

## **Copper Nanoparticles Supported on Highly Nitrogen-Rich Containing Covalent Organic Polymers as Heterogeneous Catalyst for *ipso*-Hydroxylation of Phenyl boronic acid to Phenol**

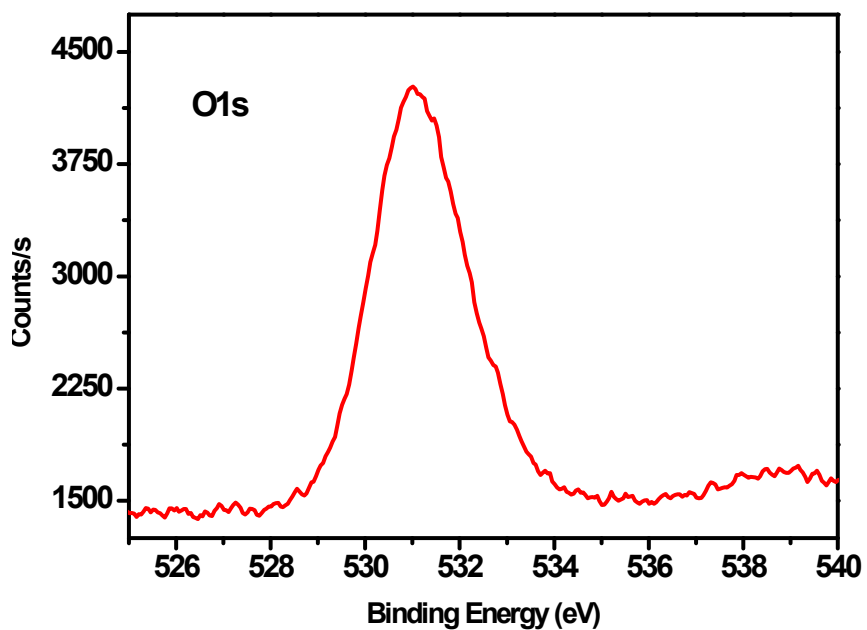
Velu Sadhasivam,<sup>b</sup> Muniyasamy Harikrishnan,<sup>a</sup> Ganesan Elamathi,<sup>a</sup> Rajendran Balasaravanan,<sup>a</sup> Sepperumal Murugesan<sup>a</sup> and Ayyanar Siva\*<sup>a</sup>

<sup>a</sup>Supramolecular and Organometallic Chemistry Lab, Department of Inorganic Chemistry, School of Chemistry, Madurai Kamaraj University, Madurai-625 021, Tamil Nadu, India.

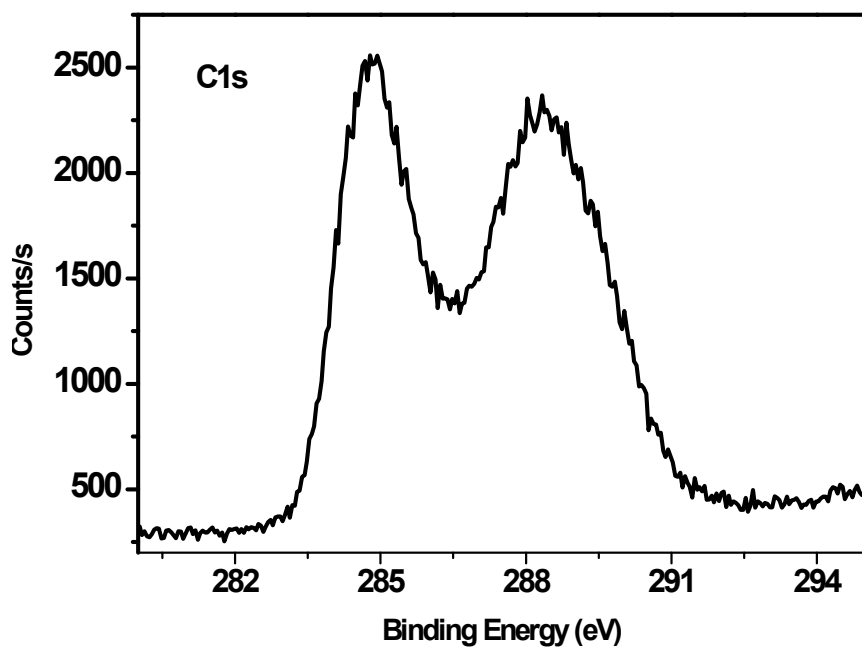
<sup>b</sup>Department of chemistry, V. M. K. V Engineering College, Vinayaga Mission's Research Foundation (Deemed to be university), Salem, Tamil Nadu, India.

E-mail: drasiva@gmail.com (Ayyanar Siva)

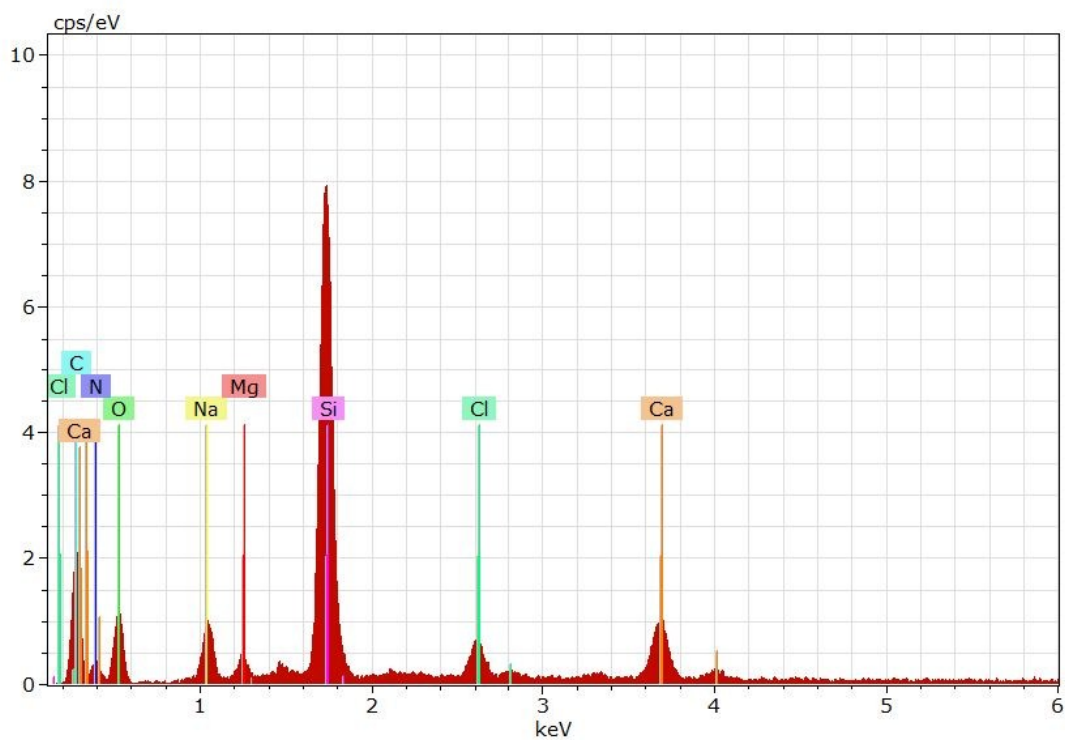
<b>S. No</b>	<b>Content</b>	<b>Page No</b>
1	XP spectrum of compound Cu@TCOP for C1s and O1s	2
2	EDS spectrum of compound TCOP and Cu@TCOP	3
3	Elemental analysis of TCOP and Cu@TCOP	4
4	SEM images of TCOP and Cu@TCOP	5
5	<sup>1</sup> H NMR and <sup>13</sup> C NMR spectra of all the phenols derivatives	6



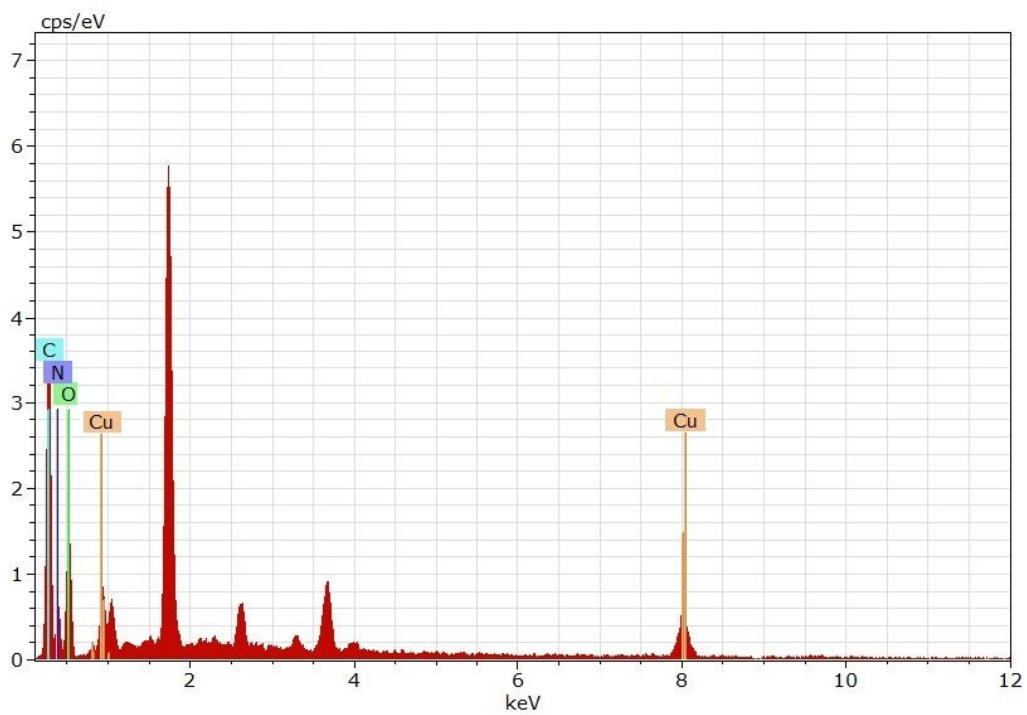
**Figure S1.** XP spectra of compound Cu@TCOP for O1s.



**Figure S2.** XP spectra of compound Cu@TCOP for C1s.



**Figure S3.** EDS spectra of compound TCOP.



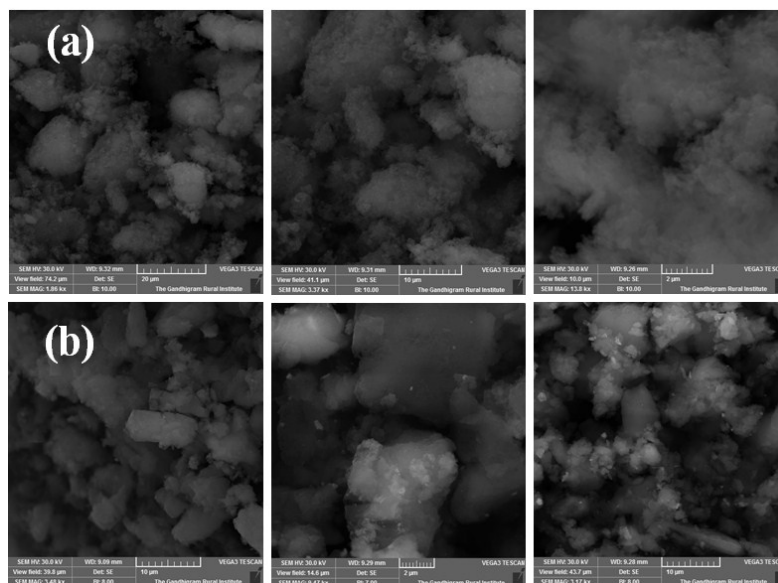
**Figure S4.** EDS spectra of compound Cu@TCOP.

**Table 1.** Elemental atomic weight % of compound TCOP.

<b>Elem: At No</b>	<b>Series</b>	<b>Unn. C</b> <b>[Wt %]</b>	<b>Norm. C</b> <b>[Wt %]</b>	<b>Atom. C</b> <b>[at. %]</b>	<b>(1 Sigma)</b> <b>[Wt %]</b>
C-6	K- Series	39.90	39.95	49.74	7.06
N-7	K-Series	25.65	25.68	24.00	5.03
O-8	K-Series	14.86	14.88	15.89	4.19
Si-14	K-Series	12.09	12.11	6.45	0.58
Na-11	K-Series	3.37	3.38	2.20	0.30
Mg-12	K-Series	2.10	2.02	0.84	0.08
Ca-20	K-Series	2.03	2.00	0.88	0.05
<b>Total</b>		<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	

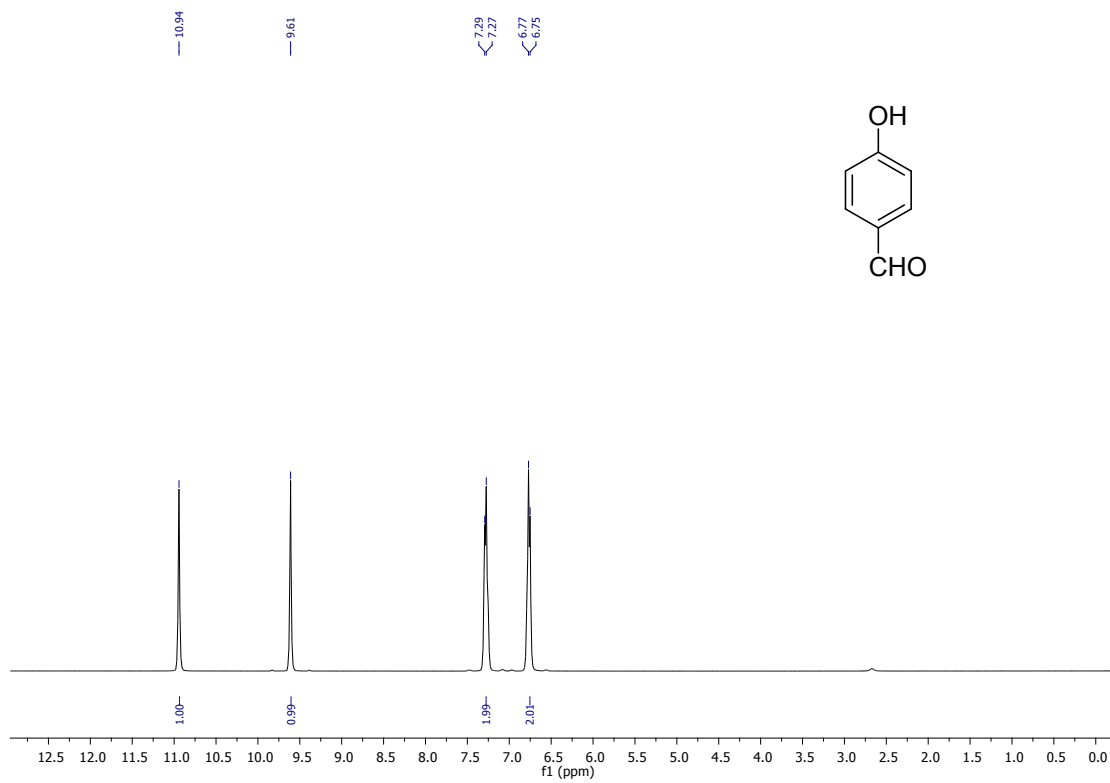
**Table 2.** Elemental atomic weight % of compound Cu/TCOP.

<b>Elem: At No</b>	<b>Series</b>	<b>Unn. C</b> <b>[Wt %]</b>	<b>Norm. C</b> <b>[Wt %]</b>	<b>Atom. C</b> <b>[at. %]</b>	<b>(1 Sigma)</b> <b>[Wt %]</b>
C-6	K- Series	44.10	43.84	52.76	8.54
N-7	K-Series	31.15	29.98	25.17	5.92
O-8	K-Series	21.51	20.71	19.11	5.39
Cu -29	K-series	3.34	3.29	0.71	0.12
<b>Total</b>		<b>100.00</b>	<b>100.00</b>	<b>100.00</b>	

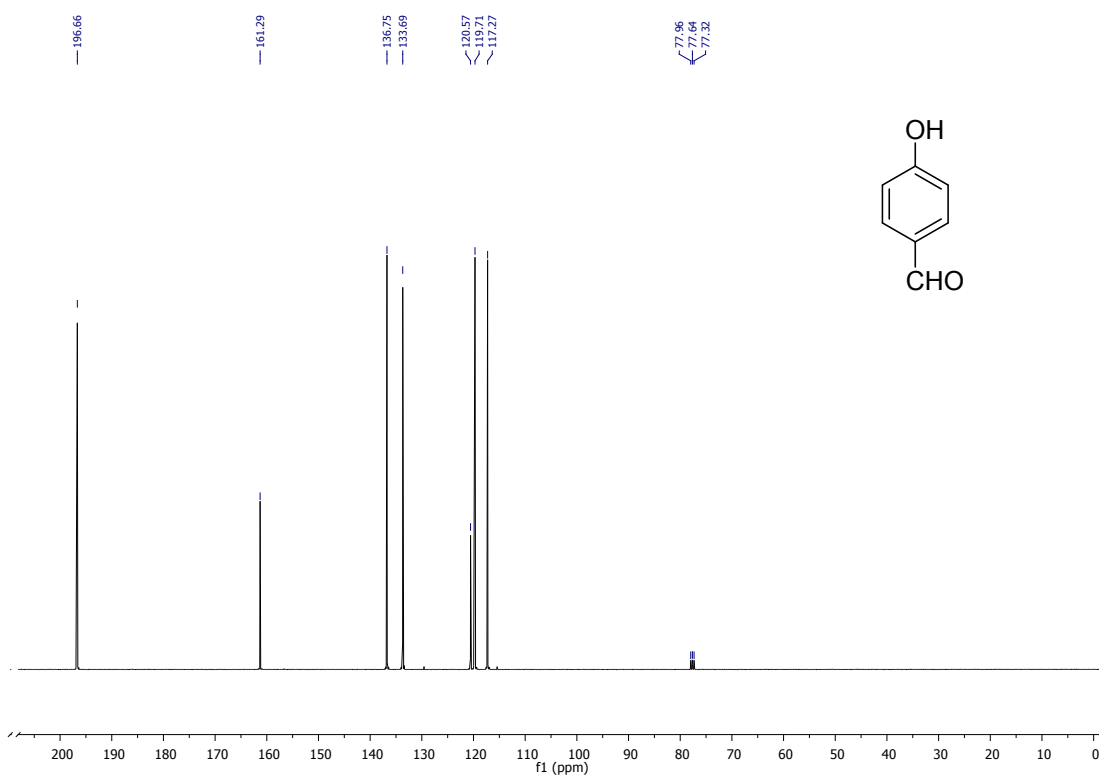


**Figure S5.** SEM images of compound (a) TCOP (b) Cu/TCOP.

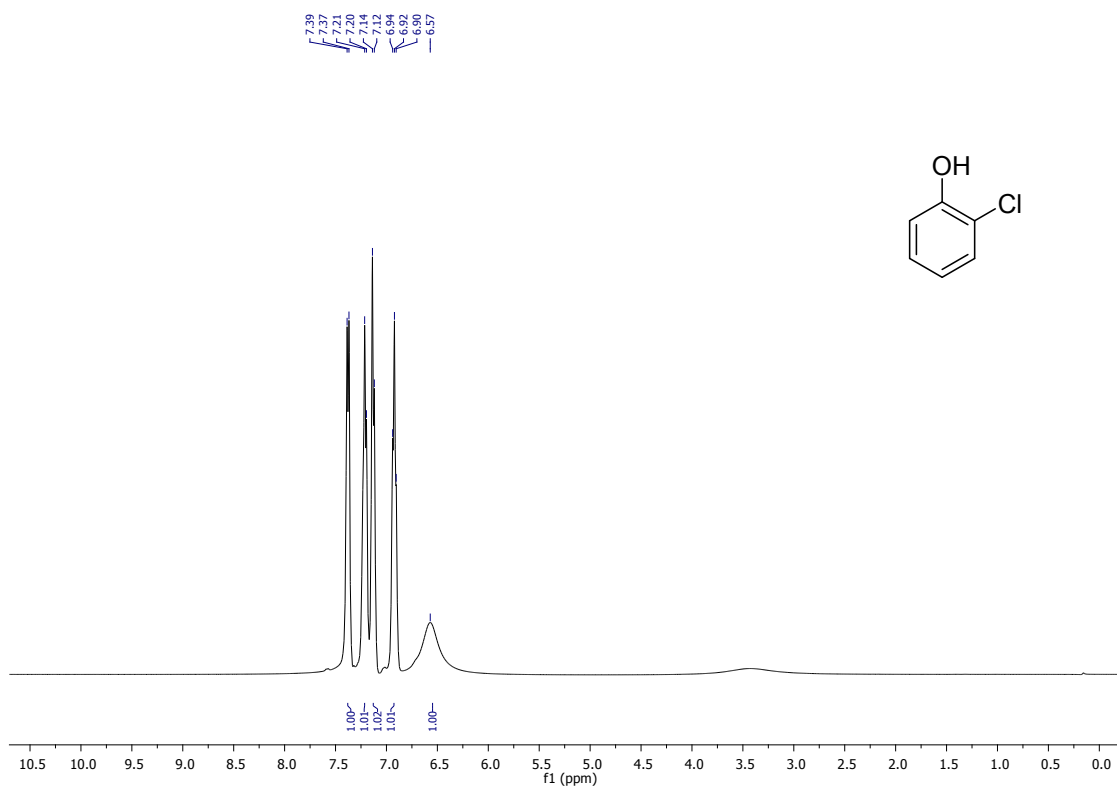
**<sup>1</sup>H NMR characterization data for all the biphenyl coupled products.**



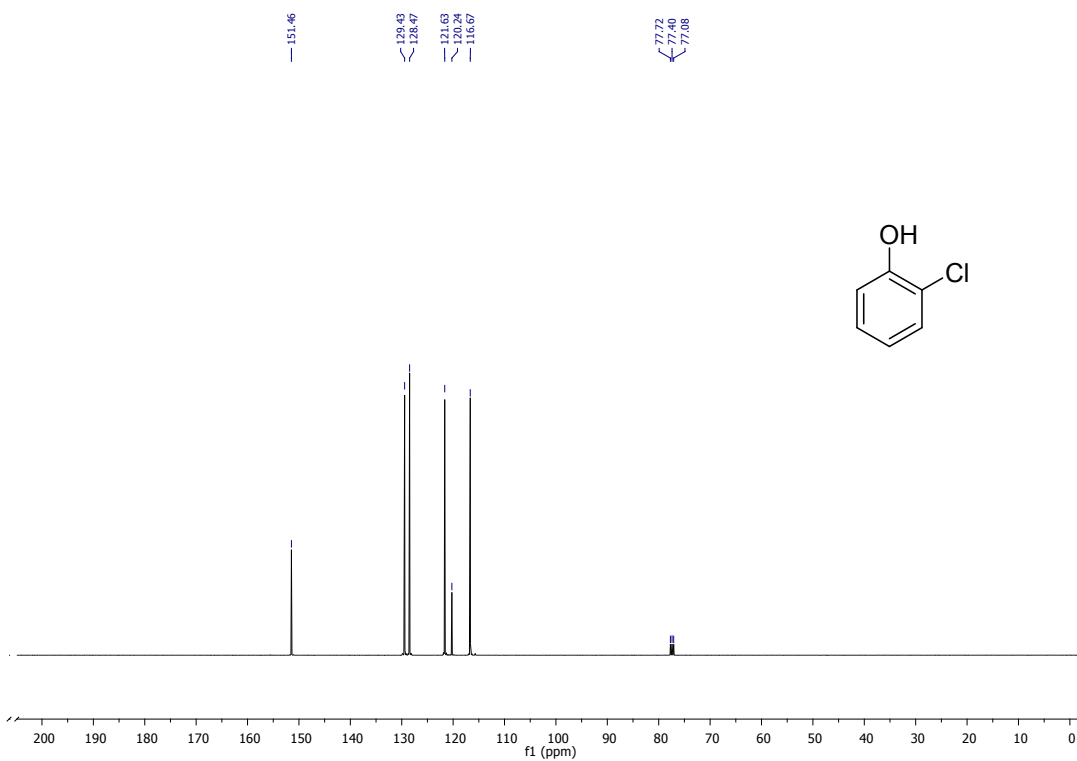
**Figure S6:** <sup>1</sup>H NMR spectrum of compound 4-hydroxy benzaldehyde.



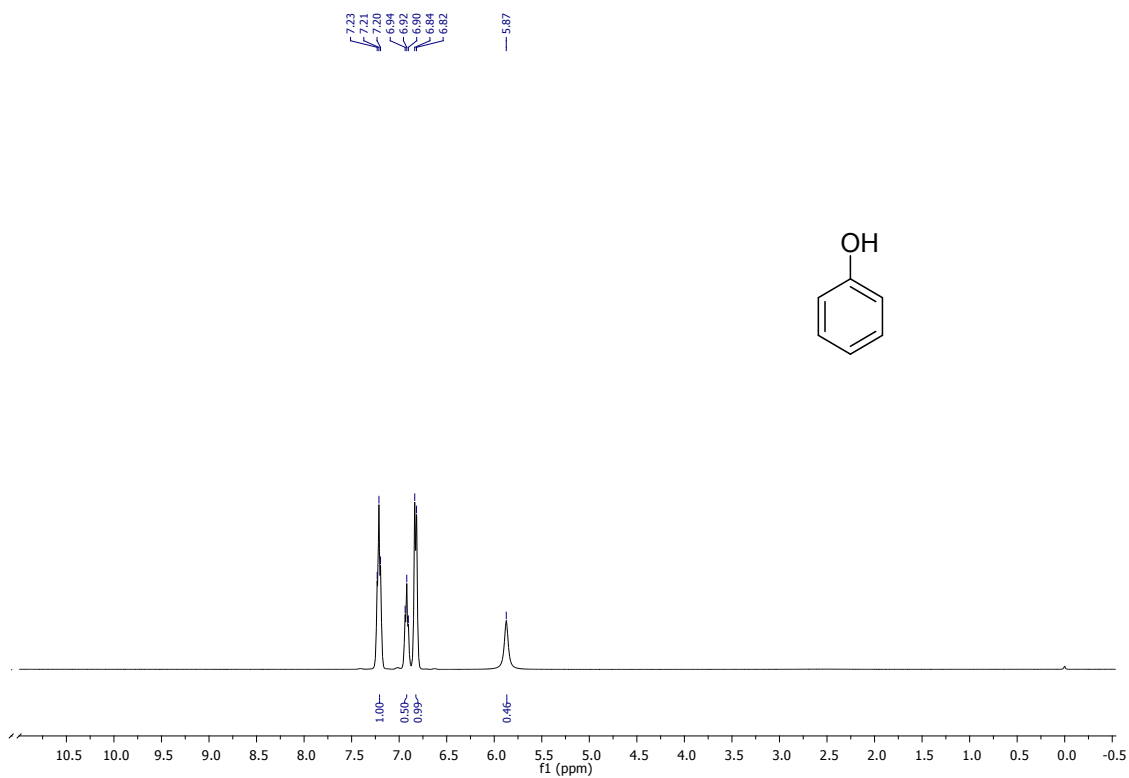
**Figure S7:** <sup>13</sup>C NMR spectrum of compound 4-hydroxy benzaldehyde.



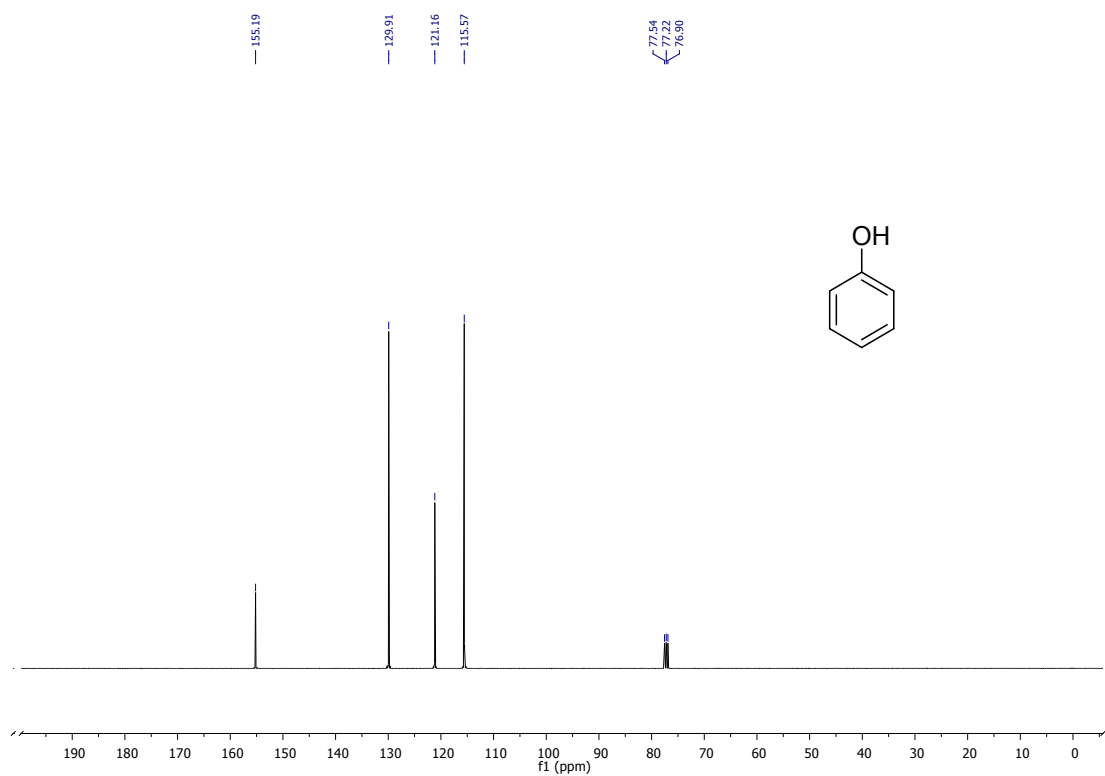
**Figure S8:**  $^1\text{H}$  NMR spectrum of compound 2-chloro phenol.



**Figure S9:**  $^{13}\text{C}$  NMR spectrum of compound 4-chloro phenol.

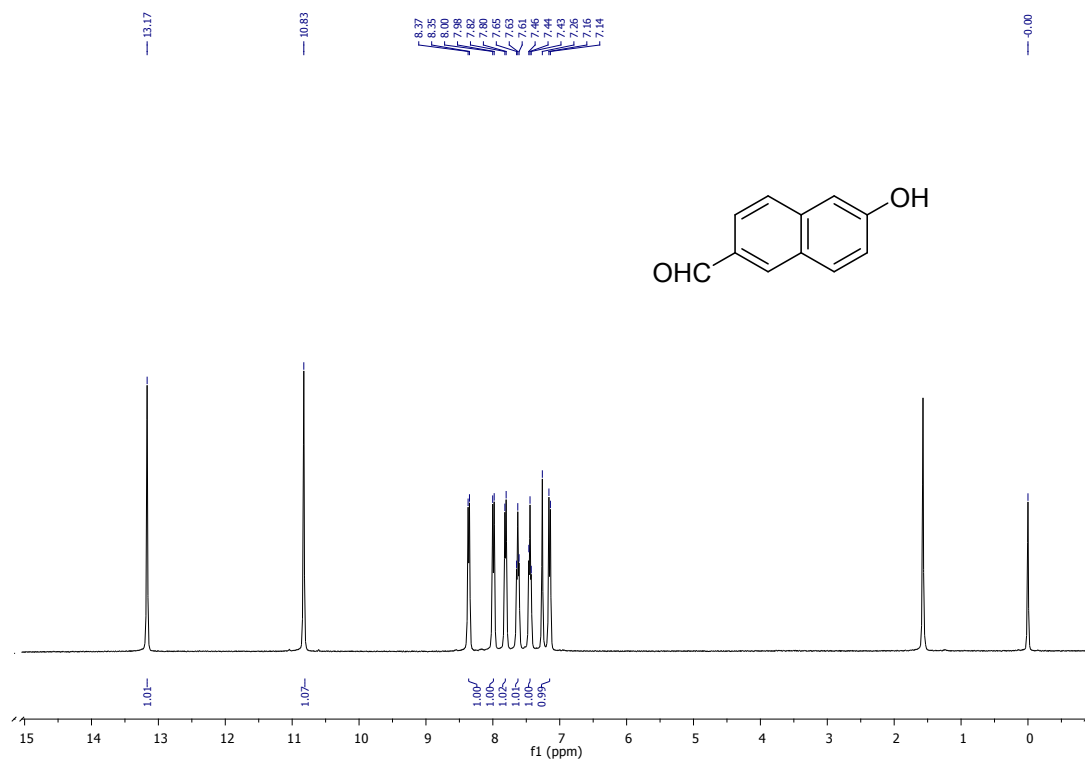


**Figure S10:**  $^1\text{H}$  NMR spectrum of compound phenol.

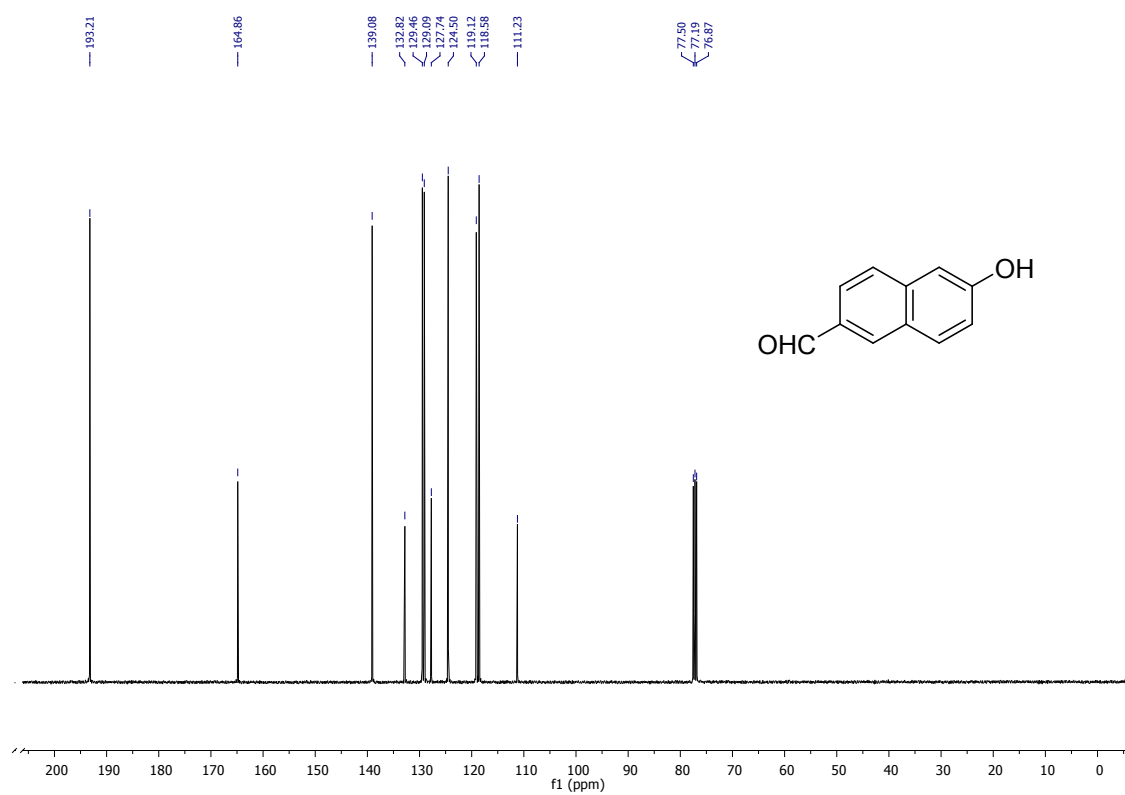


**Figure S11:**  $^{13}\text{C}$  NMR spectrum of compound phenol.

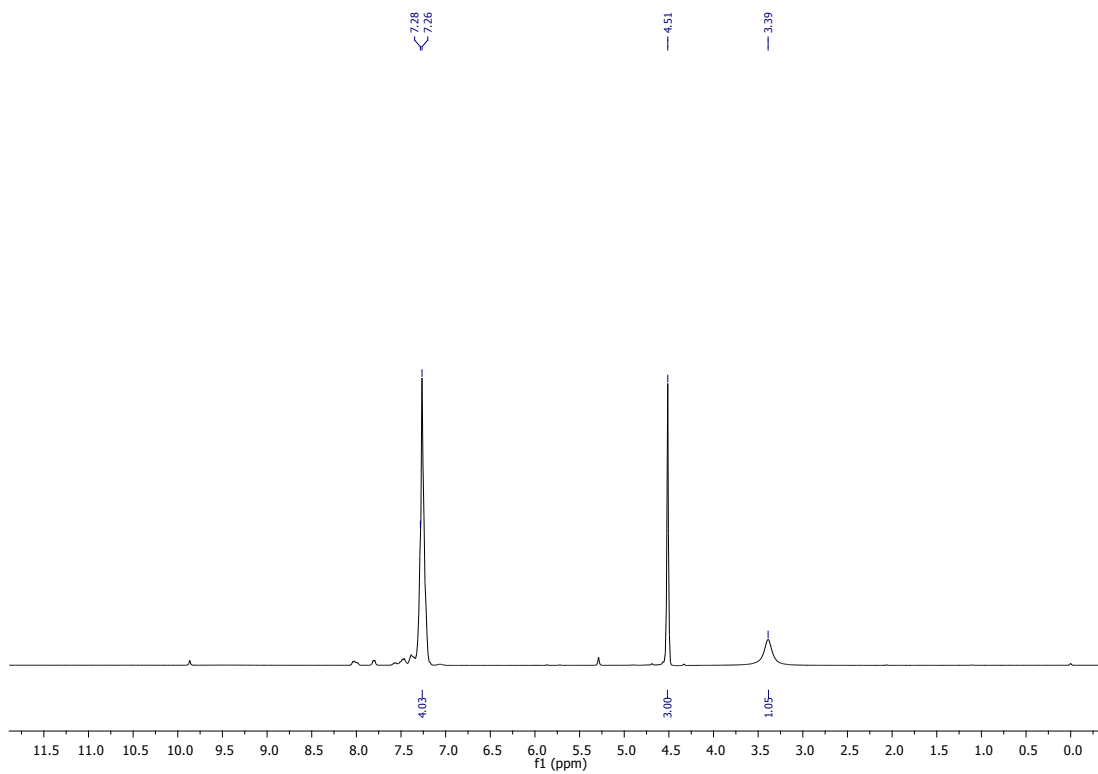




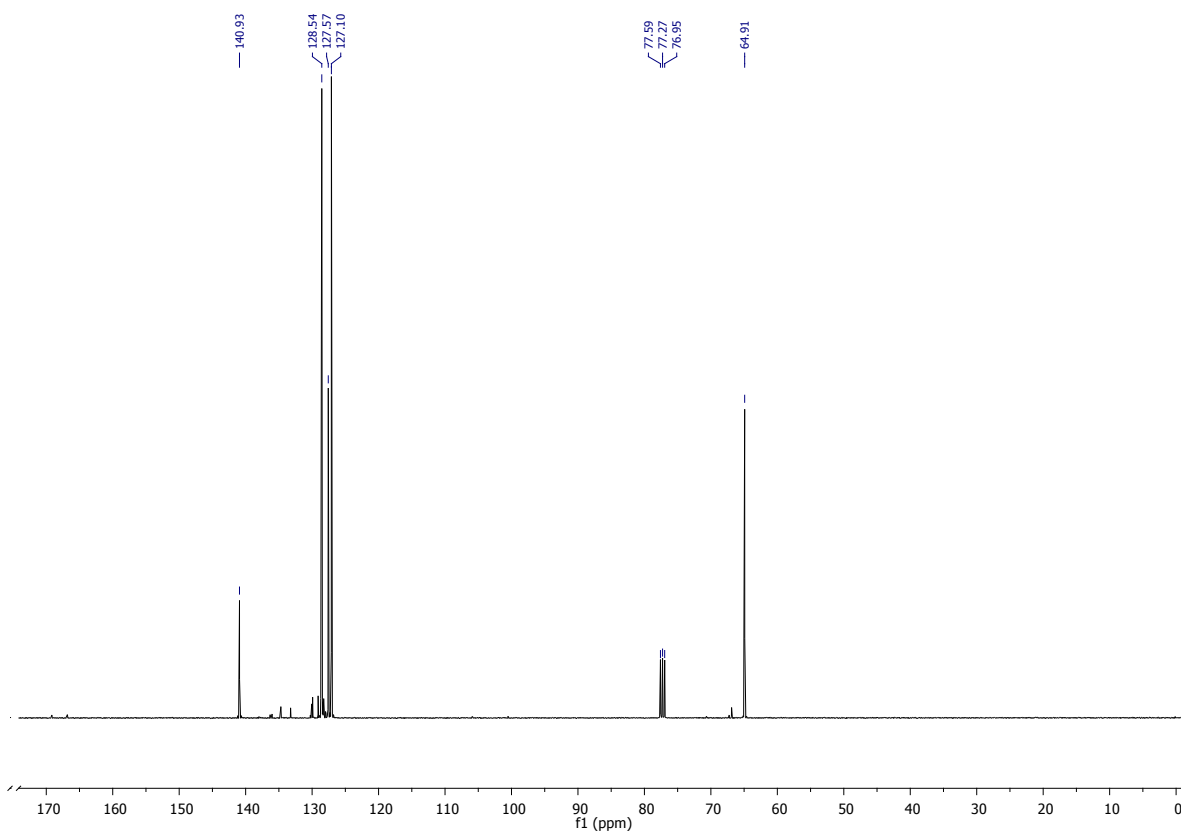
**Figure S12:** <sup>1</sup>H NMR spectrum of compound 2-Naphthol.



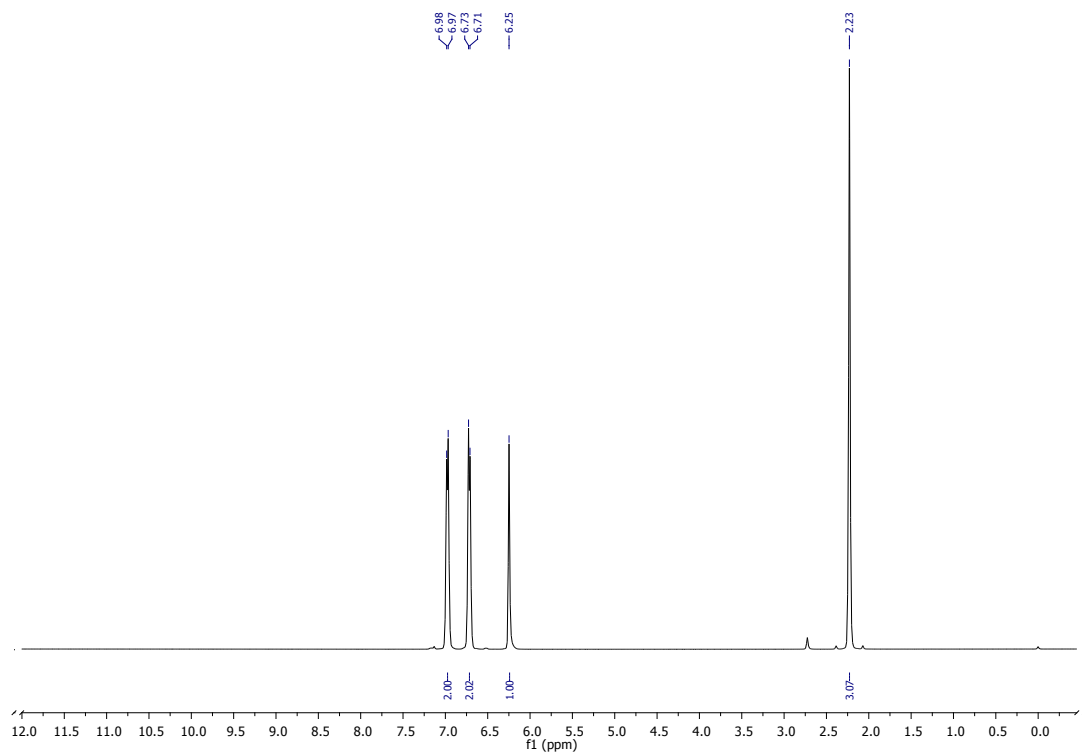
**Figure S13:** <sup>13</sup>C NMR spectrum of compound 2-Naphthol.



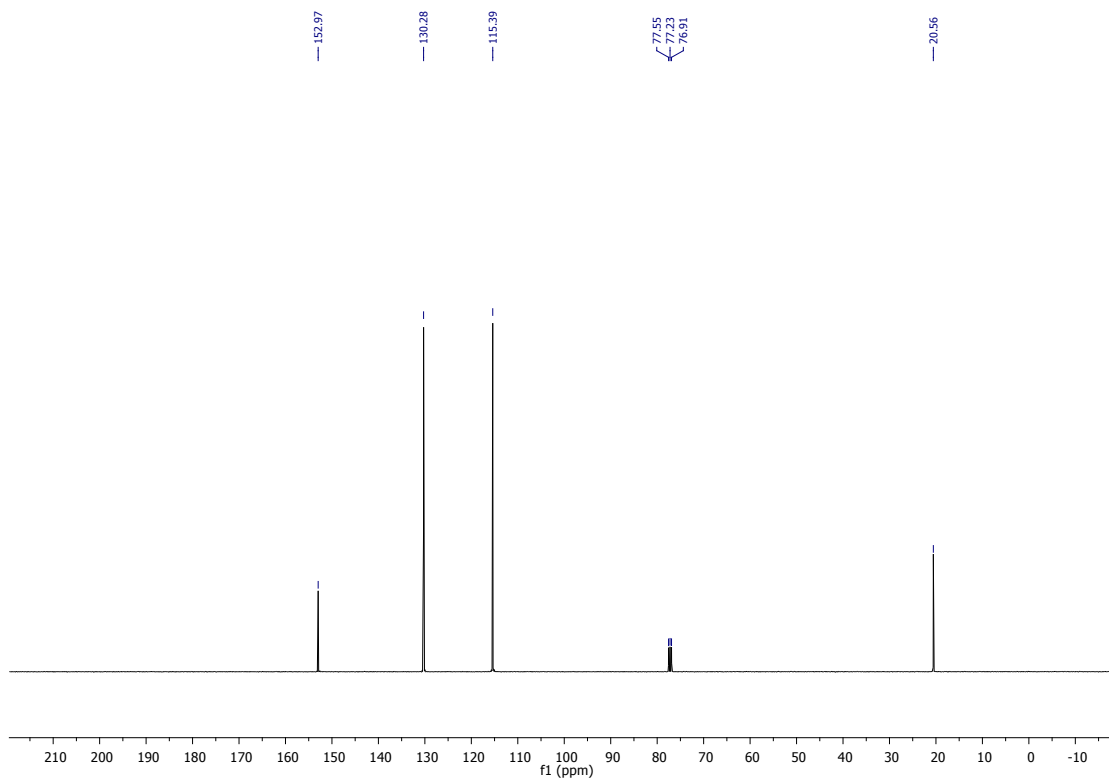
**Figure S14:** <sup>1</sup>H NMR spectrum of compound 4-Methoxy Phenol.



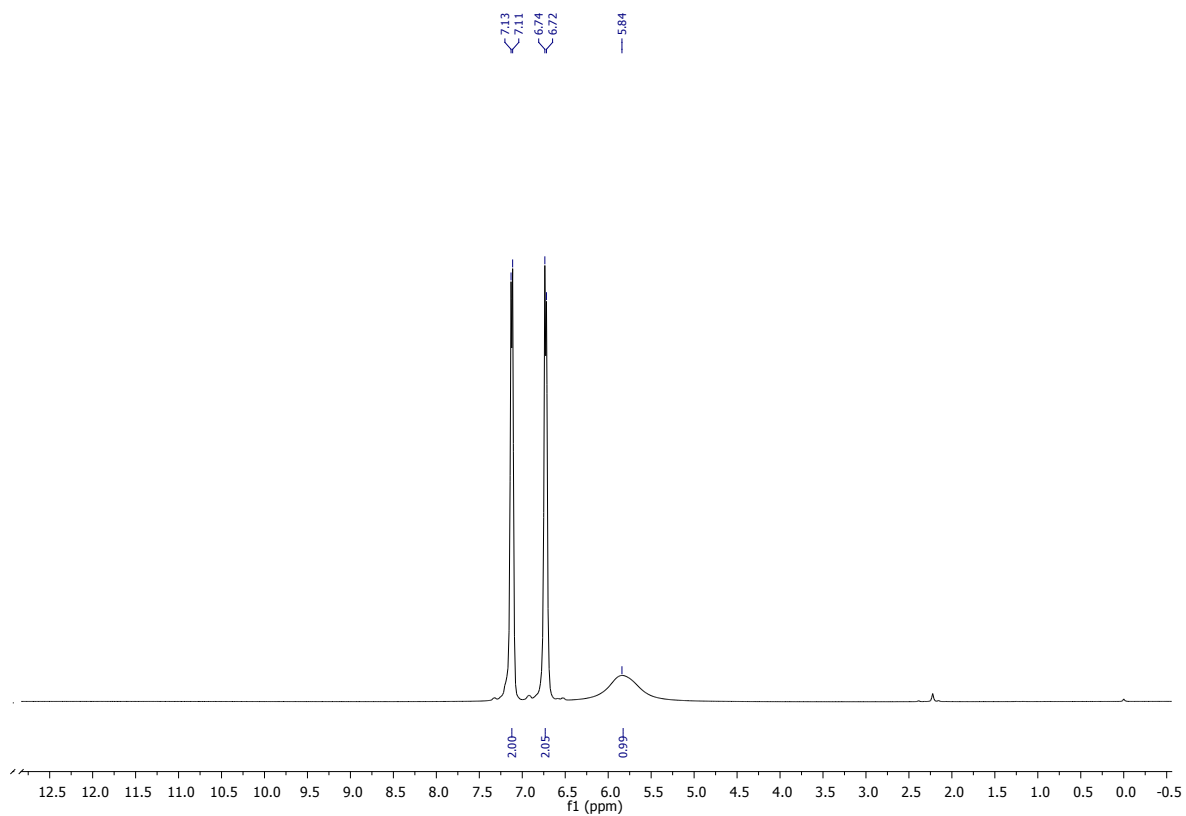
**Figure S15:** <sup>13</sup>C NMR spectrum of compound 4-Methoxy Phenol.



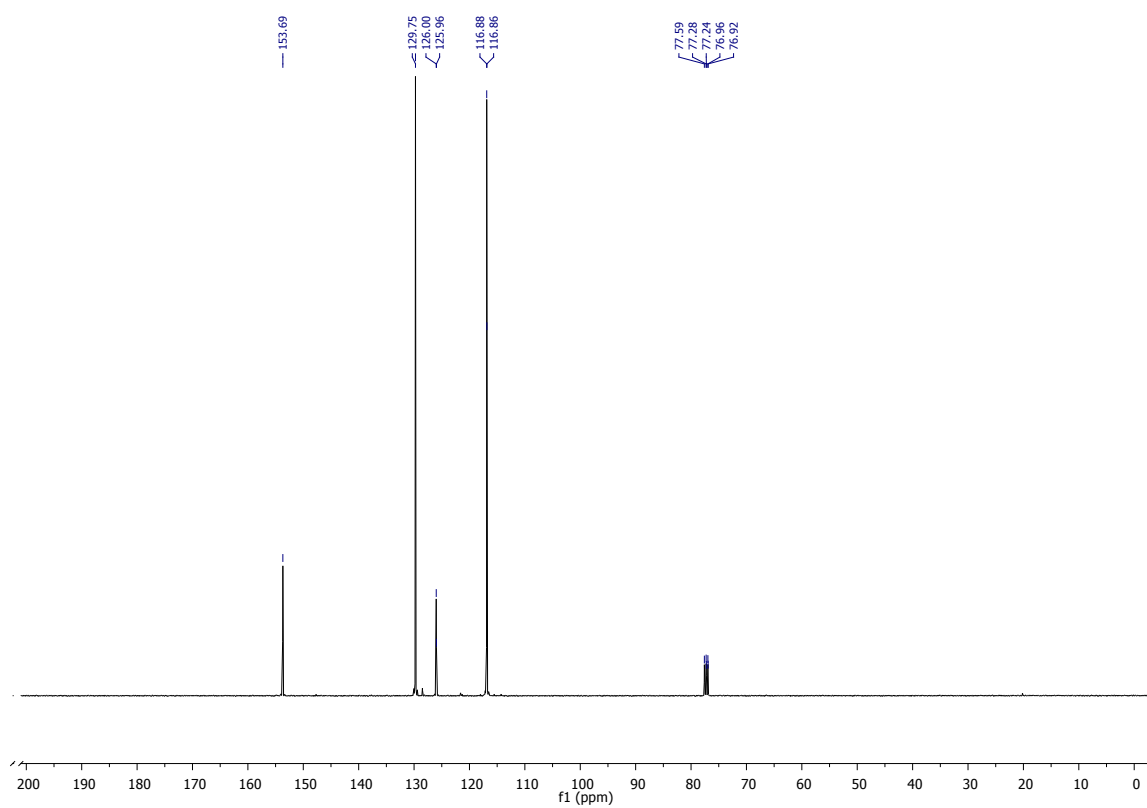
**Figure S16:** <sup>1</sup>H NMR spectrum of compound 4-Methyl Phenol.



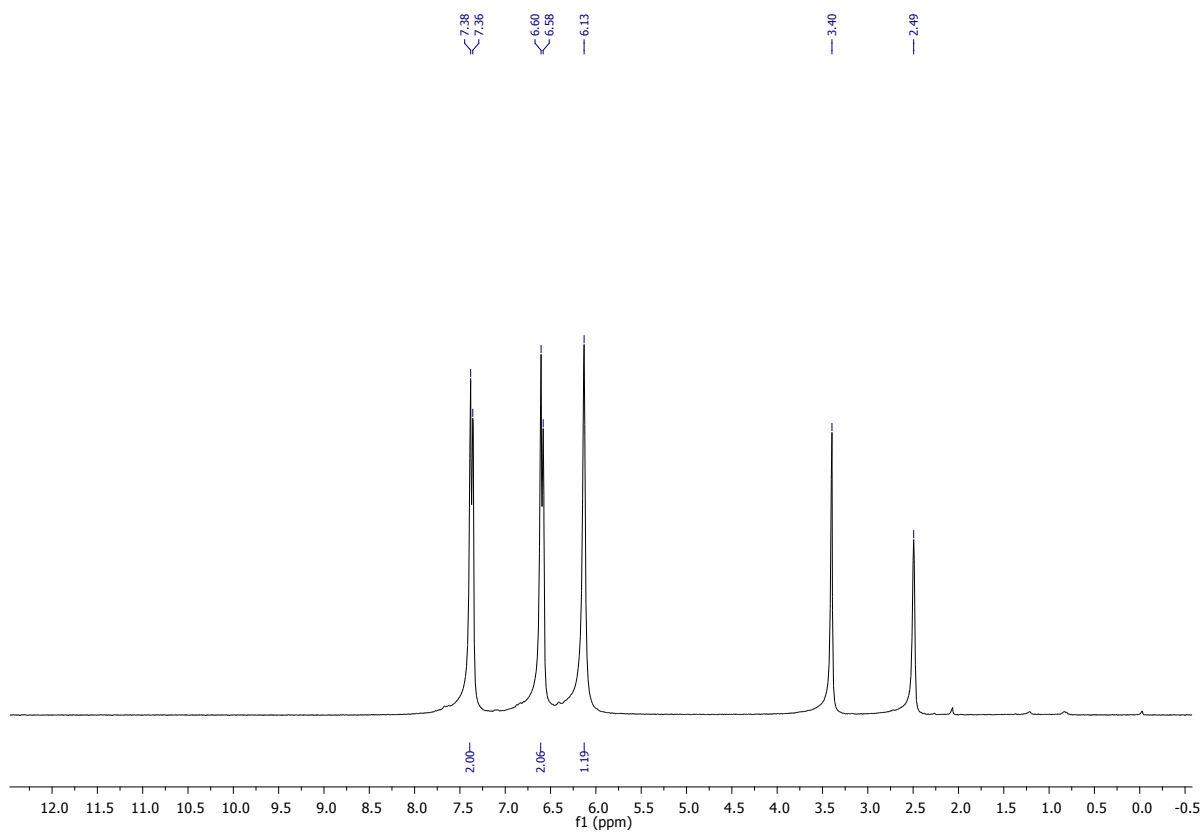
**Figure S17:** <sup>13</sup>C NMR spectrum of compound 4-Methyl Phenol.



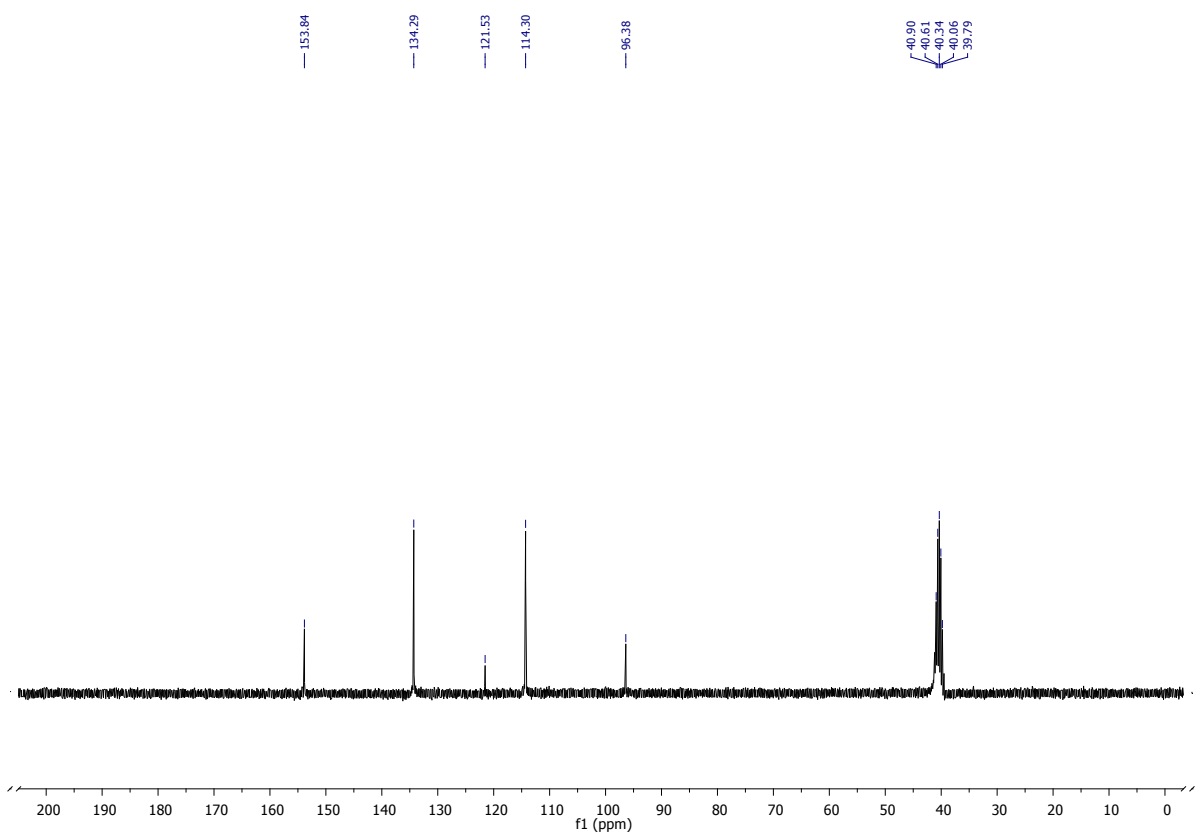
**Figure S18:**  $^1\text{H}$  NMR spectrum of compound 4-Chlorophenol.



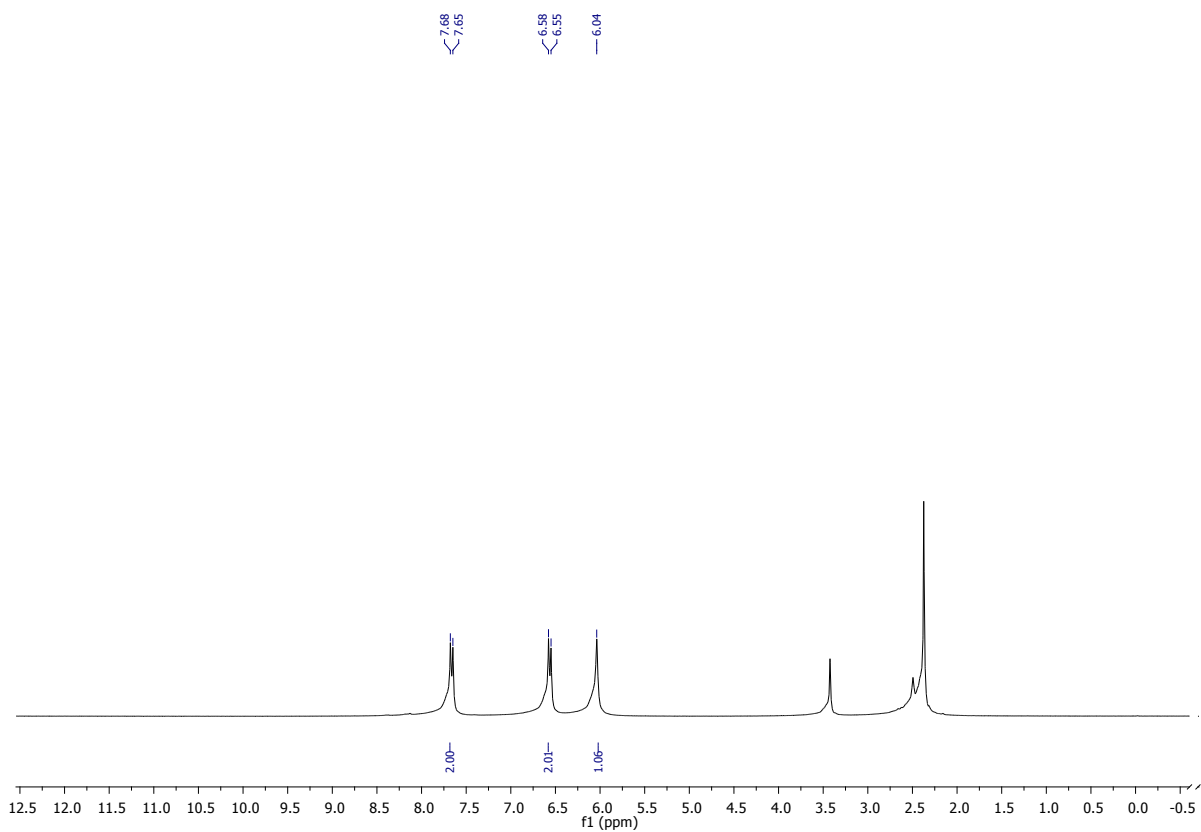
**Figure S19:**  $^{13}\text{C}$  NMR spectrum of compound 4-ChloroPhenol.



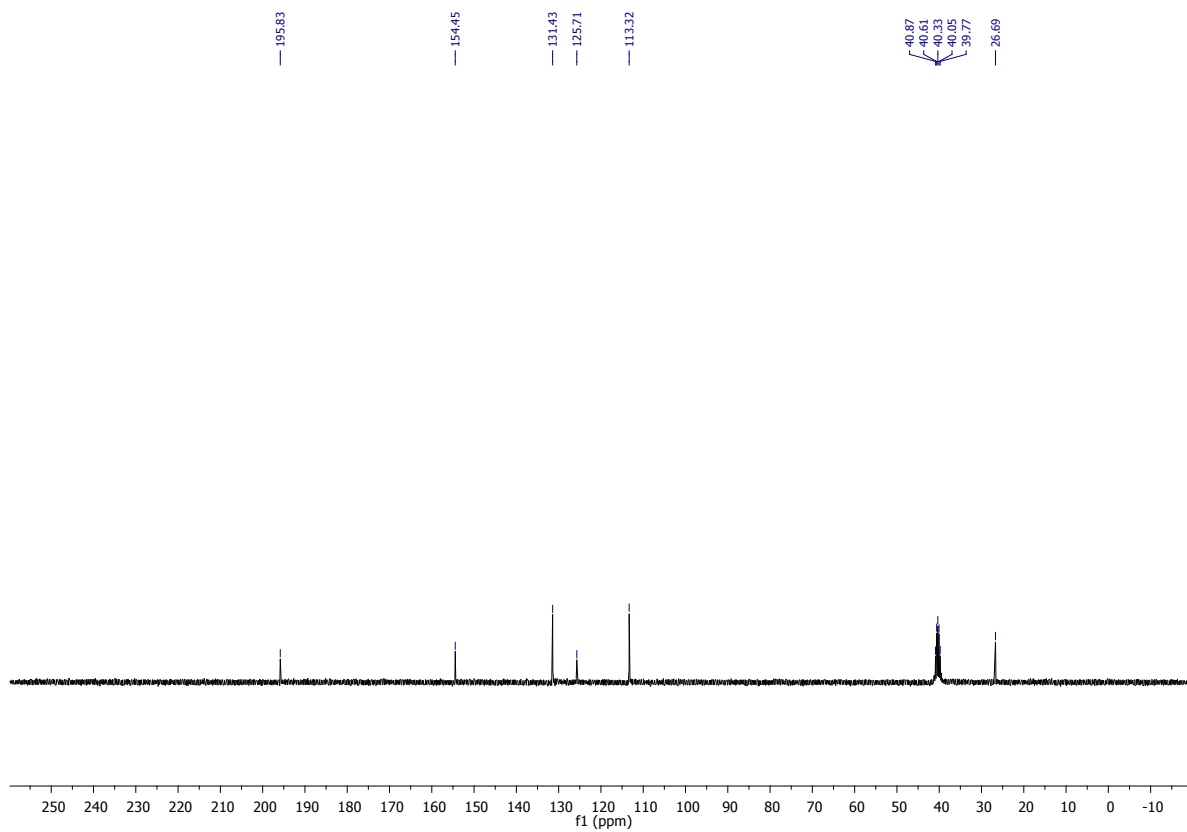
**Figure S20:**  $^1\text{H}$  NMR spectrum of compound 4-hydroxy benzonitrile.



**Figure S21:**  $^{13}\text{C}$  NMR spectrum of compound 4-hydroxy benzonitrile.

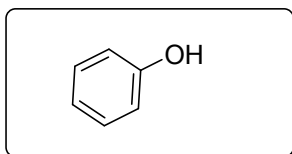


**Figure S22:**  $^1\text{H}$  NMR spectrum of compound 4-hydroxyacetophenone.

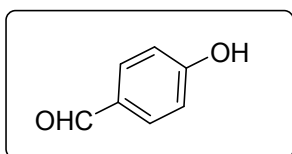


**Figure S23:**  $^{13}\text{C}$  NMR spectrum of compound 4-hydroxyacetophenone.

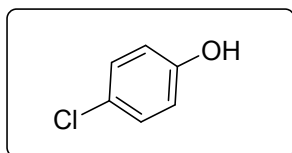
## <sup>1</sup>H and <sup>13</sup>C NMR data for all the derivatives



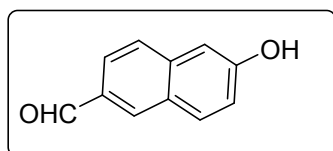
**Phenol 1a** Compound **1a** was prepared according to the general procedure to give a colourless viscous liquid with 99% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.21 (t, *J* = 7.0 Hz, 2H), 6.92 (t, *J* = 6.8 Hz, 1H), 6.83 (d, *J* = 7.5 Hz, 2H), 5.87 (s, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 155.19 (s), 129.91 (s), 121.16 (s), 115.57 (s), 77.54 (s), 77.22 (s), 76.90 (s).



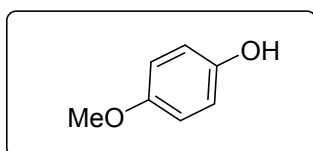
**4-hydroxyl benzaldehyde (1b)** Compound **1b** was prepared according to the general procedure to give a colourless solid with 92% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 10.94 (s, 1H), 9.61 (s, 1H), 7.28 (d, *J* = 7.4 Hz, 2H), 6.76 (d, *J* = 7.5 Hz, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 196.66 (s), 161.29 (s), 136.75 (s), 133.69 (s), 120.57 (s), 119.71 (s), 117.27 (s), 77.96 (s), 77.64 (s), 77.32 (s).



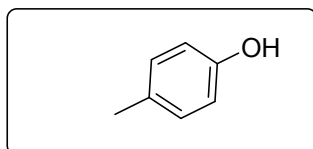
**4-chlorophenol (1c)** Compound **1c** was prepared according to the general procedure to give a light brown solid with 95% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.12 (d, *J* = 7.4 Hz, 2H), 6.73 (d, *J* = 7.5 Hz, 2H), 5.84 (s, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 153.69 (s), 129.75 (s), 125.98 (d, *J* = 4.0 Hz), 116.87 (d, *J* = 2.5 Hz), 77.59 (s), 77.26 (d, *J* = 4.0 Hz), 76.94 (d, *J* = 3.9 Hz).



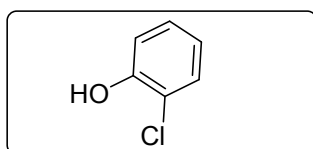
**6-hydroxy-2-naphthaldehyde (1d)** Compound **1d** was prepared according to the general procedure to give a colourless solid with 80% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 13.17 (s, 1H), 10.83 (s, 1H), 8.36 (d, *J* = 8.3 Hz, 1H), 7.99 (d, *J* = 8.9 Hz, 1H), 7.81 (d, *J* = 7.7 Hz, 1H), 7.63 (t, *J* = 7.5 Hz, 1H), 7.44 (t, *J* = 7.2 Hz, 1H), 7.15 (d, *J* = 8.9 Hz, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 193.21 (s), 164.86 (s), 139.08 (s), 132.82 (s), 129.46 (s), 129.09 (s), 127.74 (s), 124.50 (s), 119.12 (s), 118.58 (s), 111.23 (s), 77.50 (s), 77.19 (s), 76.87 (s).



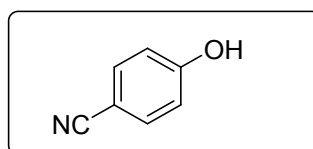
**4-methoxyphenol (1e)** Compound **1e** was prepared according to the general procedure to give a colourless solid with 85% yield.  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  140.93 (s), 128.54 (s), 127.57 (s), 127.10 (s), 77.59 (s), 77.27 (s), 76.95 (s), 64.91 (s).



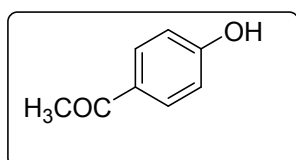
**p-cresol (1f)** Compound **1f** was prepared according to the general procedure to give a colourless viscous liquid with 87% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  6.98 (d,  $J = 7.2$  Hz, 2H), 6.72 (d,  $J = 6.8$  Hz, 2H), 6.25 (s, 1H), 2.23 (s, 3H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  152.97 (s), 130.28 (s), 115.39 (s), 20.56 (s).



**2-chlorophenol (1g)** Compound **1g** was prepared according to the general procedure to give a colourless solid with 80% yield.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.38 (d,  $J = 7.4$  Hz, 1H), 7.21 (d,  $J = 7.0$  Hz, 1H), 7.13 (d,  $J = 7.4$  Hz, 1H), 6.92 (t,  $J = 6.9$  Hz, 1H), 6.57 (s, 1H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  151.46 (s), 129.43 (s), 128.47 (s), 121.63 (s), 120.24 (s), 116.67 (s), 77.72 (s), 77.40 (s), 77.08 (s).



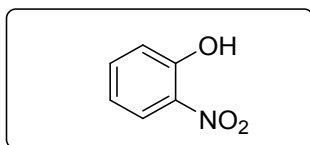
**4-hydroxybenzonitrile (1h)** Compound **1h** was prepared according to the general procedure to give a light brown colour solid with 78% yield.  $^1\text{H}$  NMR (300 MHz, DMSO)  $\delta$  7.37 (d,  $J = 7.0$  Hz, 2H), 6.59 (d,  $J = 7.0$  Hz, 2H), 6.13 (s, 1H).  $^{13}\text{C}$  NMR (75 MHz, DMSO)  $\delta$  153.84 (s), 134.29 (s), 121.53 (s), 114.30 (s), 96.38 (s), 44.93 – 42.02 (m), 41.18 (s), 40.76 (d,  $J = 21.4$  Hz), 40.20 (d,  $J = 20.9$  Hz), 39.79 (s).



**1-(4-hydroxyphenyl)ethan-1-one (1i)** Compound **1i** was prepared according to the general procedure to give a light yellow solid with 72% yield.  $^1\text{H}$  NMR (300 MHz, DMSO)  $\delta$  7.66 (d,  $J = 8.5$  Hz, 2H), 6.56 (d,  $J = 8.6$  Hz, 2H), 6.04 (s, 1H).  $^{13}\text{C}$  NMR (75 MHz, DMSO)  $\delta$  195.83 (s), 154.45 (s), 131.43 (s),



125.71 (s), 113.32 (s), 40.87 (s), 40.47 (d,  $J = 20.8$  Hz), 39.91 (d,  $J = 21.4$  Hz), 39.71 – 39.53 (m), 26.69 (s).



**2-Nitro phenol 1j** Compound **1j** was prepared according to the general procedure to give a light yellow solid with 54% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  10.59 (s, 1H), 8.11 (d,  $J = 8.3$  Hz, 1H), 7.59 (dd,  $J = 10.8, 4.2$  Hz, 1H), 7.16 (d,  $J = 8.2$  Hz, 1H), 7.00 (t,  $J = 7.5$  Hz, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  155.13 (s), 137.57 (s), 133.69 (s), 125.07 (s), 120.24 (s), 119.97 (s), 77.39 (s), 77.07 (s), 76.75 (s).