**Supporting Information** 

## A quasi-solid composite electrolyte with dual salts for dendrite -

## free lithium metal batteries

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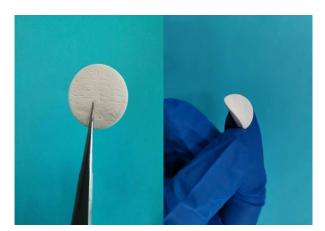


Fig. S1 Photos of the prepared electrolyte membrane with 30wt% Mg(ClO<sub>4</sub>)<sub>2</sub>.

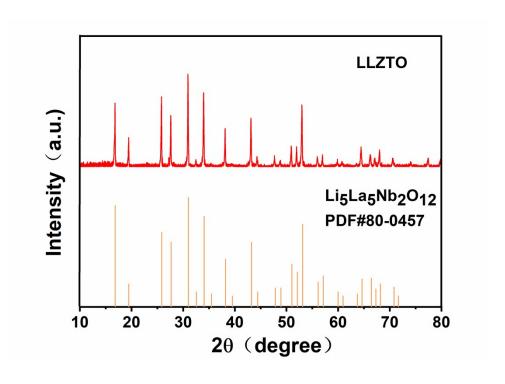


Fig. S2 XRD pattern of the as-prepared LLZTO powders.

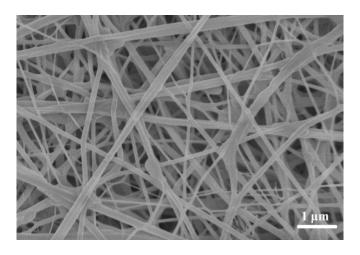
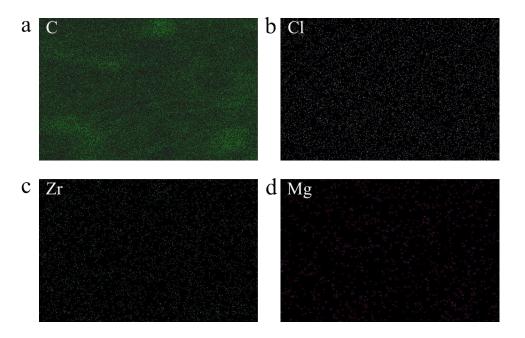
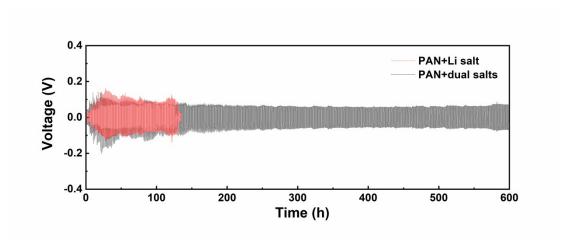


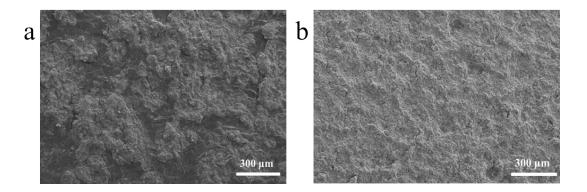
Fig. S3 SEM image of the surface of the electrolyte membrane.



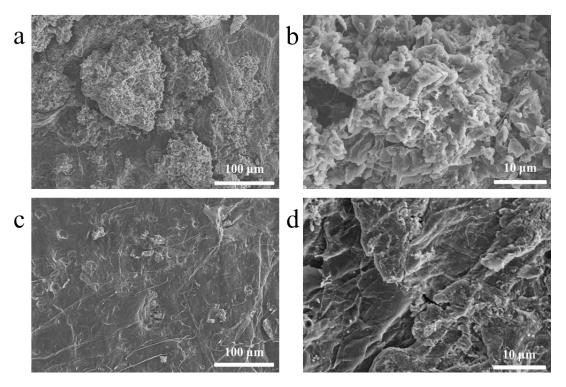
**Fig. S4** EDS elemental mappings of the electrolyte membrane. (a) C, (b) Cl, (c) Zr, and (d) Mg.



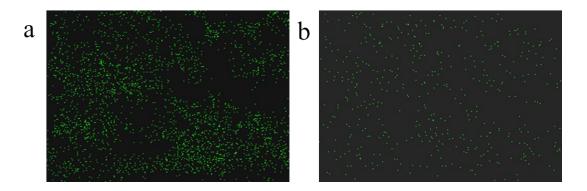
**Fig. S5** Voltage profile of the lithium plating/stripping cycling in the symmetrical Li|HSE|Li cells with magnesium salt added electrolyte and electrolyte without magnesium salt.



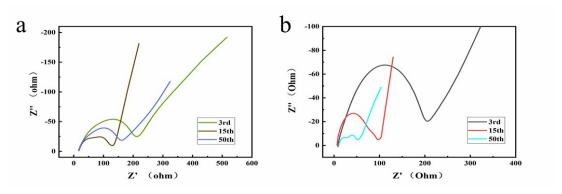
**Fig. S6** (a) SEM image of the surface of metal lithium electrode in LiFePO<sub>4</sub>|HSE|Li cells without magnesium salt added electrolyte after 200 cycles. (b) SEM image of the surface of metal lithium electrode in LiFePO<sub>4</sub>|HSE|Li cells with magnesium salt added electrolyte after 200 cycles.



**Fig. S7** (a,b) SEM image of the surface of metal lithium electrode in Li|HSE|Li cells without magnesium salt added electrolyte after tested at different current densities from 0.045~0.36 mA cm<sup>-2</sup> for 400 h. (c,d) SEM image of the surface of metal lithium electrode in Li|HSE|Li cells with magnesium salt added electrolyte after tested at different current densities from 0.045~0.36 mA cm<sup>-2</sup> for 400 h.



**Fig. S8** EDS elemental mappings of the Li electrode of the cell with the magnesium salt added electrolyte. (a) F, and (b) Mg.



**Figure S9.** Nyquist impedance spectra of the cells after different cycles: (a) without magnesium salt added electrolyte and (b) with magnesium salt electrolyte.