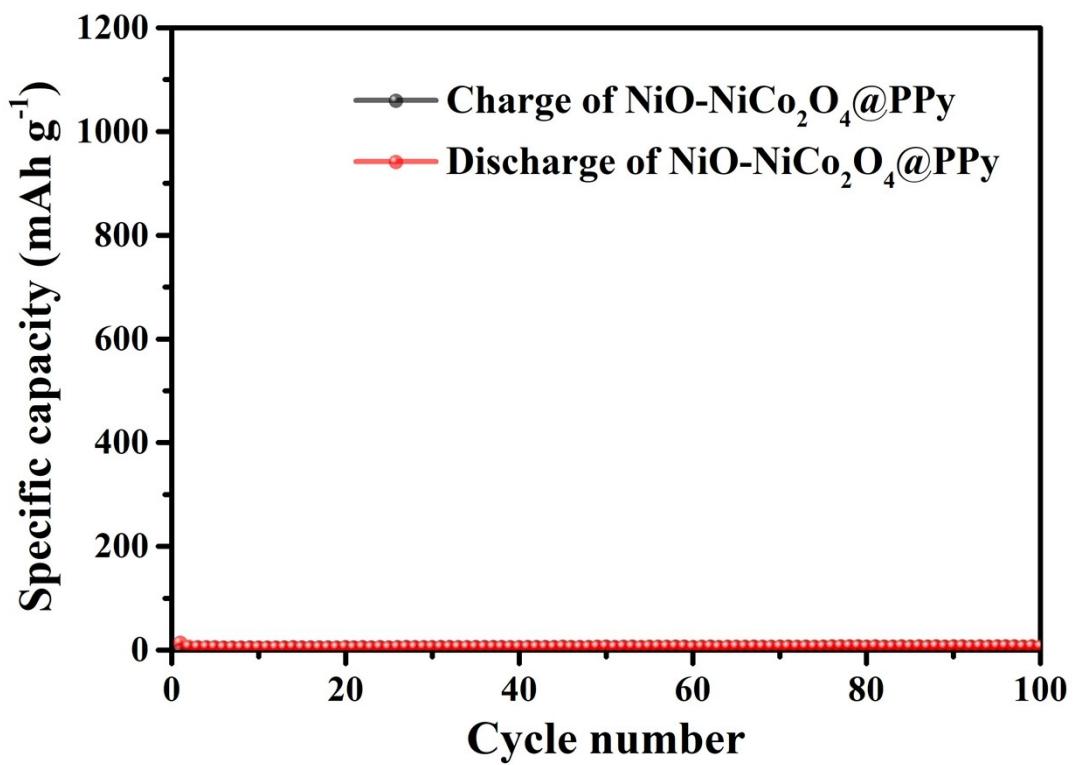
**Figure S6.** (a) C 1s, (b) O 1s, (c) Co 2p and (d) Ni 2p spectra of S/NiO-NiCo<sub>2</sub>O<sub>4</sub>@PPy.



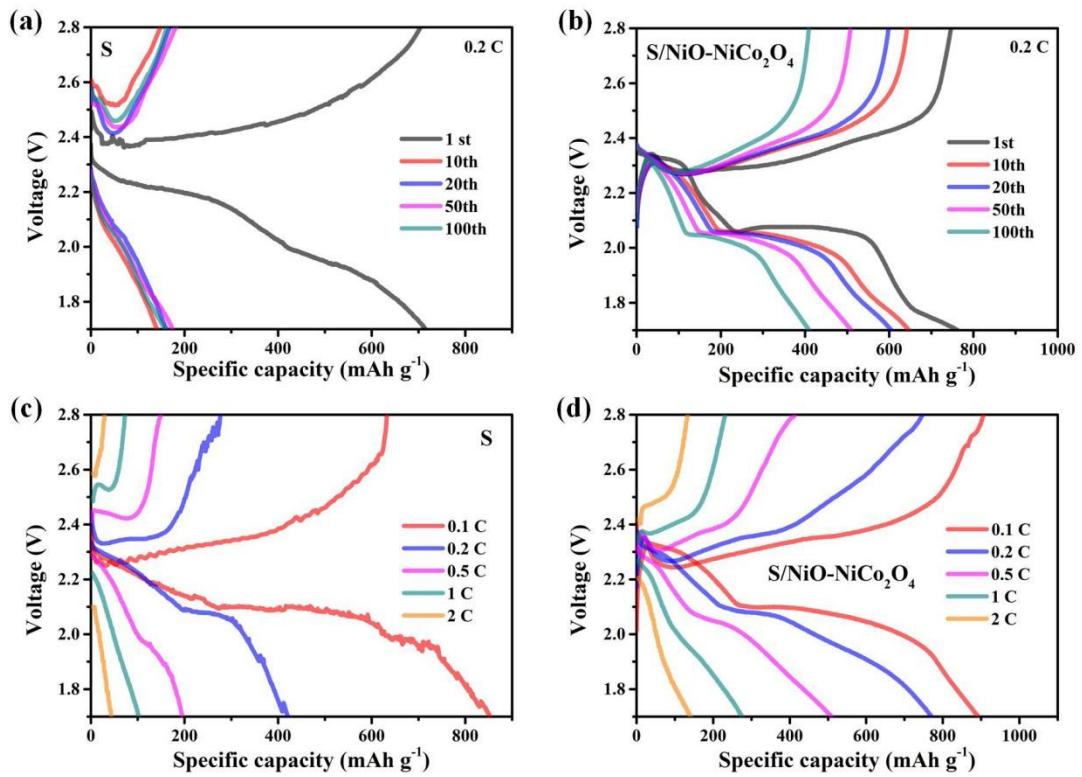
**Figure S7.** (a) Nitrogen adsorption-desorption isotherms and (b) pore size distribution curve of  $\text{NiO-NiCo}_2\text{O}_4@\text{PPy}$  composite.



**Figure S8.** CV curves of (a) S and (b) S/NiO-Ni $\text{Co}_2\text{O}_4$ .

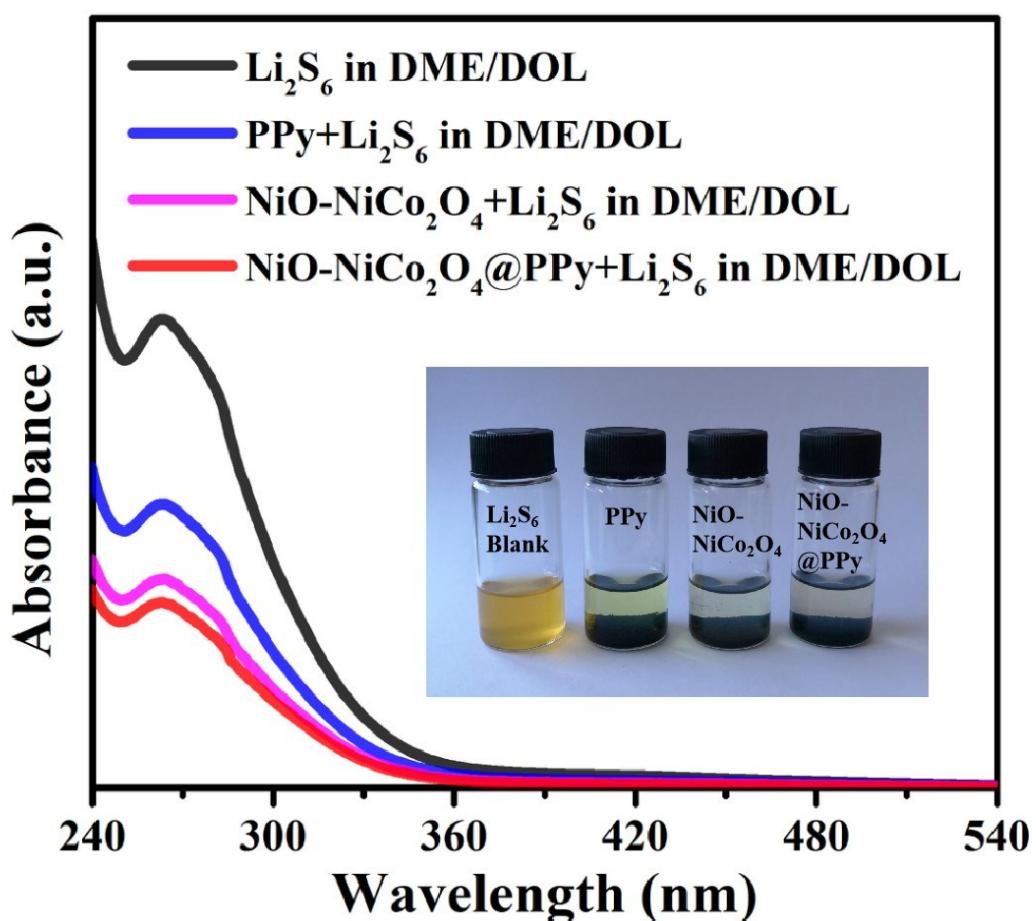


**Figure S9.** Cycling performance of  $\text{NiO}-\text{NiCo}_2\text{O}_4@\text{PPy}$  at  $335 \text{ mA g}^{-1}$ .

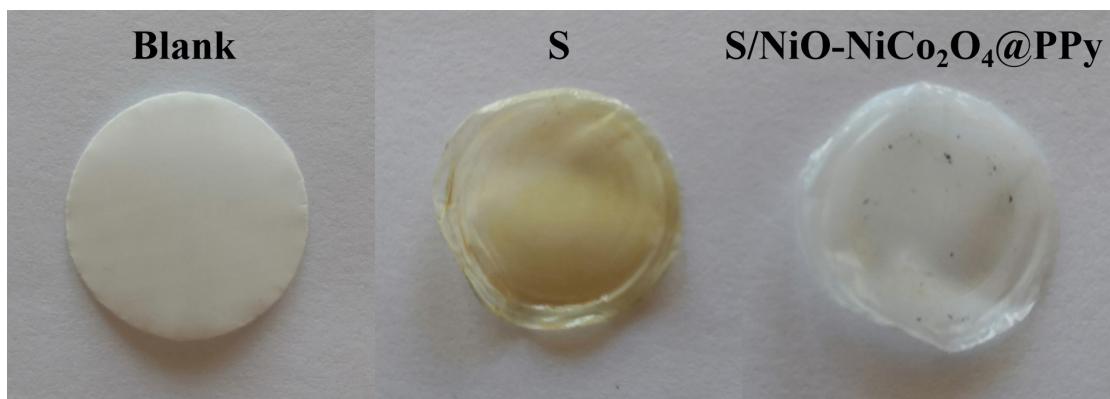


**Figure S10.** Charge-discharge profiles of (a) S and (b) S/NiO-NiCo<sub>2</sub>O<sub>4</sub> at 0.2 C.

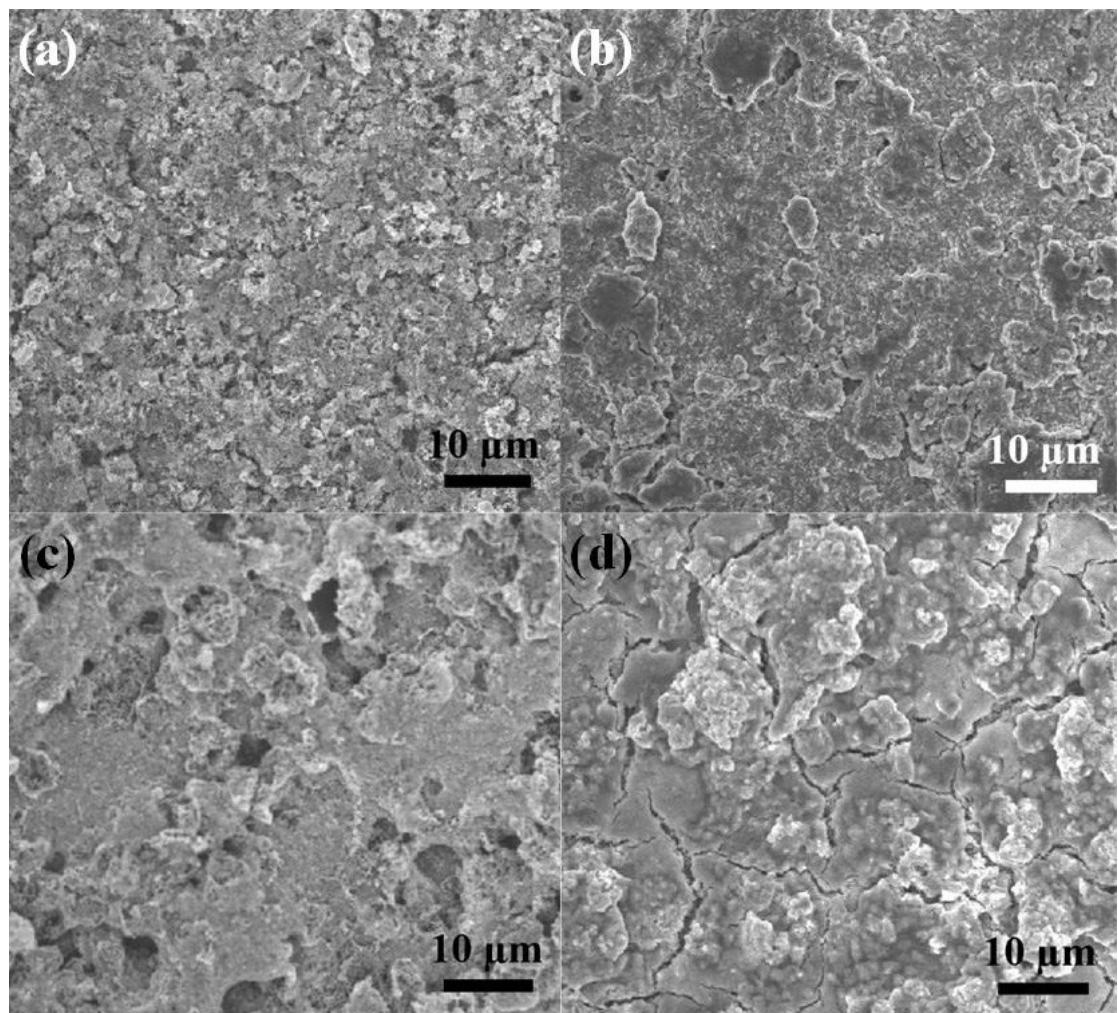
Charge-discharge profiles of (c) S and (d) S/NiO-NiCo<sub>2</sub>O<sub>4</sub> at various current densities from 0.1 to 2 C in the voltage range of 1.7-2.8 V.



**Figure S11.** UV-Vis spectra of  $\text{Li}_2\text{S}_6$  in DME/DOL, PPy+ $\text{Li}_2\text{S}_6$  in DME/DOL, NiO-Ni $\text{Co}_2\text{O}_4$ + $\text{Li}_2\text{S}_6$  in DME/DOL, and NiO-Ni $\text{Co}_2\text{O}_4$ @PPy+ $\text{Li}_2\text{S}_6$  in DME/DOL. Inset: photographs of sealed vials of  $\text{Li}_2\text{S}_6$  in DOL/DME and  $\text{Li}_2\text{S}_6$  in DOL/DME after being in contact with PPy, NiO-Ni $\text{Co}_2\text{O}_4$ , and NiO-Ni $\text{Co}_2\text{O}_4$ @PPy for 5 h.



**Figure S12.** Photographs of separators of batteries with pure sulfur and S/NiO-NiCo<sub>2</sub>O<sub>4</sub>@PPy after cycling, compared to photograph of separator before cycling (blank).



**Figure S13.** SEM images of S/NiO-NiCo<sub>2</sub>O<sub>4</sub> cathode (a) before cycle and (b) after 100 cycles at 0.2 C. SEM images of S cathode (c) before cycle and (d) after 100 cycles at 0.2 C.

**Table S1.** Comparison of the related electrodes with this work.

Materials	Capacity (mAh g <sup>-1</sup> )	Current density	Sulfur content	Areal loading (mg cm <sup>-2</sup> )	Cycle	Ref.
<b>S-PPy-based sandwich electrode</b>	383	0.1 C	—	—	500	[1]
<b>S/PPy</b>	613	0.1 C	—	—	50	[2]
	475	1 C	—	—	50	
<b>NiCo<sub>2</sub>O<sub>4</sub>/S</b>	601	0.5 C	27%	—	200	[3]
<b>S-Co<sub>3</sub>O<sub>4</sub> nanotubes</b>	538	0.2 C	78%	—	100	[4]
<b>Co<sub>3</sub>O<sub>4</sub>/S/ACNTs</b>	496	0.5 C	58.73%	1.5	550	[5]
<b>S/NiO-NiCo<sub>2</sub>O<sub>4</sub>@PPy</b>	641	0.2 C	61.5%	2	100	This work
	411	1 C			200	

## References

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