Supporting Information

Fabrication of Dual Catalytic Microcapsules by Mesoporous Graphitic Carbon Nitride (mpg-C₃N₄) Nanoparticles-Enzyme Conjugates Stabilized Emulsions.

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The N₂ adsorption- desorption isotherms of the mesoporous graphitic carbon nitride have been conducted to estimate the specific surface area, and pore size distribution. The analysis shows that the mpg-C₃N₄ high specific surface area (110 m²/g) as compared to the bulk-gC₃N₄ (20 m²/g).

The Average crystallite size is calculated by the Scherrer equation

 $D = (K \times \lambda) / (\beta \cos \theta).$

Where, D is the mean size of the ordered (crystalline) domains at particular angle, K is a dimensionless shape factor (0.9), λ is the X-ray wavelength, β is the line broadening at Full Width at Half Maxima (FWHM and θ is the Bragg angle. Using this equation, and the average of all the D values has been calculated to be 4.9 nm.



Figure S1: Powder XRD pattern of mpg-C₃N₄ NPs



Figure S2: TEM images of (a) MCM 41 (b) freshly prepared mpg-C₃N₄ NPs



Figure S3 : BET studies of mpg-C $_3N_4$ and bulk gC $_3N_4$



Figure S4: mpg-C₃N₄ pore size distribution

B. Zeta Potential Titration

Stock Solution:- mpg-C₃N₄ NPs = 1mg/mL; Lipase: 1 mg/mL.

Total Volume for titration: 800 μ L.

Titration with varying volume ratio of mpg-C $_3N_4$ and lipase

Volume ratio	Zeta Potential (mV)
1:1.5	-8.21±0.56
1:1	-3.27±0.375
1:0.7	-0.556±0.16
1:0.6	3.79±0.42
1:0.5	3.95±0.366



Figure S5: Zeta potential values of NP: Enzyme complexes formed at varied mole ratio of NP with enzyme.



Figure S6: Absorption spectra of RhB with irradiation time over (a) mpg-C₃N₄MCs and (b) dispersed solution of mpg-C₃N₄.



Figure S7: (a) UV-Vis diffuse reflectance Plot (b) $(\alpha hv)^2 vs$ photon energy (Eg) plot for the band gap determination of mpg-C₃N₄